

Surface modification via 'lock and key' specific self-assembly of polyhedral oligomeric silsesquioxane (POSS) derivatives to modified gold surfaces

Eunhee Jeoung, Joseph B. Carroll and Vincent M. Rotello*

Department of Chemistry, The University of Massachusetts, Amherst, MA 01003, USA.

E-mail: rotello@chem.umass.edu; Fax: 413 545 4490; Tel: 413 545 4865

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Diaminopyridine (DAP) functionalized POSS derivatives self-assemble on thymine functionalized monolayers on gold surfaces affording hybrid inorganic/organic surfaces.

Surface modification through non-covalent interactions provides a powerful tool for the control of both the chemical and physical properties of surfaces. A number of techniques have been employed to modify these properties, including Langmuir-Blodgett deposition,¹ electrostatic interactions,² ionic coordination³ and interfacial hydrogen bonding.⁴

The use of designed recognition elements provides a means for enhancing the specificity, and hence the versatility of surface modification procedures. While interfacial binding using engineered recognition elements has been used in sensor systems,⁵ this approach has not been widely applied to surface modification strategies.⁶ To provide a system for exploring recognition-mediated surface modification, we have synthesized polyhedral oligomeric silsesquioxanes (POSSs) functionalized with specific recognition elements. POSS derivatives feature a variety of attributes that make them attractive as surface modification agents.⁷ These three-dimensional, inorganic/organic hybrids are ~1.5 nm in diameter, presenting a large cross-section for efficient surface coverage. POSS derivatives are stable to a variety of conditions, making them useful modifiers for polymeric⁸ and ceramic⁹ systems, as well as applications including nonablatives, atomic-oxygen resistance, and thermal protection. In this communication, we report the recognition-mediated modification of thymine terminated self-assembled monolayers (SAMs) on gold surfaces using complementary diaminopyridine-functionalized POSS derivatives† (POSS-DAP)¹⁰ (Fig. 1).

Thymine-functionalized SAM **1** was prepared by immersing gold films in a 10 mM ethanolic solution of a 1:3 ratio of thymine thiol¹¹/octanethiol for 16 h at room temperature. Films were then washed with a hexane/chloroform (1:1) solution to remove any free thiol. Interaction of surface **1** with POSS-DAP

was explored by immersing the surfaces in a 5 mM solution of diaminopyridine POSS-DAP in hexane/chloroform (1:1) for 10 minutes, and drying *in vacuo*. To confirm that the POSS-DAP interacts with the monolayer through specific hydrogen bonding, control surface **2** featuring the n-methylthymine functionality, and octanethiol surface **3** were prepared and studied in similar fashion.

Specific POSS-DAP deposition on surface **1** was confirmed by X-ray photoelectron spectroscopy (XPS). The characteristic peak corresponding to Si was present after deposition of the POSS-DAP on the Au surface (Fig. 2b). In contrast, XPS of the control surfaces **2** and **3** showed no measurable Si present after dipping in POSS-DAP solution. Further confirmation of effective surface modification was provided by the dramatic decrease of the Au peak after POSS-DAP deposition onto surface **1**, with very little or no change observed in the Au peaks of control surfaces **2** and **3** (Fig. 2a). Ellipsometric measurements at multiple positions of POSS-DAP-modified surface **1** provided a thickness for the POSS layer of $18 \pm 3 \text{ \AA}$, indicating that POSS-DAP was deposited as a monolayer rather than as

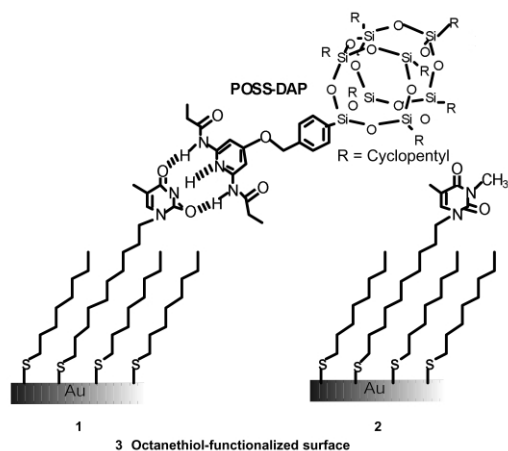


Fig. 1 Schematic illustration of the binding of POSS-DAP to thymine-functionalized SAM **1**, n-methylthymine-functionalized SAM **2** and octanethiol SAM **3**.

