## An enantiocontrolled synthesis of a key intermediate to (+)-lactacystin

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An asymmetric synthesis of a key intermediate 16 to (+)-lactacystin 1 has been established starting from epoxide 2 *via* intramolecular mercurioamidation of allylic trichloroacetimidate 4 and concomitant addition-reduction of ester 13 by Pr<sup>i</sup> MgBr, in which reduction of the intermediate ketone proceeded with complete stereoselectivity.

Since neurotrophic factors are responsible for the survival and function of neurons,<sup>1</sup> they might be useful in the treatment of various nerve diseases.<sup>2</sup> Omura et al. screened a number of microbial culture samples to isolate the first non-protein neurotrophic agent (+)-lactacystin 1 from *Streptomyces* sp. OM-6519.3 Its structure, elucidated by NMR spectroscopy and X-ray crystallographic analysis, is composed of (R)-N-acetylcysteine and a unique pyroglutamic acid via a thioester linkage.4 (+)-Lactacystin inhibits cell proliferation, induces neuritogenesis and increases the intracellular cAMP level transiently in the Neuro 2A neuroblastoma cell line.3,5 Its intriguing structural features as well as potential therapeutic utility have engendered considerable interest in the fields of synthetic and medicinal chemistry. Here we describe a stereoselective synthetic route to (+)-lactacystin.<sup>6-9</sup> The key steps of our synthesis comprise tertiary amination of the olefinic double bond in allylic trichloroacetimidate 4 via mercurioamidation.<sup>10</sup> facile differentiation of the hydroxymethyl groups in 10 by ring formation and diastereoselective derivatization of ester 13 into alcohol 14.

The known epoxide  $2^{11} [\alpha]_D^{20} - 24.7$  (*c* 1.15, CHCl<sub>3</sub>), was treated with LDA to give allylic alcohol **3**,  $[\alpha]_D^{21}$  +10.4 (*c* 1.44, CHCl<sub>3</sub>), in 91% yield (Scheme 1). Only the primary hydroxy group of 3 was functionalized to a trichloroacetimidate. The crude monoimidate 4 was subjected to intramolecular mercurioamidation using mercuric trifluoroacetate with K<sub>2</sub>CO<sub>3</sub> to furnish a 1:1 diastereomeric mixture of oxazolines 5 in 92% overall yield after aqueous KBr work-up. Since oxidative demercuration<sup>12</sup> of **5** using  $O_2$  failed under a variety of reaction conditions, it was attempted by exposing 5 to TEMPO in the presence of LiBH<sub>4</sub> to provide the oxidized products 6 in 78% yield. The secondary hydroxy groups of 6 were protected with MeOCH<sub>2</sub>Cl (MOMCl) and then the silyl groups were removed to afford the corresponding primary alcohols in 84% overall yield. While PDC oxidation of the alcohols in DMF was sluggish, they were efficiently oxidized to carboxylic acids 8 in 78% yield by Swern oxidation<sup>13</sup> followed by KMnO<sub>4</sub> oxidation.<sup>14</sup> Complete hydrolysis and the ensuing cyclization were effected by heating 8 at reflux with ethanolic HCl in AcOH. The 2,2,6,6-tetramethylpiperidyl (TEMP) groups of the generated pyrrolidinones 9 were reductively cleaved in situ by adding zinc to the hot reaction mixture to produce trihydroxy pyrrolidinone **10**,  $[\alpha]_D^{19}$  +9.5 (*c* 0.95, MeOH), in 72% overall yield from **8**.

For the appropriate elaboration of the  $\alpha$ -hydoxymethyl groups in **10**, it was chemoselectively reacted with acetone under acidic conditions to give a 7:1 mixture of acetonides **11** and **12** in 95% combined yield (Scheme 2). After chromatographic separation, the primary alcohol **11**,  $[\alpha]_D^{20} + 31.4$  (*c* 1.10, CHCl<sub>3</sub>), was oxidized under Swern conditions and the resulting aldehyde reacted with Pr<sup>i</sup>MgBr under various reaction conditions to furnish a 1:1 mixture of alcohols **14** and **15** along with an appreciable amount of the reduced starting alcohol **11**.

