Biosynthesis of phytyl side-chain of chlorophyll *a***: apparent reutilization of carbon dioxide evolved during acetate assimilation in biosynthesis of chloroplastidic isoprenoid**

Kensuke Nabeta,*† Tatsuto Saitoh, Kadzuya Adachi and Kaori Komuro

Department of Bioresource Science, Obihiro University of Agriculture and Veterinary Medicine, Obihiro 080, Japan

Equal incorporation of either acetyl methyl or carboxy carbons into all carbon atoms of the phytyl side-chain of chlorophyll *via* **doubly labelled acetyl CoA, with complete** loss of methyl hydrogens, indicated that $CO₂$ evolved from **the carboxy carbon during acetate assimilation through the TCA cycle may be reutilized** *via* **the reductive pentose phosphate cycle followed by the glycolic pathway, while incorporation of [2-13C]glycerol and [6,6-2H2]glucose into the phytyl chain demonstrated the simultaneous operation of two distinct pathways, both the mevalonate pathway and the 1-deoxy-D-xylulose mediating pathways (non-mevalonate pathway) in isopentenyl diphosphate formation within liverwort chloroplasts.**

The labelling pattern of the phytyl side-chain of chlorophyll was determined by incorporation studies using 2H- or 13C-labelled acetates, glycerol and glucose into cultured cells of liverwort, *Heteroscyphus planus*. The most unexpected feature of the labelling pattern was that either acetyl methyl or carboxy carbons were equally incorporated into all carbon atoms of the phytyl side-chain *via* doubly labelled acetyl CoA with complete loss of methyl hydrogens. The labelling pattern of the phytyl side-chain incorporating $[2^{-13}C]$ glycerol and $[6,6^{-2}H_2]$ glucose

Fig. 1 Labelling patterns of the phytyl side-chain of chlorophyll *a* and sesquiterpenes incorporating 13C labelled acetates, [2-13C]glycerol and [6,6-2H2]glucose in cultured cells of *H. planus*

together with ²H- and ¹³C-labelled mevalonate (MVA)^{1,2} revealed the simultaneous operation of two distinct pathways, the classical acetate/MVA (mevalonate) pathway³ and the novel 1-deoxy-p-xylulose (non-mevalonate) pathway,⁴ in isopentenyl diphosphate (IPP) formation within liverwort chloroplasts.

Cell cultures of *Heteroscyphus planus* were grown in MSK-4 medium⁵ (6–8 \times 75 ml), and fed with sodium [1-¹³C]-, [2-¹³C]-, [1,2-¹³C₂]-, [2,2,2-²H₃, 1-¹³C]- and [2,2,2-²H₃, 2-¹³C]-acetate, $(> 99$ atom%, 0.5 mmol), $[2⁻¹³C]$ glycerol (60 atom%, 0.5 mmol in 1 ml of 10% aq. EtOH) and $[6,6-2H_2]$ glucose (20 atom%, 11.1 mmol) under continuous light at 25 °C.6 Chlorophyll *a* was isolated and hydrolyzed, and the isolated phytol was acetylated by the procedure reported previously.

The ¹³C enrichment [sixteen ¹³C-enriched peaks with doublets due to 13 C $^{-13}$ C coupling (C-1–C-2, C-3–C-20, C-5–C-6, C-7–C-19, C- 9–C-10, C-11–C-18, C-13–C-14 and C-15–C-17) and four intense singlet peaks (C-4, C-8, C-12 and C-16) as shown in Fig. 1 and Table 1] in the phytols incorporating $[1-13C]$ -, $[2-13C]$ - and $[1,2-13C_2]$ -acetates was identical, indicating that the doubly 13C-labeled acetyl CoA was formed from either two carboxy carbons of [1-13C]acetate or two methyl carbons of [2-13C]acetate. Transposition of 13C label between the C-4 and C-5 carbons in IPP was also observed. The average intensity of the ${}^{13}C-{}^{13}C$ coupled peaks relative to the intense center peak in phytol incorporating $[1,2^{-13}C_2]$ acetate (observed

Table 1 13C enrichment of phytyl side-chain of chlorophyll *a* incorporating 13C-labelled acetates

