

# Self-assembly of a lacunary $\alpha$ -Keggin undecatungstophosphate into a three-dimensional network linked by *s*-block cations†‡

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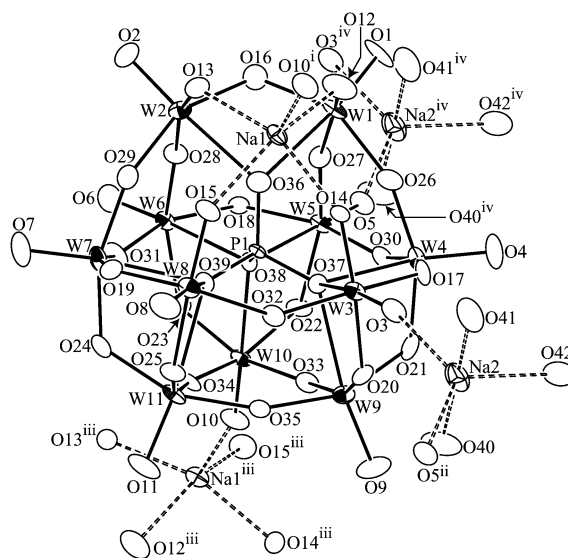
The crystal structure of  $[(\text{CH}_3)_4\text{N}]_4\text{Na}_2\text{H}[\alpha\text{-PW}_{11}\text{O}_{39}]\cdot 8\text{H}_2\text{O}$  has been determined by using the Weissenberg camera at the BL04B2 beamline of SPring-8, revealing that it contains a disorder-free lacunary  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anion with one sodium cation embedded into its lacuna that links the oxometalate building blocks into a one-dimensional chain, which is then woven into a three-dimensional latticework by another  $\text{Na}^+$ .

Recent interest in synthetic chemistry of polyoxometalates (POMs) largely focuses on rational syntheses of larger complexes, having either extended<sup>2–9</sup> or discrete<sup>2,10–15</sup> structures. Most of their syntheses utilize the strategy of self-assembly of building blocks. As the linker atoms to bridge building blocks, *f*-block elements have been widely used to connect complete<sup>2,3,10</sup> or lacunary<sup>2,4,11</sup> POMs. Other linkers hitherto reported include *d*-block<sup>2,5–7,12–14</sup> and, less frequently, *p*-block<sup>15</sup> elements. Assembly of organic derivatives of POMs via their organic moieties has also been reported.<sup>8</sup> However, *s*-block elements in POM crystals have been paid much less attention as linker atoms assisting the self-assembly of POM building blocks. Exceptions include, for example,  $[\text{Et}_3\text{NH}][_3\text{NaMo}_8\text{O}_{26}]$ , where a  $\text{Na}^+$  cation links complete  $\beta$ -octamolybdate anions into a zigzag chain.<sup>9</sup> Here we report the first extended POM structure wherein sodium, an *s*-block element, links lacunary Keggin anions into infinite linear chain, which exemplifies a new mode of self-assembly of POMs.

During an attempt to introduce Ti atoms into lacunary tungstophosphates, crystals of  $[(\text{CH}_3)_4\text{N}]_4\text{Na}_2\text{H}[\alpha\text{-PW}_{11}\text{O}_{39}]\cdot 8\text{H}_2\text{O}$  (**1**) were obtained.† It is a polymorphic modification of the compound reported by Fuchs and co-workers, in which the anion was fully disordered.<sup>16</sup> Since efforts to grow larger crystals had been unsuccessful, a single crystal X-ray diffraction experiment was carried out at a synchrotron facility. A colorless crystal measuring  $0.070 \times 0.015 \times 0.010$  mm was mounted on the imaging plate Weissenberg camera installed in the BL04B2 beamline of SPring-8.<sup>1</sup> 35 diffraction images were obtained in the oscillation mode with high-energy X-rays ( $\lambda = 0.3282$  Å). The procedure for the data reduction was the same as that described in ref. 1. The structure was solved by direct methods and refined by the use of *SHELX97*.<sup>17</sup>¶

