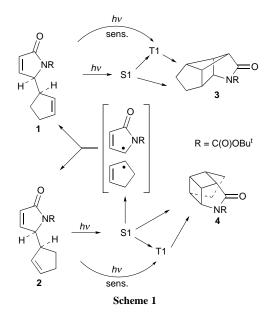
Diastereomer-differentiating photoisomerization of 5-(cyclopent-2-en-1-yl)-2,5-dihydro-1*H*-pyrrol-2-ones

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The diastereometric l- and u-dihydropyrroles 1 and 2 photoisometrize to azatetracyclodecanones 3 and 4, respectively.

According to the reactant-based Izumi-Tai classification,1 reactions wherein a l(ike) and an u(nlike) diastereomer exhibit differing behaviour are termed 'diastereomer differentiating'. Here we report the first example of a photochemical reaction where two diastereomeric hexa-1,5-dienes undergo photoisomerization to a crossed and a straight cycloadduct,² respectively. The 1:1 diastereomeric mixture of tert-butyl 5-(cyclopent-2-en-1-yl)-2-oxo-2,5-dihydro-1H-pyrrole-1-carboxylates 1 and 2 was available in 36% yield by alkylation of tert-butyl 2-tert-butyldimethylsilyloxy-1H-pyrrole-1-carboxylate³ with 3-bromocyclopentene. Separation was achieved by chromatography (SiO₂, Et₂O-hexane 1:1) affording the *u*-diastereomer **1**, mp 86 °C, whose structure was confirmed by X-ray analysis, and the *l*-diastereomer 2, mp 104 °C. Both (CD₃)₂CO-sensitized⁴ (300 nm) or direct (254 nm) irradiation of 1 afford the Boc-protected 5-azatetracyclo[4.4.0^{2,8}.0^{3,7}]decan-4-one 3 (mp 94 °C) in nearly quantitative yield. In contrast,



sensitized irradiation of 2 affords the Boc-protected 4-azatetracyclodecan-3-one 4 (oil) selectively (9:1) in 75% yield, but on direct irradiation a 1:1 mixture of 3 and 4 is formed. In both sets of experiments the conversion of 2 to product(s) proceeds slower than that of 1.

The fact that in the sensitized runs both diastereomers undergo regioselective intramolecular [2 + 2]cycloaddition but on direct irradiation only **1**, and not **2**, exhibit this behaviour can be explained in terms of differential ratios of efficiencies of cycloadduct formation *vs.* cleavage to a radical pair for singlet and triplet excited states of **1** and **2**, respectively. The results indicate that only for singlet excited **2** does this racemization, which causes cleavage, become competitive, most probably reflecting the higher activation energy barrier in the formation of a bicyclo[2.2.0]hexane unit, as in **4**, compared to the less strained⁵ bicyclo[2.1.1]hexane moiety of **3**. This differential behaviour is expected to be typical for all compounds containing a 3-(cyclopent-2-en-1-yl)cyclopentene framework.

The new compounds 1–4 have been fully characterized and have spectroscopic properties compatible with the structures assigned.

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Notes and References

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