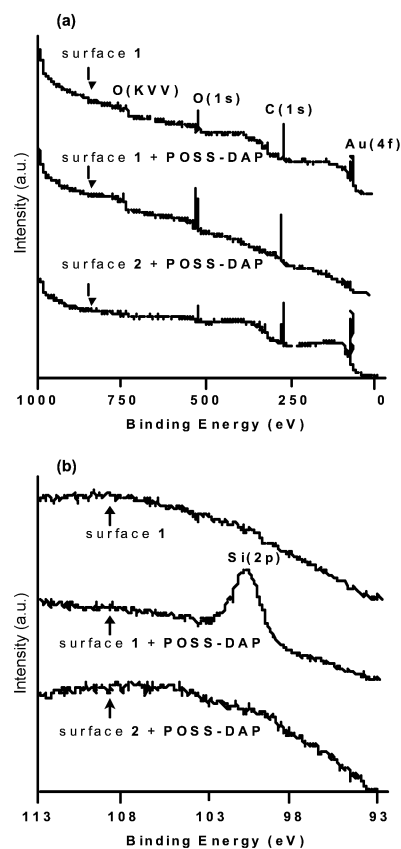


Fig. 2 (a) XPS survey scans of surface **1**, surface **1** + POSS-DAP, and surface **2** + POSS-DAP recorded with take-off angles of 15°. (b) XPS spectra for surface **1**, surface **1** + POSS-DAP, and surface **2** + POSS-DAP showing the Si region.

island aggregates or multilayers. Finally, grazing angle FT-IR measurements of the **POSS-DAP**-modified surface **1** shows an intense peak at 1100 cm^{-1} arising from the asymmetric Si-O-Si stretching vibration as well as additional bands in the $1700\text{--}1300\text{ cm}^{-1}$ region after **POSS-DAP** surface modification (Fig. 3).

Functional demonstration of surface modification was provided using dynamic contact angle measurements with water (Table 1). After **POSS-DAP** deposition, the contact angle of surface **1** dramatically increases, indicating the formation of a much more hydrophobic surface. In contrast, control surfaces **2** and **3** show a negligible change in contact angle after exposure to **POSS-DAP**. The small hysteresis changes after **POSS-DAP** deposition indicate that surface **1** is uniformly modified with POSS functionality.¹²

In summary, we have demonstrated highly specific 'lock and key' functionalization of SAM surfaces through controlled

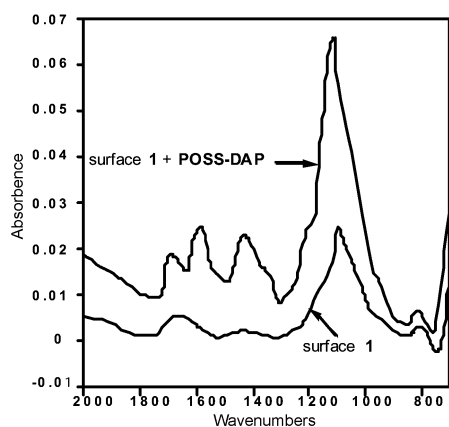


Fig. 3 Grazing angle (86 degree incident) reflection absorption infrared spectra of surface **1** before (bottom) and after (top) deposition of **POSS-DAP**.

Table 1 Dynamic water contact angles for surfaces **1–3** before and after POSS deposition^a

Substrate	Advancing (°)	Receding (°)	Hysteresis (°)
Surface 1	74.5	64.8	9.7
Surface 1 + POSS-DAP	89.8	76.7	13.1
Surface 2	73.0	63.2	9.7
Surface 2 + POSS-DAP	74.6	63.1	11.5
Surface 3	91.5	77.8	13.7
Surface 3 + POSS-DAP	91.3	75.8	15.4

^a Values reported here are an average of at least six measurements, and are $\pm 2^\circ$.

host-guest interactions. Using this strategy, we were able to control both the chemical and physical nature of the surface through deposition of POSS-based surface modification agents. This strategy enhances the versatility of current functionalization strategies by exploiting the diversity of currently available host-guest dyads¹³ to allow orthogonal deposition of surface modifiers. Research in this direction is currently underway, and will be reported in due course.

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Notes and references

† For **POSS-DAP**: δ_{H} (CDCl₃, 200 MHz): 7.80–7.55 (m, 4H), 7.50–7.3 (m, 4H), 2.00–1.3 (m, 46H), 1.18 (t, 6H), 0.94 (brq, 6H). MALDI MS: (M + H)⁺ 1226. FT-IR(neat): 2949, 2864, 2360, 1697, 1584, 1438, 1111, 668 cm^{-1} . Mp > 350 °C.

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