Two free radical routes for the preparation of novel difluoromethylene-linked serine-*O***-glycopeptide analogues**

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The first examples of *gem***-difluoromethylene-linked analogues of serine-derived glycopeptides have been prepared in a highly stereoselective manner using two complementary free radical chain reactions.**

The importance of the carbohydrate moiety in molecular recognition processes involving glycoproteins1,2 and the considerable potential of glycosylated peptide drugs³ have stimulated intense current research interest in the synthesis of modified glycopeptides, where the customary *O*- or *N*-glycosidic bond linking the amino acid component to the sugar portion is replaced by a carbon-based unit of greater stability and resistance towards enzymatic deglycosylation. Thus, methods have been described, both for the synthesis of *C*-glycosyl amino acids in which the anomeric carbon is directly bound to the α -carbon of the amino acid^{4–8} and for methylene-bridged glycopeptides $9-14$ in which the additional methylene group can serve as a heteroatom mimic. Even the latter, however, lead to derivatives with a truncated chain when compared with the corresponding serine-*O*-glycopeptides.

Our objective in this area was to achieve site specific replacement of the exocyclic anomeric oxygen atom in a serinederived glycopeptide fragment by a difluoromethylene group. This tetrahedral and electronegative unit has been shown to be of particular value as an oxygen atom replacement in several difluoromethylene phosphonate analogues of naturally occurring phosphates.15

As shown in Scheme 1, our intention was to capitalise on the chemical versatility of readily prepared carbohydrate *gem*difluoro enol ethers¹⁶ **1**, either as radicophiles¹⁷ for the addition of an amino acid-derived radical to give the anomeric radical **2** (route A), or *via* their thiophenol adducts **3**, 18 as apposite precursors for difluoroalkyl radical trapping by a dehydroalanine derivative to give the carbon centred radical **4** around the amino acid unit (route B).

Thus, in a preliminary study, slow addition of tributylstannane and AIBN as initiator to a refluxing benzene solution of carbohydrate 5 and the iodoalanine reagent^{19,20} 6 developed by Baldwin21,22 afforded out first example of a difluoromethylenelinked serine analogue **7**, albeit in a disappointing 14% yield (Scheme 2). Efforts to optimise this reaction or to extend it to other *gem*-difluoro enol ether substrates were unsuccessful.

We reasoned that a much more compact and conformationally restricted radical reagent was required, and elected to prepare the cyclic bis(trifluoromethyl)oxazolidinone derivative **10** which should find general use as a convenient bromoalanine reagent. Thus, as shown in Scheme 3, protection of aspartic acid **8** with gaseous hexafluoroacetone23 was followed by conversion to the acid chloride24 **9** which was transformed *via* the acylthiohydroxamate to the desired bromide **10** using the radical variant of the Hünsdieker reaction.²⁵

Addition of **10** to *gem*-difluoro enol ethers under tributylstannane-mediated conditions proceeded in acceptable yields with a range of substrates (Scheme 4), with abstraction of the hydrogen atom occurring exclusively from the less hindered convex face^{14,26} in the case of the gulose $5 \rightarrow 11$ and erythrose $12 \rightarrow 13$ derivatives. Reaction of the ribose derivative 14 however produced an equimolar mixture of both anomers **15**, and on replacement of the methyl ether by the bulky *tert*butyldimethylsilyl group, no addition products were observed.

The utility of the labile bis(trifluoromethyl)oxazolidinone protecting group was demonstrated by simple treatment of **11** with l-phenylalanine *tert*-butyl ester to afford the dipeptide **16** (Scheme 5).

Finally, we have also carried out a preliminary study of the alternative approach outlined in Scheme 1 route B, which effectively circumvents problems created by the steric bulk of the carbohydrate. The chiral dehydroalanine derivative (2*R*)-*N*benzoyl-2-(*tert*-butyl)-4-methylidene-1,3-oxazolidin-5-one **17** prepared from (S) -alanine and used by Beckwith¹³ as a radicophile, was accordingly selected to ensure diastereoselectivity at the α -carbon atom of the amino acid unit. In the final analysis, we also elected to use the phenylselanyl congeners of our previously described18 thiophenol adducts in order to

Scheme 2 *Reagents and conditions*: i, AIBN, Bu₃SnH (slow addition), benzene, reflux

Scheme 3 Reagents and conditions: i, (CF₃)₂CO, DMSO, 50 °C; ii, SOCl₂, reflux; iii, 2-sulfanylpyridine 1-oxide sodium salt, THF, -15 °C; iv, BrCCl₃, *hv* (tungsten)

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facilitate generation of the required difluoroalkyl radicals. The results for the three carbohydrates **18**–**20** in Scheme 6 indicate that this tributylstannane-mediated coupling gives consistently higher yields than those obtained by the previous method. The successful coupling of the hindered ribose derivative **20** is particularly noteworthy since the alternative coupling strategy had been unsuccessful. From a stereochemical standpoint, the ratio of the two diastereoisomers produced in each reaction is consistent with the results obtained by Beckwith¹³ with the same chiral auxiliary.

