

# Catalytic asymmetric hydroboration–amination

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## A one-pot asymmetric synthesis of primary amines from vinylarenes *via* catalytic hydroboration is described.

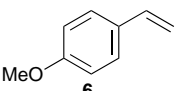
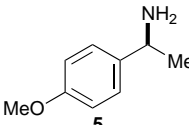
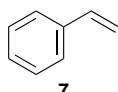
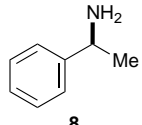
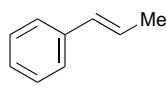
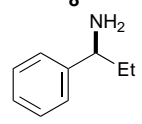
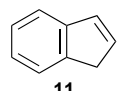
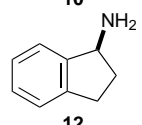
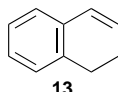
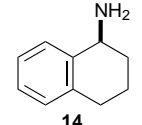
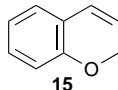
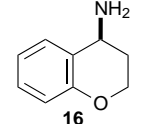
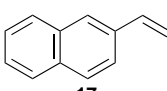
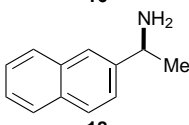
Despite considerable efforts to find alternatives, catecholborane is the most useful borane and rhodium complexes are the most useful catalysts in catalytic hydroboration.<sup>1</sup> Synthetic application of the resulting catecholborane adducts is however limited to oxidation with basic H<sub>2</sub>O<sub>2</sub>, and hence the asymmetric version<sup>2</sup> is in turn limited to the synthesis of secondary alcohols, for which many excellent alternative routes are already available.<sup>3</sup> Given the wealth of functional group transformations that can be performed with alkylboranes,<sup>4</sup> there is an incentive for developing further synthetic outlets for the catalysed case. Our initial efforts were directed to the reaction of isolated 1-phenylethylcatecholboronates† with electrophilic aminating agents such as CIMgN(Me)OSiMe<sub>3</sub>, and although some success was achieved in controlling enantioselectivity, the chemoselectivity for secondary amine formation over alcohol was moderate.<sup>5</sup>

In considering a direct route from alkylcatecholboronates to amines, we were guided by prior results from Brown's research in which alkylboronates, which do not react directly with the standard aminating agent H<sub>2</sub>NOSO<sub>3</sub>H,<sup>6</sup> were converted by a practical, albeit multistep, route into amines *via* the more reactive dialkylborinate esters. In contrast, excess R<sub>2</sub>Zn is reported to effect clean transmetalation of a range of dialkylvinylboranes,<sup>7a</sup> trialkylboranes,<sup>7b</sup> alkylboronates and alkyldithio-boronates,<sup>7c</sup> thus providing a route to less accessible alkenyl and alkylzinc compounds. Reaction intermediates have not been reported, although it is likely that the trialkylborane is an intermediate in boronate ester transmetalations (*vide infra*).

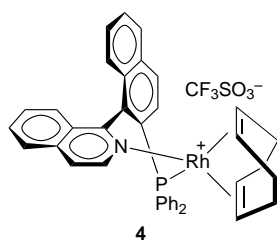
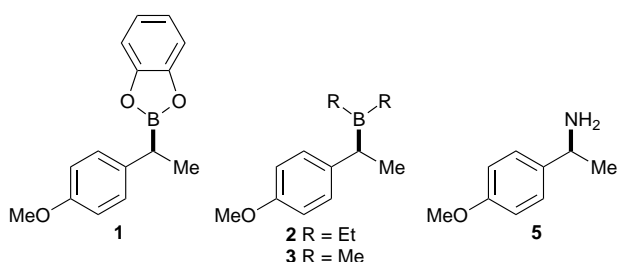
Our own initial experiments were conducted in toluene solution by reacting an isolated sample of catecholboronate ester **1** with 1 equiv. of Et<sub>2</sub>Zn; we were delighted to observe that a clean and quantitative displacement took place, resulting in the transfer of two ethyl groups to boron to form **2** and removal of the Zn from the reaction medium as an insoluble catecholate.<sup>8</sup> The resulting borane was characterized by <sup>1</sup>H NMR [δ<sub>H</sub>(C<sub>7</sub>D<sub>8</sub>) 1.20 (q), 0.95(t)] and MS (*m/z* 205). Oxidation of the borane **2**

under standard conditions (H<sub>2</sub>O<sub>2</sub>, NaOH) demonstrated that the configurational integrity of the stereogenic centre had been maintained. At this point we became aware of an early report that primary alkylcatecholboronates react with 2 equiv. of RMgBr by complete displacement of the catechol residue giving the mixed trialkylborane.<sup>9</sup> In accord with this, complete displacement of catecholate without precipitation was observed when the reaction between MeMgCl and borane **1** was carried

Table 1 Hydroboration–amination of vinylarenes<sup>a</sup>

Reactant	Product	Regio-selectivity (%) <sup>b</sup>	Isolated yield (%)	Ee (%) <sup>c</sup>
		> 98	56	98
		> 98	54	87
		> 98	50	90
		> 98 <sup>d</sup>	61	77
		96	51	97
		98 <sup>e</sup>	64	89
		92 <sup>f</sup>	62	90

