## Preparation and X-ray analysis of novel carbonyltungsten(0) complexes of diphosphinidenecyclobutenes

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A sterically protected diphosphinidenecyclobutene 5 bearing a bulky 2,4,6-triisopropylphenyl group is prepared and allowed to react with W(CO)<sub>5</sub>(thf) to give a doubly tungstencoordinated complex 8, with chelate- and  $\pi$ -type coordination; the structure of the complex is confirmed by X-ray analysis.

Sterically protected phosphorus-containing multiple bonded compounds with bulky substituents are currently of interest. We have been interested in cumulated and conjugated systems<sup>1,2</sup> involving phosphorus atom(s) in low coordination states, as well as diphosphenes<sup>3</sup> and phosphaethenes.<sup>4</sup> Recently, Appel *et al.*,<sup>5</sup> Märkl *et al.*<sup>6</sup> and ourselves<sup>7–9</sup> have reported the preparation and isolation of diphosphinidenecyclobutenes (**1a–c**, Scheme 1) utilizing 2,4,6-tri-*tert*-butylphenyl as a protecting group (Ar), as well as their *E*/*Z* isomerization about the P=C bonds.<sup>7,10</sup> We have also reported on their transition metal complexes, where metals are Cr, Mo, W,<sup>8,11,12</sup> Pt and Pd.<sup>13</sup> We now report a new type of tungsten carbonyl complex ligated with diphosphinidenecyclobutenes carrying the 2,4,6-triisopropylphenyl group<sup>14</sup> (Tip) as well as the Ar group.

ArP  

$$ArP$$
  
 $R$   
 $1a R = SiMe_3$   
 $1b R = Ph$   
 $1c R = H$   
Scheme 1 Ar =  $C_6H_2Bu^t_3$ -2,4,6

According to a slightly modified procedure employed for the preparation of **1b**, (2-phenylethynyl)(2,4,6-triisopropylphenyl)phosphinous halides **3** were prepared by the reaction of 2-phenylethynylphosphinous dichloride **2**<sup>15</sup> with 2,4,6-triisopropylphenylmagnesium bromide in thf at -78 °C. Then **3** was allowed to react with Bu<sup>4</sup>Li at -78 °C or with Zn powder at room temp. to give diphosphane **4** as a diastereomeric mixture (D,L:*meso* = *ca*. 1:1) and the mixture was refluxed in thf to give (*E*,*Z*)-1,2-diphenyl-3,4-bis(2,4,6-triisopropylphenylphosphinidene)cyclobutene **5**.† It should be noted that only a trace amount of (*E*,*E*)-**5** was observed by <sup>31</sup>P NMR spectroscopy.

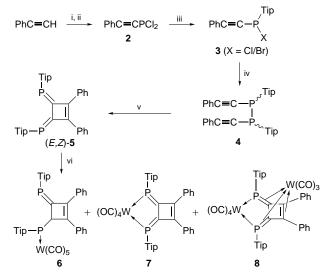
In the <sup>1</sup>H NMR spectrum of **5** in  $[^{2}H_{8}]$  toluene, there appeared six types of isopropyl peaks at 295 K. At higher temperatures, only two sets due to the *ortho*-isopropyl groups became coalesced, but the other two sets due to the other *ortho*isopropyl groups remained magnetically non-equivalent. The remainder of the isopropyl groups assigned to the *para*isopropyl groups remained unchanged even at 375 K. The fact suggests that one of the Tip groups, probably (*E*)-Tip, taking the steric congestion into account, starts to rotate at 323 K, while (*Z*)-Tip appears not to rotate even at 375 K.