	Enrichment (atom% excess)		
Carbon	$[1-13C]$ acetic acid	$[213C]$ acetic acid	$[1,2^{-13}C]$ acetic acid ^a
$C-1$	1.87	1.18	0.34
$C-2$	1.61	0.83	0.69
$C-3$	1.47	0.77	0.44
$C-4$	3.14	1.94	1.31
$C-5$	1.94	1.10	0.53
$C-6$	1.64	1.67	0.67
$C-7$	2.94	2.15	1.32
$C-8$	2.77	2.37	0.85
$C-9$	1.87	1.58	0.67
$C-10$	1.74	1.45	0.20
$C-11$	1.80	1.50	0.66
$C-12$	2.40	1.84	0.41
$C-13$	1.89	1.39	0.77
$C-14$	2.91	2.83	1.09
$C-15$	2.93	2.45	1.45
$C-16$	3.18	3.30	1.80
$C-17$	2.41	2.52	1.03
$C-18$	2.45	2.96	1.32
$C-19$	3.12	2.96	1.36
$C-20$	1.49	1.14	0.59
Average	2.28	1.90	0.89

*a J*_{13C}–13_C/Hz: C-1–C-2, C-3–C-20, C-5–C-6, C-7–C-19, C-9–C-19. C-11–C-18, C-13–C-14, C-15–C-17 have been previously reported. C-3–C- $4 = 34.2$, C-7–C-8 = 34.2, C-11–C-12 = 35.4 and C-15–C-16 = 35.4.

Fig. 2 Pathway for $[1,2^{-13}C_2]$ acetate from $[1^{-13}C]$ acetate *via* the TCA cycle, the reductive pentose phophate cycle and the glycollate pathway

relative intensity: 0.185) was much lower (\times 4.5) than that estimated (0.82) on the basis of the natural abundance of 13 C (1.08%) and 13C enrichment (0.89 atom% excess) indicating that reformed acetate, rather than intact acetate, was incorporated into MVA. The α -2H and β -2H isotopic peaks⁶ were not observed in phytol incorporating $[2,2,2^{2-2}H_3, 2^{-13}C]$ - and $[2,2,2^{-2}H_3, 1^{-13}\text{C}]$ -acetates, respectively, indicating the complete loss of 2H of the acetate during formation of $[1,2^{-13}C_2]$ acetate.

The pathway leading to two contiguously labelled acetate (or three contiguously labelled propionate) molecules from [2-¹³C]acetate, which was detected in a rare actinomycetes, *Actinomadura azurea*,7 and cell cultures of the vascular plants *Zea mays*8 and *Morus alba*9 was reasonably explained by the participation of the tricarboxylic acid cycle (TCA cycle). Formation of doubly labeled acetate from [1-¹³C]acetate, however, has not yet been reported. The results of this as well as earlier studies in which we examined the incorporation of [2-13C]glycine into the phytyl moiety of chlorophyll *a*1,2 suggest that the 13C-labelled carbon dioxide evolved during [1-13C]acetate assimilation through the TCA cycle is reutilized *via* the reductive pentose phosphate cycle. Carbon-13 labelled $CO₂$ was incorporated into ribulose 1,5-bisphosphate at C-1 by the mechanism shown in Fig. 2. The 13C label at C-1 in ribulose 1,5-bisphosphate was further translocated into the methylene carbon (C-2) of glycine *via* the glycolic pathway. The doubly labelled acetyl-CoA was formed from two C-2 carbons of the endogenously formed [2-13C]glycine *via* the glycolic pathway. There is indirect evidence that reassimilation of $CO₂$ liberated during acetate photoassimilation could be reutilized in glycolate production in *Chlorella*10 other eukarytic algae and vascular plants.11

Label was detected at C-2, C-3, C-6, C-7, C-10, C-11, C-14 and C-15 of the phytyl side-chain incorporating [2-13C]glycerol, all of which showed $J_{13_c-13_c}$ coupling $(J_{2,3} = 73.2,$ $J_{6,7}$ = 34.8, $J_{10,11}$ = 34.8 and $J_{14,15}$ = 34.2 Hz). When $[6,6-2H₂]$ glucose was added, the label was incorporated into C-1 (δ_{D} 4.15) and C-20 (δ_{D} 1.66) together with three methylene carbons (C-5, C-9 and C-13, δ_{D} 1.35–1.55, unresolved) and three methyl carbons (C-17, C-18, C-19, $\delta_{\rm D}$ 0.8–0.9, unresolved). The labelling pattern indeed confirmed the operation of the non-mevalonate pathway in biosynthesis of the phytyl sidechain. We found that the phytyl side-chain was also formed from MVA.1,2 Thus, it is suggested that biosynthesis of all compounds derived from geranylgeranyl diphosphate within liverwort chloroplasts proceeds *via* both the classical mevalonate pathway^{1,2,12} and the novel non-mevalonate pathway. The simultaneous operation of the mevalonate and the nonmevalonate pathways has also been detected in microorganisms without organelles.¹³The labelling pattern of β -barbatene, a predominant sesquiterpene hydrocarbon in *H. planus*, 14 incorporating labelled glycerol and glucose revealed that biosynthesis of the cytoplasmic terpenoids proceeds *via* the mevalonate pathway but not *via* the non-mevalonate pathway.

