Synthesis of taurospongin A: a potent inhibitor of DNA polymerase and HIV reverse transcriptase, using π -allyltricarbonyliron lactone complexes

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The synthesis of taurospongin A has been achieved using, as a key step, a π -allyltricarbonyliron lactone complex to control a highly stereoselective addition of a methyl group to a carbonyl unit located in the side chain of the complex.

Taurospongin A (1) (Fig. 1) is an unusual acetylene containing natural product consisting of two fatty acid residues and a taurine unit. This compound was isolated from a purple coloured sponge *Hippospongia sp.* in 1997 and shown to be a potent inhibitor of DNA polymerase β and HIV reverse transcriptase.¹

One elegant synthesis of **1** has already been reported in which the three stereogenic centres were created in three separate reagent controlled processes.² Here we report a conceptually different approach whereby a π -allyltricarbonyliron lactone complex³ is used to transfer chiral information, through tethering, to achieve a highly stereoselective addition of a methyl group to a carbonyl unit appended to the side-chain of



Fig. 1 Taurospongin A (1): R = H. Taurospongin A methyl ester (20): R = Me.

the complex. This general process had been established earlier by our group to install 1,5-, 1,7- and 1,5,7-hydroxylated stereogenic centres in alkyl chains.⁴

The synthesis begins from ethyl (3*R*)-hydroxybutanoate **2** by a known sequence⁵ of reactions comprising *tert*-butyldimethylsilyl protection, DIBAL reduction, Swern oxidation, Wittig coupling and then a further reduction to afford the allylic alcohol **3** in good overall yield. Compound **3** was subjected to the Sharpless asymmetric epoxidation procedure,⁶ which gave **4** with 93% d.e. again in good yield (Scheme 1). Following oxidation of **4**, to an intermediate aldehyde, reaction with the diethyl phosphonate derivative **5** gave **6** as the precursor for the iron carbonyl chemistry. The phosphonate **5**, in turn, was prepared from β -propiolactone by ring opening with methoxide followed by protection with *tert*-butyldimethylsilyl chloride and coupling with diethylmethylphosphonate in the usual way.

Next the alkenyl epoxide **6** was reacted with diiron nonacarbonyl in THF⁷ to give the π -allyltricarbonyliron lactone complexes **7** and **8** in 14 and 48% yields respectively. These complexes were readily separated by silica gel chromatography.

The major *endo*-complex **8** was then reacted with trimethylaluminium to give the tertiary alcohol **9** in 81% yield and excellent stereocontrol owing to the preferred s-*cis* conformation of the carbonyl group in the appended side chain.^{4b} Following our previously established reductive decomplexation protocol⁸ using sodium acetoxyborohydride followed by hydrogenation, **9** was smoothly converted to the protected polyol **10**



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Scheme 1 *Reagents and conditions*: (a) TBSCl, imidazole, DMF, 0 °C, 100%; (b) DIBAL-H, PhMe, 0 °C, 83%; (c) (COCl)₂, DMSO, NEt₃, CH₂Cl₂, -78 °C; (d) Ph₃PCHCO₂Me, CH₂Cl₂, rt, 78% for two steps; (e) DIBAL-H, PhMe, 0 °C, 82%; (f) Ti(O*i*-Pr)₄, D-diisopropyl tartrate, *t*-BuOOH, 4 Å MS, CH₂Cl₂, -20 °C, 82%; (g) (COCl)₂, DMSO, NEt₃, CH₂Cl₂, -78 °C, 88%; (h) **5**, NaH, THF, 76%; (i) Fe₂(CO)₉, THF, rt, **7** 14%, **8** 48%; (j) AlMe₃, CH₂Cl₂, 0 °C, 81%; (k) NaBH(OAc)₃, THF, rt; (l) H₂, Pd/C, EtOAc, rt, 81% for two steps; (m) Ac₂O, DMAP, NEt₃, CH₂Cl₂, 0 °C, 98%; (n) TBAF, AcOH, THF, 0 °C, 81%.

in 81% overall yield. Lastly this was acetylated using acetic anhydride and 4-dimethylaminopyridine (DMAP) and deprotected selectively using buffered tetra(*n*-butyl)ammonium fluoride (TBAF) to give the known² taurospongin fragment **11** (Scheme 1). The spectral and optical rotational data were identical to the sample previously synthesised by Jacobsen *et al.*²

For the synthesis of the other fatty acid side chain (Scheme 2), octadec-1-yne **12** was deprotonated with *n*-butyllithium and coupled with 2-(3-bromopropoxy)tetrahydro-2*H*-pyran followed by acidic work-up to give the alcohol **13**. This compound was selectively hydrogenated with the Lindlar catalyst and oxidised with periodinane⁹ to the aldehyde **14**, which was homologated with the Ohira reagent **15**¹⁰ to give the enyne **16** in 84% yield over three steps. This compound was extended further by deprotonation with *n*-butyllithium and alkylation with 2-(3-bromopropoxy)tetrahydro-2*H*-pyran to give a product (82% yield) which, upon direct oxidation with Jones reagent, gave the required acid **17** in 92% yield (46% overall yield from the acetylene **12**).

