

# Synthesis of amino acid derivatives *via* enantio- and diastereoselective Pd-catalyzed allylic substitutions with a non-stabilized enolate as nucleophile†

Thomas D. Weiß,<sup>a</sup> Günter Helmchen<sup>\*a</sup> and Uli Kazmaier<sup>b</sup>

<sup>a</sup> Organisch-Chemisches Institut, Universität Heidelberg, Im Neuenheimer Feld 270, 69120, Heidelberg, Germany. E-mail: en4@ix.urz.uni-heidelberg.de; Fax: +49 6221 4205; Tel: +49 6221 8401

<sup>b</sup> Institut für Organische Chemie, Universität des Saarlandes, Am Stadtwald, Geb. 33/2, 66123, Saarbrücken, Germany

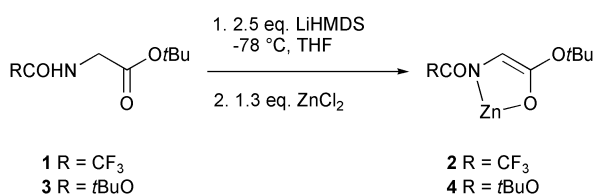
Received (in Cambridge, UK) 18th April 2002, Accepted 1st May 2002

First published as an Advance Article on the web 15th May 2002

Diastereomer ratios of up to 95 : 5 and enantiomeric excesses of up to 95% were achieved in Pd-catalyzed asymmetric allylic substitutions with zinc enolates of glycine esters as nucleophiles; a remarkable effect of the ligand on the diastereoselectivity of the substitution was found.

Pd-catalyzed asymmetric allylic substitution is a valuable synthetic method.<sup>1</sup> Most C-nucleophiles in use are conjugated carbanions prepared from CH-acidic compounds with  $pK_a < ca. 20$ , *i.e.* malonates and  $\beta$ -keto esters. With symmetric nucleophiles, giving rise to only one stereogenic center in the coupling step, good yields and high enantioselectivities can now be consistently obtained. With unsymmetrical C-nucleophiles, *e.g.*  $\beta$ -keto esters or imines of amino acid esters, mixtures of diastereomers are usually generated because of configurational lability of the products. An exception was recently reported by Trost *et al.* who used substituted azlactones as nucleophiles which gave rise to  $\alpha$ -alkylated  $\gamma,\delta$ -unsaturated amino acids with a high degree of stereoselectivity.<sup>2</sup>

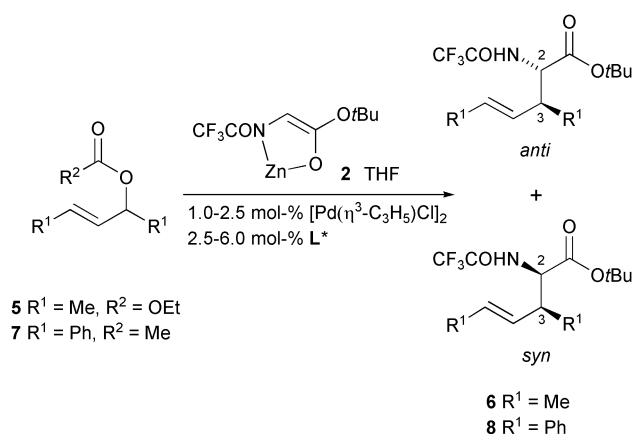
The epimerization problem does not exist for products generated with more basic carbanions such as enolates of esters and ketones; however, these were found to be problematic because of side reactions, in particular elimination and cyclopropane formation.<sup>3</sup> To overcome this problem, considerable effort has been expended on varying the counter ion of these enolates. Some success was obtained with tin and boron enolates.<sup>4</sup> In our laboratory, successful alkylations were achieved with the Zn enolates **2** and **4** (Scheme 1),<sup>5</sup> which gave excellent degrees of diastereoselectivity and made valuable,  $\gamma,\delta$ -unsaturated amino acids available (Scheme 2).