An asymmetric unit of **1** includes an  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anion with the lacunary  $\alpha$ -Keggin structure, two  $\text{Na}^+$  cations, four tetramethylammonium cations, eight water molecules of crystallization, plus one proton that could not be located by this structure analysis. As depicted by Fig. 1, one of the two independent  $\text{Na}^+$  cations, Na1, is embedded into the lacuna of the  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anion with four shorter Na–O contacts (to O12–O15, 2.210–2.312 Å) plus one longer contact (to O36, 2.711 Å). Na1 is also coordinated by a terminal O atom, O10, of

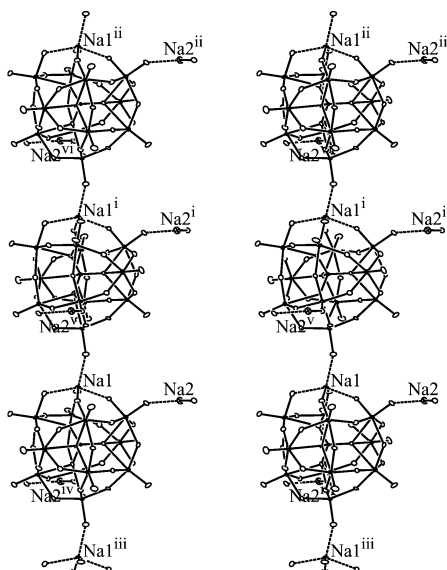
an adjacent  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anion at 2.269 Å, to complete a square pyramidal (disregarding the Na1...O36 contact) or distorted octahedral (Na1...O36 contact taken into account) coordination. Through the linkage at Na1, the lacunary  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anions form a one-dimensional chain structure running along the crystallographic *a* + *c* direction as illustrated in Fig. 2. The lacunary anions in this chain are related by a glide plane, just like those in  $[\text{NEt}_3\text{H}]_5[\text{XCo}^{\text{II}}\text{W}_{11}\text{O}_{39}]\cdot 3\text{H}_2\text{O}$  (*X* = P or As).<sup>5</sup> Similar chain structures,  $(\text{Et})_{8n}[\text{PMnW}_{11}\text{O}_{39}]_n\cdot 2n\text{H}_2\text{O}$ <sup>6</sup> and  $[\text{Co}(\text{dpa})_2(\text{OH}_2)_2]_2[\text{Hdpa}][\text{PCoW}_{11}\text{O}_{39}]$ ,<sup>7</sup> have also been reported. As this linkage aligns the  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  building blocks almost parallel with respect to their dipoles, the resultant linear chain is highly polarized. And all these polarized chains run parallel to each other, giving a highly polarized nature to the whole crystal. Fig. 1 also illustrates that another  $\text{Na}^+$  cation, Na2, is coordinated by two terminal O atoms of two  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anions related by a lattice translation,  $(a \pm b)/2$ . With three O atoms of water molecules of crystallization, Na2 completes a trigonal bipyramidal coordination with the POM moieties as equatorial ligands. A similar linkage where



**Fig. 1** An ORTEP<sup>18</sup> drawing of the  $[\text{PW}_{11}\text{O}_{39}]^{7-}$  anion and its associated  $\text{Na}^+$  cations. Displacement ellipsoids are scaled to enclose 50% probability levels. Selected distances (Å): Na1–O12–W1–O27–W5–O22–W10–O34–W11–O25–W8–O15–Na1: 2.312, 1.730, 2.081, 1.788, 1.941, 1.906, 1.885, 1.924, 1.825, 2.056, 1.738, 2.292; Na1–O13–W2–O28–W6–O23–W10–O33–W9–O20–W3–O14–Na1: 2.305, 1.761, 2.067, 1.812, 1.977, 1.889, 1.890, 1.934, 1.863, 2.052, 1.732, 2.210; W1–O16–W2–O29–W7–O24–W11–O35–W9–O21–W4–O26–W1: 1.893, 1.939, 1.904, 1.897, 1.912, 1.925, 1.889, 1.934, 1.873, 1.935, 1.829, 1.965; W3–O17–W4–O30–W5–O18–W6–O31–W7–O19–W8–O32–W3: 1.986, 1.870, 1.901, 1.915, 1.931, 1.910, 1.929, 1.885, 1.899, 1.952, 1.940, 1.873; Na1–O10<sup>i</sup>: 2.269, Na1–O36: 2.711, Na2–O3: 2.342, Na2–O5<sup>ii</sup>: 2.395, Na2–O41: 2.413, Na2–O40: 2.38, Na2–O42: 2.28 Å. Standard uncertainties associated with these distances are 0.010–0.016 Å for W–O and 0.013–0.03 Å for Na–O. Symmetry codes: (i)  $x+1/2, -y+1/2, z+1/2$ ; (ii)  $x+1/2, y-1/2, z$ ; (iii)  $x-1/2, -y+1/2, z-1/2$ ; (iv)  $x-1/2, y+1/2, z$ .

