

# Lifting the lid on metatungstate. $^1\text{H}$ and $^{183}\text{W}$ NMR study of the six electron reduced anion $[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{34}(\text{OH})_3]^{3-}$

Colette Boskovic,<sup>a</sup> Maruse Sadek,<sup>\*b</sup> Robert T. C. Brownlee,<sup>b</sup> Alan M. Bond<sup>c</sup> and Anthony G. Wedd<sup>\*a</sup>

<sup>a</sup> School of Chemistry, University of Melbourne, Parkville, Victoria 3052, Australia.  
E-mail: t.wedd@chemistry.unimelb.edu.au

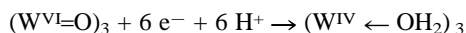
<sup>b</sup> Department of Chemistry, La Trobe University, Bundoora, Victoria 3083, Australia.  
E-mail: m.sadek@latrobe.edu.au

<sup>c</sup> Department of Chemistry, Monash University, Clayton, Victoria 3168, Australia

Received (in Cambridge, UK) 4th January 1999, Accepted 15th February 1999

The distribution of the eleven protons present in a six-electron reduced form of metatungstate,  $[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{34}(\text{OH})_3]^{3-}$ , in dry  $\text{CD}_3\text{CN}$  is mapped by  $^1\text{H}$  and  $^{183}\text{W}$  NMR, allowing assessment of structural changes which accompany reduction.

The four trinuclear capping units of Keggin polyoxo anions  $[\text{X}^n\text{W}^{\text{VI}}_{12}\text{O}_{40}]^{n-8}$  ( $T_d$  point symmetry; see framework in Fig. 1) are linked to define a tetrahedral cavity encapsulating atom X.<sup>1</sup> Six-electron reduction leads, for  $\text{X}^n = 2\text{H}^{\text{I}}, \text{B}^{\text{III}}, \text{Si}^{\text{IV}}$ , to the 'poly-browns'.<sup>1,2</sup> X-Ray crystallographic and aqueous solution NMR data suggest effective  $C_{3v}$  point symmetry for these species, consistent with localisation of the added electrons in a single trinuclear cap whose terminal oxo ligands are protonated to aqua ligands.<sup>3-6</sup>



The salt  $(\text{NH}_4)_4[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{35}(\text{OH})_2]$  **I** was synthesised by controlled potential electrolysis in aqueous HCl and converted to  $(\text{NBu}_4)_3[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{34}(\text{OH})_3]$  **II** by phase transfer.<sup>7</sup> In  $\text{D}_2\text{O}$ , **I** exhibits a single  $^1\text{H}$  NMR resonance (two equivalent internal hydrogen atoms) and three  $^{183}\text{W}$  resonances (3:3:6) consistent with effective  $C_{3v}$  point symmetry.<sup>2-4,8</sup> However, in dry  $\text{CD}_3\text{CN}$ , **II** exhibits five  $^1\text{H}$  resonances (a–e) integrating for eleven protons and seven  $^{183}\text{W}$  resonances (A–G) integrating for three  $\text{W}^{\text{IV}}$  and nine  $\text{W}^{\text{VI}}$  centres (Fig. 2, Table 1).

The intensities of the  $^1\text{H}$  (4:2:1:2:2) and  $^{183}\text{W}$  (1:2:2:2:2:2:1) resonances imply the presence of a mirror plane consistent with effective  $C_s$  point symmetry. It is apparent that proton exchange rates are reduced significantly in dry  $\text{CD}_3\text{CN}$ , allowing detection of surface-bound protons and differentiation of inequivalent tungsten sites. What is the driving force responsible for the effective change from  $T_d$  to  $C_s$

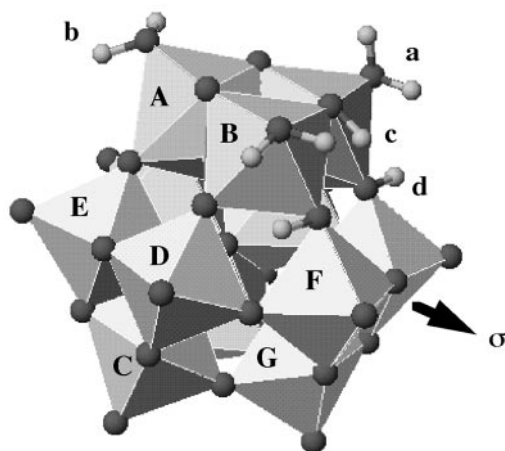


Fig. 1 Polyhedral representation of  $[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{34}(\text{OH})_3]^{3-}$ . W atoms are centred in each 'octahedron' of O atoms.