The final steps of the synthesis were achieved by a modification of the literature route.² Diol **11** was transformed to the corresponding acid *via* oxidation with TEMPO, KBr, NaHCO₃ and NaOCl, and converted to the allyl ester **18** by alkylation with allyl bromide in the presence of Hünigs' base (*i*-Pr₂NEt). Removal of the *tert*-butyldimethylsilyl protecting group from **18** using HF–pyridine was successful and the product was then coupled directly with the acid **17** using 1,3-diisopropylcarbodiimide (DIC), Hünigs' base and DMAP to give **19**.

Lastly, removal of the allyl protecting group using standard procedures with $Pd(PPh_3)_4$ and pyrrolidine gave an intermediate carboxylic acid that was activated by forming *N*-hydroxysuccinic ester with DCC and finally coupled with taurine (H₂NCH₂CH₂SO₃H) to give natural product taurospongin A **1** (Scheme 3).¹ The NMR data for **1** was consistent with the previously synthesised material.^{2†} For full characterization, our synthetic sample was treated with diazomethane to give the corresponding methyl ester **20**,¹ which turned out to be in complete agreement with previously reported data of the samples derived either from isolated or synthetic natural products.^{1,2}

In summary, we report new syntheses of the two fatty acid components that are then used to construct taurospongin A, an



Scheme 2 Reagents and conditions: (a) *n*-BuLi, *n*-C₁₆H₃₃I, HMPA, THF, $-20 \text{ °C} \rightarrow \text{rt}$, then 6 M H₂SO₄, 0 °C, 73%; (b) H₂, Lindlar cat., quinoline; (c) Dess–Martin periodinane, CH₂Cl₂, rt; (d) **15**, K₂CO₃, MeOH, 84% for three steps; (e) *n*-BuLi, 2-(3-bromopropoxy)tetrahydro-2*H*-pyran, HMPA, THF, $-20 \rightarrow 0 \text{ °C}$, 82%; (f) Jones reagent, acetone, 0 °C $\rightarrow \text{rt}$, 92%.



Scheme 3 Reagents and conditions: (a) cat. TEMPO, Aliquat[®] 336, KBr, aq. NaOCl, aq. NaHCO₃, CH₂Cl₂; (b) allyl bromide, *i*-Pr₂NEt, CH₂Cl₂, 87% for two steps; (c) HF·py, THF; (d) **17**, DIC, *i*-Pr₂NEt, DMAP, CH₂Cl₂, 63% for two steps; (e) Pd(PPh₃)₄, pyrrolidine, CH₂Cl₂; (f) *N*-hydrox-ysuccinimide, DCC, 1,4-dioxane; (g) taurine, NEt₃, 1,4-dioxane, H₂O, 97% for three steps; (h) CH₂N₂, ether, 75%.

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Notes and references

† The NMR data for 1: $\delta_{\rm H}$ (600 MHz, CDCl₃) 0.88 (3H, t, J = 7.0 Hz), 1.18 (3H, br s), 1.23–1.33 (33H, m), 1.42–1.60 (4H, m), 1.67–1.68 (1H, m), 1.91–1.95 (1H, m), 1.99–2.04 (2H, m), 2.05 (3H, s), 2.14–2.22 (4H, m), 2.31 (2H, br s), 2.44–2.46 (4H, m), 3.13 (2H, br s), 3.66 (2H, br s), 4.70 (1H, br s), 4.91–4.95 (2H, m), 5.35–5.44 (2H, m), 7.74 (1H, br s); $\delta_{\rm C}$ (150 MHz, CDCl₃) 14.1, 14.8, 19.1, 19.6, 20.0, 21.2, 22.7, 26.1, 26.8, 27.3, 29.3–31.9, 34.3, 34.9, 35.2, 40.4, 42.2, 46.5, 50.0, 68.5, 71.1, 71.8, 78.2, 80.7, 127.8, 131.2, 171.0, 171.6, 173.2. Measured *α*_D of synthetic product 1, [*α*]_D²⁵ –3.5 (*c* 0.37, CHCl₃) has opposite sign to the reported data, [*α*]_D²⁷ +2.4 (*c* 0.2), although the *α*_D of methyl ester 20, [*α*]_D²⁵ –1.4 (*c* 0.49, CHCl₃) is consistent with the reported data, [*α*]_D²⁷ –1.4 (*c* 0.78, CHCl₃, ref. 1) and [*α*]_D –3.00 (*c* 0.53, CHCl₃, ref. 2).

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