Synthesis of ¹³C-labelled, bicyclic mimetics of natural enediynes

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Received (in Cambridge, UK) 13th September 2002, Accepted 9th October 2002 First published as an Advance Article on the web 23rd October 2002

Using a versatile synthesis with $^{13}CH_3PPh_3I$ and $CH_3^{13}CO_2Et$ as ^{13}C sources, the first examples of nine-membered chromophores which have been differentially labelled with ^{13}C in their carbocyclic enediyne cores are described.

With their complex antitumour behaviours (nucleic acid/protein damage, *p*-diradical/*p*-quinone formation) and complex structures, the chemistry and biology of the enediyne-class of antitumour antibiotics has fascinated and challenged researchers for well over a decade now.^{1–3} Yet, despite several mechanistic studies, it can be argued that the reactive diradical intermediate which results from the cycloaromatisation of a natural enediyne has never been proven unambiguously.³ To address this issue, we have completely re-formulated our former synthesis of the core structure (1) of the kedarcidin chromophore (like kedarcidin which exhibits paramagnetic behaviour)^{3b,4} and herein provide carbobicyclic models (2, 3) labelled with ¹³C at either the 3 or 6 position for the precise characterisation of *p*-benzyne and other radical species through EPR measurements (Scheme 1).

In order to incorporate ¹³C into the said strategic positions of the enediyne **1** in an atom-economical fashion, the goal here was to devise a strategy that would not only be flexible to both isotopomers **2** and **3** but also be amenable to the late stage introduction of carbon-13. However, reliable synthetic routes to epoxybicyclo[7.3.0]dodecenediyne frameworks are exceedingly difficult to realise due to product instability.^{2,5} For the structural case at hand, we elected to pursue a new approach through the ynone **4** (Scheme 1). Apart from the practical availability of the iodocyclopentene **5**⁵ ($\alpha/\beta = 5$ at C9), two points should be made: (1) the ynone **4** was anticipated to be more stable than those that had been used previously,^{5,6} and (2) the coupling of **5** with acetylenic fragments like **6** or **7**⁷ was envisaged to give broad scope in carbon-13 labelling studies.



After extensive studies with non-labelled components, a streamlined synthesis of the C3, carbon-13 labelled system 2 was realised as follows. First, the ¹³C-labelled component 6 was prepared from 8⁸ through an efficient, gram-scale synthesis via compounds 9 and 10 (Scheme 2). Here, it should be noted that the Wittig reaction between the erythrulose derivative 8 (2 equiv.) and ¹³CH₃PPh₃I (1 equiv.)⁹ occurred in quantitative yields and that the stereochemistry of the C4 position was superfluous to our planned synthesis of 2, since it would be oxidatively destroyed in the formation of the ynone 4 (*cf.* Scheme 1).

As shown in Scheme 3, using well-established Sonogashiratype conditions on 5 and ¹³C-labelled $6^{2c,5,10}$ the diol 11 was obtained in 72% yield as a four-component diastereomeric mixture at C4 ($\alpha/\beta = 1:1.3$) and C9 ($\alpha/\beta = 5$ at C9. Oxidative cleavage of 11 with NaIO₄ then cleanly furnished the desired ynone 4 ($\alpha/\beta = 5$ at C9) which proved to be stable to both silicagel chromatography and storage. Treatment of 4 with propargyl zinc bromide and TBS ether protection (under anionic conditions)¹¹ permitted the nitrile 12 to be isolated as its major diastereomer (4 α , 9 α) in 46% overall yield.[†] DIBAL reduction of the nitrile 12 afforded the cyclisation precursor 13⁺ in excellent yield, which was immediately subjected to the typical CeCl₃-mediated acetylide cyclisation protocol at -25 °C to generate the *trans*-diol 14 in 45–52% isolated yield.^{2c,4,5} Mesylation of 14 followed by global desilylation then afforded the epoxide 15 in 76% yield over two steps. Finally, 15 was silvlated selectively at the C11-allylic alcohol and then, in a one-pot reaction, mesylated at 5 °C and treated with DBU at -40 °C to produce the fully-fledged, ¹³C-labelled epoxydienediyne 2 in high purities and in yields of about 45-50%.‡

