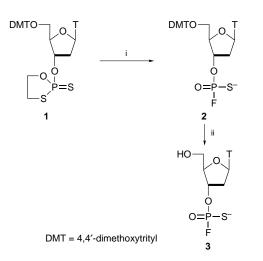
Synthesis and chemical and enzymatic reactivity of thymidine 3'-O- and 5'-O-phosphorofluoridothioates

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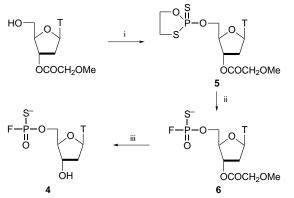
5'-O- or 3'-O-Protected thymidine 3'-O- or 5'-O-(2-thiono-1,3,2-oxathiaphospholanes) react with triethylammonium fluoride in a presence of DBU and furnish, after deprotection, thymidine 3'-O- and 5'-O-phosphorofluoridothioates; the latter undergoes stereoselective hydrolysis by *snake venom* phosphodiesterase.

Within the course of our studies on the development of oxathiaphospholane methodology for phosphorylation,1 phosphorothioylation² and phosphorodithioylation³ of biomolecules we have found that the endocyclic P-S bond of a 2-thiosubstituted 1,3,2-oxathiaphospholane ring attached to nucleoside moiety can be easily broken in the presence of DBU by fluoride ions leading, after spontaneous episulfide elimination, 3'-O-methoxyacetyl nucleoside 5'-O-phosphoroto fluoridothioates or 5'-O-dimethoxytrilyl nucleoside 3'-O-phosphorofluoridothioates.^{4,5} Thus, 5'-O-dimethoxytritylthymidine 3'-O-(2-thiono-1,3,2-oxathiaphospholane) 1 (ratio of diastereomers ca. 1:1), after treatment with triethylammonium fluoride[‡] in a presence of DBU gives 5'-O-dimethoxytritylthymidine 3'-O-phosphorofluoridothioate 2§ in 78% yield (Scheme 1). The dimethoxytrityl protecting group was removed by the treatment of 2 with 80% AcOH. The final product, thymidine 3'-O-phosphorofluoridothioate $3, \P$ was purified by anion-exchange chromatography on Sephadex A-25 using triethylammonium hydrogen carbonate buffer (pH 7.5, 0.02-0.5 M) as eluent. Pure **3** has also been obtained from **1** in 68% yield in a one-pot synthesis. Reaction of 1 with triethylammonium fluoride in a presence of DBU is, unlike the 1,3,2-oxathiaphospholane ring-opening condensation with alcohols2 or amines,6 a non-stereospecific process. Starting from a mixture of partially separated diastereomers of 1 [ratio 73:27, $\delta_{\rm P}$ (CD_3CN) , 105.78 and 105.83] diastereomers of 2 were obtained in a ratio of 54:46 [$\delta_{\rm F}$ (CD₃CN), -30.06 and -29.96]. The epimerisation at phosphorus was not unexpected in the light of



Scheme 1 Reagents and conditions: i, Et₃NHF, DBU, MeCN, 15 min; ii, 80% AcOH, 1 h

