Photochemical cleavage of duplex DNA by *N***-benzoyloxy-2-thiopyridone linked to 9-aminoacridine**

Emmanuel A. Theodorakis,* Xin Xiang and Petra Blom

Department of Chemistry and Biochemistry, University of California, San Diego, 9500 Gilman Drive, La Jolla, California 92093-0358, USA

Upon illumination *N***-aroyloxy-2-thiopyridones induce nonspecific single-strand nicks in duplex DNA at micromolar concentrations.**

Synthetic reagents that cleave DNA are of great interest as tools in biochemical sciences. The design of such molecules constitutes a timely and challenging research topic and has led to the development of both sequence-specific DNA cleavers1 and DNA footprinting reagents.2

Our studies are directed towards the development of conceptually new approaches to DNA cleavage, induced by photoactivation of *N*-aroyloxy-2-thiopyridone derivatives, such as **1** (Scheme 1).3,4 To this end, we were guided by the operational advantages offered by the photoinducible DNA cleavage using the *N*-benzoyloxy-2-thiopyridone **4**.5 These

molecules could be ideal nucleic acid cleavers since they possess the following characteristics: a purely organic structure, facile preparation, prolonged stability in the absence of light, and well documented radical chemistry.3 The observed nucleic acid strand scission occurs upon a simple irradiation ($\lambda > 350$ nm), presumably *via* the generation of aroyloxyl radicals **2** and without the need of a metal or external oxidants.

Herein we describe our preliminary data on a novel family of photoactivated DNA-cleavers represented by **5**. The reagent design is based on linking a DNA photocleaving ligand to the 9-aminoacridine *via* a polymethylene chain. The aminoacridinyl group could assure high affinity for duplex DNA *via* intercalation,6 while the thiopyridone entity could account for the DNA cleavage. Indeed, upon irradiation ($\lambda > 350$ nm), compound **5** produces single strand breaks in duplex DNA with no intrinsic sequence selectivity.

The synthesis of compound **5** commences with bromide **6**, itself readily available by the condensation of 5-bromovaleroyl chloride with methyl *p*-amino benzoate. Azidation of **6**,

Scheme 2 Reagents and conditions: i, NaN₃ (2.0 equiv.), 18-crown-6 (0.05) equiv.) DMF, 25 °C, 3 h, 91%; ii, KOH (2.0 equiv.), THF-H₂O (1:1), 25 °C, 24 h, 98%; iii, 10% Pd/C (0.1 equiv.), H2, MeOH, 24 h, 89%; iv, 9-chloroacridine (1.0 equiv.), PhOH, 110 °C, 1 h, 90%; v, 2-mercaptopyridine *N*-oxide (1.0 equiv.), EDC (1.0 equiv.), DMF, 25 °C, 1 h, 43%

(Scheme 2), followed by saponification of the methyl ester moiety and reduction of the azido group gave rise to amino acid **9** in 79% overall yield, through intermediates **7** and **8**. Coupling of **9** with 9-chloroacridine followed by esterification with 2-mercaptopyridine *N*-oxide using 1-(3-dimethylaminopropyl)- 3-ethylcarbodiimide hydrochloride (EDC) afforded conjugate **5** in 38% overall yield.

We further compared the visible light photolysis of **5** and **4** in the presence of supercoiled circular ϕ X174 DNA (Fig. 1). Control experiments indicated that both light and the thiopyridone derivatives **5** or **4** are necessary for the DNA cleavage (lanes 1–3). Complete inhibition of the cleavage by glutathione (lanes 11, 12) further supports the notion that free radicals are responsible for the DNA damage. Moreover, compound **5**

Fig. 1 Concentration-dependent photocleavage of ϕ X174 DNA induced by **4** and **5**. The ϕ X174 DNA (50 μ M/base pair) was incubated for 1 h at 25 °C with 4 or 5 (30 mm Tris-HCl, 20 mm NaCl), then subjected to irradiation at 4 °C (lanes 1 and 4–12) with one lamp (GE 300 W) placed at approximately 20 cm from the samples. The results were analysed on 1% agarose gel (Trisacetate buffer) stained with ethidium bromide. Lane 1: ϕ X174DNA (control); lane 2: DNA and 60 μ M of 5 (no *hv*); lane 3: DNA and 1 mm of **4** (no *hy*) lanes $4-7$: DNA and $\overline{5}$ at concentrations of $5: 5, 10, 30$ and 60 μ m. respectively; lanes 8–10: DNA and **4** at concentrations of **4**: 30, 60 and 1.0 mm, respectively; lane 11: DNA, 60 mm of **5**, and 3.0 mm of glutathione; lane 12: DNA, 1.0 mm of **4** and 3.0 mm of glutathione.

