## Diastereoselective iodoamidation of 3-acetyloxybut-1-enylamines: simple synthesis of a precursor of aza sugars involving a pyrrolidine ring

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## 3-Acetyloxybut-1-enylamines 3–9 were easily transformed using iodine to pyrrolidine derivatives 3a–9a, precursors for aza sugars, *via* a diastereoselective iodoamidation.

Since the discovery that polyhydroxylated pyrrolidines are potent glycosidase inhibitors with potential therapeutic utility in the treatment of various diseases such as diabetes, <sup>1</sup> cancer<sup>2</sup> and viral infections, <sup>3</sup> much attention has been concentrated on the development of convenient and efficient routes to these compounds. In general, synthetic routes to aza sugars require azide displacement/reduction and intramolecular N-alkylative cyclisation with protecting group manipulation, <sup>4</sup> starting from carbohydrates or non-carbohydrates. Here we report the highly diastereoselective iodoamidation of 3-acetlyoxybut-1-enylamines 1 for the preparation of pyrrolidine derivatives 2 (*e.g.* anisomycine, <sup>5</sup> 3,4-dihydroxyprolinol, <sup>6</sup> swainsonine <sup>7</sup> and lentiginosine <sup>8</sup>) (Scheme 1).

The requisite substrates 3-9 were prepared easily by the usual method from commercially available L-tyrosin, L-phenyl-alanine and L-serine. † Fortunately, each diastereomeric allylic alcohol given by the Grignard reaction could be isolated in pure form by column chromatography. We chose the 9-phenyl-fluoren-9-yl (Pf) group for protection of the amine since this protecting group has been shown to inhibit deprotonation at the  $\alpha$ -position of  $\alpha$ -amino aldehydes.  $\alpha$ -Amino aldehydes having the Pf group are very stable to Grignard reaction conditions.

Compound **3** was treated with  $I_2$  under biphasic conditions (aq. NaHCO<sub>3</sub>-THF-Et<sub>2</sub>O = 2:1:1) at room temperature for 3 h to give the all *trans* pyrrolidine **3a** (*vide infra*) as the sole product in high yield *via* a diastereoselective iodoamidation. Although THF, MeOH,  $CH_2Cl_2$  and MeCN have been found to be acceptable solvents for iodoamidation, these solvents required much longer reaction times and resuted in 35–50% recovery of the starting material. As shown in Table 1, the optimum reaction conditions involved biphasic conditions to improve reactivity and affording pyrrolidine **3a**.

Treatment of **4** under the same conditions afforded a 25:1 ratio of the *cis* and *trans* isomers of pyrrolidine **4a**. Compounds **5** and **6** were cyclized according to these standard conditions to give the expected the corresponding pyrrolidines *trans*-**5a** and *cis*-**6a** in high yield, respectively. Compound **7** was also exposed to the same reaction conditions to give *cis*-**7a** in a ratio of 20:1 in 66% yield. In the case of starting materials *trans*-**8** and *cis*-**9**,‡ this solvent mixture was not suitable. Thus,

Scheme 1

compounds **8** and **9** were treated with  $I_2$  in THF to give the corresponding *trans*-**8a** and *cis*-**9a** in 92% (based on 65% conversion) and 88% (based on 80% conversion) yields, respectively (Table 2). The structures of all pyrrolidines **3a**-**9a** were confirmed by their characteristic spectroscopic data.§

The relative stereochemistries of the products  $\bf 3a$  and  $\bf 4a$  were determined from their  $^1H$  NMR spectra based on the coupling constant values and 2D NOE experiments. For *trans-3a*, proton  $H_3$  (t like,  $J_{2,3}$ ,  $J_{3,4}=5.1$  Hz) adjacent to the acetyloxy group gave weak correlation with protons ( $H_2$  and  $H_4$ ) adjacent to the p-tolylmethyl and iodine groups, but strong NOE cross peaks were observed between  $H_2$ – $H_4$  and  $H_2$ – $H_{5b}$ , thereby allowing one to assign its relative stereochemistry. In pyrrolidine cis- $\bf 4a$ , proton  $H_3$  (t like,  $J_{2,3}$ ,  $J_{3,4}=6.9$  Hz) adjacent to the acetyloxy group displayed strong mutual correlation with the protons ( $H_2$  and  $H_4$ ) adjacent to the p-tolylmethyl and iodine groups, thereby verifying the structure of cis- $\bf 4a$  as shown in Fig. 1.

