## **Green light-emitting diodes from poly(2-dimethyloctylsilyl-1,4=phenylenevinylene)**

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A novel silyl-substituted solvent processable poly(1,4-phe**nylenevinylene) (PPV) derivative, poly(2-dimethyloctylsilyl-1,4-phenylenevinylene) (DMOS-PPV) is synthesized by the dehydrohalogenation route from 2-dimethyloctylsilyl-1,4 bis(bromomethyl)benzene, and the light-emitting properties of the polymer are studied; single layer electroluminescent devices (ITO/polymer/Ca or Al) exhibit an emission maximum at 520 nm with internal quantum efficiency in the**  range  $0.2 - 0.3\%$ .

Light-emitting polymers have been extensively investigated in recent years since the Cambridge group first reported a green light-emitting diode (LED) using  $poly(1, 4$ -phenylenevinylene) (PPV) as an emitting layer.<sup>1-4</sup> Organic polymer LEDs have many advantages for the development of a large-area visible light-emitting display, because of the good processability, low operation voltage, fast response time and colour tunability over the full visible range by control of the HOMO-LUMO bandgap of the emissive layer. PPV has been most widely used as the emissive layer for the light-emitting diodes, and has been prepared through a thermal elimination process from a water-5 or organic-soluble precursor polymer.<sup>6-8</sup> Several organic solvent soluble PPV derivatives have been developed in order to improve processability.<sup>9-11</sup> Recently, Zhang et al. reported the improved quantum efficiency in green polymer light-emitting diodes with a silyl-substituted soluble PPV derivative, poly(2- PPV).<sup>11</sup> They reported<sup>12</sup> that  $\overline{CS}$ -PPV showed high quantum efficiency with an air-stable aluminium electrode by adding an electron transporting molecular dopant, 2-(4-biphenyl)- 5-(4-tert-butylphenyl)- 1,3,4-0xadiazole (PBD). **<sup>13314</sup>** cholestanoxy-5-thexylsilyl-1,4-phenylenevinylene) (CS-

The effects of silicon substitution on the luminescence properties are of interest in the field of polymer LEDs, and here we report the synthesis of a new silyl-substituted soluble PPV derivative, poly(2-dimethyloctylsilyl-1,4-phenylenevinylene) (DMOS-PPV). Single layer electroluminescent (EL) devices have been fabricated using this polymer as the emissive layer. The synthetic route is outlined in Scheme 1. Silylation of the Grignard reagent derived from **1** afforded the silyl derivative **2**  which after radical bromination gave the dibromo-compound **3.1.** Dehydrohalogenation condensation polymerization afforded DMOS-PPV 4.15,16

DMOS-PPV 4 is completely soluble in common organic solvents such as chloroform, tetrahydrofuran and toluene without evidence of gel formation.

Fig. 1 shows the UV-VIS, photoluminescence (PL) and EL spectra of the DMOS-PPV film. DMOS-PPV 4 shows a slightly narrower absorption band compared with PPV. The absorption maximum and edge of the DMOS-PPV are at ca. 414 and *500*  nm, respectively, at room temperature. These positions are blueshifted compared with those of the unsubstituted PPV (420 and 530 nm, respectively), presumably owing to the steric effect of the bulky dimethyloctylsilyl group.

DMOS-PPV 4 shows an emission maximum at ca. 520 nm which corresponds to the green region. The absolute photoluminescence quantum efficiency for a solid film of DMOS-PPV was 60%. By comparison, the reported PL efficiencies of PPV and MEH-PPV are 27 and  $15\%$ , respectively.<sup>17</sup>

Fig. 2 shows the current density-electric field characteristics measured for a typical ITO/DMOS-PPV/Al device with film thickness of 700  $\hat{A}$ . The forward current density increases with increasing forward bias field and the curve shows typical diode



**Fig. 1 UV-VIS (crosses), PL (open circles) of DMOS-PPV film and EL**  (solid circles) of the ITO/DMOS-PPV/Al device

characteristics. The voltage dependence of emission intensity from the device shows that light emission becomes observable at a bias of about 15 V at a current density of 0.93 mA  $cm^{-2}$ . The devices showed reproducible internal quantum efficiencies of 0.2% (ca. 0.05% external efficiency). $\ddagger$ 

Fig. 3 shows the current density-field characteristics of an ITO/DMOS/Ca device (700 A thickness). The threshold voltage of the device was about 11 V at a current density of 1.8  $mA \, cm^{-2}$ . The measured maximum internal quantum efficiency of the diode was 0.3% *(ca.* 0.1% external efficiency). These values compare favourably with the efficiency of the single layer green polymer LED reported by Son et al.<sup>8</sup>

Recently we have fabricated multilayer EL devices with DMOS-PPV and various charge-transporting materials, such as **2-(4-biphenylyl)-5-(4-tert-butylphenyl)-** 1,3,4-0xadiazole (PBD) or poly(aromatic oxadiazo1e)s. These devices showed highly improved quantum efficiencies using an aluminium cathode.

The DMOS-PPV film has good processability and the high quantum efficiency may make it a good candidate for application in polymer LEDs.