Scheme 1

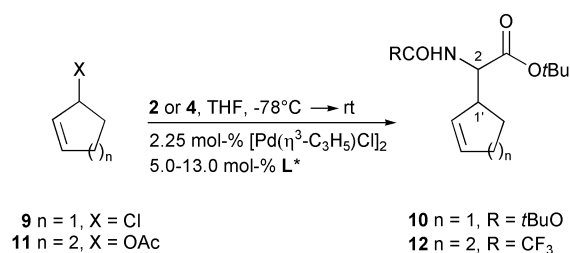
With our group's background on asymmetric allylic substitutions,<sup>6</sup> we of course initiated an investigation on an enantioselective version of this reaction.<sup>7</sup> Related work was recently reported by Braun *et al.* who found examples of highly enantio- and diastereoselective reactions of ketone enolates with 1,3-diphenylallyl acetate<sup>8</sup> and Trost *et al.* who obtained excellent results for allylic substitutions with enolates of  $\alpha$ -tetralones.<sup>9</sup>

In order to assess the scope of the reactions with nucleophiles **2** and **4**, we not only used the particularly benevolent 1,3-diphenylallyl acetate (**7**) but selected a representative set of



Scheme 2

substrates including a large and a small acyclic (Scheme 2) and two cyclic allylic derivatives (Scheme 3). As chiral ligands<sup>10</sup> we used phosphinoxazolines **L2** and **L3**, which are particularly suited for acyclic substrates as well as the cymantrene derivative **L4** and the phosphinomyrtanic acid **L5** which are better suited for cyclic compounds (Fig. 1).



Scheme 3

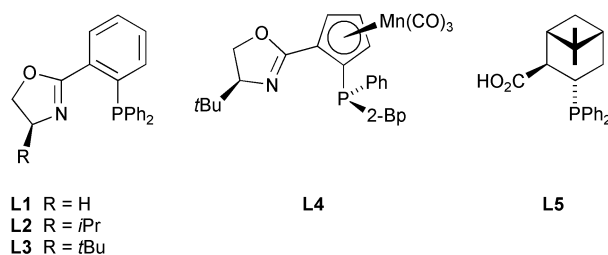


Fig. 1 Chiral ligands.

Each of the reactions yields four stereoisomers. These isomers were base-line separable by GLC on the Chiralasil-L-Val phase. Individual isomers are characterized by their GLC retention times as given in Table 1.<sup>11</sup> Absolute and relative configurations were assigned for the full set of compounds **6**, **10**

† This work is dedicated to Professor Walter Siebert on the occasion of his 65th birthday.

**Table 1** Results of allylic alkylation

Entry	Substrate	Nucleophile	Ligand	Yield (%)	dr <sup>a</sup> <i>anti</i> : <i>syn</i>	ee (%) <sup>a</sup> <i>anti</i>	ee (%) <sup>a</sup> <i>syn</i>
1 <sup>b</sup>	<b>5</b>	<b>2</b>	<b>L1</b>	75	82:18		
2 <sup>b</sup>	<b>5</b>	<b>2</b>	<b>L2</b>	85	71:29	<b>53 6c</b>	<b>19 6b</b>
3 <sup>b</sup>	<b>5</b>	<b>2</b>	<b>L3</b>	54	89:11	<b>66 6c</b>	<b>7 6b</b>
4 <sup>b</sup>	<b>5</b>	<b>2</b>	<b>L5</b>	78	51:49	<b>6 6a</b>	<b>74 6b</b>
5 <sup>c</sup>	<b>7</b>	<b>2</b>	<b>L1</b>	80	89:11		
6 <sup>c</sup>	<b>7</b>	<b>2</b>	<b>L2</b>	62	95:5	<b>94 8d</b>	<b>91 8a<sup>d</sup></b>
7 <sup>c</sup>	<b>7</b>	<b>2</b>	<b>L3</b>	71	87:13	<b>87 8d</b>	<b>61 8a<sup>d</sup></b>
8 <sup>e</sup>	<b>9</b>	<b>4</b>	<b>L1</b>	29	65:35		
9 <sup>e</sup>	<b>9</b>	<b>4</b>	<b>L4</b>	72	30:70	<b>80 10a</b>	<b>95 10d</b>
10 <sup>e</sup>	<b>9</b>	<b>4</b>	<b>L5</b>	50	29:71	<b>84 10b</b>	<b>58 10c</b>
11 <sup>f</sup>	<b>11</b>	<b>2</b>	<b>L1</b>	65	58:42		
12 <sup>f</sup>	<b>11</b>	<b>2</b>	<b>L4</b>	80	51:49	<b>75 12a</b>	<b>81 12d</b>
13 <sup>f,g</sup>	<b>11</b>	<b>2</b>	<b>L4</b>	51	48:52	<b>89 12a</b>	<b>77 12d</b>
14 <sup>f</sup>	<b>11</b>	<b>2</b>	dppp	81	82:18		
15 <sup>f</sup>	<b>11</b>	<b>2</b>	<b>L5</b>	66	20:80	<b>93 12b</b>	<b>53 12c</b>

