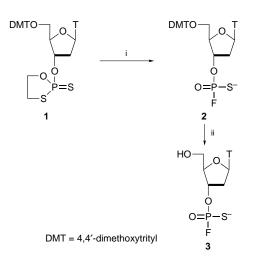
## Synthesis and chemical and enzymatic reactivity of thymidine 3'-O- and 5'-O-phosphorofluoridothioates

## Konrad Misiura, Daria Szymanowicz and Wojciech J. Stec\*†

Polish Academy of Sciences, Centre of Molecular and Macromolecular Studies, Department of Bioorganic Chemistry, Sienkiewicza 112, 90-363 Lódź, Poland

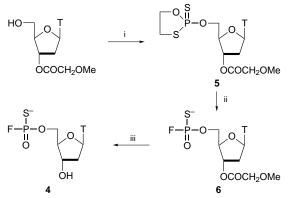
5'-O- or 3'-O-Protected thymidine 3'-O- or 5'-O-(2-thiono-1,3,2-oxathiaphospholanes) react with triethylammonium fluoride in a presence of DBU and furnish, after deprotection, thymidine 3'-O- and 5'-O-phosphorofluoridothioates; the latter undergoes stereoselective hydrolysis by *snake venom* phosphodiesterase.

Within the course of our studies on the development of oxathiaphospholane methodology for phosphorylation,1 phosphorothioylation<sup>2</sup> and phosphorodithioylation<sup>3</sup> of biomolecules we have found that the endocyclic P-S bond of a 2-thiosubstituted 1,3,2-oxathiaphospholane ring attached to nucleoside moiety can be easily broken in the presence of DBU by fluoride ions leading, after spontaneous episulfide elimination, 3'-O-methoxyacetyl nucleoside 5'-O-phosphoroto fluoridothioates or 5'-O-dimethoxytrilyl nucleoside 3'-O-phosphorofluoridothioates.<sup>4,5</sup> Thus, 5'-O-dimethoxytritylthymidine 3'-O-(2-thiono-1,3,2-oxathiaphospholane) 1 (ratio of diastereomers ca. 1:1), after treatment with triethylammonium fluoride<sup>‡</sup> in a presence of DBU gives 5'-O-dimethoxytritylthymidine 3'-O-phosphorofluoridothioate 2§ in 78% yield (Scheme 1). The dimethoxytrityl protecting group was removed by the treatment of 2 with 80% AcOH. The final product, thymidine 3'-O-phosphorofluoridothioate  $3, \P$  was purified by anion-exchange chromatography on Sephadex A-25 using triethylammonium hydrogen carbonate buffer (pH 7.5, 0.02-0.5 M) as eluent. Pure **3** has also been obtained from **1** in 68% yield in a one-pot synthesis. Reaction of 1 with triethylammonium fluoride in a presence of DBU is, unlike the 1,3,2-oxathiaphospholane ring-opening condensation with alcohols2 or amines,6 a non-stereospecific process. Starting from a mixture of partially separated diastereomers of 1 [ratio 73:27,  $\delta_{\rm P}$  $(CD_3CN)$ , 105.78 and 105.83] diastereomers of 2 were obtained in a ratio of 54:46 [ $\delta_{\rm F}$  (CD<sub>3</sub>CN), -30.06 and -29.96]. The epimerisation at phosphorus was not unexpected in the light of



Scheme 1 Reagents and conditions: i, Et<sub>3</sub>NHF, DBU, MeCN, 15 min; ii, 80% AcOH, 1 h