As indicated in our retrosynthetic analysis (Scheme 1), the incorporation of carbon-13 into the C6-position would necessitate a simple modification of the aforementioned route. Instead of using propargylic zinc, carbon-13 incorporation was thus achieved by the anionic-addition of labeled ethyl acetate (CH₃¹³CO₂Et) onto the enantiopure ynone **4**, that was derived from the readily available synthon **7**⁷ (Scheme 3). After *O*-TBS silylation, **17** was subsequently obtained as a 4:1 α/β mixture at C4 in a 66% yield over two steps. Following the chemoselective reduction of the ester functionality in the presence of the nitrile group and re-oxidation, the dibromoalkene **18** was generated



Scheme 2 Reagents and conditions: (a) ${}^{13}CH_3PPh_3I$, NaN(TMS)₂, THF, $-78 \,^{\circ}C$ to rt, 1 h, quantitative; (b) OsO₄ (0.01 eq.), NMO (2.5 eq.), acetone–H₂O (4:1), 12 h, 86%; (c) SO₃·Py (3 eq.), Et₃N (15 eq.), DMSO, 10 min, rt; (d) TBSOTf (3 eq.), 2,6-lutidine (4 eq.), CH₂Cl₂, 1 h, rt; (e) CBr₄ (3 eq.), Ph₃P (6 eq.), CH₂Cl₂, 1 h rt; (f) *n*-BuLi (2.2 eq.), THF, $-78 \,^{\circ}C$, 1 h; (g) TBAF (3 eq.), THF, 1 h, rt, 63% for five steps.



Scheme 3 *Reagents and conditions*: (a) 5 (1 eq.), Pd₂(dba)₃·CHCl₃ (0.05 eq.), CuI (0.1 eq.), DMF–*i*-Pr₂NEt (2:1), rt, 1 h, 72% for **11**, 40% for **16** (after crystallisation); (b) NaIO₄ (5 eq.), CH₃OH–H₂O (4:1), rt, 1 h, 84%; (c) Zn (3 eq.), propargyl bromide (3 eq.), THF, -15 °C, 1 h; (d) TBSCl (1.2 eq.), KH (3 eq.), 18-crown-6 (0.05 eq.), THF, 0 °C, 5 min, 46% for two steps; (e) DIBAL (1.5 eq.), CH₂Cl₂, -78 °C, 1 h; (f) CeCl₃ (25 eq.), LiN(TMS)₂ (26 eq.), THF, -25 °C to rt, 1 h, 45–52% for **14**, or 68–72% for **21**, for two steps; (g) MsCl (3 eq.), Et₃N (6 eq.), CH₂Cl₂, rt, 1 h; (h) TBAF (5 eq.), THF, rt, 10 min, 70–76% for **15** or **22**, for two steps; (i) TBSOTf (3 eq.), 2,6-lutidine (6 eq.), CH₂Cl₂, -78 °C, 10 min; (j) MsCl (5 eq.), Et₃N (10 eq.), CH₂Cl₂, 5 °C, 30 min, then DBU (2 eq.), CD₂Cl₂, -40 °C, then rt, 30 min, 45–50% for **2** or **3**, for two steps; (k) CH₃¹³CO₂Et, LiN(SiMe₃)₂ (1.1 eq.), THF, -78 °C, 1 h, 78%; (l) TBSOTf (3 eq.), CH₂Cl₂, rt, 10 h, 85%; (m) DIBAL (4 eq.), THF, -78 to -25 °C, 2 h, 70%; (n) Dess–Martin periodinane (1.5 eq.), CH₂Cl₂, 1 h, 80%; (o) CBr₄ (3 eq.), Ph₃P (6 eq.), CH₂Cl₂, 0 °C, 30 min, then SiO₂ separation; (p) *n*-BuLi (2 eq.), THF, -78 °C, 58% for two steps.

and isolated in enantiopure form after silica-gel chromotography. Completion of this oxidation-homologation sequence then gave the desired C6–carbon-13 labelled alkyne **19** in 22% yield over six steps from **4**. Lastly, under a parallel reaction sequence to that described for **2**, the targeted epoxyenediyne **3** was generated from **19** (*via* the C6-labelled intermediates **20–22**) in a 22% overall yield over six steps.[‡]

In closing, we should first point out that for all isotopomers 2 and 3 the nine-membered cyclization steps to 14 or 21 have been reliably performed in yields that reflect the diastereomeric purity of the cyclization precursors 13 or 20, *i.e.* 45 to 72%. Second, the routes described herein are highly expeditious and practical, particulary considering the high lability of the compounds involved. For example, the unlabelled enediyne 1⁴ can now be prepared in a 10–15% overall yield from enantiopure 7¹¹ over the shortest 10-step sequence. Comprehensive EPR characterisation studies on the radical cycloaromatised states of 1–3 are currently being finalised and will be detailed elsewhere.

