## Non-linear optical properties of the inorganic cluster $[(\eta-C_5H_5)C_0Fe_2Se(CO)_6]$

### Sudeep Banerjee,<sup>a</sup> G. Ravindra Kumar,<sup>\*a</sup><sup>†</sup> Pradeep Mathur<sup>\*b</sup><sup>‡</sup> and P. Sekar<sup>b</sup>

<sup>a</sup> Tata Institute of Fundamental Research, Homi Bhabha Road, Mumbai 400 005, India

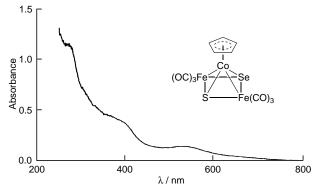
<sup>b</sup> Department of Chemistry, Indian Institute of Technology, Powai, Mumbai 400 076, India

# Non-linear refraction and absorption have been studied in a mixed-metal, mixed chalcogenide cluster. The cluster displays optical limiting significantly superior to that displayed by $C_{60}$ .

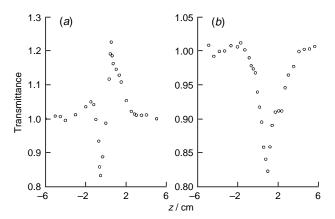
The relentless search for materials with good non-linear optical response has led to impressive strides in understanding existing materials as well as the design and synthesis of new types of molecules.<sup>1</sup> In particular, there has been great interest in finding better non-linear media for the process of optical limiting.<sup>2</sup> Organic systems like dyes, porphyrins and fullerenes have been explored in considerable detail.<sup>2–4</sup> Some transition-metal cluster compounds have also been investigated for their potential for response comparable to or exceeding that of C<sub>60</sub>.<sup>5–8</sup>

Cluster compounds which contain one or more bridging chalcogen ligands have recently been attracting considerable attention.<sup>9</sup> It is possible to design the construction of certain classes of mixed-metal clusters in which different combinations of chalcogen ligands can be incorporated.<sup>10</sup> Recently we have synthesised and structurally characterised the first example of a mixed-metal cluster containing both S and Se ligands  $[(\eta-C_5H_5)CoFe_2(CO)_6(\mu_3-S)(\mu_3-Se)].^{11}$  Here, we report on its optical limiting characteristics. Among the key requirements for a good optical limiter are (*a*) large transmission at low intensities followed by low transmission at high intensities and (*b*) a low threshold for the onset of non-linear attenuation of the input beam.<sup>2</sup> We show that, on both these counts, the present cluster displays superior performance compared to other materials.

The linear absorption spectrum of a solution of  $[(\eta-C_5H_5)Co-Fe_2SSe(CO)_6]$  in *n*-hexane is shown in Fig. 1. There is weak absorption in the visible and near-IR regions. Commonly used laser wavelengths (Nd-YAG, dye and diode) exist in these regions and hence limiting applications are important here. The non-linear optical properties of the solution were measured by the Z-scan technique<sup>12</sup> at 1064 and 532 nm using an Nd-YAG laser emitting 35 ps pulses at a 10 Hz repetition rate. Typical Z-scans are shown in Fig. 2. The large peak to valley difference (nearly 40%) in the closed aperture case [Fig. 2(*a*)] indicates considerable non-linear refraction and the occurrence of the peak at positive *z* indicates positive non-linearity or selffocussing. The non-linear refraction coefficient  $\gamma$  can be extracted from the difference in transmission between the peak and the valley.<sup>12</sup> Table 1, which summarises the experimental parameters and the extracted non-linear coefficients, indicates that our  $\gamma$  value at 532 nm is two orders of magnitude larger than the values reported by Shi and coworkers for other inorganic clusters at the same wavelength.<sup>13</sup> The  $\gamma$  at 1064 nm for our



**Fig. 1** Linear absorption spectrum of  $[(\eta-C_5H_5)CoFe_2SSe(CO)_6]$  in *n*-hexane. The inset shows the structure of this cluster.