We are now able to obtain a novel class of tungsten complex by the reaction of the diphosphinidenecyclobutene **5** with W(CO)<sub>5</sub>(thf)<sup>16</sup> in refluxing thf.<sup>‡</sup> The reaction gave **8** together with **6** and **7** (Scheme 2).<sup>†</sup> The <sup>31</sup>P NMR chemical shift of **8** appeared at  $\delta_P$  57, indicating that the phosphorus atom is strongly shielded compared with the value in the free ligand of either  $\delta_P$  164 for (*E*,*E*)-**5** or  $\delta_P$  169 and 183 for (*E*,*Z*)-**5**. The structure of **8** was confirmed by the X-ray analysis§ to reveal a novel coordination type. Fig. 1 depicts an ORTEP drawing of the molecular structure for the bis-tungsten complex **8**. There are observed two WCO moieties, one is tetracarbonyl with chelate type coordination, the other is tricarbonyl with  $\pi$ -coordination. The two phosphorus atoms coordinate doubly on both W(CO)<sub>4</sub> and W(CO)<sub>3</sub> adopting sp<sup>3</sup> type configuration. The exocyclic P=C bonds are bent strongly from the ring plane [interplanar angle between the plane P(1)P(2)C(3)C(4) and the plane C(1)–C(4) 32.3°] to adopt a good interaction with the  $6\pi$ system to coordinate on W(CO)<sub>3</sub>, thus exhibiting a sharp contrast to the regular group 6 metal tetracarbonyl or palladium complexes of diphosphinidenecyclobutenes, where the system is planar with all sp<sup>2</sup> type configuration as exemplified in W,<sup>11</sup> Mo,<sup>12</sup> and Pd complexes.<sup>13</sup>

The P=C bond lengths of **8** [av. 1.763(10) Å] are elongated upon coordination compared to those of (E,E)-**1a** [1.677(6) Å]<sup>12</sup> and (E,E)-**1b** [1.690(8) Å].<sup>5</sup> The bond length C(1)–C(2) of **8** [1.47(1) Å] is fairly elongated compared to the values of 1.400(9) Å and 1.380(9) Å for (E,E)-**1a** and (E,E)-**1b**, respectively. Thus, the interaction between the W(CO)<sub>3</sub> group and the ligand is apparently strong, since little change in bond lengths has been observed upon coordination with the chelate.<sup>8,11–13</sup>

Furthermore, in the case of compound (E,E)-1c, carrying the two Ar groups and no substituents on the cyclobutene ring, a doubly coordinated complex 9 was formed under similar conditions using W(CO)<sub>4</sub>(MeCN)<sub>2</sub> (Scheme 3).<sup>16</sup> Complex 9 seems to have been formed *via* 10 as an intermediate, since 10 reacted with W(CO)<sub>4</sub>(MeCN)<sub>2</sub> to give 9.¶ It appeared to require less hindered environment than 1a or 1b to obtain double coordination products such as 8 or 9.

Although Rau and Behrens reported on the formation and the structural analysis of  $(\eta^6$ -dimethylenecyclobutene)tricar-



Scheme 2 Tip =  $C_6H_2Pr^i_3$ -2,4,6. *Reagents and conditions*: i, NEt<sub>3</sub>; ii, PCl<sub>3</sub>; iii, (Tip)MgBr; iv, Bu<sup>i</sup>Li or Zn; v, reflux in thf; vi, W(CO)<sub>5</sub>(thf), reflux in thf.

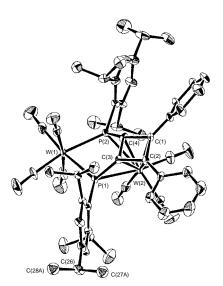
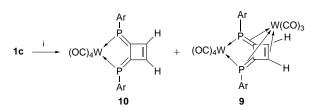


Fig. 1 The molecular structure of bis-tungsten complex **8** showing the atomic labelling scheme; only atoms with dominant occupancy factors are displayed. Important bond lengths (Å) and angles (°): W(1)–P(1) 2.493(3), W(1)–P(2) 2.490(3), W(2)–P(1) 2.679(3), W(2)–P(2) 2.664(3), W(2)–C(1) 2.35(1), W(2)–C(2) 2.321(10), W(2)–C(3) 2.24(1), W(2)–C(4) 2.18(1), P(1)–C(3) 1.788(10), P(2)–C(4) 1.737(9), C(1)–C(2) 1.47(1), C(1)–C(4) 1.49(1), C(2)–C(3) 1.47(1), C(3)–C(4) 1.47(1); P(1)–W(1)–P(2) 73.22(9), P(1)–W(2)–P(2) 67.60(8), C(2)–C(4) 9.0(8), C(1)–C(2) 89.2(8).