A postdoctoral CREST fellowship (to P. D.) from the Japan Science and Technology Corporation (JST) and a fellowship (to T. M.) from the Japanese Society for the Promotion of Science for Young Japanese Scientists are gratefully acknowledged.

Notes and references

†12 and 13 were isolated and used as $(4\alpha, 9\alpha)$: $(4\beta, 9\beta) \ge 10:1$ mixtures.

 \ddagger *NMR data* for **2** and **3** (500 MHz, CD₂Cl₂, ref. 5.32 ppm for ¹H and 53.10 ppm for ¹³C).

2: ¹H NMR: δ 0.11 (s, 3H), 0.11 (s, 3H), 0.90 (s, 9H), 1.37 (s, 3H), 1.46 (s, 3H), 1.86 (dd, 1H, *J* 15.0, 2.5 Hz), 2.71 (dd, 1H, *J* 15.0, 7.3 Hz), 3.75 (d, 1H, *J* 1.5 Hz), 3.80 (dd, 1H, *J* 8.3, 6.7 Hz), 4.17 (dd, 1H, *J* 8.3, 6.7 Hz), 4.63 (ddt, *J* 6.7, 1.5 Hz, ³*J*_{C-H} 5.0 Hz), 5.00 (ddd, *J* 7.3, 2.5 Hz, ⁵*J*_{C-H} 1.5 Hz), 5.99 (dt, *J* 1.5 Hz, ³*J*_{C-H} 12.5 Hz), 6.40 (br d, *J* 2.0 Hz).

 $^{13}\mathrm{C}$ NMR: δ –5.41, –5.34, 17.65, 25.10, 25.17, 25.74, 38.60, 49.81, 68.82 (d, $J_{\mathrm{C-C}}$ 1.4 Hz), 69.77 (d, $J_{\mathrm{C-C}}$ 2.8 Hz), 72.07, 75.07 (d, $J_{\mathrm{C-C}}$ 3.9 Hz), 88.38 (d, $J_{\mathrm{C-C}}$ 2.9 Hz), 97.18 ($^{13}\mathrm{C}$ label), 102.42 (d, $J_{\mathrm{C-C}}$ 182.6 Hz), 104.52, 110.02, 119.70 (d, $J_{\mathrm{C-C}}$ 1.9 Hz), 124.63 (d, $J_{\mathrm{C-C}}$ 11.4 Hz), 143.00 (d, $J_{\mathrm{C-C}}$ 89.6 Hz), 145.47 (d, $J_{\mathrm{C-C}}$ 4.3 Hz).

3: ¹H NMR: δ 0.11 (s, 3H), 0.11 (s, 3H), 0.90 (s, 9H), 1.37 (s, 3H), 1.46 (s, 3H), 1,86 (dd, 1H, *J* 15.0, 2.5 Hz), 2.72 (dd, 1H, *J* 15.0, 7,3 Hz), 3.75 (dd, 1H, *J* 1.5 Hz, ²*J*_{C-H} 3.7 Hz), 3.80 (dd, 1H, *J* 8.5, 6.7 Hz), 4.17 (dd, 1H, *J* 8.5, 6.5 Hz), 4.63 (bt, 1H, *J* 6.5 Hz), 5.00 (dt, *J* 7.3, 2.5 Hz), 5.99 (br d, *J* 1.5 Hz), 6.40 (d, *J* 2.5 Hz).

