A facile synthesis of β -(4-azuleno[1,2-*b*]thienyl) α , β -unsaturated ketones

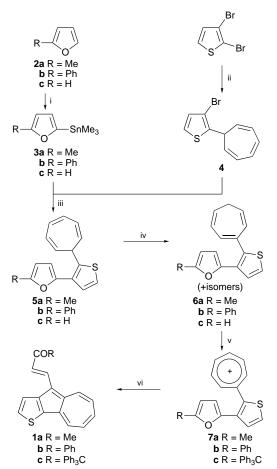
Naoki Kusuhara,^a Yuka Sugano,^a Hideki Takagi,^a Hideyoshi Miyake^b and Kimiaki Yamamura^{*a}

^a Department of Chemistry, Faculty of Science, Kobe University, Nada, Kobe 657, Japan

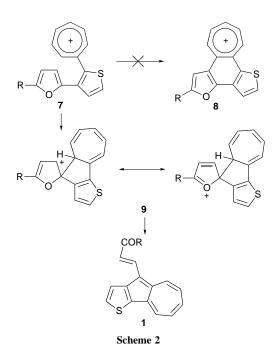
^b Department of Biofunctional Chemistry, Faculty of Agriculture, Kobe University, Nada, Kobe 657, Japan

The facile and novel synthesis of β -(4-azuleno[1,2-*b*]thienyl) α , β -unsaturated ketones 1 from 2-tropylio-3-(2-furyl)thiophenes 7 is reported; a mechanism involving a novel ring cyclization by intramolecular attack of the tropylium ion at the 2-position of the furan ring is postulated.

The azulenoid heterocyclic aromatic compounds which consist of the azulene ring condensed with heterocyclic aromatic nuclei are of interest not only for their physical and chemical properties, but also their pharmaceutical and physiological activity. Several azulene derivatives condensed with heteroaromatics such as thiophene, furan, pyrrole and pyridine rings are known.¹ Most of them were prepared by the heteroaromatization of the azulene derivatives, and were restricted to the parent compounds and alkyl-, aryl- or alkoxycarbonyl-substituted derivatives. Herein we disclose a novel and facile synthesis of azuleno[1,2-*b*]thiophenes having an α,β -unsatu-



Scheme 1 Reagents and conditions: i, BuLi, Me₃SnCl, THF, -78 °C; ii, BuLi, C₇H₇⁺, THF, -78 °C to room temp; iii, PdCl₂(Ph₃P)₂, THF, reflux; iv, xylene, reflux; v, Ph₃C⁺BF₄⁻, CH₂Cl₂, room temp; vi, CH₂Cl₂ or CH₂ClCH₂Cl



rated ketone group from the corresponding 2-tropylio-3-(2-furyl)thiophenes **7**.

The synthetic sequence leading to 1 from the readily available 2-substituted furans 2 is depicted in Scheme 1.† 2-Cycloheptatrienyl-3-(2-furyl)thiophenes 5 were prepared by the palladium(II)-catalysed Stille reaction² of 2-trimethylstannylfurans 3 with 2-cycloheptatrienyl-3-bromothiophene 4. Thermal sigmatropic rearrangement of 5 to 6 and subsequent hydride abstraction using triphenylmethyl tetrafluoroborate gave 2-tropylio-3-(2-furyl)thiophenes 7 in 65-85% yields. ‡ It might be expected that the intramolecular electrophilic substitution reaction of the tropylium moiety with the furan ring would afford the tetracyclic carbocations 8, which are isoelectronic with triphenylene.³ However, when a CH₂Cl₂ solution of **7a**§ was refluxed for 7 h, neutral red-black prisms were obtained instead of the expected carbocations 8a. The structure of these prisms was confirmed to be 4-(4-azuleno[1,2-b]thienyl)but-3-ene-2-one **1a** via its spectral data,⁵ together with elemental analysis. In the same manner, the chalcone-type compound 1b and triphenylmethyl derivative 1c were obtained as green-black needles (43%) and black needles (36%), respectively, from the corresponding tropylio derivatives 7b and 7c.

A possible mechanism for the formation of β -(4-azuleno-[1,2-*b*]thienyl) α , β -unsaturated ketones **1** from the corresponding 2-tropylio-3-(2-furyl)thiophenes **7** is illustrated in Scheme 2. The intramolecular electrophilic attack of the tropylium ion on the 2-poisition of the furan ring gives the spiro-type intermediate **9**, which can be converted to **1** by a ring-opening reaction. Although it is well-known that the 2-position of the furan ring is much more reactive towards electrophiles than the 3-position and that furans are readily ring-opened under acidic conditions,⁴ this is to the best of our knowledge the first

Chem. Commun., 1997 1951

tropylium ion-mediated furan ring-opening reaction to yield α,β -unsaturated carbonyl derivatives of azuleno[1,2-*b*]thiophene compounds. Since 2-substituted furans **2** are readily available and the procedure is simple and easy, it is considered that this is a valuable general synthetic route to various β -(4-azuleno[1,2-*b*]thienyl) α,β -unsaturated ketones.