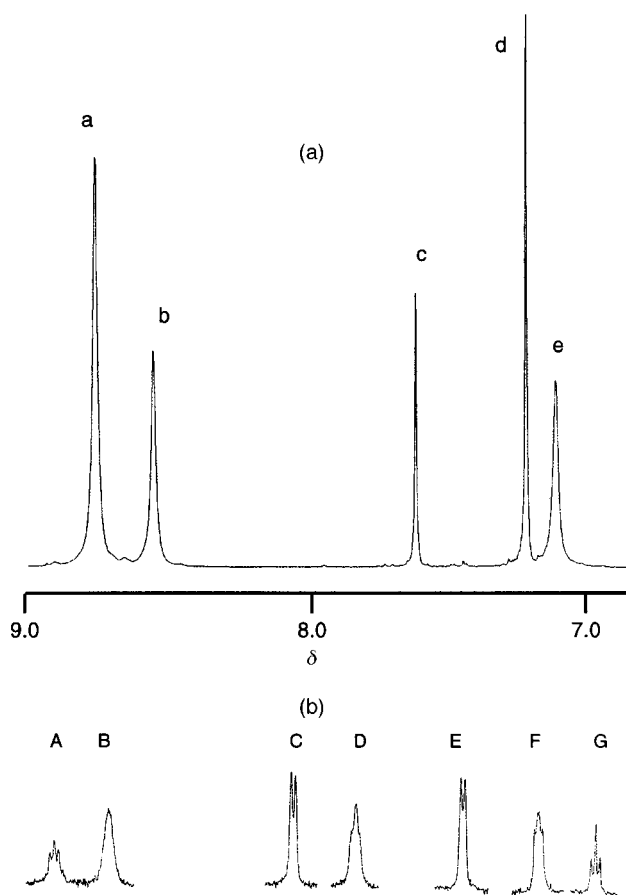


Fig. 2 NMR spectra of **II** (0.15 M) in  $\text{CD}_3\text{CN}$  (anion resonances only) (a)  $^1\text{H}$ , (b)  $^{183}\text{W}$  (99 atom%  $^{183}\text{W}$ ).

symmetry that accompanies the six electron reduction process?

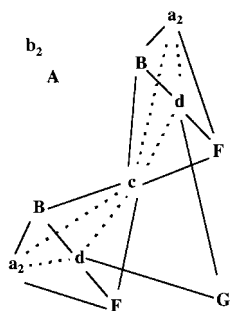
$\text{H}_2\text{O}$  addition experiments assign resonance e to the slowly exchanging internal protons  $\text{H}_e$  and indicate that the protons  $\text{H}_a$  and  $\text{H}_b$  are in chemically similar sites.<sup>10</sup> Their relative intensities  $a : b = 4 : 2$  allow assignment to the  $\text{W}^{\text{IV}}$ -aqua ligand protons. Tungsten resonances A and G of intensity 1 must be assigned to the unique  $\text{W}^{\text{IV}}$  and  $\text{W}^{\text{VI}}$  atoms lying on the mirror plane. The five resonances B–F of intensity 2 are due to five pairs of W atoms related by the mirror plane. B must be assigned to  $\text{W}^{\text{IV}}$  and so the fragments  $\text{W}_A\text{O}(\text{H}_b)_2$  and  $\{\text{W}_B\text{O}(\text{H}_a)_2\}_2$  are identified (Fig. 1).

A sample of **II** was enriched to 99 atom%  $^{183}\text{W}$ . Its  $^{183}\text{W}$  NMR spectrum featured resolved fine structure (13–18 Hz) assigned as  $^2J_{\text{WOW}}$  coupling characteristic of corner-sharing WOW bridges [Fig. 2(b)].<sup>11</sup> Assigned resonances A and G are triplets consistent with corner-sharing to two equivalent W

**Table 1** NMR data for  $(\text{NBu}_4)_3[(\text{H})_2\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{34}(\text{OH})_3]$  **II** (0.15 M,  $\text{CD}_3\text{CN}$ )

Nucleus	Resonance	$\delta$	$W_{1/2}^a/\text{Hz}$	Relative intensity
$^1\text{H}$	$\text{H}_a$	8.76	7.7	3.9
	$\text{H}_b$	8.55	7.3	2.0
	$\text{H}_c$	7.61	2.5	1.0
	$\text{H}_d$	7.22	2.4	2.0
	$\text{H}_e$	7.11	8.6	2.1
$^{183}\text{W}$	$\text{W}_A$	1413.2	4.4	1
	$\text{W}_B$	1402.7	5.7	2
	$\text{W}_C$	-42.6	5.7	2
	$\text{W}_D$	-64.5	7.1	2
	$\text{W}_E$	-128.7	4.5	2
	$\text{W}_F$	-171.0	3.4	2
	$\text{W}_G$	-187.4	5.2	1

<sup>a</sup> For natural abundance samples.



