Zipping up 'the crushed fullerene' $C_{60}H_{30}$: C_{60} by fifteen-fold, consecutive intramolecular H_2 losses[†]

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MALDI TOF-MS of tribenzo[l:l':l'']benzo[1,2-e:3,4-e':5,6-e'']triacephenanthrylene (1a, C₆₀H₃₀) gives C₆₀⁺ by multiple intramolecular cyclodehydrogenation reactions.

The proposal that laser evaporation of graphite gives the fullerene C_{60} ,¹ followed by its isolation from arc-evaporated graphite soot² in sufficient quantities has had considerable impact. For example, the chemical modification of C_{60} has become a subject of intense research.³ However, despite many efforts,⁴ rational syntheses of any of the hitherto isolated fullerenes from unimolecular precursors still remain elusive. Only recently mass spectral (FT-ICR LD⁵ and LD-TOF MS⁶) evidence has been reported showing that C_{60} '+/ C_{60} '- are accessible from the transient $C_{60}H_6$ cyclophane-like polyyne generated *in situ* in the mass spectrometer.

In a different approach inspired by the observation that in suitable flames C_{60} and C_{70} are formed besides the common (non)-alternant polycyclic aromatic hydrocarbon (PAH) effluents,⁷ tribenzo[*1*:*1'*:*1''*]benzo[*1*,2-*e*:3,4-*e'*:5,6-*e''*]triacephenanthrylene [**1a**, $C_{60}H_{30}$, C_3 ; ΔH_{f}^0 (AM1) 344.6 kcal mol⁻¹, Chart 1], was put forward as an unimolecular C_{60} precursor.^{8–10}



Chart 1

Notice that **1a** has the exact carbon atom topology of the C₆₀ Schlegel diagram (Fig. 1). Unfortunately, the initial synthesis of **1a** by S₈-mediated cyclotrimerization of 4,5-dihydro[*I*]benzacephenanthrylene [**2**, C₁, Chart 1]^{10,11}† lacked proper *regio*chemical control.^{8,12} Scrutiny of the cyclotrimerization mechanism using AM1¹³ results strongly indicated that instead of **1a**, its unsymmetrical isomer tribenzo[*I*:*I*':*I*"]benzo[1,2-*e*:3,4*e*':6,5-*e*"]triacephenanthrylene [**1b**, C₁; ΔH_f^0 (AM1) 345.3 kcal

† Electronic supplementary information (ESI) available: AM1 optimised geometries, heats of formation, first ionisation potentials, Schlegel diagrams and MALDI TOF-MS spectrum of 1a⁺⁺. See http://www.rsc.org/ suppdata/cc/b1/b110587f/ mol⁻¹] will be the major $C_{60}H_{30}$ product (statistical *ratio* **1a** : **1b** 1 : 3).⁸ Whilst **1a** can be converted into C_{60} via fifteen-fold, consecutive intramolecular H_2 losses accompanied by ring closures, **1b** will yield either a curved $C_{60}H_{12}$ or $C_{60}H_{10}$ PAH under identical conditions, viz. in the absence of rearrangements of its carbon skeleton. MALDI TOF-MS (positive-ion mode)‡ of the S₈-cyclotrimerization reaction mixture containing **1a**/**1b** corroborated these contentions. At a high laser fluence of *ca*. 440 µJ pulse⁻¹ consecutive intramolecular H_2 losses from m/z 750 ($C_{60}H_{30}$ ⁻⁺) down to m/z 732/730 ($C_{60}H_{12}$ ⁻⁺/ $C_{60}H_{10}$ ⁻⁺) were clearly discernible. However, no evidence for C_{60} ⁻⁺ (m/z 720) formation was found.⁸⁺

Here we report that fifteen-fold, consecutive intramolecular cyclodehydrogenations occur if pure **1a**, prepared *via* triple Pd-catalysed arylation of *syn*-5,10,15-tris[2-(1-bromo)naphthylmethyl]truxene,^{9,16} is subjected to similar MALDI TOF-MS conditions.

