

### 3-Component palladium–indium mediated diastereoselective cascade allylation of imines with allenes and aryl iodides†

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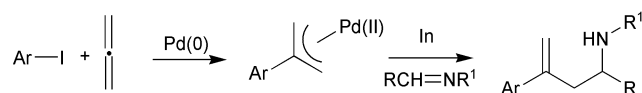
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A new palladium–indium diastereoselective cascade allylation of imines using allenes and aryl iodides is described; *N*-tosyl and *N*-aryl homoallyl amines were obtained in moderate to good yields and excellent diastereoselectivity was observed when enantiomerically pure *N*-*tert*-butanesulfinyl imines were employed in the cascade.

Allylation of carbonyl compounds and imines giving the corresponding homoallylic alcohols or amines is an important synthetic transformation and a number of reagents have been developed for this purpose.<sup>1</sup> Indium has emerged as the metal of choice to mediate the reaction because of its environmentally benign properties allied with a high degree of chemo-, regio- and diastereo-selectivity especially in aqueous media.<sup>2–5</sup> This



Scheme 1

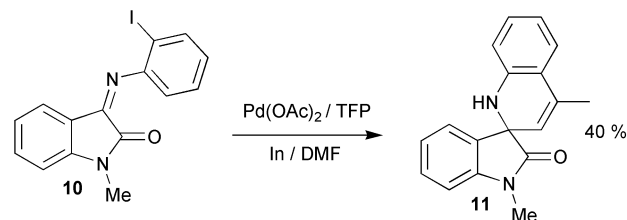
Table 1 Cascade allylation of isatin imines

ArI	Imine	Product	Yield (%)
			61
			51
			50
			53

† Electronic supplementary information (ESI) available: experimental details. See <http://www.rsc.org/suppdata/cc/b2/b202940e/>

process requires no activation of the electrophiles and produces fewer side products than other main group allyl derivatives. Recently we reported the multi-component palladium–indium mediated allylation of aldehydes with allenes<sup>6,7</sup> and identified four synthetic variants of this process depending on whether the Pd and In steps are inter- or intra-molecular.

We now describe reactions of allyl indium species, generated by transmetalation<sup>8–10</sup> of  $\pi$ -allyl palladium(II) complexes arising from aryl iodides and allenes, with imines to afford


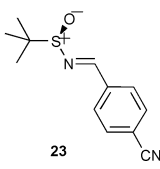
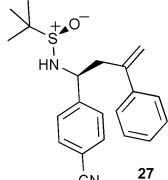

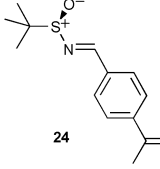
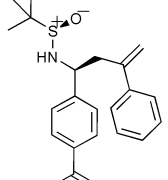

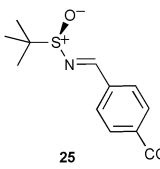
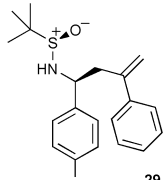

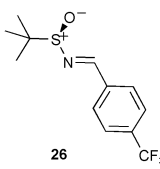
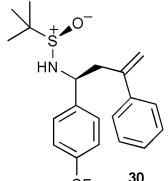

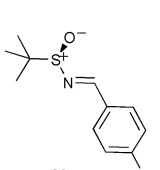
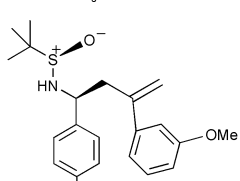


Scheme 2

Table 2 Cascade allylation of aryl aldimines

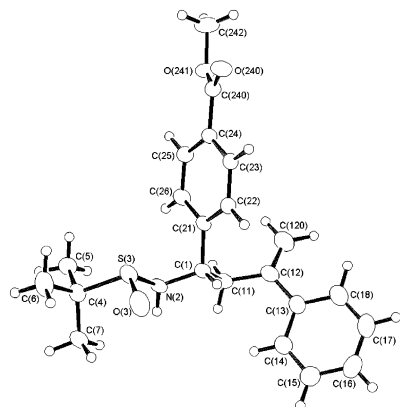
ArI	Imine	Product	Yield (%)
			61
			51
			50
			53
			53

**Table 3** Enantioselective cascade allylation of (*R*)-*N*-*tert*-butanesulfinyl imines

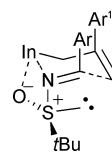
ArI	Imine <sup>a</sup>	Product	Yield (%)
			52
			45
			56 <sup>b</sup>
			56
			54 <sup>c</sup>