<sup>a</sup> The designations **a**, **b**, **c**, **d** are defined by the GLC retention times as follows: Chrompack Chirasil-L-Val; He; **6**: 80 °C,  $t_R$  = 17.0 min ((2*R*,3*R*)-**6a**), 17.9 min ((2*R*,3*S*)-**6b**), 22.9 min ((2*S*,3*S*)-**6c**), 24.5 min ((2*S*,3*R*)-**6d**). **8**: 165 °C,  $t_R$  = 49.2 min ((2*R*,3*R*)-**8a**), 52.3 min ((2*S*,3*S*)-**8b**), 56.2 min ((2*R*,3*S*)-**8c**), 64.5 min ((2*S*,3*R*)-**8d**); **10**: 95 °C,  $t_R$  = 12.8 min ((2*R*,1'*R*)-**10a**), 14.7 min ((2*S*,1'*S*)-**10b**), 15.4 min ((2*R*,1'*S*)-**10c**), 19.7 min ((2*S*,1'*R*)-**10d**). **12**: 95 °C,  $t_R$  = 25.0 min ((2*R*,1'*R*)-**12a**), 28.5 min ((2*S*,1'*S*)-**12b**), 30.1 min ((2*R*,1'*S*)-**12c**), 36.9 min ((2*S*,1'*R*)-**12d**). <sup>b</sup> 3.0 mol % of [Pd( $\eta^3$ -C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub>. <sup>c</sup> 1.2 mol % of [Pd( $\eta^3$ -C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub>. <sup>d</sup> The absolute configuration is assigned on the basis of analogy. <sup>e</sup> 0.5 mol % of [Pd( $\eta^3$ -C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub>. <sup>f</sup> 2.25 mol % of [Pd( $\eta^3$ -C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub>. <sup>g</sup> Al(O*i*Pr)<sub>3</sub> was used instead of ZnCl<sub>2</sub>.

and **12** by saponification, subsequent iodolactonization and X-ray crystal structure analysis. For **8** the absolute configuration of the *anti*-diastereomer was assigned *via* bromolactonization, while the absolute configuration of *syn*-**8** is tentative.

The reaction of substrate **5** with enolate **2** was previously carried out with PPh<sub>3</sub> as ligand which gave rise to a diastereomer ratio (dr) of 92:8 in favor of the *anti* diastereomer.<sup>5</sup> With the achiral phosphinoxazoline **L1** a dr of 82:18 was obtained (Table 1, entry 1). The chiral ligands **L2** and **L3** gave similar dr and ee of up to 66% for the *anti* product (*S,S*)-**6** (entries 2, 3). Chiral ligand **L5** gave only dr = 1:1 but 74% ee for the *syn* product (*R,S*)-**6** (entry 4). The steric course and the degree of enantioselectivity with respect to C(3) is very similar to that of the corresponding reactions of malonate enolates.<sup>12</sup> Thus, in accordance with our previous findings, the reactions of the non-stabilized enolate **2** closely resemble those of stabilized enolates. The dependence of the dr on the ligand is remarkable and indicates that both enantio- and diastereoselectivities of the substitutions can be controlled externally, *i.e.* by the ligand.

As anticipated, higher levels of selectivity were achieved with 1,3-diphenylallyl acetate (**7**) as substrate. With the achiral ligand **L1** a dr of 89:11 (80 % yield) (entry 5), with the chiral ligands **L2** and **L3** diastereoselectivity of up to 95:5 and ee of up to 94% could be obtained (entries 6, 7). Recrystallization of the crude product from diethyl ether-*n*-hexane gave diastereo- and enantiomerically pure major isomer (2*S*,3*R*)-**8**.