At a laser fluence of *ca*. 12 μ J pulse⁻¹ the precursor ion **1a**⁺⁺ (C₆₀H₃₀⁺⁺, *m/z* 750) undergoes three-fold H₂ losses giving C₆₀H₂₈⁺⁺, C₆₀H₂₆⁺⁺ and C₆₀H₂₄⁺⁺ [*m/z* 748 (11%), 746 (3%) and 744 (2%)].‡§ No skeletal fragmentation of **1a** occurs. At a laser fluence of *ca*. 440 μ J per pulse, however, cyclodehydrogenation products spanning the mass range *m/z* 750 (C₆₀H₃₀⁺⁺) down to *m/z* 720 (C₆₀⁺⁺) are unequivocally identified [Fig. 2; *m/z* 748 (85%), 746 (72%), 744 (43%), 742 (23%), 740 (10%), 738 (6%), 736 (5%), 734 (4%), 732 (3%), 730 (2%), 728 (1%), 726 (2%), 724 (2%), 722 (6%) and 720 (5%)].‡§The stepwise character of the multiple cyclodehydrogenations is substantiated by the following additional observations:

i) At low laser fluences (*ca.* 12 μ J pulse⁻¹) decacyclene [**3**, C₃₆H₁₈ (*D*₃)^{8,14,15,†}] and benzo[1,2-*e*:3,4-*e*':6,5-*e''*]triacephenanthrylene [**4**, C₄₈H₂₄ (*C*₃)^{16†}] like **1a** undergo stepwise threefold intramolecular overall H₂ losses from C₃₆H₁₈⁺⁺ and C₄₈H₂₄⁺⁺ down to C₃₆H₁₂⁺⁺ [*m*/*z* 448 (10%), 446 (4%) and 444 (1%)]^{14,15} and C₄₈H₁₈⁺⁺ [*m*/*z* 598 (7%), 596 (3%) and 594



Fig. 1 Projection of 1a on the C₆₀ Schlegel diagram.[†]

370

(1%)], respectively.§ In contrast at a laser fluence of 440 μ J per pulse **3**⁺⁺ in line with previous observations¹⁵ fragments extensively, while **4**⁺⁺ still gives C₄₈H₂₂⁺⁺, C₄₈H₂₀⁺⁺ and C₄₈H₁₈⁺⁺.

Only in the case of $1a^{++}$ are consecutive cyclodehydrogenations spanning the mass range m/z 750 down to m/z 720 (C₆₀⁺) observed.§

ii) The *post-source-decay* (PSD)¹⁷ spectra, in which **1a**⁺⁺ is isolated prior to TOF-MS analysis, measured at different laser fluences, are nearly identical to the corresponding *in-source* spectra.[†] Similar observations were made for **3**⁺⁺ and **4**⁺⁺. Hence, intramolecular H₂ losses accompanied by ring closures originate from the precursor ions of **1a**, **3** and **4**, respectively.