† High-Energy X-ray Structure Determinations Using Synchrotron Radiation. Part 3. Part 2: reference 1.

‡ Electronic supplementary information (ESI) available: ORTEP plot of the asymmetric unit of **1** and rotatable 3-D crystal structure diagrams in CHIME format. See <http://www.rsc.org/suppdata/cc/b2/b209076g/>



**Fig. 2** A stereoscopic diagram of the infinite  $(\text{Na}^+\cdots[\text{PW}_{11}\text{O}_{39}]^{7-})_{\infty}$  chain structure running along the  $a + c$  axis, viewed along the  $b$  axis. Symmetry codes: (i)  $x+1/2, -y+1/2, z+1/2$ ; (ii)  $x+1, y, z+1$ ; (iii)  $x-1/2, -y+1/2, z-1/2$ ; (iv)  $x-1/2, y+2/1, z$ ; (v)  $x, -y, z+1/2$ ; (vi)  $x+1/2, y+1/2, z+1$ .

octahedrally coordinated  $\text{Na}^+$  links POMs has been reported.<sup>12</sup> Through the connection at Na2, the  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anions are linked into secondary one-dimensional chain structure running alternately along the  $a + b$  and  $a - b$  directions. The chains running along  $a + b$  are arranged parallel to each other into a layer parallel to the 001 plane without any interactions with each other. A layer above it accommodates chains running along  $a - b$ , which are perpendicular to the chains in the layer below. Thus the interaction at Na2 links the  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anions into alternating perpendicular arrays of parallel chains like the logs in a log cabin (see Fig. S1†). The interaction at Na1 are running through the eclipsed points in this arrangement.

These cation linkages are pinning down the orientation of the  $[\alpha\text{-PW}_{11}\text{O}_{39}]^{7-}$  anion, which had been disordered in other salts with  $s$ -block cations,<sup>16,19</sup> and thus the effect of the incorporation of the  $s$ -block element into the lacuna of the anion is clearly observed. The  $\text{W}-\text{O}(\text{-Na1})$  distances (1.730 to 1.761 Å) are shorter than the other  $\text{W}-\mu_2\text{-O}$  distances, to compensate for the smaller  $\text{Na1}-\text{O}$  bond valences. Their *trans* influences propagate to the W atoms four bonds away from Na1 in the  $\text{NaW}_5\text{O}_6$  closed loops<sup>20</sup> ( $\text{Na1}-\text{O12}-\text{W1}-\text{O27}-\text{W5}-\text{O22}-\text{W10}-\text{O34}-\text{W11}-\text{O25}-\text{W8}-\text{O15}-\text{Na1}$  and  $\text{Na1}-\text{O13}-\text{W2}-\text{O28}-\text{W6}-\text{O23}-\text{W10}-\text{O33}-\text{W9}-\text{O20}-\text{W3}-\text{O14}-\text{Na1}$ ), exhibiting the alternating long and short  $\text{W}-\text{O}$  distances. Average distances for the  $\text{Na}-\text{O}-\text{W}-\text{O}-\text{W}-\text{O}-\text{W}$  sequence are 2.28(5), 1.74(1), 2.06(1), 1.82(3), 1.94(2) and 1.89(1) Å, respectively. The average  $\text{W}-\text{O}$  distances in the other two  $\text{W}_6\text{O}_6$  closed loops is 1.91(3) Å. Although  $\text{Na}^+$  cations in lacunae of a lacunary POM have already been structurally characterized,<sup>13</sup> the present study is the first to reveal that the structural distortion caused by the incorporation of  $\text{Na}^+$  propagates over the whole POM framework. These structural features may provide the closest approximation to isolated (if there are any) lacunary Keggin  $[\text{XM}_{11}\text{O}_{39}]^{n-}$  anions, since other disorder-free structures incorporate  $f$ -block<sup>4,11</sup> or  $d$ -block<sup>7</sup> atoms that would more tightly bind to the lacunae than  $\text{Na}^+$ .