published earlier results on the stereochemistry of nucleophilic substitution at phosphorus by fluoride ion.<sup>7,8</sup> Also, thymidine 5'-O-phosphorofluoridothioate 4 has been synthesized according to the reaction sequence presented in the Scheme 2. Phosphitylation of 3'-O-methoxyacetylthymidine with N,Ndiisopropylamino-1,3,2-oxathiaphospholane<sup>2</sup> in a presence of 1H-tetrazole, followed by sulfurization yielded oxathiaphospholane 5. Reaction of 5 with triethylammonium fluoride-DBU furnished intermediate 3'-O-methoxyacetylthymidine 5'-Ophosphorofluoridothioate 6 which subsequently was deprotected with a concentrated solution of ammonia providing, after purification on Sephadex A-25, the final product 4\*\* in 75% yield. The phosphorofluoridothioate monoesters 3 and 4 were hydrolytically stable even under basic conditions (conc. ammonia, room temp., 1 h), as proven by <sup>31</sup>P NMR assay. Similarly, the resistance of 3 and 4 towards methanolysis, attempted in the presence of triethylamine or pyridine, has been observed. Attempts at internucleotide bond formation in reaction of 2 with 3'-O-acetylthymidine in the presence of strong bases such as ButOK, DBU and 2-tert-butylimino-2-diethylamino-1,3-dimethyl-1,3,2-diazaphosphinane (BEMP) have failed. Reactions were performed in DMF and their progress was followed by <sup>31</sup>P NMR spectroscopy. Under these conditions, even after 18 h, formation of dithymidylyl (3',5') phosphorothioate  $(T_{PS}T)$  was not observed. Instead, both 3 and 4 underwent intramolecular cyclization in the presence of an excess of ButOK (five-fold molar excess) leading to thymidine cyclic (3',5')phosphorothioate (cTMPS). Similarly, as previously observed<sup>9</sup> for Bu<sup>t</sup>OK-catalyzed cyclization of nucleoside 5'-O-p-nitrophenyl phosphophosphorothioates, in the reactions of rofluoridothioates 3 and 4  $[S_P]$ -cTMPS<sup>10</sup> was also formed preferentially (ratio of  $[S_P]$ -:  $[R_P]$ -cTMPS ca. 2:1). Yields of Sephadex-purified cTMPS obtained from 3 and 4 were 57 and 38%, respectively. The lack of formation of  $T_{PS}T$  in Bu<sup>t</sup>OK assisted condensation of 2 with 3'-O-acetylthymidine was rather unexpected in the light of the results of von Tigerstrom and Smith<sup>11</sup> on effective formation of T<sub>P</sub>T and other medium-sized oligothymidylates in the reaction of protected thymidine 3'-Ophosphorofluoridate with 3'-O-acetylthymidine.



Scheme 2 Reagents and conditions: i, N,N-diisopropylamino-1,3,2-oxathiaphospholane, 1*H*-tetrazole, 2 h, then elemental sulfur, 18 h; ii, Et<sub>3</sub>NHF, DBU, CH<sub>2</sub>Cl<sub>2</sub> MeCN, 15 min; iii, conc. NH<sub>4</sub>OH, EtOH, 2.5 h

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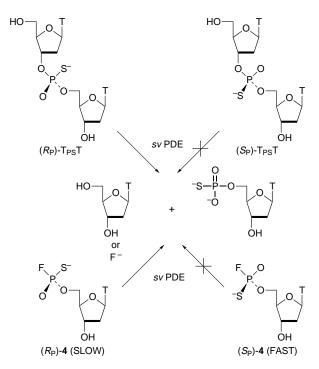


Fig. 1 Tentative assignment of absolute configuration of diastereomers of 4

In the light of earlier results on an enzymatic cleavage of thymidine 5'-O- or 3'-O-phosphorofluoridate assisted by *snake* venom (sv PDE)<sup>12,13</sup> and *spleen*<sup>12</sup> phosphodiesterases (*spleen* PDE) it was tempting to study the thio analogues **4** and **3** as substrates for these enzymes. From the pioneering work of Eckstein<sup>14</sup> and Benkovic<sup>15</sup> demonstrating the stereoselectivity of *sv* PDE towards P-chiral diesters of phosphorothioic acid it was of interest to check if this enzyme can discriminate between the diastereomers of **3** or **4**.

5'-O-Phosphorofluoridothioate 4 was incubated with sv PDE<sup>††</sup> and the progress of the enzymatic digestion was analyzed by RP-HPLC. It was found that sv PDE, if added to a diastereomeric mixture of 4, causes the stereoselective hydrolysis of the P-F bond of slow-eluted 4 leaving the fast-eluted diastereomer intact. Also, in the case of diastereomeric mixture of 3<sup>++</sup> only slow-eluted 3 underwent hydrolysis in a presence of sv PDE, albeit the reaction proceeded much slower than that observed for slow-eluted 4. Interestingly, under analogous conditions, the rate of hydrolysis of slow-4 by sv PDE was similar to that obtained during digestion of dithymidylyl (3',5') phosphate (T<sub>P</sub>T). In the presence of spleen PDE both diastereomers of 3 were hydrolyzed while both diastereomers of 4 were resistant to this enzyme.\*\* It was also found that 3 and 4 have no inhibitory activity§§ towards either phosphodiesterase. Results on the use of 4 for inhibition of thymidylate synthase will be published separately.16