We present here evidence that $CO₂$ evolved from acetate assimilation is reutilized to biosynthesize isoprenoids in chloroplasts. The enzymes involved in the TCA cycle, the reductive pentose cycle and the glycolic acid pathway are separately localized in mitochondria, chloroplasts and peroxisomes, respectively. Thus reconstruction of acetate in chloroplasts requires consideration of the flux of $CO₂$ and the intermediates through the TCA cycle in mitochondria.

We are grateful to Dr H. Koshino (The Institute of Physical and Chemical Research) for 2H NMR analyses and to Professors H. Seto (Tokyo University) and K. Kakinuma (Tokyo Institute of Technology) for a generous gift of $[6,6^{-2}H_2]$ glucose. These investigations were supported by the Suhara Memorial Foundation and Grants-in-aid for Scientific Research (A. No. 08306021 and C. No. 08660125) from the Ministry of Education, Science and Culture, Japan.

Notes and References

† E-mail: knabeta@obihiro.ac.jp

- 1 K. Nabeta, T. Kawae, T. Kikuchi, T. Saitoh and H. Okuyama, *J. Chem. Soc., Chem. Commun.*, 1995, 2539.
- 2 K. Nabeta. T. Kawae, T. Saitoh and T. Kikuchi, *J. Chem. Soc., Perkin Trans. 1*, 1997, 261.
- 3 L. Ruzicka, A. Eschenmoser and H. Heusser, *Experimentia*, 1953, **9**, 357; S. L. Spurgeon and J. W. Porter, *Biosynthesis of Isoprenoid Compounds*, ed. J. W. Porter and S. L. Spurgeon, Wiley, New York, 1981; vol. 1, pp. 1–47; N. Quereshi and J. W. Porter, in *Biosynthesis of Isoprenoid Compounds*, ed. J. W. Porter and S. L. Spurgeon, Wiley, New York, 1981; vol. 1, pp. 47–93.
- 4 T. Duvold, J.-M. Bravo, C. Pale-Grosdemage and M. Rohmer, *Tetrahedron Lett.,* 1997, **38**, 4769; D. Arigoni, S. Sagner, C. Latzel, W. Eisenreich, A. Bacher and M. H. Zank, *Proc. Natl. Acad. Sci. USA*, 1997, **94**, 10 600.
- 5 R. Takeda and K. Katoh, *Planta*, 1981, **151**, 525; K. Nabeta, K. Katayama, S. Nakagawara and K. Katoh, *Phytochemistry*, 1993, **32**, 117.
- 6 J. C. Vederas, *Nat. Prod. Rep.*, 1987, **4**, 277 and references cited therein.
- 7 M. Ubukata, J. Uzawa and K. Isono, *J. Am. Chem. Soc.*, 1984, **106**, 2213.
- 8 D. J. Aschworth, R. Y. Lee and D. O. Adams, *Plant. Physiol.*, 1987, **85**, 463.
- 9 Y. Hano, T. Nomura and S. Ueda, *Chem. Pharm. Bull.*, 1989, **37**, 554; Y. Hano, A. Ayukawa, T. Nomura and S. Ueda, *J. Am. Chem. Soc.*, 1994, **116**, 4189.
- 10 M. J. Merrett and K. H. Goulding, *Planta*, 1967, **75**, 275.
- 11 W. Wiessner, *Photosynthesis II. Photosynthetic Carbon Metabolism and Related Processes*, ed. M. Gibbs and E. Latzko, Springer-Verlag, Berlin, 1979, pp. 181–189.
- 12 K. Nabeta, T. Ishikawa, T. Kawae and H. Okuyama, *J. Chem. Soc., Chem. Commun.*, 1995, 681.
- 13 H. Seto, H. Watanabe and K. Furihata, *Tetrahedron Lett.*, 1996, **37**, 7979.
- 14 K. Nabeta, K. Komuro, T. Utoh, H. Tazaki and H. Koshino, *Chem. Commun.*, 1998, 169.

Received in Cambridge, UK, 24th December 1997; 7/09274A