*Chem. Commun***., 1997 1463**

cleaves DNA at concentrations as low as 5×10^{-6} M, while at 6×10^{-5} M complete conversion of the form I to form II is observed. In comparison, derivative **4** produces similar results at 1×10^{-3} M concentration. This substantial increase in efficiency of cleavage is attributed to the intercalating properties of 9-aminoacridine.6

An autoradiogram illustrating the results obtained upon irradiation of 5 in the presense of the $5'$ -32P labelled 93-mer duplex DNA is shown in Fig. 2. Our data show that both **4** and **5** generate identical DNA ladders upon irradiation and the photocleavage is indisputably neither base- nor sequencespecific. Furthermore, comparison of lanes 8–12 indicate that **5** is more efficient than **4** in cleaving DNA and can accurately cut the duplex at concentrations as low as 5×10^{-6} m. This increase in efficiency of cleavage is attributed to the presence of the

Fig. 2 Autoradiogram of a 10% denaturing polyacrylamide gel showing photocleavage of 5'-32P end-labeled Sal I / Sph I restriction fragment of pBR322 duplex DNA (93-mer), induced by **4** and **5**. DNA was incubated for 1 h at 25 °C with compounds **4** or **5** in buffered solution (30 mm Tris-HCl, 20 mm NaCl) and then irradiated (with one GE 300 W lamp placed 20 cm from the samples) for another 2 h at 4° C (lanes 3,4 and 7–13). The resulted solution was treated with piperidine (1 m) at 90 °C for 30 min, followed by EtOH precipitation (lanes 8–13). Lane 1: DNA cut by PIe I (14 base pairs) and EcoN I (28 base pairs); lane 2: DNase footprinting; lane 3: DNA (control); lane 4: DNA irradiated and piperidine treated without **4** or **5**; lane 5: DNA and 200 μ m of **4** (no *hv*); lane 6: DNA and 60 μ M of **5** (no *hv*); lane 7: DNA and 5 mm of **5** (no piperidine treatment); lanes 8–10: DNA and **5** at concentrations of 5: 10 and 20 μ M respectively; lanes 11,12: DNA and 4 at concentrations of 4:40 and 60 μ m respectively; lane 13: DNA, 4 (60 μ m) and 9-aminoacridine (30 μ M).

9-aminoacridinyl group. We ruled out the possibility that the 9-aminoacridinyl group enhances cleavage by altering the conformation of the DNA, since less cleavage was detected when 9-aminoacridine was added as an external intercalator (compare lines 12,13). Interestingly, the cleavage is more enhanced upon subsequent treatment with piperidine at 90 °C for 30 min without any change in sequence or base specificity (lines 7,8). Based on the above data we believe that in the case of **5** the DNA cleavage is performed by the intercalation complex and is probably mediated by aroyloxyl radicals.

It is evident from the above studies that the *N*-aroyloxy-2-thiopyridones **1** can induce non-specific single strand nicks in duplex DNA in a light-dependent reaction. The efficiency and/ or selectivity of the cleavage could be tuned by the proper choice of the DNA recognition element. In addition the light intensity that is responsible for the photoactivation could be tuned by structurally modifying the thiopyridone core. Thus, the *N*-benzoyloxy-2-thiopyridone moiety can be used for the design of new DNA photocleaving reagents with potential use as 'photofootprinting reagents' or as 'site-directed photonucleases'. Studies across these lines are now under investigation in our laboratories.

We thank Dr T. Li and Professors Y. Tor and K. C. Nicolaou of this department for useful discussions and the Donors of The Petroleum Research Fund, administered by the American Chemical Society, for partial support of this research.