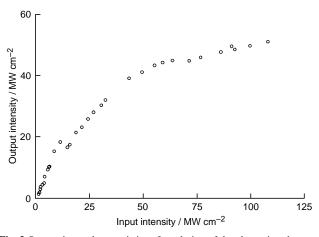
**Fig. 2** Results of a Z-scan at 532 nm. A lens with 50 cm focal length was used, giving a focussed spot radius of about 25  $\mu$ m. The sign convention for z is as defined in ref. 12. z = 0 is the focus. The fluence at focus is 0.80 J cm<sup>-2</sup>. (a) Non-linear refraction, (b) non-linear absorption.

**Table 1** Experimentally determined values for  $[(\eta-C_5H_5)CoFe_2SSe(CO)_6]$  and those reported for other materials. The maximum (at focus) intensity in our *Z*-scan experiment was 23 GW cm<sup>-2</sup>, corresponding to a maximum fluence of 0.8 J cm<sup>-2</sup>. The sample concentration was 50 µmol ml<sup>-1</sup>;  $\gamma$  is the non-linear refractive index coefficient (see ref. 12).  $F_t$  is the limiting threshold fluence at 532 nm

Material	$\gamma$ (532 nm)/ $m^2~W^{-1}$	$\gamma (1064 \text{ nm})/m^2 \text{ W}^{-1}$	$F_{\rm t}$ (with ap.)/ J cm^{-2}	$F_{\rm t}$ (no. ap.)/ J cm <sup>-2</sup>	Ref.
$[(\eta - C_5H_5)CoFe_2SSe(CO)_6]$	$2.7  imes 10^{-14}$	$-5.8 \times 10^{-18}$	$2.1 \times 10^{-3}$	$11.0 \times 10^{-3}$	This work
$C_{60}$	$-9  imes 10^{-17}$	_	_	$175 \times 10^{-3}$	14, 15
[MoOS <sub>3</sub> (CuNCS) <sub>3</sub> ] <sup>2-</sup>	$-2.3  imes 10^{-16}$		_	7.0	13, 16
$[Mo_8O_8Cu_{12}S_{24}]^{4-}$	$-3.5  imes 10^{-16}$	_	_		13
$[Cu_4(SPh)_6]^{2-}$	$5.0  imes 10^{-17}$	_	_		13

cluster is also of a significant magnitude (no other data exist for a comparison to be made at this wavelength). Note that the sign of the non-linearity changes at 1064 nm, where the non-linear refraction is found to be due to self-defocussing. Moving on to the open aperture data [Fig. 2(*b*)], it is evident that there is nonlinear absorption, indicated by the dip in the transmission as the sample is moved towards the focus. The non-linear absorption of about 16% at the maximum fluence of 0.80 J cm<sup>-2</sup>, is significant considering the low fluence levels we have used. The solvent *n*-hexane did not give any signals. It is also clear from Table 1 that the  $\gamma$  is considerably superior to that measured for C<sub>60</sub>.<sup>14</sup>

The significant non-linear refraction as well as non-linear absorption displayed by this cluster have been used to design optical limiters with and without apertures. Fig. 3 shows the measurements of transmission at 532 nm as a function of input energy density (fluence) with an aperture of 1 mm in the output beam. The input pulse was focussed by a lens with a focal length of 50 cm. The concentration of the sample was adjusted such that the transmission of the limiter at low fluences is quite large (80%). The transmission starts decreasing as the input increases to a fluence of 1.0 mJ cm<sup>-2</sup> (28 MW cm<sup>-2</sup>) and comes down to 30% at 3.5 mJ cm<sup>-2</sup>. The onset of non-linear transmission in other inorganic clusters has been found to occur at values that are two orders of magnitude larger than that for the present cluster.<sup>7</sup> The benchmark limiting material,  $C_{60}$ , also shows onset of non-linear transmission at 80 mJ cm<sup>-2.15</sup> The threshold for limiting,<sup>2</sup> defined as the fluence at which the transmission decreases to half of the value at low input fluences, is found to be 2.1 mJ cm<sup>-2</sup> for our cluster. This value is about 250 times smaller than the smallest value obtained for other inorganic



**Fig. 3** Output–input characteristics of a solution of the cluster in *n*-hexane at 532 nm with an aperture of 1 mm in the transmitted beam. Linear (low intensity) transmission of the limiter was 80%. The sample was contained in a 1 mm path length quartz curvette.

clusters studied recently.<sup>16</sup> Even without an aperture (the open aperture case), our limiting threshold is 11 mJ cm<sup>-2</sup>, which is a factor of over 45 times smaller compared to the other clusters<sup>16</sup> and about 15 times smaller than that for C<sub>60</sub>. An additional advantage of our cluster is that it has a large transmission at low input intensities, a characteristic required for an ideal optical limiter.

