

A highly enantioselective synthesis of cyclic α -amino acids involving a one-pot, single catalyst, tandem hydrogenation–hydroformylation sequence†

Euneace Teoh, Eva M. Campi, W. Roy Jackson and Andrea J. Robinson*

Centre for Green Chemistry and School of Chemistry, PO Box 23, Monash University, Victoria 3800, Australia. E-mail: A.Robinson@sci.monash.edu.au; Fax: 61 03 9905 4597; Tel: 61 03 9905 4553

Received (in Corvallis, OR, USA) 10th January 2002, Accepted 12th March 2002

First published as an Advance Article on the web 5th April 2002

Tandem enantioselective hydrogenation followed by a hydroformylation–cyclisation sequence leading to cyclic α -amino acids with ee's >95% can be achieved in a single pot, one catalyst system by successive reactions of prochiral dienamide esters with H₂ followed by H₂/CO using Rh(*t*)-DuPHOS.

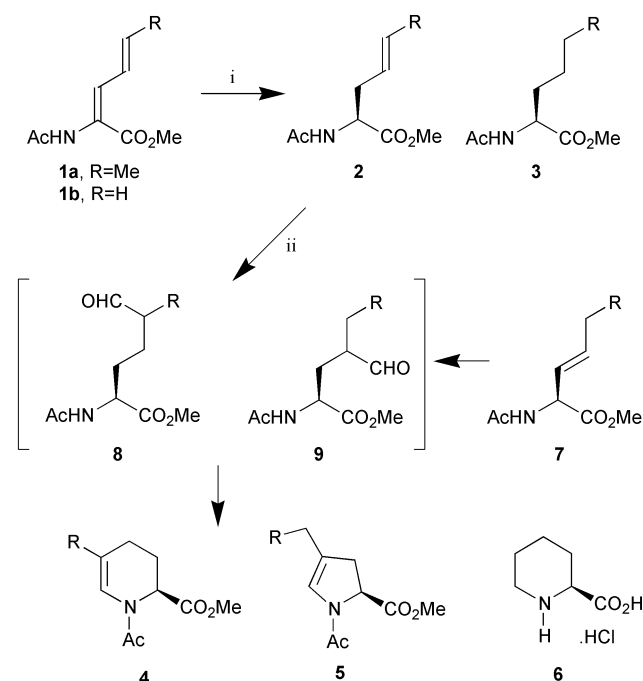
Cyclic amino acids are of increasing biological importance because of their relationship to naturally occurring biological molecules, *e.g.* the izidine alkaloids¹ (piperidines), kainic acid analogues² (pyrrolidines), and their use in peptidomimetics.³ Recent syntheses of pipercolic acid derivatives have involved aldol condensation–reductive amination of aspartate β -aldehyde,⁴ intramolecular cyclohydrocarbonylation of unsaturated amides⁵ and ruthenium catalysed ring closing metathesis.⁶ Both of these latter preparations start with allylglycine which is available in both enantiomeric forms.⁷ In spite of these recent advances, there appears to be a demand for facile routes to this class of compound as other general syntheses are relatively limited in scope. In this communication we describe a general synthesis of either enantiomer of piperidino- and pyrrolidino-based α -amino acids by a sequence which, in its final form, involves a single pot, one catalyst reaction sequence. Such tandem sequences are becoming increasingly important in organic synthesis and have recently been used for isomerisation–carbonylation of alkenes.⁸ Domino reactions employing hydroformylation–condensation–hydrogenation sequences have also recently been reported.⁹

Rhodium catalysed hydrogenation of the prochiral dienamide ester **1a** under conditions reported by Burk¹⁰ gave the allylglycine derivative **2a** in high enantiomeric excess (>95%) and yield (98%) (Scheme 1). Either enantiomer of **2a** could be obtained by the use of (*R,R*)- or (*S,S*)-Et-DuPHOS with the (*R,R*)-catalyst giving the (*R*)-enantiomer of **2a**.^{10,11} Excellent regioselectivity was observed with <5% of the saturated product **3a** being produced. The unsubstituted analogue **1b** was also hydrogenated under milder conditions to give **2b** with excellent regioselectivity but with a slightly increased amount of over-reduced material **3b** (5–15%). The enantioselectivity was again >95% as shown by chiral gas chromatography (Scheme 1).