published earlier results on the stereochemistry of nucleophilic substitution at phosphorus by fluoride ion.^{7,8} Also, thymidine 5'-O-phosphorofluoridothioate 4 has been synthesized according to the reaction sequence presented in the Scheme 2. Phosphitylation of 3'-O-methoxyacetylthymidine with N,Ndiisopropylamino-1,3,2-oxathiaphospholane² in a presence of 1H-tetrazole, followed by sulfurization yielded oxathiaphospholane 5. Reaction of 5 with triethylammonium fluoride-DBU furnished intermediate 3'-O-methoxyacetylthymidine 5'-Ophosphorofluoridothioate 6 which subsequently was deprotected with a concentrated solution of ammonia providing, after purification on Sephadex A-25, the final product 4** in 75% yield. The phosphorofluoridothioate monoesters 3 and 4 were hydrolytically stable even under basic conditions (conc. ammonia, room temp., 1 h), as proven by ³¹P NMR assay. Similarly, the resistance of 3 and 4 towards methanolysis, attempted in the presence of triethylamine or pyridine, has been observed. Attempts at internucleotide bond formation in reaction of 2 with 3'-O-acetylthymidine in the presence of strong bases such as ButOK, DBU and 2-tert-butylimino-2-diethylamino-1,3-dimethyl-1,3,2-diazaphosphinane (BEMP) have failed. Reactions were performed in DMF and their progress was followed by ³¹P NMR spectroscopy. Under these conditions, even after 18 h, formation of dithymidylyl (3',5') phosphorothioate $(T_{PS}T)$ was not observed. Instead, both 3 and 4 underwent intramolecular cyclization in the presence of an excess of ButOK (five-fold molar excess) leading to thymidine cyclic (3',5')phosphorothioate (cTMPS). Similarly, as previously observed⁹ for Bu^tOK-catalyzed cyclization of nucleoside 5'-O-p-nitrophenyl phosphophosphorothioates, in the reactions of rofluoridothioates 3 and 4 $[S_P]$ -cTMPS¹⁰ was also formed preferentially (ratio of $[S_P]$ -: $[R_P]$ -cTMPS ca. 2:1). Yields of Sephadex-purified cTMPS obtained from 3 and 4 were 57 and 38%, respectively. The lack of formation of $T_{PS}T$ in Bu^tOK assisted condensation of 2 with 3'-O-acetylthymidine was rather unexpected in the light of the results of von Tigerstrom and Smith¹¹ on effective formation of T_PT and other medium-sized oligothymidylates in the reaction of protected thymidine 3'-Ophosphorofluoridate with 3'-O-acetylthymidine.



Scheme 2 Reagents and conditions: i, N,N-diisopropylamino-1,3,2-oxathiaphospholane, 1*H*-tetrazole, 2 h, then elemental sulfur, 18 h; ii, Et₃NHF, DBU, CH₂Cl₂ MeCN, 15 min; iii, conc. NH₄OH, EtOH, 2.5 h

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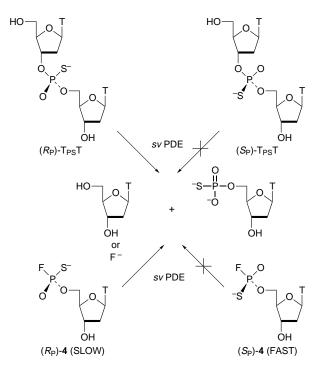


Fig. 1 Tentative assignment of absolute configuration of diastereomers of 4

In the light of earlier results on an enzymatic cleavage of thymidine 5'-O- or 3'-O-phosphorofluoridate assisted by *snake* venom (sv PDE)^{12,13} and *spleen*¹² phosphodiesterases (*spleen* PDE) it was tempting to study the thio analogues **4** and **3** as substrates for these enzymes. From the pioneering work of Eckstein¹⁴ and Benkovic¹⁵ demonstrating the stereoselectivity of *sv* PDE towards P-chiral diesters of phosphorothioic acid it was of interest to check if this enzyme can discriminate between the diastereomers of **3** or **4**.

5'-O-Phosphorofluoridothioate 4 was incubated with sv PDE^{††} and the progress of the enzymatic digestion was analyzed by RP-HPLC. It was found that sv PDE, if added to a diastereomeric mixture of 4, causes the stereoselective hydrolysis of the P-F bond of slow-eluted 4 leaving the fast-eluted diastereomer intact. Also, in the case of diastereomeric mixture of 3⁺⁺ only slow-eluted 3 underwent hydrolysis in a presence of sv PDE, albeit the reaction proceeded much slower than that observed for slow-eluted 4. Interestingly, under analogous conditions, the rate of hydrolysis of slow-4 by sv PDE was similar to that obtained during digestion of dithymidylyl (3',5') phosphate (T_PT). In the presence of spleen PDE both diastereomers of 3 were hydrolyzed while both diastereomers of 4 were resistant to this enzyme.** It was also found that 3 and 4 have no inhibitory activity§§ towards either phosphodiesterase. Results on the use of 4 for inhibition of thymidylate synthase will be published separately.16