In conclusion, we have found that the 1,3,2-oxathiaphospholane ring can be opened in the presence of DBU by fluoride anion leading to the appropriate phosphorofluoridothioates. The nucleoside 5'-O- or 3'-O-phosphorofluoridothioates obtained can be used in studies of the mode of action of nucleolytic enzymes. Comparative topological analysis of diastereomers of  $T_{PS}T$  and 4 undergoing sv PDE- assisted hydrolysis allows the tentative assignment of absolute configuration of the sloweluted 4 as  $R_P$  (Fig. 1). Spleen and sv PDE-assisted hydrolysis of P-F bonds in compounds 3 and 4 is in agreement with earlier findings12,13 that these enzymes split nucleoside 3'-O- or 5'-Ophosphorofluoridate, respectively, giving rise to the appropriate nucleoside phosphates. From this perspective our data on enzymatic hydrolyses of 3 and 4 disagree with the results of Dabkowski et al.,17 who characterized thymidine 3'-O-phosphorofluoridate as the product of spleen PDE-assisted degradation of thymidin-3'-yl 2'-deoxyadenosin-5'-yl phosphorofluoridate. Adenosine 5'-O-phosphorofluoridate was found as the product of *sv* PDE-assisted hydrolysis of the same substrate. Besides the hydrolytic instability of the P–F bond of dinucleoside (3',5')phosphorofluoridates in buffered aqueous media<sup>18</sup> yielding appropriate phosphates, even if thymidine 3'-O-phosphorofluoridate or adenosine 5'-O-phosphorofluoridate were the respective products of PDE-catalyzed hydrolyses, they would necessarily undergo further enzymatic degradation to phosphomonoesters.

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## Notes and References

† E-mail: wjstec@bio.cbmm.lodz.pl

<sup>‡</sup> Triethylammonium fluoride (1 M solution in THF) was obtained by mixing triethylamine tris(hydrofluoride) (1 equiv.) with triethylamine (2 equiv.).

\$\$ Compound **2** consists of a 1 : 1 mixture of diastereomers,  $\delta_{\rm P}({\rm CD}_3{\rm CN}, 81 {\rm MHz})$  54.52 ( ${}^{1}J_{\rm P-F}$  1043 Hz), 54.58 ( ${}^{1}J_{\rm P-F}$  1046 Hz); m/z (-FAB) 641.4 (M<sup>++</sup> -1).

¶ Compound **3** consists of a mixture of diastereomers (ratio 48:52),  $\delta_P(D_2O, 81 \text{ MHz}) 53.72$ ,  $({}^1J_{P-F} 1053 \text{ Hz})$ ,  $53.74 ({}^1J_{P-F} 1055 \text{ Hz})$ ;  $\delta_F(D_2O, 188 \text{ Mz}) - 31.4 ({}^1J_{P-F} 1043 \text{ Hz})$ ,  $-31.2 ({}^1J_{P-F} 1046 \text{ Mz})$ ; m/z (-FAB) 339.1 (M<sup>++</sup> -1).

|| Compound **5** was obtained as a mixture of diastereomers,  $\delta_P(CDCl_3, 81 \text{ MHz})$  106.64, 106.82 (ratio 1 : 1); m/z (+FAB) 453.2 (M<sup>++</sup> +1).

\*\*Compound **4** consists of mixture of diastereomers (ratio 59:41),  $\delta_{P}(D_{2}O, 81 \text{ Mz})$  54.17, 54.20 ( ${}^{1}J_{P-F}$  1053 Hz);  $\delta_{F}(D_{2}O, 188 \text{ Mz})$  -35.6, -35.7 ( ${}^{1}J_{P-F}$  1053 Hz); m/z (-FAB) 339.2 (M<sup>++</sup> -1).

†† The reaction mixture consists of 0.1 mM 4 or 3, 100 mM Tris–HCl pH 8.0, 20 mM MgCl<sub>2</sub>, and *sv* PDE (0.01 U ml<sup>-1</sup>); 37 °C; incubation time: 0.5 h for 4 and 16 h for 3.

<sup>‡‡</sup> The reaction mixtures consist of 0.1 mM **3** (or **4**), 50 mM acetate buffer pH 5.0, and *spleen* PDE (0.15 U ml<sup>-1</sup>), 37 °C; incubation time: 1 h for **3** and 16 h for **4**.

§§ Enzymatic digestions were performed under conditions mentioned above.  $T_PT$  and 3 (or 4) were used at equimolar concentrations (0.1 mM).

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