The authors thank Miss Masuko Nishinaka for elemental analyses. The present work was supported in part by Grant-in-Aid for Scientific Research (C) (No. 08640685) from the Ministry of Education, Science, Sports and Culture, Japan.

Footnotes and References

* E-mail: kyama@kobe-u.ac.jp

[†] All new compounds were fully characterized by mass and ¹H NMR spectroscopy and elemental analysis.

[‡] With **6c**, triphenylmethyl tetrafluoroborate acts as both a hydride abstraction reagent and an electrophile, and triphenylmethyl-substituted tropylium ion derivative **7c** was obtained.

§ *Selected data* for **7a**: ($\delta_{\rm H}$ CD₂Cl₂, 250 MHz) 2.22 (3 H, s, CH₃), 6.20 (1 H, d, furan-H), 6.75 (1 H, d, furan-H), 7.43 (1 H, d, thiophene-H), 7.95 (1 H, d, thiophene-H), 8.68–8.83 (4 H, m, tropylium-H), 8.93–9.00 (2 H, m, tropylium-H). For **1a**: $\delta_{\rm H}$ (CD₃Cl, 250 MHz) 2.46 (3 H, s, CH₃), 6.90 (1 H, d, olefinic-H), 7.26 (2 H, dd, seven-membered ring-H), 7.62 (1 H, dd, seven-membered ring-H), 7.62 (1 H, dd, seven-membered ring-H), 7.89 (1 H, d, d, hiophene-H), 8.28 (1 H, d, olefinic H), 8.29 (1 H, d, seven-membered ring-H), 8.49 (1 H, d, seven-membered ring-H), 8.49 (1 H, d, seven-membered ring-H), 8.20 (CD₃Cl, 75 MHz) 198.2 (C=O), 150.1

(quaternary), 143.6 (quaternary), 137.3, 135.7 (quaternary), 134.8, 134.0, 133.7, 133.1 (quaternary), 131.9, 126.4, 125.4, 123.3, 119.8, 116.4 (quaternary), 27.8 (CH₃). Calc. for $C_{16}H_{12}SO: C$, 76.16; H, 4.79. Found: C, 76.30; H, 4.63%.

- K. Matsui and T. Nozoe, *Chem. Ind.*, 1960, 1302; T. Nozoe, S. Seto and S. Matsumura, *Chem. Ind.*, 1961, 1715; T. Nozoe and K. Kikuchi, *Bull. Chem. Soc. Jpn.*, 1963, **36**, 633; S. Hayashi, S. Kurokawa, M. Okano and T. Matsuura, *Tetrahedron Lett.*, 1967, 3443; L. L. Replogle, K. Katsumoto and H. L. Ammon, *J. Am. Chem. Soc.*, 1968, **90**, 1086; T. Nozoe, K. Takase, T. Nakazawa, S. Sugita and M. Saito, *Bull. Chem. Soc. Jpn.*, 1974, **47**, 1750; K. Kohara, H. Tachibana, Y. Masuyama and Y. Otani, *Bull. Chem. Soc. Jpn.*, 1979, **52**, 1549; K. Yamane, K. Fujimori and T. Takeuchi, *Bull. Chem. Soc. Jpn.*, 1981, **54**, 2537; T. Morita, T. Nakadate and K. Takase, *Heterocycles*, 1981, **15**, 835; K. Fujimori, T. Fujita, K. Yamane, M. Yasunami and K. Takase, *Chem. Lett.*, 1983, 1721.
- J. K. Stille, Angew. Chem., Int. Ed. Engl., 1986, 25, 508; T. R. Bailey, Tetrahedron Lett., 1986, 27, 4407; A. Dondoni, A. R. Mastellari, A. Medict, E. Negrini and P. Pedrini, Synthesis, 1986, 757; V. N. Kalinin, Synthesis, 1992, 413; T. N. Mitchell, Synthesis, 1992, 803.
- 3 K. Yamamura, H. Miyake, S.Nakatsuji and I. Murata, *Chem. Lett.*, 1992, 1213.
- 4 M. V. Sargent and T. M. Cresp, *Comprehensive Organic Chemistry*, Pergamon, Oxford, 1979, vol. 4, p. 693.

Received in Cambridge, UK, 30th June 1997; 7/004583B