Ojima and co-workers have previously reported preparation of pipercolic acid derivatives from protected allylglycines closely related to **2b**.⁵ Reactions of **2b** using rhodium–phosphine catalysts with H₂/CO under conditions previously used in our work¹² gave very good isolated yields (*ca.* 70%) of the cyclic amino acid derivatives **4b** and **5b** (Table 1, entries 1 and 2) (Scheme 1). The compounds were readily separated and chiral hplc showed that the enantiomeric excess had not been compromised. Confirmation that the (*R*)-configuration was present in samples of **4b** prepared using (*R,R*)-Et-DuPHOS was obtained by conversion of **4b** to (*R*)-pipercolic acid hydrochloride **6** and comparison of optical rotation values.¹³

The isolated piperidine **4**:pyrrolidine **5** ratio originates from the regioselectivity exhibited during hydroformylation. The

observed selectivity (<2:1 for **4b**:**5b**), however, was considerably lower than expected given that hydroformylation of a terminal alkene using PPh₃ as ligand would be expected to yield a piperidine:pyrrolidine ratio of *ca.* 2:1, whereas the bulkier BIPHEPHOS ligand should bias this ratio to *ca.* 9:1.^{12,14} An



Scheme 1 (i) **1a**: (*S,S*)-Et-DuPHOS-Rh(*t*), H₂ (90 psi), 2 h, PhH, rt; **1b**: (*S,S*)-Et-DuPHOS-Rh(*t*), H₂ (30 psi), 3 h, PhH, rt. (ii) H₂/CO (1:1), 80–400 psi, 80–100 °C, 20–72 h. [Rh(OAc)₂]₂, PPh₃ or BIPHEPHOS with substrate:Rh(*t*):phosphine ratio of 100:1:2.

Table 1 Rhodium catalysed hydroformylation of chiral enamides **2a** (R = Me) and **2b** (R = H)^a

Entry	R	Catalyst System ^b	H ₂ /CO (psi)	T (°C)	Time (h)	Product Ratio (4 : 5)	Yield (%) ^c	% ee (4/5) ^g
1	H	A	400	80	72	50:50	73	--/87
2	H	B	400	80	20	63:37	66	88/--
3	H	B	100	80	20	71:29	54 ^d	--/87
4	H	B	80	80	20	78:22	— ^e	
5	H	B	80	80	72	66:34	75	99/--
6	Me	A	400	80	20	67:33	45	91/98
7	Me	B	400	80	20	100:0	37 ^f	
8	Me	B	400	100	72	91:9	81	97/--

^a Reaction conditions: Substrate (0.3 mmol) : [Rh(OAc)₂]₂ : PPh₃ or BIPHEPHOS ratio = 100:1:2 in benzene (5–10 ml) with H₂/CO (1:1).

^b Catalyst code: A = Rh-PPh₃, B = Rh-BIPHEPHOS. ^c Isolated yield of cyclic products **4** and **5** after chromatography. ^d Crude product contained *ca.* 20% isomerised alkenamide **7**. ^e Crude product contained *ca.* 50% isomerised alkenamide **7**. ^f Aldehydes **8a** and **9a** (*ca.* 1:1) also obtained. ^g -- Signifies that enantioselectivity was not assessed.