Scheme 3 Ar =  $C_6H_2Bu^{t_3}$ -2,4,6. *Reagents and conditions*: i, W(CO)<sub>4</sub>-(MeCN)<sub>2</sub>, thf, in the dark.

bonylchromium( $_0$ )<sup>17</sup> and ( $\eta^6$ -diisopropylidenecyclobutene)tricarbonylchromium( $_0$ ),<sup>18</sup> this is the first example in the diphosphinidenecyclobutene–tungsten carbonyl system, where the ligand is doubly coordinated to tungsten.

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## **Footnotes and References**

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† Selected spectroscopic data (NMR spectra were recorded on either a Bruker AC200P, a JEOL  $\alpha$ -500, or a Bruker AM600 spectrometer): (*E*,*Z*)-5: <sup>1</sup>H NMR (600 MHz, at 295 K, CDCl<sub>3</sub>): δ 0.92 (6 H, d, J 6.8 Hz, o-CHMe), 1.00 (6 H, d, J 6.8 Hz, o-CHMe), 1.20 (6 H, d, J 6.9 Hz, p-CHMe2), 1.25 (6 H, d, J 6.9 Hz, p'-CHMe2), 1.32 (6 H, d, J 6.8 Hz, o'-CHMe), 1.39 (6 H, d, J 6.8 Hz, o'-CHMe), 2.75 (1 H, spt, J 6.9 Hz, p-CHMe2), 2.75 (1 H, spt, J 6.9 Hz, p'-CHMe<sub>2</sub>), 3.21 (2 H, m, o-CHMe<sub>2</sub>), 3.72 (2 H, m, o'-CHMe<sub>2</sub>); <sup>31</sup>P{<sup>1</sup>H} NMR (81 MHz, CDCl<sub>3</sub>):  $\delta$  168.6, 182.8 (AB, *J* 17.8 Hz). 8: dark violet crystals, mp 183 °C (decomp.); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.18 (6 H, d, J 6.6 Hz, o-CHMe), 1.31 (6 H, d, J 6.9 Hz, p-CHMe), 1.31 (12 H, d, J 6.9 Hz, o-, p-CHMe), 1.41 (6 H, d, J 6.7 Hz, o-CHMe), 1.43 (6 H, d, J 6.6 Hz, o-CHMe), 2.94 (2 H, spt, J 6.9 Hz, p-CHMe2), 4.35 (2 H, br s, o-CHMe<sub>2</sub>), 4.46 (2 H, spt, J 6.7 Hz, o-CHMe<sub>2</sub>); <sup>31</sup>P{<sup>1</sup>H} NMR (81 MHz, CDCl<sub>3</sub>):  $\delta$  56.7 (satellite d,  ${}^{1}J_{PW}$  226 Hz); IR (KBr):  $\nu/cm^{-1}$  2033, 1994, 1979, 1923, 1913, 1878, UV–VIS (CH<sub>2</sub>Cl<sub>2</sub>):  $\lambda_{max}(\log \varepsilon)$  254 (sh, 4.87), 272 (sh, 4.75), 321 (4.37), 456 (4.04), 603 nm (sh, 3.51); MS (FAB): m/z 1234 (M<sup>+</sup>), 1206 (M<sup>+</sup>-CO). 9: red needles, mp 180-181 °C (decomp.); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 1.19 (18 H, s, p-Bu<sup>t</sup>), 1.65 (18 H, s, o-Bu<sup>t</sup>), 1.75 (18 H, s, *o*-Bu<sup>t</sup>), 4.66 (2 H, m, P=C–CH);  ${}^{31}P{}^{1}H{}$  NMR (81 MHz, CDCl<sub>3</sub>):  $\delta$ 64.6 (satellite d, <sup>1</sup>J<sub>PW</sub> 227.6 Hz); IR (KBr): v/cm<sup>-1</sup> 2033, 1994, 1942, 1900, 1865, 604 cm<sup>-1</sup>; UV–VIS (*n*-hexane):  $\lambda_{max}(\log \varepsilon)$  218 (4.88), 236 (4.71), 380 (3.94), 456 (3.66), 505 (sh, 3.54), 598 nm (3.30); MS (FAB): m/z 1166 (M<sup>+</sup>), 1138 (M<sup>+</sup> - CO), 968 (M<sup>+</sup> - 7CO - 2). **10**: orange powder, mp 198–199.5 °C (decomp.); <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>): δ 1.34 (18 H, s, *p*-Bu<sup>i</sup>), 1.63 (36 H, s, *o*-Bu<sup>i</sup>), 6.45 (2 H, dd, <sup>3</sup>J<sub>PH</sub> = <sup>4</sup>J<sub>PH</sub> 3.6 Hz, P=C–CH); <sup>31</sup>P{<sup>1</sup>H} NMR (81 MHz, CDCl<sub>3</sub>): δ 154.1 (satellite d, <sup>1</sup>J<sub>PW</sub> 250.1 Hz); UV–VIS (*n*-hexane):  $\lambda_{max}(\log \varepsilon)$  242 (4.79), 319 (4.12), 413 (4.09), 482 nm (3.45); MS (FAB): m/z 898 (M<sup>+</sup>), 870 (M<sup>+</sup> - CO).