**Fig. 3** Map of dipolar HH (----) and scalar WH (—) connectivities to protons  $\text{H}_a$ – $\text{H}_d$  in **II**. Longer range interactions  $bd$  and  $Ad$  are also observed but are omitted for clarity.

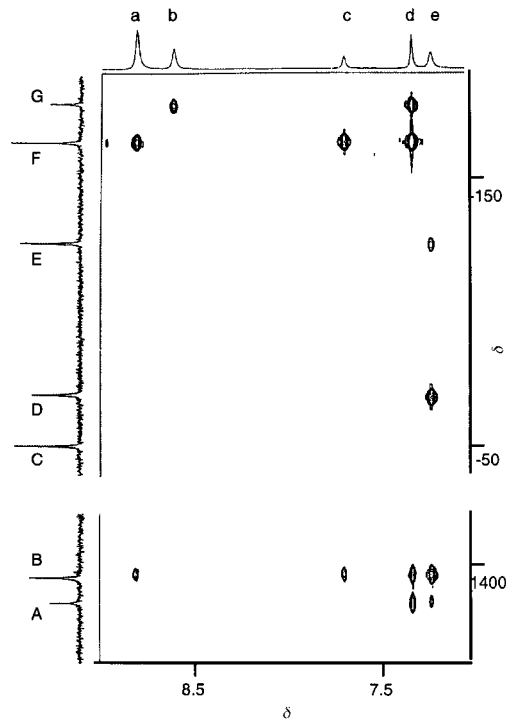
atoms (Fig. 1). On the other hand, C and E are doublets. A COSY experiment displayed a  $\text{W}^{\text{VI}}\text{W}^{\text{VI}}$  cross-peak CG which allows assignment of C to the pair of tungsten atoms corner-sharing to unique atom  $\text{W}_G$  and E to the pair of tungsten atoms corner-sharing with unique atom  $\text{W}_A$  (Fig. 1). The only other  $\text{W}^{\text{VI}}\text{W}^{\text{VI}}$  COSY crosspeak detected was DF and tungsten atoms  $\text{W}_D$  and  $\text{W}_F$  were assigned on the basis of multiple WOW corner-sharing interactions [Figs. 1, 2(b); *vide infra*: ref. 12].

Assignment and location of the three protons  $\text{H}_c$  and  $\text{H}_d$  were addressed initially by a HH ROESY experiment. Cross peaks  $ac$ ,  $ad$ ,  $bd$  and  $cd$  were detected. These dipolar interactions are mapped in Fig. 3. No interactions were observed between internal protons  $\text{H}_c$  and any of the external protons  $\text{H}_a$ – $\text{H}_d$ .

Mapping of scalar interactions utilised WH HMQC experiments.  $^{183}\text{W}$ – $^1\text{H}$  coupling was not detected in the 1D spectra (*cf.* Fig. 2) implying that  $^2J_{\text{WOH}} < W_{1/2} = 2$ – $9$  Hz, allowing constraints to be set on mixing times. A spectrum is shown in Fig. 4 and the scalar connectivities are included in Fig. 3.<sup>12</sup>

The mapping indicates that resonances c and d may be assigned to three hydroxyl protons associated with the reduced  $\text{W}^{\text{IV}}_3(\text{AB}_2)$  and unique  $\text{W}^{\text{VI}}_3(\text{GF}_2)$  caps (Figs. 1, 3). Hydrogen oxide ligands  $\text{H}_3\text{O}_2^-$  ligands which bridge pairs of metal atoms have been identified previously,<sup>13</sup> but similar stereochemistry does not appear possible here.