† Electronic supplementary information (ESI) available: experimental information. See <http://www.rsc.org/suppdata/cc/b2/b200374k/>

explanation for this anomaly became apparent after conducting the hydroformylation experiments under milder conditions. Reaction of **2b** at 80 °C with 100 psi of H₂/CO over 20 h (entry 3) gave a modest yield (54%) of **4b** and **5b** and ca. 20% of the isomerised alkenamide **7**. Significantly, Rh(I)-BIPHEPHOS catalysed hydroformylation of **7** would favour reaction remote from the bulky amino acid moiety resulting in pyrrolidine **5b** after cyclisation and dehydration (Scheme 1). This secondary route to **5b** explains the higher than expected amounts of pyrrolidine in reactions involving **2b**. Under lower hydroformylation pressure (80 psi) an even greater percentage of **7** (ca. 50%) was observed after 20 h (entry 4) and extended reaction over 72 h gave significant amounts of **5b** (entry 5).

Hydroformylation of the homologue **2a** at 80 °C with 400 psi H₂/CO using PPh₃ as ligand gave **4a** and **5a** in a 2:1 ratio (entry 6) whilst a similar reaction using BIPHEPHOS gave a significant amount of the uncyclised aldehydes **8a** and **9a** together with piperidine **4a** (entry 7). Increasing the temperature and reaction time led to complete conversion and isolation of cyclic compounds **4a** and **5a** in high yield and in a ratio of 9:1.

A tandem reaction was then investigated which capitalized on the disparate operating conditions of the two Rh-DuPHOS and Rh-BIPHEPHOS catalysts. Under mild reaction pressures, only the Rh-DuPHOS catalyst would be expected to facilitate hydrogenation of the enamide substrates **1** ensuring the maintenance of high enantioselectivity. Hence, in the presence of both Rh-DuPHOS and Rh-BIPHEPHOS, hydrogenation of **1b** (rt, 30 psi of H₂, 3 h) followed by hydroformylation (80 °C, 80 psi of H₂/CO) gave **4b** and **5b** in a 2:1 ratio and 63% isolated yield. Importantly, the enantiomeric excess of **4b** and **5b** was found to be >95% (entry 9, Table 2). A similar reaction

sequence involving **1a** with Rh-DuPHOS and Rh-PPh₃ gave **4a** and **5a** in ca. 1:1 ratio in 81% isolated yield (entry 10). The ee of **4a** and **5a** were also shown to be ≥ 95% by chiral hplc.

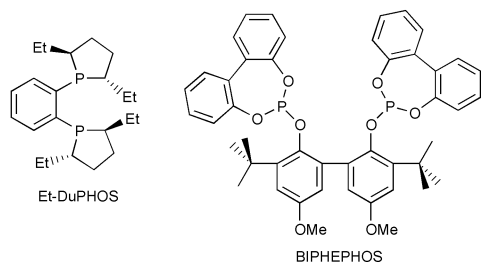
Next we investigated whether a single catalyst, namely Rh(I)-DuPHOS, could be employed to facilitate both hydrogenation and hydroformylation. To our knowledge, the use of Rh-DuPHOS as a hydroformylation catalyst has not been previously reported. Reaction of dienamide **1b** in the presence of Rh-DuPHOS alone, initially with H₂ at ambient temperature and then with H₂/CO (80 psi) at 80 °C, gave a product containing significant amounts of the isomerised alkene **7** (entry 11). Reaction using a higher pressure of H₂/CO (400 psi) gave complete conversion to **4b** and **5b** in ca. equimolar ratio in 91% isolated yield with excellent enantioselectivity (entry 12). It appears that the Rh-DuPHOS system is slightly less efficient for hydroformylation than the Rh-BIPHEPHOS system (compare entries 5 and 11). A similar diminution in rate was observed for a reaction of **1a** in the presence of Rh-DuPHOS where hydroformylation under conditions giving complete conversion using Rh-PPh₃ (entry 6, Table 1; entry 10, Table 2) gave substantial amounts of aldehydes (**8a** and **9a**) (entry 13). A reaction under more forcing conditions gave complete conversion to the piperidine **4a** which was isolated as the sole product in 58% yield and with 96% ee (entry 14).

These results clearly establish that it is possible to obtain cyclic α-amino acids in good to very good yields and with excellent enantioselectivity using a single catalyst in a single pot *via* a tandem reaction sequence.