A common feature observed in **1** and  $[\text{Et}_3\text{NH}]_3[\text{NaMo}_8\text{O}_{26}]$ ,<sup>9</sup> where linkages of POM building blocks by  $s$ -block elements are observed, is that the crystals are double salts of  $s$ -block cations and organic cations. As the charge of the  $s$ -block cations is only +1, simple salts of highly charged POMs with solely the  $s$ -block cations would require a considerable number of cations, which would then lead to complicated intermolecular interactions.

Introduction of organic cations reduces the number of  $s$ -block cations and its relevant intermolecular interactions, leading to, at least in these examples, well organized and easily distinguishable extended three-dimensional structure. This double salt strategy might be applied for designing various kinds of ionic solids, especially those with highly charged ions.

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## Notes and references

§ The crystal used for the X-ray diffraction experiment was obtained as follows: 5.00 g of  $\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$  and 0.90 g of  $\text{NaH}_2\text{PO}_4$  were dissolved in 20 mL of  $\text{H}_2\text{O}$ . After adjusting its pH to 6.0 with hydrochloric acid, 0.15 mL of  $\text{TiCl}_4$  was added to the filtrate. The mixture was refluxed for 1 h. After cooling the solution to room temperature, the solution was filtered and 1.00 g of  $(\text{CH}_3)_4\text{NCl}$  was added. After keeping the solution at room temperature, colorless thin-plate crystals were obtained. Found: C, 6.30; H, 2.25; N, 2.05%. Calcd for  $\text{C}_{16}\text{H}_{65}\text{N}_4\text{O}_{47}\text{Na}_2\text{PW}_{11}$ : C, 6.07; H, 2.07; N, 1.77%.  $v_{\text{max}}/\text{cm}^{-1}$  1078(m), 1042(m), 947(vs), 892(sh), 858(s), 803(vs), 762(vs), 515(w). This compound was confirmed to be also obtainable by a similar procedure without  $\text{TiCl}_4$ .

¶ Crystal data for **1**:  $\text{C}_{16}\text{H}_{65}\text{N}_4\text{O}_{47}\text{Na}_2\text{PW}_{11}$ ,  $M = 3165.0$ , monoclinic,  $a = 18.494(1)$ ,  $b = 18.102(1)$ ,  $c = 19.214(1)$  Å,  $\beta = 106.206(3)^\circ$ ,  $V = 6176.8(6)$  Å<sup>3</sup>,  $T = 153$  K, space group  $Cc$  (no. 9),  $Z = 4$ ,  $\mu(0.3282 \text{ Å}) = 2.73 \text{ mm}^{-1}$ ,  $D_x = 3.403 \text{ Mg m}^{-3}$ , 25381 reflections measured, 14592 unique ( $R_{\text{int}} = 0.0391$ ) which were used in all calculations. The final  $wR(F^2) = 0.1385$  for all data and  $R = 0.0543$  for 13509 reflections with  $F^2 > 2\sigma(F^2)$ . CCDC reference number 194132. See <http://www.rsc.org/suppdata/cc/b2/b209076g/> for crystallographic data in CIF or other electronic format.

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