Allylic alkylations of cyclopent- and hex-2-enyl acetate (**9** and **11**) (Scheme 3) lead to cyclopent- and cyclohexenylglycines **10** and **12**, respectively, which display antibacterial activity and, therefore, are of particular interest.<sup>13</sup> For cyclopentenyl chloride (**9**) as substrate and nucleophile **4** the achiral ligand **L1** induced an *anti*-*syn* ratio of 65:35 (entry 8). The chiral ligands **L4** and **L5** gave rise to a reversal in the *anti*-*syn* ratio and enantioselectivity of up to 84% for *anti* and 95% ee for *syn* product **10** (entries 9,10). For the homologous cyclohex-2-enyl system the phosphinoxazoline ligands induced *ca.* 1:1 *anti*-*syn* ratios (entries 11–13) with ee of up to 89% for *anti* and 77% for *syn* isomers, respectively (**L4**) (entry 13). Again, a remarkably strong influence of the ligand on the *anti*-*syn* ratio was found: with the *P,P*-ligand dppp (1,3-bis(diphenylphosphanyl)propane), *anti*-*syn* ratio of 82:18 resulted (entry 14), whilst ligand **L5** gave rise to an *anti*-*syn* ratio of 20:80 and ee of 93% (*anti*) and 53% (*syn*) (entry 15). The reversed steric course with ligands **L4** and **L5** is also found for reactions with stabilized enolates.

In conclusion, we have shown that Pd-catalyzed allylic substitutions with glycine ester enolate derivatives **2** and **4** can

yield derivatives of  $\gamma,\delta$ -unsaturated amino acids with moderate to high levels of diastereo- and enantioselectivity. Generally, the steric course of the reactions of these non-stabilized enolates at the allylic carbon corresponds to that found with stabilized enolates. The diastereoselectivity of the substitution is strongly dependent on the ligand employed and can be controlled *via* ligand screening. This is an advantage but presently precludes proposal of a simple predictive model.

This work was supported by the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie.

## Notes and references

- Recent review: B. M. Trost and C. Lee in *Catalytic Asymmetric Synthesis*, 2nd edn., ed. I. Ojima, Wiley-VCH, New York, 2000, pp. 593–649.
- B. M. Trost and X. Ariza, *J. Am. Chem. Soc.*, 1999, **121**, 10727–10737.
- H. M. R. Hoffmann, A. R. Otte, A. Wilde, S. Menzer and D. J. Williams, *Angew. Chem.*, 1995, **107**, 73–76; H. M. R. Hoffmann, A. R. Otte, A. Wilde, S. Menzer and D. J. Williams, *Angew. Chem., Int. Ed. Engl.*, 1995, **34**, 100–103.
- E. Negishi and F. Liu, in *Metal catalyzed Cross-coupling Reactions*, 1st edn., ed. F. Diederich and P. J. Stang, Wiley-VCH, New York, 1998, pp. 1–47.
- (a) U. Kazmaier and F. L. Zumpe, *Angew. Chem.*, 1999, **111**, 1572–1574; U. Kazmaier and F. L. Zumpe, *Angew. Chem., Int. Ed.*, 1999, **38**, 1468–1470; (b) U. Kazmaier and F. L. Zumpe, *Angew. Chem.*, 2000, **112**, 808–811; U. Kazmaier and F. L. Zumpe, *Angew. Chem., Int. Ed.*, 2000, **39**, 802–804; (c) U. Kazmaier and F. L. Zumpe, *Eur. J. Org. Chem.*, 2001, 4067–4076.
- Review: G. Helmchen and A. Pfaltz, *Acc. Chem. Res.*, 2000, **33**, 336–345.
- For a preliminary report of our results see: G. Helmchen, U. Kazmaier, Th. D. Weiß and F. L. Zumpe, Abstract of Poster P 186, 11th European Symposium on Organic Chemistry ESOC 11, Göteborg, Sweden, July 23–28, 1999.
- M. Braun, F. Laichner and T. Meier, *Angew. Chem.*, 2000, **112**, 3637–3640; M. Braun, F. Laichner and T. Meier, *Angew. Chem., Int. Ed.*, 2000, **39**, 3494–3497.
- B. M. Trost and J.-P. Surivet, *J. Am. Chem. Soc.*, 2000, **122**, 6291–6292.
- Review: G. Helmchen, *J. Organomet. Chem.*, 1999, **576**, 203–214.
- Satisfactory analytical and spectroscopic data were obtained for all new compounds.
- Likewise, 1,3-diphenylallyl acetate (**7**) furnishes the same configuration at the allylic center in Pd-BINAP-catalyzed reactions with ketone enolates (*cf.* ref. 8) and malonate enolates: U. Bremberg, M. Larhed, C. Moberg and A. Hallberg, *J. Org. Chem.*, 1999, **64**, 1082–1083.
- U. Cramer, A. G. Rehfeldt and F. Spener, *Biochemistry*, 1980, **19**, 3074–3080.