Sequential ring closure and [2,3]-sigmatropic rearrangement reactions: an approach to the synthesis of C-19 oxygenated cyathane-type diterpenoids

Edward Piers* and Katherine Louise Cook

Department of Chemistry, University of British Columbia, 2036 Main Mall, Vancouver, British Columbia, Canada V6T 121

Preparation of the tertiary allylic alcohols 17–22, followed by subjection of these substances to the Still–Mitra [2,3]-sigmatropic rearrangement sequence, provides the functionalised bicyclo[4.3.0]nonanes 23–26; acquisition of 26 points to a strategy for the synthesis of C-19 oxygenated cyathane-type diterpenoids.

Members of the cyathane family of diterpenoids share the tricyclic carbon skeleton shown in 1.[†] Although many of the cyathanes are highly oxygenated, particularly on ring C, very few have oxygen functions associated with the C-3 isopropyl group. Examples of C-19 hydroxylated compounds belonging to this class include cyathin A_4 (2),³ sarcodonin G (3)^{1b} and sarcodonin A (4).^{1b} Although the configuration at C-18 of 2 was not determined,³ an X-ray crystallographic analysis of a derivative of sarcodonin G established that this natural product and, presumably, the related substance sarcodonin A, possess the absolute configurations shown in formulas 3 and 4, respectively.^{1b}

From a synthetic viewpoint, the presence of the C-18 stereogenic centre in substances such as 2-4 translates into a complexity not present in cyathanes that possess at C-3 an unoxygenated isopropyl group. On the other hand, the fact that 2-4 also contain a C-3-C-4 double bond suggests that a synthetic approach to the ring A functionality of these materials could involve a [2,3]-sigmatropic rearrangement strategy. For example, successful bond reorganisation of each of the isomeric anions of general structure 5 and 7 would provide a product 6 having the correct relative configurations at C-9 and C-18

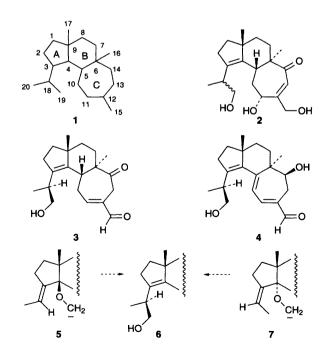
(cyathane numbering). We report herein the results of a study that shows the viability of this strategy.[‡]

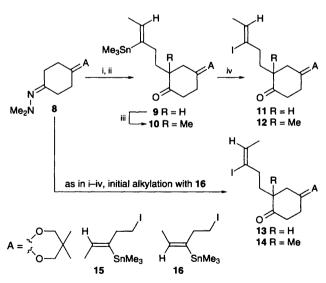
The keto alkenyl iodides 11-14 employed in this study were prepared as shown in Scheme 1.§ Alkylation⁵ of the hydrazone 8 with the iodide 15.¶ followed by cleavage of the hydrazone function,⁵ provided (78%) the keto trimethylstannane 9. Methylation of 9 afforded 10 (66%). Iodostannylation of 9 and 10 to give 11 and 12 was readily achieved in yields exceeding 90%. In a similar fashion, employing the iodide 16¶ as the initial alkylating agent, the substrates 13 and 14 were also prepared from the hydrazone 8 (Scheme 1). The reaction yields were similar to those involved in the conversion of 8 into 11 and 12.

Treatment of the keto iodide 11 with BuLi (4 equiv.)) afforded in 84% yield a mixture of two products, 17 and 18, in a ratio of about 20:1 (Scheme 2). Flash chromatography⁷ of this mixture on silica gel allowed the isolation of pure 17 and 18. BuLi-mediated ring closure of the substrate 13 (geometric isomer of 11) produced a mixture of the alcohols 19 and 20 (ratio $\sim 4:1$) in 78% yield. Separation of these substances was achieved by flash chromatography.⁷

In contrast to the substrates 11 and 13, which produced mixtures of diastereoisomers containing a preponderance of the *trans*-fused products 17 and 19, respectively, BuLi-mediated cyclisation of substances 12 and 14 gave, in each case, a single bicyclic product. Based on an examination of molecular models, it was not surprising to find (*vide infra*) that both of the products 21 and 22 are *cis*-fused (Scheme 2). The isolated yields of these materials were 58 and 72%, respectively.

Sequential treatment⁸ of the tertiary allylic alcohol **17** with KH and iodomethyl(tributyl)stannane, followed by addition of BuLi, provided, in 63% yield, the bicyclic alcohol **23**** (Scheme 2). As expected,⁸ the same product **23** (56% yield) was



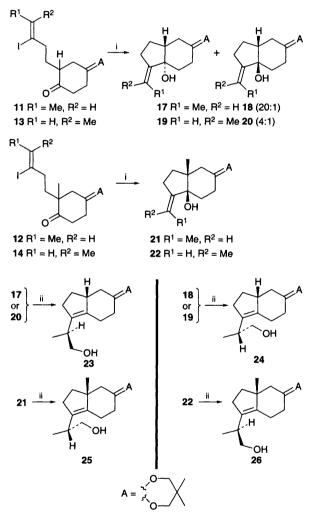


Scheme 1 Reagents and conditions: i, BuLi, THF, 0 °C; DMPU, 15, 35 °C; ii, NaIO4, THF, H₂O, KH₂PO₄-NaOH-H₂O buffer (pH 7.2), 40 °C; iii, KOBu¹, THF, HMPA, 60 °C; MeI; iv, I₂, CH₂Cl₂, room temp.

derived from subjection of the substrate 20 to a similar reaction protocol. On the other hand, the allylic alcohols 18 and 19 (epimers of 17 and 20, respectively) were transformed into 24, which is epimeric with 23.