Based on the coupling constant values in the high-field <sup>1</sup>H NMR spectra of *trans-***3a** and *cis-***4a**, the stereochemistries of *trans-***5a** (H<sub>3</sub>, t like,  $J_{2,3}$ ,  $J_{3,4} = 5.1$  Hz) and *cis* **6a** (H3, t like,  $J_{2,3}$ ,  $J_{3,4} = 6.9$  Hz) could be determined from each coupling constant value. The stereochemistries of *trans-***8a** (H3, dd,  $J_{2,3} = 4.8$ ,  $J_{3,4} = 3.4$  Hz) and *cis* **9a** (H3, dd,  $J_{2,3} = 2.4$ ,  $J_{3,4} = 4.2$  Hz) were also confirmed using coupling constant values.

Although numerous construction methods for the electrophilic cyclisation have been developed,  $^{12}$  the closest literature precedent to this haloamidation has been independently studied by the groups of Takahata $^{13}$  and Yoshida. $^{14}$  Takahata has shown that iodine-induced lactamization of  $\gamma$ , $\delta$ -unsaturated thioimidates proceeds regioselectively to provide  $\gamma$ -lactams. Yoshida has reported that N-(p-tolylsulfonyl)pent-4-enylamines were subjected to stereoselective haloamidation to afford mainly cis substituted pyrrolidines. Although these methodologies have been proven to be useful protocols, they are of limited use for the direct synthesis of polyhydroxylated aza sugars because these reactions proceed via 5-exo-trig cyclisation. Thus, we are the first to observe chiral induction on the pyrrolidine ring through an diastereoselective iodoamidation and to succeed in

Table 1 Solvent effects in stereoselective iodoamidationa

MeO	OAC NH Pf 3	MeO.	AcQ N Pf
Solvent	I <sub>2</sub> /equiv.	t/h	Yield $(\%)^b$
THF	3.0	15	$48^c$
MeOH	3.0	18	$45^d$
CH <sub>2</sub> Cl <sub>2</sub>	3.0	6	$50^e$
MeCN	3.0	6	$60^f$
Biphaseg	3.0	3	92

- <sup>a</sup> All reactions were carried out under room temperature. <sup>b</sup> Isolated yield.
- <sup>c</sup> 48% Recovery of starting material. <sup>d</sup> 40% Recovery of starting material.
- e 50% Recovery of starting material. f 35% Recovery of starting material.
- <sup>g</sup> NaHCO<sub>3</sub>-THF-Et<sub>2</sub>O = 2:1:1.

Table 2 Diastereoselective iodoamidation of 3-acetoxybut-1-enylamines with iodine

Substrates <sup>a</sup>	Conditions	Products <sup>b</sup>	Yield <sup>c</sup>
OAc NHPf	I <sub>2</sub> ,biphase, <sup>d</sup> room temp. 3 h	AcO, I	92% (sole)
OAC  NHPf  A R = p-tolyl	I <sub>2</sub> ,biphase, room temp. 3.5 h	R N Pf	90% (25 : 1)
OAC NHPf	I <sub>2</sub> ,biphase, room temp. 3 h	AcO <sub>N</sub> Pf	90% (62 : 1)
OAc NHPf	I <sub>2</sub> ,biphase, room temp. 3 h	AcO N Pf	90% • (5 : 1)
MOMO NHPf	I <sub>2</sub> ,biphase, room temp. 10 h	MOMO N Pf	66% (21 : 1)
Me O NHPf	I <sub>2</sub> ,THF, room temp. 10 h	HO N Pf	92% <sup>e</sup> (20 : 1)
Me Me NHPf 9	I <sub>2</sub> ,THF, room temp. 10 h	HO N Pf	88% <sup>f</sup> (sole)

