# **Cyclophyne chemistry: synthesis and study of an octacobalt complex of** [8.8]paracyclophaneoctayne

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### Use of a palladium/cobalt/copper alkyne reaction strategy provides the cobalt-stabilized complex of a highly strained cyclophane.

The recent discovery of the fullerenes1 and related carbon nanotubes<sup>2</sup> has prompted a flurry of activity directed toward the synthesis and study of curved, fully conjugated  $\pi$ -systems.<sup>3</sup> In a similar regard, there have been numerous attempts to prepare related belt-shaped molecules possessing conjugated  $\pi$ -electron circuits; the various belt- and hoop-shaped compounds studied include cyclacenes,<sup>4</sup> [n]cyclocarbons,<sup>5</sup> cyclic paraphenylenes<sup>6</sup> and others.7 Recently, two groups reported the successful preparation of fully conjugated, belt-shaped compounds. Oda and co-workers produced cyclic paraphenylacetylenes containing six and eight subunits.8 Herges and co-workers isolated a 'picotube' based on four anthracene units.9 Our recent work on polyacetylenic macrocycles<sup>10</sup> spurred an interest in systems possessing the cyclic  $\pi$ -orbital motif. We report herein our work toward the synthesis of [8.8]paracyclophaneoctayne 1, a cyclophane-cyclocarbon hybrid nicknamed a cyclophyne.11



An octacobalt complex such as 2 was envisioned for the immediate precursor to the target macrocycle as: (i) the carbon skeleton could be assembled quickly and efficiently using known methodology, and (ii) the alkyne C=C-C angles in this type of (u-acetylene)dicobalt hexacarbonyl complex are reduced to ca. 140°, thus allowing the complex to participate in reactions that might be difficult or otherwise impossible for the rigid, unbound molecule to undergo. The synthesis of 2 is depicted in Scheme 1. Palladium-catalysed alkynylation of 1,4-diiodobenzene with excess triisopropylsilylbutadiyne  $3^{12}$ furnished a pale yellow tetrayne, which in turn reacted with 2 equiv. of octacarbonyldicobalt to provide the sterically favoured tetracobalt complex 4.<sup>+</sup> Attempted desilvlation with tetrabutylammonium fluoride led to complete decomposition of 4. Ligand exchange<sup>5a</sup> with bis(diphenylphosphino)methane (dppm) afforded a bridged complex (85% yield) stable to fluoride ion; desilylation now proceeded in high yield (> 95%). Eglinton coupling<sup>13</sup> of the terminal acetylenes provided fine, deep maroon crystals of  $2^{\dagger}$  in 47% yield. Spectroscopic data supported formation of the strain-free dimeric structure.

In their work with the hexacobalt complex of [18]cyclocarbon, Diederich *et al.* were unable to free  $C_{18}$  of the protecting groups by mild oxidation.<sup>5a,14</sup> With this problem in mind, we prepared model complex **5**† (Scheme 2) in order to investigate various methods to remove the cobalts. Of the numerous reagents and conditions explored, only oxidation by molecular iodine was found to regenerate butadiyne **6** cleanly and in excellent yield.

Application of the iodine-promoted decomplexation method to **2** provided dppm as the sole organic-soluble component along with a copious amount of insoluble material. Analysis of an aliquot of the crude reaction mixture by <sup>1</sup>H NMR showed a broad signal at  $\delta$  7.6 attributable to the phosphine ligand; no other aromatic peaks were observed. Although disappointing, this result is not wholly unexpected. A survey of the current literature reveals that, despite significant efforts on the part of several research groups, systems containing contiguous, bent triynes and higher polyynes have proven elusive.<sup>5</sup> Only strained monoyne<sup>8,15</sup> and diyne<sup>16</sup> derivatives are known. The cyclocarbons represent the ultimate bent polyyne macrocycles; C<sub>18</sub> is predicted to exhibit C=C–C bond angles near 160°, leading to a strain energy of around 72 kcal mol<sup>-1</sup>.<sup>5b</sup> Cyclophyne **1** should be similarly strained as the triple bond angles are calculated to



Scheme 1 Reagents and conditions: i, 1,4-diiodobenzene, Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>, CuI, Et<sub>3</sub>N, 73%; ii, Co<sub>2</sub>(CO)<sub>8</sub>, Et<sub>2</sub>O, 66%; iii, dppm, toluene, 85%; iv, Bu<sub>4</sub>NF, THF, > 95%; v, Cu(OAc)<sub>2</sub>·H<sub>2</sub>O, py, 47%



Scheme 2 Reagents and conditions: i, CBr<sub>4</sub>, PPh<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 85%; ii, phenylacetylene, CuCl, NH<sub>2</sub>OH·HCl, PrNH<sub>2</sub>, MeOH, H<sub>2</sub>O, 82%; iii, Co<sub>2</sub>(CO)<sub>8</sub>, Et<sub>2</sub>O, 76%; iv, dppm, toluene, 87%; v, I<sub>2</sub>, THF, 85%

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be between 161.7 and 163.6°.<sup>17</sup> Whether the failure to prepare these two classes of macrocycles is due to the extreme reactivity of the distorted polyyne moiety or to the lack of a viable synthetic route is not certain. Thus isolation and characterization of smaller bent hexatriyne- and octatetrayne-containing systems should help answer these questions. To this end, we are currently studying dehydrobenzoannulenes possessing higher polyynes distorted on one or more edge.<sup>18</sup> These results will be reported in due course.

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#### Footnotes

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† Selected spectral data (<sup>1</sup>H NMR: 300.13 MHz, <sup>13</sup>C NMR: 75.5 MHz). For **2**  $\delta_{\rm H}$  (CD<sub>2</sub>Cl<sub>2</sub>) 7.93 (8 H, s), 7.53 (16 H, m), 7.45–7.12 (64 H, m), 3.72 (4 H, dt), 3.39 (4 H, dt);  $\delta_{\rm C}$  (CD<sub>2</sub>Cl<sub>2</sub>) 206.61, 201.59, 142.67, 137.32, 135.03, 132.85, 131.78, 130.45, 130.03, 129.56, 128.89, 128.72, 100.11, 87.19, 83.50, 65.60, 37.73;  $\nu_{\rm max}$  (KBr)/cm<sup>-1</sup> 3033, 2154, 2089, 2055, 2023 For **4**  $\delta_{\rm H}$  (CDC<sub>13</sub>) 7.65 (4 H, s), 1.14 (42 H, s);  $\delta_{\rm C}$  (CDCl<sub>3</sub>) 138.36, 129.77, 105.30, 104.03, 88.93, 69.59, 18.65, 11.37;  $\nu_{\rm max}$  (KBr)/cm<sup>-1</sup> 3069, 3056, 2130, 2092, 2038, 2006, 1976. For **5**  $\delta_{\rm H}$  (CD<sub>2</sub>Cl<sub>2</sub>) 7.87 (2 H, d), 7.51 (4 H, m), 7.42–7.16 (19 H, m), 3.66 (1 H, dt), 3.36 (1 H, dt), 1.12 (21 H, s);  $\delta_{\rm C}$  (CD<sub>2</sub>Cl<sub>2</sub>) 207.25, 201.38, 143.91, 139.00, 133.44, 131.37, 130.64, 129.87, 129.39, 128.97 (2), 128.94, 128.64, 127.11, 110.33, 99.04, 96.89, 65.36, 34.80, 19.21, 12.16;  $\nu_{\rm max}$  (KBr)/cm<sup>-1</sup> 3056, 2102, 2062, 2027, 2004, 1975.

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