The  $\text{C}_s$  point symmetry of **II** is imposed by the positions of the three hydroxyl protons  $\text{H}_c$  and  $\text{H}_d$  (Fig. 1). The electronic feature which drives this localisation of protons is the localisation of the six added electrons in the reduced cap  $\text{AB}_2$ . In six-electron reduced  $\text{K}_5[\text{B}^{\text{III}}\{\text{W}^{\text{IV}}_3(\text{OH}_2)_3\}\text{W}^{\text{VI}}_9\text{O}_{37}]\cdot 13.5\text{H}_2\text{O}$ , the  $\text{W}^{\text{IV}}$ – $\text{W}^{\text{IV}}$  bond lengths [mean  $2.543(3)$  Å] in the reduced cap are *ca.*  $0.77$  Å shorter than the  $\text{W}^{\text{VI}}\dots\text{W}^{\text{VI}}$  separations in the oxidised caps.<sup>6</sup> The resulting mismatch in dimensions causes the boron atom to move *ca.*  $0.4$  Å within the internal  $\text{BO}_4$  unit and away from the reduced cap but  $\text{C}_{3v}$  symmetry is retained. The internal cavity in **II** features two hydrogen atoms only and this anion accommodates the mismatch in a different way: three WOW links are converted to longer  $\text{W}(\text{OH})\text{W}$  links [Fig. 1]. Ongoing work will provide more details.



**Fig. 4**  $^{183}\text{W}$ – $^1\text{H}$  HMQC spectrum of **II** (0.15 M, natural abundance) in  $\text{CD}_3\text{CN}$  ( $t_m = 60$  ms;  $128 \times 4$  K, 256 transients per increment;  $25^\circ\text{C}$ ).

A. M. B. and A. G. W. acknowledge support from the Australian Research Council. Associate Professor Richard Robson is thanked for stimulating discussions.

## Notes and references

- M. T. Pope, *Heteropoly and Isopoly Oxometalates*, Springer-Verlag, New York, 1983.
- J. P. Launay, *J. Inorg. Nucl. Chem.*, 1976, **38**, 807.
- K. Pieprgrass and M. T. Pope, *J. Am. Chem. Soc.*, 1987, **109**, 1586.
- K. Pieprgrass and M. T. Pope, *J. Am. Chem. Soc.*, 1989, **111**, 753.
- Y. Jeannin, J. P. Launay, M. A. Seid Sedjadi, *Inorg. Chem.*, 1980, **19**, 2933.
- T. Yamase and E. Ishikawa, *J. Chem. Soc., Dalton Trans.*, 1996, 1619.
- A solution of ammonium metatungstate in HCl was reduced to the  $6e^-$  level.<sup>3</sup> The ammonium salt was isolated after volume reduction and precipitation with MeOH. Phase transfer<sup>4</sup> mediated via a solution of  $\text{NBu}_4\text{Br}$  in  $\text{CH}_2\text{Cl}_2$  led to the  $\text{NBu}_4\text{Br}$  salt.
- J. P. Launay, M. Boyer and F. Chauveau, *J. Inorg. Nucl. Chem.*, 1976, **38**, 243.
- $^{183}\text{W}$  spectra were acquired on a Bruker DRX500 spectrometer equipped with a broadband probe at  $30^\circ\text{C}$ . Spectra were acquired at 20.837 MHz using a  $60^\circ$  pulse of  $26$   $\mu\text{s}$ . Spectra were processed with a line broadening of 1 Hz. COSY 90 spectra were acquired using a  $90^\circ$  pulse of  $39$   $\mu\text{s}$ . Experiments over the  $\text{W}^{\text{VI}}$  and  $\text{W}^{\text{IV}}$  chemical shift regions were acquired separately owing to their large separation. Gradient non-phase sensitive HMQC experiments were acquired on a Bruker DRX400 spectrometer equipped with an inverse probe at  $25^\circ\text{C}$ . The probe was tuned to 400.130 MHz for  $^1\text{H}$  and 16.671 MHz for  $^{183}\text{W}$ .  $90^\circ$  pulses of 13.5, 15.3 and 25.7  $\mu\text{s}$  for  $^1\text{H}$  and 44  $\mu\text{s}$  for  $^{183}\text{W}$  were employed.
- $\text{H}_c$  shifts by  $< 0.1$  ppm ( $W_{1/2}$ , 5–9 Hz) in the range 0–100 equiv. added  $\text{H}_2\text{O}$ . Resonances  $\text{H}_a$ ,  $\text{H}_b$  and  $\text{H}_e$ ,  $\text{H}_d$  broaden beyond detectability at  $\approx 100$  and 25 equiv. added  $\text{H}_2\text{O}$ , respectively.
- J. Lefebvre, F. Chauveau, P. Doppelt and C. Brevard, *J. Am. Chem. Soc.*, 1981, **103**, 4589. Coupling characteristic of edge sharing is not resolved (*ref.* 4); the situation is also complicated by long range WW coupling.
- The assignments of  $\text{W}_D$  and  $\text{W}_F$  are strengthened by the absence of Dd and the presence of Fd and Gd crosspeaks.
- M. Ardon and A. Bino, *Struct. Bonding (Berlin)*, 1987, **65**, 1.