Scheme 4 Reagents and conditions: i, AIBN, Bu₃SnH (slow addition), benzene, reflux. *a* Obtained as a 1 : 1 mixture of diastereoisomers at the anomeric centre.

Scheme 5 *Reagents and conditions*: i, l-phenylalanine *tert*-butyl ester, $Et₂O$, room temp., 24 h

Scheme 6 *Reagents and conditions*: i, AIBN, Bu₃SnH (slow addition), benzene, reflux

In summary, the two complementary free radical routes described herein provide a convenient approach for the synthesis of a wide range of *C*-glycosylated amino acid units in which the difluoromethylene group serves not only as a potential heteroatom mimic but also contributes to providing the correct number of atoms in the crucial linkage between the carbohydrate and amino acid components of the structures.

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References

- 1 H. Lis and N. Sharon, *Eur. J. Biochem*., 1993, **218**, 1.
- 2 J. Montreuil, in *Comprhensive Biochemistry*, ed. A. Neuberger and L. L. van Deene, Elsevier, Amsterdam, 1982, vol. 19 BII, 1.
- 3 J. F. Fisher, A. W. Harrison, G. L. Bundy, K. F. Wilkinson, B. D. Rush and M. J. Ruwart, *J. Med. Chem*., 1991, **34**, 3141.
- 4 G. Rassu, F. Zanardi, L. Battistini and G. Casiraghi, *Tetrahedron: Asymmetry*, 1995, **6**, 371.
- 5 L. Colombo, G. Casiraghi and A. Pattalis, *J. Org. Chem*., 1991, **56**, 3897.
- 6 G. Casiraghi, L. Colombo, G. Rassu and P. Spanu, *J. Org. Chem*., 1991, **56**, 6523.
- 7 P. Garner and J. M. Park, *J. Org. Chem*., 1990, **55**, 3772.
- 8 A. G. M. Barrett and S. A. Lebold, *J. Org. Chem*., 1990, **55**, 3853. 9 C. R. Bertozzi, P. D. Hoepricht and M. D. Berdnarski, *J. Org. Chem*.,
- 1992, **57**, 6092. 10 M. K. Gurjar, A. S. Mainkar and M. Syamala, *Tetrahedron: Asymmetry*,
- 1993, **4**, 2343.
- 11 G. Simchen and E. Pürkner, *Synthesis*, 1990, 524. 12 H. Kessler, V. Wittmann, M. Köck and M. Kottenhahn, *Angew. Chem.*,
- *Int. Ed. Engl*., 1992, **31**, 902.
- 13 J. R. Axon and A. L. J. Beckwith, *J. Chem. Soc., Chem. Commun*., 1995, 549.
- 14 B. Giese, *Angew. Chem., Int. Ed. Engl*., 1989, **28**, 969.
- 15 G. M. Blackburn, *Chem. Ind. (London)*, 1981, 134; G. M. Blackburn, D. E. Kent and F. Kolkmann, *J. Chem. Soc., Perkin Trans*., 1984, 1119; R. D. Chambers, R. Jaouhari and D. O'Hagen, *Tetrahedron*, 1989, **45**, 5101.
- 16 J. S. Houlton, W. B. Motherwell, B. C. Ross, M. J. Tozer, D. J. Williams and A. M. Z. Slawin, *Tetrahedron*, 1993, **49**, 8087.
- 17 T. F. Herpin, W. B. Motherwell and M. J. Tozer, *Tetrahedron: Asymmetry*, 1994, **5**, 2269.
- 18 W. B. Motherwell, B. C. Ross and M. J. Tozer, *Synlett*, 1989, 68.
- 19 J. A. Bajgrowicz, A. El Hallaoui, R. Jacquier, C. Pigierre and
- P. Viallefont, *Tetrahedron*, 1985, **41**, 1833. 20 R. F. W. Jackson, N. Wishart, A. Wood, K. James and M. J. Wythes,
- *J. Org. Chem*., 1992, **57**, 3397. 21 J. E. Baldwin, R. E. Adlington, D. J. Birch, J. A. Crawford and
- J. B. Sweeney, *J. Chem. Soc., Chem. Commun*., 1986, 1339. 22 J. E. Baldwin, R. Fieldhouse and A. T. Russel, *Tetrahedron Lett*., 1993,
- **34**, 5491. 23 K. Burger, H. Neuhauser and M. Rudolph, *Chem.-Ztg*., 1990, **114**,
- 249 24 K. Burger, M. Gold, H. Neuhauser and M. Rudolph, *Chem.-Ztg*., 1991, **115**, 77.
- 25 D. H. R. Barton, Y. Hervé, P. Potier and J. Thierry, *Tetrahedron*, 1988, **44**, 5479; D. H. R. Barton, D. Crich and W. B. Motherwell, *Tetrahedron*, 1985, **41**, 3901.
- 26 D. H. R. Barton, S. D. Gero, Quiclet-Sire and M. Samadi, *Tetrahedron: Asymmetry*, 1994, **5**, 2123.
- 27 A. L. J. Beckwith, S. G. Pyne, B. Dikic, C. L. L. Chai, P. A. Gordon, B. W. Skelton, M. J. Tozer and A. H. White, *Aust J. Chem*., 1993, **46**, 1425.

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