**Fig. 2** Current density-electric field and light intensity-voltage (inset) characteristics of ITO/DMOS-PPV/Al device



**Fig.** 3 Current density-electric field and light intensity-voltage (inset) characteristics of ITO/DMOS-PPV/Ca device

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## **Footnotes**

t *Selected spectroscopic data.* 2, IH NMR (CDC13,200 MHz) **6** 7.28 **(1 H,**  s), 7.09 (2 H, s), 2.43 (3 H, s), 2.34 (3 H, s), 1.41-1.14 (12 H, m), 0.97-0.75 *(5* H, m), 0.33 (6 H, s); **3,** 'H NMR (CDC13, 200 MHz) **6** 7.48 (1 H, s), 7.43 (2 H, s), 4.61 (2 H, s), 4.48 (2 H, s), 1.43-1.19 (12 H, m), 0.99-0.81 (5 H, m), 0.42 (6 H, s); DMOS-PPV 4, GPC (polystyrene standard) showed a  $M_w$ of  $1.1 \times 10^6$  and polydispersity index of 7.2. (Found: C, 77.5; H, 10.05. Calc. C, 79.34; H, 10.36%). FTIR (NaCl)  $v_{max}/cm^{-1}$  2955, 2922, 2853, 1729, 1468, 1377, 1251, 1137, 1066,961, 836.

**4** The internal efficiency is a factor of *2n2* larger than the external efficiency where  $n$  is the refractive index of the emissive layer.<sup>18</sup>

## **References**

- 1 J. H. Burroughes, D. D. C. Bradley, A. R. Brown, R. N. Marks, K. Mackay, R. H. Friend, P. L. Bum and **A.** B. Holmes, *Nature,* 1990, 347, 539.
- 2 N. C. Greenham, **S.** C. Moratti, D. D. C. Bradley, R. H. Friend and A. B. Holmes, *Nature,* 1993, 365, 628.
- 3 G. Gustafsson, Y. Cao, G. M. Treacy, F. Klavetter, N. Colaneri and A. J. Heeger, *Nature,* 1992, 357 477.
- 4 P. L. Bum, A. B. Holmes, A. Kraft, D. D. C. Bradley, **A.** R. Brown,
- R. H. Friend and R. W. Gymer, *Nature,* 1992,356,47.
- *5*  R. A. Wessling, *J. Polym. Sci., Polym. Symp.,* 1985, 72, *55.*
- 6 F. Louwet, D. Vanderzande, J. Gelan and J. Mullens, *Macromolecules,*  1995,28, 1330.
- 7 P. L. Bum, D. D. C. Bradley, R. H. Friend, D. A. Halliday, A. B. Holmes, R. W. Jackson and A. Kraft, *J. Chem. SOC., Perkin Trans. I,* 1992, 3225.
- 8 **S.** Son, A. Dodabalapur, **A.** J. Lovinger and M. E. Galvin, *Science,* 1995, **269,** 376.
- 9 F. Wudl, P. M. Allemand, G. Srdanov, Z. Ni and D. McBranch, in *Materials for Nonlinear Optics: Chemical Perspectives,* ed. **S.** R. Marder, J. E. Sohn and G. D. Stucky, **ACS** *Symp. Ser.,* 1991,45, 683.
- 10 F. Wudl, **S.** Hoger, C. Zhang, K. Pakbaz and **A.** J. Heeger, *Polym. Prepr.,* 1993, 34, 197.
- 11 **S.** Hoger, J. J. McNamara, **S.** Schricker and F. Wudl, *Chem. Muter.,*  1994, **6,** 171.
- 12 C. Zhang, **S.** Hoger, K. Pakbaz, F. Wudl and **A.** J. Heeger, *J. Electron. Muter.,* 1994, 23, 453.
- 13 P. L. Bum, A. B. Holmes, A. Kraft, **A.** R. Brown, D. D. C. Bradley and R. H. Friend, *Muter. Res. SOC. Symp. Proc.,* 1992,247,447.
- 14 A. R. Brown, D. D. C. Bradley, J. H. Burroughes, R. H. Friend, N. C. Greenham, P. L. Bum, A. B. Holmes and A. Kraft, *Appl. Phys. Lett.,* 1992, **61,** 279.
- 15 H. G. Gilch and W. L. Wheelwright, *J. Polym. Sci. A-1,* 1966, **4,**  1337.
- 16 W. **S.** Swatos and B. Gordon, *Polym. Prepr.,* 1990,31, 505.
- 17 N. C. Greenham, I. D. W. Samuel, G. R. Hayes, R. T. Philips, **Y. A.** R. R. Kessener, **S.** C. Moratti, **A. B.** Holmes and R. H. Friend, *Chem. Phys. Lett.,* 1995, 241, 89.
- 18 N. C. Greenham, R. H. Friend and D. D. C. Bradley, *Adv. Muter.,* 1994, **6,** 491.
- 19 D. **A.** Fletcher, R. F. McMeeking and D. Parkin, *J. Chem. 1nf. Comput. Sci.,* 1996, 36, 746.

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