 $^a$  All allylic alcohols are enantiomeric pure.  $^b$  The stereochemistry was signed by  $^1$ H NMR and 2D NOE experiments.  $^c$  Isolated yields.  $^d$  3 equiv. of I<sub>2</sub>, aq. NaHCO<sub>3</sub>—THF–Et<sub>2</sub>O = 2:1:1.  $^e$  Based on 65% conversion of starting material and 35% cleavage of the isopropylidene group of 8.  $^f$ Based on 80% conversion of starting material and 20% cleavage of the isopropylidene group of 9.

Fig. 1 NOE interactions derived from NOESY experiments.

using a strong electron-donating group, 9-phenylfluoren-9-yl (Pf), on an amine moiety.

In conclusion, we found that optically active starting materials 1 as chiral building blocks are converted easily to pyrrolidine derivatives 2 via a diastereoselective iodoamidation. These species should be valuable for the total synthesis of polyhydroxylated aza sugars having a pyrrolidine ring and may be suitable for substitution with various nucleophiles (NaN<sub>3</sub>, amines, alcohols, thiols and Grignard compounds), giving novel aza sugar derivatives. Thus, we are currently investigating the preparation of all six diasteromers of anisomycin and other 3,4-dihydroxyprolinols.

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## Notes and references

 $\dagger$  Starting materials 3–9 were prepared by a sequential reaction, namely, free animo acids were treated with TMSCl in MeOH to give methyl esters, the amino groups of which were protected with PfBr in CH<sub>2</sub>Cl<sub>2</sub>. The methyl esters were then subjected to reduction and Swern oxidation to afford aldehydes, which were reacted with vinylmagnesium bromide to give separable allylic alcohols.

‡ The relative stereochemistries of the corresponding acetonides *trans-8* (J = 8.1 Hz for the proton on oxygen) and *cis-9* (J = 1.0 Hz for the a proton on oxygen) were confirmed by coupling constant analysis.

§ Selected data for 3a: colorless prisms, mp 67–68 °C;  $[\alpha]_D^{22}$  +33.0 (c 1.3, CHCl<sub>3</sub>);  $\delta_H$  (500 MHz, CDCl<sub>3</sub>) 1.61 (3H,s), 2.37 (1H, dd, J 4.0, 13.6), 2.44 (1H, dd, J 6.7, 10.1), 2.67 (1H, dd, J 9.7, 13.6), 2.73 (1H, m), 3.00 (1H, dd, J 2.7, 10.2), 3.28 (1H, dt, J 4.3, 5.1), 3.71 (3H, s), 4.56 (1H, t-like, J 2.3, J 3.4 5.1), 6.72 (2H, d, J 8.7), 6.89 (2H, d, J 8.7), 7.23–7.34 (13H, m). For 4a: colorless prisms, mp 70–71 °C;  $[\alpha]_D^{20}$  —22.9 (c 1.2, CHCl<sub>3</sub>);  $\delta_H$  (500 MHz, CDCl<sub>3</sub>): 2.08 (3H, s), 2.19 (1H, dd, J 3.4, 14.1), 2.65 (1H, dd, J 3.7, 9.2), 2.86 (1H, dd, J 10.9, 14.1), 3.17 (1H, dd, J 9.2, 11.2), 3.43 (1H, ddd, J 3.5, 7.2, 10.8), 3.55 (1H, ddd, J 3.7, 7.2, 8.1), 5.17 (1H, tlike, J 2.3, J 3,4 6.9), 6.65 (2H, d, J 8.7), 6.87 (2H, d, J 8.7), 7.24–7.75 (13H, m).

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