We thank Monash University and the Centre for Green Chemistry for provision of a postgraduate award (to ET), the Australian Research Council for its financial support of this research and Johnson Matthey Pty Ltd for the loan of rhodium.

Table 2 Rhodium catalysed tandem hydrogenation and hydroformylation of dienamides **1a** (R = Me) and **1b** (R = H)^a

Entry	R	Catalyst System ^b	H ₂ /CO (psi)	T (°C)	Time (h)	Product		
						Ratio (4 : 5)	Yield (%) ^c	% ee (4/5)
9	H	B + C	80	80	72	67:33	63	95/95
10	Me	A + C	400	80	20	56:44	81	95/99
11	H	C	80	80	72	74:26	— ^d	
12	H	C	400	80	72	54:46	91	95/99
13	Me	C	400	80	20	83:17	35 ^e	99/99
14	Me	C	800	150	72	100:0	58	96/— ^f



^a Reaction conditions: Substrate (0.3 mmol) : Rh-Et-DuPHOS: Rh[(OAc)₂]₂:PPh₃ or BIPHEPHOS ratio = 100:1:1:2 in benzene (5–10 ml) with H₂ (30 psi, 3 h for **1b** and 90 psi, 2 h for **1a**) at ambient temperature followed by H₂/CO (1:1). ^b Catalyst code: A = Rh-PPh₃, B = Rh-BIPHEPHOS, C = Rh-Et-DuPHOS. ^c Isolated yield of cyclic products **4** and **5** after chromatography. ^d Crude product contained ca 40% isomerised alkenamide **7**. ^e Aldehydes **8a** and **9a** also obtained. ^f Enantioselectivity was not assessed.

Notes and references

- H. Takahara, H. Bandoh and T. Momose, *Tetrahedron*, 1993, **49**, 11205.
- A. F. Parsons, *Tetrahedron*, 1996, **52**, 4149.
- S. Hanessian, G. McNaughton-Smith, H.-G. Lombart and W. D. Lubell, *Tetrahedron*, 1997, **53**, 12789.
- M. E. Swarbrick, F. Gosselin and W. D. Lubell, *J. Org. Chem.*, 1999, **64**, 1993.
- I. Ojima, M. Tzamarioudaki and M. Eguchi, *J. Org. Chem.*, 1995, **60**, 7078.
- K. C. M. F. Tjen, S. S. Kinderman, H. E. Schoemaker, H. Hiemstra and F. P. J. T. Rutjes, *J. Chem. Soc., Chem. Commun.*, 2000, 699.
- F. P. J. T. Rutjes and H. E. Schoemaker, *Tetrahedron Lett.*, 1997, **38**, 677.
- R. I. Pugh, E. Drent and P. G. Pringle, *J. Chem. Soc., Chem. Commun.*, 2001, 1476.
- B. Breit and S. K. Zahn, *Angew. Chem., Int. Ed.*, 2001, **40**, 1910 and references cited therein.
- M. J. Burk, J. G. Allen and W. F. Kiesman, *J. Am. Chem. Soc.*, 1998, **120**, 657.
- M. J. Burk, J. E. Feaster, W. A. Nugent and R. L. Harlow, *J. Am. Chem. Soc.*, 1993, **115**, 10125.
- J. J. Carbo, F. Maseras, C. Bo and P. W. N. M. van Leeuwen, *J. Am. Chem. Soc.*, 2001, **123**, 7630; D. J. Bergmann, E. M. Campi, W. R. Jackson and A. F. Patti, *J. Chem. Soc., Chem. Commun.*, 1999, 1279; D. J. Bergmann, E. M. Campi, W. R. Jackson, Q. J. McCubbin and A. F. Patti, *Tetrahedron*, 1997, **53**, 17449.
- The found and literature value of [α]_D for (*S*)-**6** (+10.0 °C, c = 2, H₂O) were in agreement.
- A. G. Billig, D. R. Abatjoglou and D. R. Bryant, U.S. Patent 4769498, 1988; G. D. Cuny and S. L. Buchwald, *J. Am. Chem. Soc.*, 1993, **115**, 2066.