<sup>a</sup> Hydroboration–amination experiments as per typical protocol (see footnote ‡). <sup>b</sup> Minor component is regioisomeric primary amine. <sup>c</sup> Ees of amines or their acetamides were analysed by GC using a 25 m CHIRAL-DEX bonded silica column at 140–150 °C, achieving baseline separation in all cases. <sup>d</sup> From isolated organoborane. <sup>e</sup> Not previously obtained in enantiomerically enriched form, [α]<sub>D</sub><sup>23</sup> –55.21 (c 0.36, CHCl<sub>3</sub>) for acetyl amino derivative; configuration assigned by comparison with (*S*)-chroman-4-ol (ref. 16) formed by hydroboration–oxidation. <sup>f</sup> Analysis by <sup>1</sup>H NMR using Eu(tfc)<sub>3</sub>.



out in thf solution, the only product being **3** [ $\delta_{\text{C}}$  13.0 ( $J_{\text{BC}}$  82 Hz)].<sup>10</sup> These displacement reactions were also shown to be occur with other arylalkylboronates prepared by catalytic hydroboration of vinylarenes with catecholborane.

With a clean route to enantiomerically enriched trialkylboranes in hand, attention was then turned to developing a synthetically useful amination procedure. It is known that in aminations with  $\text{H}_2\text{NOSO}_3\text{H}$ , methyl is a poor migrating group,<sup>11</sup> and a protocol was developed accordingly. It was established in an initial experiment that successive addition of 2 equiv. of  $\text{MeMgCl}$  and 3 equiv. of  $\text{H}_2\text{NOSO}_3\text{H}$  to an isolated sample of borane **1** (86% ee) in thf gave amine **5** in 83% ee and 65% yield. Catalytic hydroboration was then carried out for several vinylarenes in thf solution at ambient temperature, employing the rhodium complex of (*S*)-quinap (**4**) in the manner previously described.<sup>12</sup> On completion of the hydroboration reaction, 2 equiv. of  $\text{MeMgCl}$  (3 m solution in thf) were added followed after 30 min by 3 equiv. of  $\text{H}_2\text{NOSO}_3\text{H}$ . After stirring for 12 h the amine was isolated either directly, or as its acetamide ( $\text{MeCOCl}$ ) and the ee analysed as described in Table 1. As far as can be determined, the ee values reflect the enantioselectivity of the catalytic hydroboration step, since  $\text{H}_2\text{O}_2\text{-OH}^-$  oxidation at the trialkylborane stage provides a secondary alcohol whose enantiomeric purity is the same or slightly lower than that of the primary amine (within experimental error). For the acyclic side-chain amines derived from alkenes **6**, **7**, **9** and **17**, the results follow quite closely those observed previously in hydroboration-oxidation, but superior results have been obtained for the slower-reacting dihydronaphthalene **13** and 4-chromene **15** by very careful attention to catalyst purity and freshness, although indene **11** was somewhat inferior. A one-stage asymmetric synthesis of 1-aminotetralin **14** in 97% ee is of potential practical significance.

In summary, we have developed a method for the catalytic asymmetric synthesis of primary amines from alkenes that complements existing methods, which are most commonly based on imine<sup>13</sup> or enamide reduction,<sup>14</sup> or allylic alkylation procedures.<sup>15</sup> The activation of catecholborane adducts offers the possibility of further extending the range of catalytic hydroboration (*e.g.* to Suzuki couplings and other C-C bond forming reactions) and provides an incentive for us to improve the scope and selectivity of the catalytic reaction itself.

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#### Footnote

† IUPAC name: *ortho*-phenylene 1-phenylethylboronate.

‡ Freshly prepared complex **4** (4.0 mg, 0.005 mmol, 1.0 mol%) and thf (0.5 ml) were placed in a vial under argon with the alkene (0.5 mmol). Freshly distilled catecholborane (53.3  $\mu\text{l}$ , 59.5 mg, 0.5 mmol) was added with

stirring, then left for 1 h.  $\text{MeMgCl}$  (3 m in thf, 333  $\mu\text{l}$ , 1.0 mmol) was added and the solution stirred for 30 min. The reactant solution was added to pre-dried  $\text{H}_2\text{NOSO}_3\text{H}$  (169 mg, 1.5 mmol), with 0.7 ml of thf or diglyme, and stirred under argon for *ca.* 15 h.  $\text{HCl}$  (1 m, 2 ml) was added, and the mixture poured into  $\text{H}_2\text{O}$  (4 ml). The aqueous layer was extracted with  $\text{Et}_2\text{O}$  ( $3 \times 20$  ml) and then was made strongly alkaline with  $\text{NaOH}$  (1 m, 3 ml). The mixture was then extracted with  $\text{Et}_2\text{O}$  ( $3 \times 20$  ml). The  $\text{Et}_2\text{O}$  extracts were combined and dried with magnesium sulfate, and the solvent removed *in vacuo*.

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