 $^{13}\mathrm{C}$ NMR: δ –5.41, –5.34, 17.65, 25.09, 25.17, 25.74, 38.60, 49.80 (d, $J_{\mathrm{C-C}}$ 14.4 Hz), 68.82, 69.76 (d, $J_{\mathrm{C-C}}$ 2.8 Hz), 72.07, 75.07 (d, $J_{\mathrm{C-C}}$ 6.1 Hz), 88.38 ($^{13}\mathrm{C}$ label), 97.17 (d, $J_{\mathrm{C-C}}$ 6.3 Hz), 102.36, 104.55 (d, $J_{\mathrm{C-C}}$ 193.9 Hz), 110.02, 118.95 (d, $J_{\mathrm{C-C}}$ 95.8 Hz), 124.62, 143.01 (d, $J_{\mathrm{C-C}}$ 3.4 Hz), 145.46.

- Recent reviews: (a) J. W. Grissom, G. U. Gunawardena, D. Klingberg and D. Huang, *Tetrahedron*, 1996, **52**, 6453; (b) Z. Xi and I. H. Goldberg, *Compr. Nat. Prod. Chem.*, 1999, **7**, 553: enediyne-induced cleavage of RNA (c) J.-M. A. Battigello, M. Cui, S. Roshong and B. J. Carter, *Bioorg. Med. Chem.*, 1995, **6**, 839; inhibition of DNA-repair enzymes (d) W. Solomon, K. L. Colson and D. R. Schroeder, *Biochemistry*, 1995, **34**, 11591; M. Hashimoto, M. M. Greenberg, Y. W. Kow, J.-T. Hwang and R. P. Cunningham, *J. Am. Chem. Soc.*, 2001, **123**, 3161; cytotoxic action of an enediyne-derived quinone species (e) L. H. Jones, C. W. Harwig, P. Wentworth Jr., A. Simeonov, A. D. Wentworth, S. Py, J. A. Ashley, R. A. Lerner and K. D. Janda, *J. Am. Chem. Soc.*, 2001, **123**, 3607.
- 2 Latest advances in the synthesis of fully-fledged, nine-membered enediyne chromophores: (a) A. G. Myers, R. Glatthar, M. Hammond, P. M. Harrington, E. Y. Kuo, J. Liang, S. E. Schaus, Y. Wu and J.-N. Xiang, J. Am. Chem. Soc., 2002, **124**, 5380; (b) A. G. Myers, P. C. Hogan, A. R. Hurd and S. D. Goldberg, Angew. Chem., Int. Ed., 2002, **41**, 1062; (c) S. Kobayashi, S. Ashizawa, Y. Takahashi, Y. Sugiura, M. Nagaoka, M. J. Lear and M. Hirama, J. Am. Chem. Soc., 2001, **123**, 11294.
- 3 Recent study of a specialised *p*-benzyne diradical: (*a*) H. H. Wenk A. Baster, W. Sander, D. A. Hrovat and W. T. Borden, *Angew. Chem., Int. Ed.*, 2001, **40**, 2295; for direct evidence that C-1027 and kedarcidin form paramagnetic radical species, see (*b*) M. Hirama, K. Akiyama, T. Tanaka, T. Noda, K.-I. Iida, I. Sato, R. Hanaishi, S. Fukuda-Ishisaka, M. Ishiguro, T. Otani and J. E. Leet, *J. Am. Chem. Soc.*, 2000, **122**, 720.
- 4 K.-I. Iida and M. Hirama, J. Am. Chem. Soc., 1995, 117, 8875.
- 5 For routes to partly-completed, nine-membered enediynes, see: K. Toyama, S. Iguchi, H. Sakazaki, T. Oishi and M. Hirama, *Bull. Chem. Soc. Jpn.*, 2001, **74**, 997 and references therein.
- 6 G. X. Wang, S. Iguchi and M. Hirama, J. Org. Chem., 2001, 66, 2146.
- 7 S. Kobayashi, P. Das, X. Wang, T. Mita, M. J. Lear and M. Hirama, *Chem. Lett.*, 2002, 300.
- 8 K. Nakatani, K. Arai, N. Hirayama, F. Matsuda and S. Terashima, *Tetrahedron*, 1992, **48**, 633.
- 9 C. D. Poulter, M. Muehlbacher and D. R. Davis, J. Am. Chem. Soc., 1989, 111, 3740.
- 10 K. Sonogashira, Y. Tohda and N. Hagihara, *Tetrahedron Lett.*, 1975, **50**, 4467.
- 11 T. F. Braish and P. L. Fuchs, Synth. Commun., 1986, 16, 111.