‡ *Preparative procedure*: **8**: under argon, a solution of phenylethynylphosphonous dichloride **2** (356 mg, 1.8 mmol)<sup>15</sup> in thf (10 ml) was cooled at −78 °C and was allowed to react with triisopropylphenylmagnesium bromide (1.8 mmol) and the mixture was warmed to room temp. to give phenylethynylphosphinous chloride **3** (X = Cl,  $\delta_P$  40) together with the bromide **3** (X = Br,  $\delta_P$  22) (chloride : bromide = 1:0.8). The mixture of the halides **3** (X = Cl, Br) was allowed to react with Zn (261 mg, 4.0 mmol) in thf (7 ml) at room temp. for 3 h to give the corresponding diphosphane **4** as a diastereomeric mixture ( $\delta_P$  −66, −65); D,L : *meso* = *ca*. 1 : 1. It was then refluxed in thf for 24 h to give (*E*,*Z*)-**5**. The ligand (*E*,*Z*)-**5** was allowed to react with W(CO)<sub>5</sub>(thf) (1.7 mmol) in refluxing thf for 9 h. **8** (62 mg, 0.051 mmol)† was isolated in 6% yield based on **2**, together with **6** [ $\delta_P$  114 (satellite d,  ${}^1J_{PW}$  265 Hz) and 186, AB,  ${}^4J_{PP}$  8 Hz] and **7** [ $\delta_P$  139,  ${}^1J_{PW}$  251 Hz].

§ *Crystal data* for **8**: recrystallised from ethanol–benzene.  $C_{53}H_{56}O_7P_2W_2$ , M = 1234.67, monoclinic, space group  $P2_1/n$ , a = 10.241(3), b = 25.462(4), c = 20.219(3) Å,  $\beta = 92.61(2)^\circ$ , U = 5266(1) Å<sup>3</sup>, Z = 4, T = 223 K,  $D_c = 1.557$  g cm<sup>-3</sup>; 9529 reflections with  $2\theta \le 50.0^\circ$  were recorded on a four-circle diffractometer using graphite-monochromated Mo-K $\alpha$  radiation. Of these, 6693 with  $I > 3\sigma(I)$  were judged as observed. The structure was solved using SIR92.<sup>19</sup> The non-hydrogen atoms, except for C(27) and C(28), were refined anisotropically. Atoms C(27) and C(28) were disordered and refined isotropically (occupancy factor for the dominant: 0.55). Hydrogen atoms except for those on C(26)–C(28) were included, but their positions were not refined. R = 0.054,  $R_w = 0.064$ . CCDC 182/653.

¶ *Preparative procedure*: **9** (*E*,*E*)-**1c** (86 mg, 0.14 mmol) was allowed to react with 0.43 mmol of W(CO)<sub>4</sub>(MeCN)<sub>2</sub> in refluxing thf (30 ml) for 1.5 h to give 90 mg of **9** in 54% yield.† **10**: a mixture of diphosphinidenecyclobutene (*E*,*E*)-**1c** (31 mg, 0.051 mmol) and W(CO)<sub>4</sub>-(MeCN)<sub>2</sub> (40 mg, 0.11 mmol) was dissolved in thf (20 ml) and the solution was stirred at room temp. for 21 h to give **10** (20 mg, 43% yield.†

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## 28 Chem. Commun., 1998