With respect to an approach to the synthesis of C-19 oxygenated cyathane-type diterpenoids, the stereospecific transformations of 21 and 22 into the corresponding alcohols 25 and 26 are particularly important. Conversion of the starting materials into the corresponding (tributylstannyl)methyl ethers, followed by the [2,3]-signatropic rearrangement reactions, afforded the products 25 and 26 in yields of 52 and 46%, respectively. It should be noted that the Still-Mitra rearrangement⁸ has been previously achieved on tertiary allylic acyclic alcohols.9 On the other hand, the rearrangement sequences outlined above involve substrates (17-22) in which the allylic hydroxy groups are, in each case, at an angular position of a bicyclic carbon framework. Indeed, the conversions of 21 and 22 into 25 and 26, respectively, are especially noteworthy since, in these cases, the hydroxy groups of the starting materials are, in addition, adjacent to quaternary carbon centres.

The relative configurations of substances 17-26 were determined by performing two key X-ray crystallographic



Scheme 2 Reagents and conditions: i, BuLi (4 equiv.), THF, -78 °C, then H₂O; ii, KH (1.1–1.5 equiv.), THF, room temp.; 18-crown-6 (2 equiv., for substrates 18 and 20–22 only); Bu₃SnCH₂I (2–4 equiv.), room temp.; BuLi (4–6 equiv.), -78 °C to room temp., then H₂O

studies.^{††} One of these, carried out on the tertiary allylic alcohol **21** (mp 88–89 °C) obtained from cyclisation of **12**, showed that the former compound possesses a *cis*-fused ring junction. Since the Still-Mitra rearrangement of **22** provided a product diastereoisomeric with that derived from **21**, the alcohol **22** must also be *cis*-fused.^{‡‡}

The alcohol (mp 84–85 °C) derived from subjecting either 17 or 20 to the [2,3]-sigmatropic rearrangement sequence was shown by X-ray analysis to possess the relative configuration shown in formula 23. Therefore, the major product 17 obtained from BuLi-mediated ring closure of 11 must be *trans*-fused, while the minor product derived from 13 must possess the *cis*-fused structure shown in formula 20. These experiments, in turn, established the relative configuration of each of the substances 18, 19 and 24.

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Footnotes

[†] For key reports on the isolation and structural elucidation of cyathane-type terpenoids, see ref. 1. The numbering system shown in 1 is that originally suggested by Ayer (ref. 2)

[‡] For previous reports related to the synthesis of cyathane-type terpenoids, see ref. 4.

§ All new compounds reported herein exhibit spectra in accord with the assigned structures and gave satisfactory elemental (C, H) analyses and/or molecular mass determinations (high resolution mass spectrometry).

¶ The homoallylic iodides 15 and 16 were prepared by reaction (room temperature) of the corresponding alcohols (ref. 6) with $Ph_3P \cdot I_2$ in CH_2Cl_2 in the presence of Et_3N .

|| Use of fewer than 4 equiv. of BuLi gave incomplete lithium-iodine exchange. Presumably, a portion of the alkyllithium was unavailable for the exchange reaction due to complexation with the ketal function present in the substrate.

** Also obtained from this reaction sequence was a significant amount (33%) of the methyl ether of the starting material. A similar pattern was observed for the other rearrangement processes. The isolated yields of the bicyclic alcohols **24–26** were in the range 46–53%.

^{††} We thank Dr Steven J. Rettig for carrying out these X-ray crystallographic structure determinations. Details of these studies will be reported elsewhere.

 \ddagger This conclusion, along with those given in the following paragraph, is based on the well-established stereochemically related characteristics of the [2,3]-sigmatropic rearrangement process [ref. 8(*b*)].

References

- (a) W. A. Ayer and S. P. Lee, Can. J. Chem., 1979, 57, 3332; (b)
 H. Shibata, T. Tokunaga, D. Karasawa, A. Hirota, M. Nakayama,
 H. Nozaki and T. Tada, Agric. Biol. Chem., 1989, 53, 3373; (c)
 H. Kawagishi, A. Shimada, R. Shirai, K. Okamoto, F. Ojima, H. Sakamoto, Y. Ishiguro and S. Furukawa, Tetrahedron Lett., 1994, 35, 1569; (d) R. S. Compagnone and D. J. Faulkner, J. Nat. Prod., 1995, 58, 145.
- 2 W. A. Ayer and H. Taube, Tetrahedron Lett., 1972, 1917.
- 3 W. A. Ayer, L. M. Browne, J. R. Mercer, D. R. Taylor and D. E. Ward, *Can. J. Chem.*, 1978, **56**, 717.
- 4 W. A. Ayer, D. E. Ward, L. M. Browne, L. T. J. Delbaere and Y. Hoyano, Can. J. Chem., 1981, 59, 2665; D. E. Ward, Can. J. Chem., 1987, 65, 2380; K. R. Dahnke and L. A. Paquette, J. Org. Chem., 1994, 59, 885.
- 5 E. J. Corey and D. Enders, Chem. Ber., 1978, 111, 1337.
- 6 E. Piers and A. V. Gavai, J. Org. Chem., 1990, 55, 2380.
- 7 W. C. Still, M. Kahn and A. Mitra, J. Org. Chem., 1978, 43, 2923.
- 8 (a) W. C. Still and A. Mitra, J. Am. Chem. Soc., 1978, 100, 1927; (b) T. Nakai and K. Mikami, Org. React., 1994, 46, 105.
- 9 M. Balestra and J. Kallmerten, Tetrahedron Lett., 1988, 29, 6901.

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