Owing to the inefficient Grignard addition, 11 was converted into ester 13,  $[\alpha]_D^{21}$  +57.1 (c 1.70, CHCl<sub>3</sub>), in 90% yield. Subjection of 13 to 1 equiv. of Pr<sup>i</sup>MgBr provided the corresponding isopropyl ketone in 80% yield, the stereoselective reduction of which was attempted employing several reducing agents such as oxazaborolidine,<sup>15</sup> Ipc<sub>2</sub>Cl,<sup>16</sup> sodium triacetoxyborohydride,<sup>17</sup> NaBH<sub>4</sub> in the presence of diethylmethoxyborane,18 and so forth. However, the best stereoselectivity turned out to be 5:1 in favor of 14 with NaBH<sub>4</sub> in MeOH at 0 °C. Some experimentation revealed that an excess amount of Pr<sup>i</sup>MgBr reduced the generated isopropyl ketone to the alcohol 14. Accordingly, 13 was treated with > 2 equiv. of Pr<sup>i</sup>MgBr to give selectively only the desired diastereomeric alcohol 14,  $[\alpha]_{D^{20}}$  +40.5 (c 1.20, CHCl<sub>3</sub>), in 91% yield. Acidic hydrolysis of 14 yielded trihydroxy pyrrolidinone 16, mp 198-199 °C (decomp.),  $[\alpha]_D^{20}$  +16.2 (*c* 0.62, MeOH), quantitatively, the spectroscopic data of which are identical to those reported in the literature and which is a known intermediate to (+)-lactacystin 1.8,19

We have developed an enantioselective synthetic route to (+)-lactacystin 1 via several crucial steps, including animo hydroxylation of the olefinic double bond in 3, the hydrolytic cyclization of 8, and the regio- and stereo-selective functionalization of one hydroxymethyl group in 10; these should have versatility in the synthesis of its analogues.



Scheme 1 Reagents and conditions: i, LDA, THF, 0–24 °C; ii,Cl<sub>3</sub>CCN, DBU, EtCN, –78 °C; iii, Hg(O<sub>2</sub>CCF<sub>3</sub>)<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, THF, 0 °C, then aq. KBr; iv, TEMPO, LiBH<sub>4</sub>, THF, 24 °C; v, MOMCl, Pri<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>, 0–24 °C; vi, Bu<sub>4</sub>NF, H<sub>2</sub>O, THF, 45 °C; vii, (COCl)<sub>2</sub>, DMSO, Et<sub>3</sub>N; viii, 1 M KMnO<sub>4</sub>, 1.25 M NaH<sub>2</sub>PO<sub>4</sub>, Bu'OH, 24 °C; ix, conc. HCl, EtOH, AcOH, reflux, then Zn, reflux

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Scheme 2 Reagents and conditions: i, TsOH, acetone, 24 °C; ii, Jones' reagent, acetone, 0 °C; iii, CH\_2N\_2, THF, 0 °C; iv, Pr<sup>i</sup>MgBr, THF, -20 to 0 °C; v, TsOH, MeOH, 60 °C

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## **Notes and References**

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- 19 All new compounds showed satisfactory spectral data. Selected data for **16**: δ<sub>H</sub>(300 MHz, CD<sub>3</sub>OD) 0.94 (3H, d, *J* 6.7), 1.02 (3H, d, *J* 6.7), 1.11 (3H, d, J 7.5), 1.91–2.00 (1H, m), 2.79 (1H, p, J 7.5), 3.49 (1H, d, J 3.6), 3.78 (2H, s) and 4.40 (1H, d, J 7.5);  $\delta_{\rm C}$ (75.5 MHz, CD<sub>3</sub>OD) 9.5, 17.6, 22.7, 30.6, 42.6, 63.4, 70.7, 74.4, 79.1 and 181.6.

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