In conclusion, we have found that the 1,3,2-oxathiaphospholane ring can be opened in the presence of DBU by fluoride anion leading to the appropriate phosphorofluoridothioates. The nucleoside 5'-O- or 3'-O-phosphorofluoridothioates obtained can be used in studies of the mode of action of nucleolytic enzymes. Comparative topological analysis of diastereomers of $T_{PS}T$ and 4 undergoing sv PDE- assisted hydrolysis allows the tentative assignment of absolute configuration of the sloweluted 4 as R_P (Fig. 1). Spleen and sv PDE-assisted hydrolysis of P-F bonds in compounds 3 and 4 is in agreement with earlier findings12,13 that these enzymes split nucleoside 3'-O- or 5'-Ophosphorofluoridate, respectively, giving rise to the appropriate nucleoside phosphates. From this perspective our data on enzymatic hydrolyses of 3 and 4 disagree with the results of Dabkowski et al.,17 who characterized thymidine 3'-O-phosphorofluoridate as the product of spleen PDE-assisted degradation of thymidin-3'-yl 2'-deoxyadenosin-5'-yl phosphorofluoridate. Adenosine 5'-O-phosphorofluoridate was found as the product of *sv* PDE-assisted hydrolysis of the same substrate. Besides the hydrolytic instability of the P–F bond of dinucleoside (3',5')phosphorofluoridates in buffered aqueous media¹⁸ yielding appropriate phosphates, even if thymidine 3'-O-phosphorofluoridate or adenosine 5'-O-phosphorofluoridate were the respective products of PDE-catalyzed hydrolyses, they would necessarily undergo further enzymatic degradation to phosphomonoesters.

Studies presented here were financially supported by the State Committee of Scientific Research (KBN), grant no 4 PO5F 023 10.

Notes and References

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[‡] Triethylammonium fluoride (1 M solution in THF) was obtained by mixing triethylamine tris(hydrofluoride) (1 equiv.) with triethylamine (2 equiv.).

\$\$ Compound **2** consists of a 1 : 1 mixture of diastereomers, $\delta_{\rm P}({\rm CD}_3{\rm CN}, 81 {\rm MHz})$ 54.52 (${}^{1}J_{\rm P-F}$ 1043 Hz), 54.58 (${}^{1}J_{\rm P-F}$ 1046 Hz); m/z (-FAB) 641.4 (M⁺⁺ -1).

¶ Compound **3** consists of a mixture of diastereomers (ratio 48:52), $\delta_P(D_2O, 81 \text{ MHz}) 53.72$, $({}^1J_{P-F} 1053 \text{ Hz})$, $53.74 ({}^1J_{P-F} 1055 \text{ Hz})$; $\delta_F(D_2O, 188 \text{ Mz}) - 31.4 ({}^1J_{P-F} 1043 \text{ Hz})$, $-31.2 ({}^1J_{P-F} 1046 \text{ Mz})$; m/z (-FAB) 339.1 (M⁺⁺ -1).

|| Compound **5** was obtained as a mixture of diastereomers, $\delta_P(CDCl_3, 81 \text{ MHz})$ 106.64, 106.82 (ratio 1 : 1); m/z (+FAB) 453.2 (M⁺⁺ +1).

Compound **4 consists of mixture of diastereomers (ratio 59:41), $\delta_{P}(D_{2}O, 81 \text{ Mz})$ 54.17, 54.20 (${}^{1}J_{P-F}$ 1053 Hz); $\delta_{F}(D_{2}O, 188 \text{ Mz})$ -35.6, -35.7 (${}^{1}J_{P-F}$ 1053 Hz); m/z (-FAB) 339.2 (M⁺⁺ -1).

†† The reaction mixture consists of 0.1 mM 4 or 3, 100 mM Tris–HCl pH 8.0, 20 mM MgCl₂, and *sv* PDE (0.01 U ml⁻¹); 37 °C; incubation time: 0.5 h for 4 and 16 h for 3.

^{‡‡} The reaction mixtures consist of 0.1 mM **3** (or **4**), 50 mM acetate buffer pH 5.0, and *spleen* PDE (0.15 U ml⁻¹), 37 °C; incubation time: 1 h for **3** and 16 h for **4**.

§§ Enzymatic digestions were performed under conditions mentioned above. T_PT and 3 (or 4) were used at equimolar concentrations (0.1 mM).

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Received in Glasgow, UK, 12th September 1997; 7/06636H