Insight into the conversion of 1a into C_{60} was obtained using AM1¹³ calculations.[†]¶ Interestingly, the results indicate that cyclodehydrogenations occur with regio-chemical control. In propeller-like **1a** (C_3 , ΔH_f^0 344.6 kcal mol^{-1 8}) the three innermost, symmetry related C-H(a) pairs are congested due to interaction with a C-H moiety of a neighbouring unit $[H(a)\cdots H]$ 1.89 Å]. Thus, they are susceptible to undergo ring-closure giving bowl-shaped $C_{60}H_{24}$ (C_3 , ΔH_{f^0} 481.0 kcal mol⁻¹) in which three other C-H(b) pairs $[H(b)\cdots H 2.01 \text{ Å}]$ become congested (Chart 1 and Fig. 1); this will facilitate $C_{60}H_{18}$ (C_3 , $\Delta H_{\rm f}^{0}$ 625.7 kcal mol⁻¹) formation. In C₆₀H₁₈ the C-H(c) pairs $[H(c) \cdots H 2.14 \text{ Å}]$ are prone to undergo cyclodehydrogenation towards $C_{60}H_{12}(C_3, \Delta \hat{H}_f^0$ 777.2 kcal mol⁻¹). After closure of the next set of over-crowded C-H(d) bonds the ring-opened fullerene C₆₀H₆ ($C_{3\nu}$, ΔH_f^0 874.3 kcal mol⁻¹) is obtained. Closure of the latter into C_{60} (I_h , ΔH_f^0 973.4 kcal mol⁻¹) may involve three extra overall H₂ losses and ring closures, *i.e.* occur *via* ring-opened fullerenes $C_{60}H_4(C_s, \Delta H_f^0 930.5 \text{ kcal mol}^{-1})$ and $C_{60}H_2$ (C_s , ΔH_f^0 977.5 kcal mol⁻¹),†¶ However, AM1 calculations also showed that the ring-closed isomer of $C_{60}H_6(C_{3v}, \Delta H_f^0 865.3 \text{ kcal mol}^{-1}^{\dagger})$, accessible via a 2 + 2 + 2 cycloaddition, is 9 kcal mol⁻¹ more stable.¹⁸ Similarly, ringclosed analogues of ring-opened C₆₀H₄ (C_s , ΔH_f^0 901.1 and C_s , $\Delta H_{\rm f}^0$ 892.2 kcal mol⁻¹[†]) and ring-opened C₆₀H₂(C_s, $\Delta H_{\rm f}^0$ 950.0 and $C_{2\nu}$, $\Delta H_{\rm f}^0$ 931.2 kcal mol⁻¹⁺) were found to be 29.4/38.3 and 27.5/46.3 kcal mol⁻¹, respectively, more stable. Thus from **1a**, curling up and closure of the surface presumably involves three-fold 6-, 5-, 5-, and 6-membered ring closures, a 2 + 2 + 2 cycloaddition followed by three-fold overall H₂ losses. In line with the endothermic homolytic C-H bond scission energy (*ca.* 100 kcal mol⁻¹ for PAH¹⁹) excess energy (high laser fluences), is required to accomplish all ring-closures and overall H₂ losses. Whereas for the consecutive conversion of C₆₀H₃₀ into C₆₀H₂₄, C₆₀H₁₈ and C₆₀H₁₂, respectively, similar ΔH_r [= $\Delta H_f^{0}(C_{60}H_x) - \Delta H_f^{0}(C_{60}H_{x-6})$] values (136.4, 144.7 and 151.5 kcal mol⁻¹) are found, $\Delta H_r(C_{60}H_{12} \rightarrow C_{60}H_6)$ (97.1 kcal mol⁻¹) is decisively smaller.[†]These results rationalise: i) cyclodehydrogenations of 1a will be only discernible in the positive-ion mode TOF-MS spectra.8[‡] ii) The similarity of the in-source and PSD¹⁷ TOF-MS spectra of 1a. iii) The gradual



Fig. 2 MALDI TOF-MS spectrum of **1a** (laser fluence *ca.* 440 μ J pulse⁻¹). See ESI for the nearly identical *post-source-decay* (PSD)¹⁷†‡ MALDI TOF-MS spectrum of **1a**⁺⁺ (*m*/*z* 750).

decrease in intensity of the mass peaks concomitant with an increasing number of overall H₂ losses. iv) The occurrence of near zero peak intensities in the mass range m/z 730–726.

Isotope pattern analysis in the mass range m/z 720 to m/z 725 shows that at the high laser fluence (*ca.* 440 µJ pulse⁻¹) both C₆₀⁺⁺ and C₆₀H₂⁺⁺ are formed [*ratio* 1:1, natural isotope pattern: *calc.* 720 (4.9%), 721 (3.3%), 722 (6.0%), 723 (3.5%), 724 (1.1%) and 725 (0.3%); *found* 720 (5.0%), 721 (3.5%), 722 (6.0%), 723 (3.6%), 724 (2.0%) and 725 (0.9%), Fig. 2]. Just after we finished this work an independent report of the conversion of **1a** into C₆₀⁺⁺ using LDI MS appeared.²⁰

In summary, at high laser fluences **1a** is converted stepwise into C_{60} ⁺. This renders **1a** a synthon for the formation of endohedral derivatives of C_{60} ^{;21} experiments are in progress.

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Notes and references

‡ MALDI TOF-MS (positive-ion mode): Applied Biosystems/PerSeptive Biosystems Voyager-DE-RP MALDI TOF-MS [N₂ laser; λ_{exc} 337 nm (3 ns pulses)]. TOF-MS spectra were recorded in the reflection mode.⁸

Samples: $1a^9$, $3^{14,15}$ and 4^{16} were mixed with a dihydroxybenzoic acid solution (3 mg L⁻¹); 1 µL of the suspension was loaded on the Au-sample plate. At a laser fluence of *ca*. 8 µJ pulse⁻¹ 1a only gives a peak at *m/z* 750 [C₆₀H₃₀⁺ natural isotope pattern: calc. 750 (100%), 751 (67%), 752 (22%) and 723 (6%); found 750 (100%), 751 (70%), 752 (33%) and 723 (10%)]. Cyclodehydrogenation reactions are not discernible in the negative-ion mode TOF-MS spectra (see text and ref. 8).