<sup>a</sup> All imines are derived from (*R*)-(+)-*N*-*tert*-butanesulfinamide. <sup>b</sup> Based on 83% conversion of imine. <sup>c</sup> Based on 81% conversion of imine.

homoallylic amines (Scheme 1). Initially we explored the class 1 process (intermolecular  $\pi$ -allyl formation, intermolecular allylation) using isatin imines. Thus aryl/heteroaryl iodides **1**



**Fig. 1** X-Ray crystal structure of **29**.



**Fig. 2**

and **2** (1.5 mmol) react (DMF, 80 °C, 24 h, Schlenk tube) with allene (1 bar) and isatin imines **3–5** (1 mmol) in the presence of indium (1.5 mmol), Pd(OAc)<sub>2</sub> (0.1 mmol) and tris(2-furyl)phosphine (0.2 mmol) affording 3,3-disubstituted oxindoles **6–9** in 50–61% yield (Table 1). The class 2 process (intermolecular  $\pi$ -allyl formation, intramolecular allylation)(Scheme 2) proceeds *via* a  $\pi$ -allyl Pd(n) complex, transmetalation by indium and subsequent 6-*endo-trig* cyclisation followed by isomerisation of the double bond affording the spiro-oxindole **11**.

Sulfonimines **14–16** and aryl iodides **1, 12** and **13** were also successfully incorporated into the cascade yielding substituted homoallylic sulfonamides **17–21** in 60–72% yield (Table 2).

Excellent levels of diastereoselectivity are observed when enantiopure (*R*)-*N*-*tert*-butanesulfinyl imines<sup>11</sup> are employed in the cascade (Table 3). Aryl iodides (1.5 mmol), imines (1 mmol) and allene (1 bar) react in the presence of Pd(OAc)<sub>2</sub> (10 mol%), tris(2-furyl)phosphine (20 mol%) and In (1.5 mmol) in DMA at 100 °C for 48–60 h to afford a single diastereoisomer **27–31**, according to <sup>1</sup>H and <sup>13</sup>C NMR studies, in each of the cases explored so far (Table 3). The absolute stereochemistry of one of the products, **29**, has been determined by X-ray crystallography (Fig. 1) to be *S* at the new chiral centre.<sup>12</sup> To rationalise the observed stereochemistry we propose a chelation control model (Fig. 2). Similar transition states have been proposed by others.<sup>13</sup> Yanada *et al* have recently reported a related process using optically active aldimines, palladium(0) and indium(i) iodide.<sup>14</sup>

In conclusion, we have demonstrated a powerful three-component cascade reaction allowing much greater levels of molecular diversity and complexity than classical Barbier-type allylations. Further work is in hand incorporating substituted allenes and other electrophiles into these processes.

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## Notes and references

- Y. Yamamoto and N. Asao, *Chem. Rev.*, 1993, **93**, 2207.
- C. J. Li and T. H. Chan, *Tetrahedron*, 1999, **55**, 11149–11176.
- W. Lu and T. H. Chan, *J. Org. Chem.*, 2001, **66**, 3467–3473.
- H. M. S. Kumar, E. Anjaneyulu, E. J. Reddy and J. S. Yadav, *Tetrahedron Lett.*, 2000, **41**, 9311–9314.
- J. S. Yadav, A. Bandyopadhyay and B. V. S. Reddy, *Tetrahedron Lett.*, 2001, 6385–6388.
- U. Anwar, R. Grigg, M. Rasparini, V. Savic and V. Sridharan, *Chem. Commun.*, 2000, 645–646.
- U. Anwar, R. Grigg and V. Sridharan, *Chem. Commun.*, 2000, 933–944.
- S. Araki, T. Kamei, T. Hirashita, H. Yamamura and M. Kawai, *Org. Lett.*, 2000, **2**, 847–849.
- J. A. Marshall and C. M. Grant, *J. Org. Chem.*, 1999, 8214–8219.
- S. Araki, T. Hirashita, M. Kawai and H. Yamamura, *Chem. Commun.*, 2001, 387–388.
- G. Lui, D. A. Cogan, T. D. Owens, T. P. Tang and J. A. Ellman, *J. Org. Chem.*, 1999, **64**, 1278–1284.
- CCDC 181904. See <http://www.rsc.org/suppdata/cc/b2/b202940e/> for crystallographic files in .cif or other electronic format.
- V. K. Aggarwal, A. M. M. Castro, A. Mereu and H. Adams, *Tetrahedron Lett.*, 2002, 1577–1581.
- R. Yanada, A. Kaieda and Y. Takemoto, *J. Org. Chem.*, 2001, **66**, 7516–7518.