Footnote

* E-mail: etheodor@ucsd.edu

References

- 1 A. M. Maxam and W. Gilbert, *Methods Enzymol.*, 1980, **65**, 499; S. A. Kane, H. Sasaki and S. M. Hecht, *J. Am. Chem. Soc.*, 1995, **117**, 9107; T. Li, Z. Zeng, V. A. Estevez, K. U. Baldenius, K. C. Nicolaou and G. F. Joyce, *J. Am. Chem. Soc.*, 1994, **116**, 3709; J. G. Muller, P. Zheng, S. E. Rokita and C. J. Burrows, *J. Am. Chem. Soc.*, 1996, **118**, 2320; M. J. Absalon, J. W. Kozarich and J. Stubbe, *Biochemistry*, 1995, **34**, 2065.
- 2 R. P. Hertzberg and P. B. Dervan, *J. Am. Chem. Soc.*, 1982, **104**, 313; M. D. Kuwabara and D. S. Sigman, *Biochemistry*, 1987, **26**, 7234; T. D. Tullius, *Nature*, 1988, **18**, 213; N. Schmid and J.-P. Behr, *Biochemistry*, 1991, **30**, 4357; P. B. Dervan, *Nature*, 1992, **359**, 87.
- 3 J. Boivin, E. Crepon and S. Z. Zard, *Tetrahedron*, 1990, **31**, 6869; D. H. R. Barton, B. Lacher and S. Z. Zard, *Tetrahedron*, 1987, **43**, 4321; W. Adam, D. Ballmaier, B. Epe, G. N. Grimm and C. R. Saha-Moller, *Angew. Chem., Int. Ed. Engl.*, 1995, **34**, 2156; B. M. Aveline, I. E. Kochevar and R. W. Redmond, *J. Am. Chem. Soc.*, 1996, **118**, 289; K. M. Hess and T. A. Dix, *Anal. Biochem.*, 1992, **206**, 309.
- 4 For selected reports on light-induced DNA cleaving agents, see M. M. Becker and J. C. Wang, *Nature*, 1984, **309**, 682; K. Uchida, A. M. Pyle, T. Morii and J. K. Barton, *Nucleic Acids Res.*, 1989, **17**, 10 259; W. Adam, J. Cadet, F. Dall'Acqua, B. Epe, D. Ramaiah and C. R. Saha-Moller, *Angew. Chem. Int. Ed. Engl.*, 1995, **34**, 107; I. Saito, M. Takayama, T. Matsuura, S. Matsugo and S. Kawanishi, *J. Am. Chem. Soc.*, 1990, **112**, 883; A. J. Blacker, J. Jazwinski, J.-M. Lehn and F. X. Wilhelm, *J. Chem. Soc., Chem., Commun.*, 1986, 1035; J. C. Quada, M. J. Levy and S. M. Hecht, *J. Am. Chem. Soc.*, 1993, **115**, 12 171; P. A. Wender, S. M. Touami, C. Alayrac and U. C. Philipp, *J. Am. Chem. Soc.*, 1996, **118**, 6522; G. A. Neuhart, C. C. Cheng and H. H. Thorp, *J. Am. Chem. Soc.*, 1995, **117**, 1463.
- 5 E. A. Theodorakis and K. M. Wilkoxen, *Chem. Commun.*, 1996, 1927. 6 9-Aminoacridine displays strong intercalating properties onto DNA with no intrinsic sequence selectivity. For selected literature on this topic, see C. Bailly and J.-P. Henichart, *Perspect. Bioconj. Chem.*, 1991, 112; A. H. J. Wang, *Curr. Opin. Struct. Biol.*, 1992, **2**, 361; P. E. Nielsen, C. Jeppesen, M. Egholm and O. Buchardt, *Biochemistry*, 1988, **27**, 6338; A. Lorente, J. F. Espinosa, M. F. Saiz, J.-M. Lehn, W. D. Wilson and Y. Y. Zhong, *Tetrahedron Lett.*, 1996, **37**, 4417; F. C. K. Chiu, R. T. C. Brownlee, K. Mitchell and D. R. Phillips, *Bioorg. Med. Chem. Lett.*, 1995, **5**, 1689.

Received in Corvallis, OR, USA, 28th April 1997; 7/02913F