Besides pure ¹²C peaks, natural isotope peaks corresponding to the presence of single and double ¹³C isotopes are discernible.

 \P With the exception of the AM1¹³ geometry of ring-opened (5,6) C₆₀H₂, all structures were identified as minima by harmonic analysis.[†]

- 1 H. W. Kroto, J. R. Heath, S. C. O'Brien, R. F. Curl and R. E. Smalley, *Nature*, 1985, **318**, 162.
- 2 W. Krätschmer, L. D. Lamb, K. Fostiropoulos and D. R. Huffman, *Nature*, 1990, **347**, 354.
- 3 For a review: A. Hirsch, Top. Curr. Chem., 1999, 199, 1.
- 4 For a review see: G. Mehta and H. S. Prakash Rao, *Tetrahedron*, 1998, **54**, 13325 and references cited therein.
- 5 Y. Rubin, T. Parker, S. J. Pastor, S. Jalisatgi, C. Boulle and C. L. Wilkins, *Angew. Chem., Int. Ed. Engl.*, 1998, **37**, 1226; *cf.* also F. Diederich, S. W. McElvany, M. M. Ross and N. S. Goroff, *Science*, 1993, **259**, 1594 and references cited therein.
- 6 Y. Tobe, N. Nakagawa, J. Kishi, M. Sonoda, K. Naemura, T. Wakabayashi, T. Shida and Y. Achiba, *Tetrahedron*, 2001, 57, 3629.
- 7 A. L. Lafleur, J. B. Howard, K. Taghizadeh, E. F. Plummer, L. T. Scot, A. Necula and K. C. Swallow, *J. Phys. Chem.*, 1996, **100**, 17421 and references cited therein.
- 8 M. Sarobe, R. H. Fokkens, T. J. Cleij, L. W. Jenneskens, N. M. M. Nibbering, W. Stas and C. Versluis, *Chem. Phys. Lett.*, 1999, **313**, 31.
- 9 B. Gómez-Lor, Ó. de Frutos and A. M. Echavarren, *Chem. Commun.*, 1999, 2431.
- 10 Cf. also M. J. Plater, J. Chem. Soc., Perkin Trans. 1, 1997, 2897.
- 11 M. Sarobe, L. W. Jenneskens, J. Wesseling and U. E. Wiersum, J. Chem. Soc., Perkin Trans. 2, 1997, 703 and references cited therein.
- 12 Cf. also: K. Zimmerman, R. Goddard, C. Krüger and M. W. Haenel, Tetrahedron Lett., 1996, 37, 8371.
- 13 M. J. S. Dewar, E. G. Zoebisch, E. F. Healey and J. J. P. Stewart, J. Am. Chem. Soc., 1985, 107, 3902.
- 14 M. Sarobe, *Ph.D. Thesis*, Utrecht University, Utrecht, The Netherlands, 1998, Chapter 8.
- 15 S. P. Ekern, A. G. Marshall, J. Szczepanski and M. Vala, *J. Phys. Chem. A*, 1998, **102**, 3498.
- 16 Ó. de Frutos, B. Gómez-Lor, T. Granier, M. A. Monge, E. Puebla-Gutierrez and A. M. Echavarren, *Angew. Chem., Int. Ed. Engl.*, 1999, 38, 204.
- 17 B. Spengler, J. Mass Spectrom., 1997, 32, 1019.
- 18 Cf. also: Y. Rubin, Chem. Eur. J., 1997, 3, 1009.
- 19 J. Cioslowski, G. Liu, M. Martinov, P. Piskorz and D. Moncrieff, *J. Am. Chem. Soc.*, 1996, **118**, 5261 and references cited therein.
- 20 M. M. Boorum, Y. V. Vasil'ev, T. Drewello and L. T. Scott, *Science*, 2001, **294**, 828.
- 21 Cf. J.-F. Nierengarten, Angew. Chem., Int. Ed. Engl., 2001, 40, 2973 and references cited therein.