In conclusion, we have shown that the mixed-metal, mixed chalcogenide cluster  $[(\eta-C_5H_5)CoFe_2SSe(CO)_6]$  displays large optical non-linearity. We have demonstrated that it has optical limiting characteristics which are considerably superior to  $C_{60}$  and other recently investigated inorganic clusters.

### Footnotes

- † E-mail: grk@tifrvax.tifr.res.in
- ‡ E-mail: mathur@ether.chem.iitb.ernet.in

#### References

- Materials for Nonlinear Optics—Chemical Perspectives, ed. S. R. Marder, J. E. Sohn and G. D. Stucky, ACS Symp. Ser., American Chemical Society, Washington, DC, 1991, 455; J.-L. Bredas, C. Adant, P. Tackx, A. Persoons and B. M. Pierce, Chem. Rev., 1994, 94, 243.
- 2 L. W. Tutt and T. F. Boggess, Prog. Quant. Electron., 1993, 17, 299.
- 3 M. Ravikanth and G. R. Kumar, Curr. Sci., 1995, 68, 1010.
- 4 G. R. Kumar and F. A. Rajgara, *Appl. Phys. Lett.*, 1995, **67**, 3871; A. Sevian, M. Ravikanth and G. R. Kumar, *Chem. Phys. Lett.*, in the press.
- 5 L. W. Tutt and S. W. McCahon, Opt. Lett., 1990, 15, 700.
- 6 S. Shi, W. Ji, S. H. Tang, J. P. Lang and X. Q. Xin, J. Am. Chem. Soc., 1994, **116**, 3615; S. Shi, W. Ji, J. P. Lang and X. Q. Xin, J. Phys. Chem., 1994, **98**, 3750.
- 7 W. Ji, S. Shi, H. J. Du, P. Ge, S. H. Tang and X. Q. Xin, *J. Phys. Chem.*, 1995, **99**, 17 297.
- 8 W. Ji, H. J. Du, S. H. Tang and S. Shi, J. Opt. Soc. Am. B, 1995, 12, 876.
- 9 L. C. Roof and J. W. Kolis, *Chem. Rev.*, 1993, 93, 1037; K. H. Whitmire, *Coord. Chem.*, 1988, 17, 95; M. A. Ansari and F. A. Ivers, *Coord. Chem. Rev.*, 1990, 100, 223; P. Mathur, D. Chakravarty and I. J. Mavunkul, *J. Cluster. Sci.*, 1993, 4, 351.
- 10 P. Mathur, M. M. Hossain, S. B. Umbarkar, C. V.V. Satyanarayana, A. L. Rheingold, L. M. Liable-Sands and G. P. A. Yap, *Organometallics*, 1996, **15**, 1898; P. Mathur and P. Sekar, *Chem. Commun.*, 1996, 727.
- 11 P. Mathur, P. Sekar, C. V. V. Satyanarayana and M. F. Mahon, Organometallics, 1995, 14, 2115.
- 12 M. Sheik-Bahae, A. A. Said, T.-H. Wei, D. J. Hagan and E. W. Van Stryland, *IEEE J. Quant. Elec.*, 1990, 26, 760.
- 13 S. Shi, X. Zhang and X. F. Shi, J. Phys. Chem., 1995, 99, 14911.
- 14 S. Couris, E. Koudoumas, A. A. Ruth and S. Leach, J. Phys. B, 1995, 28,
- 4537.
  15 C. Li, R. Wang, Y. Song and Y. Wang, J. Opt. Soc. Am. B, 1994, 11, 1356.
- 16 D. Long, S. Shi, X. Xin, B. Luo, L. Chen, X. Huang and B. Kang, J. Chem. Soc., Dalton Trans., 1996, 2617.

Received, 4th November 1996; Com. 6/07485E