

# *N*-Allyl-*N*-*tert*-butyldimethylsilylamine for chiral ligand-controlled asymmetric conjugate addition to *tert*-butyl alkenoates†

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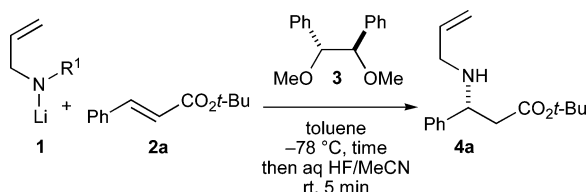
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The chiral ligand controlled asymmetric conjugate addition reaction of lithium *N*-allyl-*N*-(*tert*-butyldimethylsilyl)amide to alkenoates proceeded smoothly to give, after protodesilylation, the corresponding 3-allylaminoalkanoates with high enantioselectivities in high yields. The allyl group on the nitrogen atom was easily removable to afford 3-aminoalkanoates.

Asymmetric conjugate addition of chiral lithium amides to alkenoates has been one of the most powerful methods for the synthesis of chiral 3-aminoalkanoates as has been extensively studied by Davies and his group.<sup>1,2</sup> A chiral ligand-controlled asymmetric reaction<sup>3</sup> of an achiral lithium amide has been recently reported<sup>4</sup> as a new entry to this interesting field. However, this method relies on the use of *N*-benzy-*N*-trimethylsilylamine as a nitrogen source that requires hydrogenolysis for removal of the benzyl group. We describe herein that allylamine is also a good source of a nitrogen nucleophile in the asymmetric addition to alkenoates, and the allyl group is easily removable by the isomerisation reaction with a rhodium catalyst.<sup>5,6</sup>

We began our studies with the reaction of allylamine (Scheme 1). The lithium amide **1a** ( $R^1 = H$ ) was *in situ* prepared from allylamine with butyllithium in the presence of chiral ligand **3** in toluene and was then treated with *tert*-butyl cinnamate **2a**. Although the reaction proceeded within a half hour at  $-78^\circ C$  to give **4a** in 64% isolated yield, the enantioselectivity was unfortunately almost marginal (Table 1, entry 1).

The reaction of lithium *N*-allyl-*N*-trimethylsilylamide **1b** ( $R^1 = TMS$ , 3 equiv) was effected by **3** (3.6 equiv) at  $-78^\circ C$  to give, after purification by silica gel column chromatography, **4a** with 69% ee in 94% yield (entry 2). The absolute configuration and enantioselectivity of **4a** were determined by converting to **5** with the



**Scheme 1** Asymmetric reactions of lithium allylamides **1** with *tert*-butyl cinnamate **2a** by the mediation of **3**.

**Table 1** Asymmetric addition of lithium allylamides **1** to cinnamate **2a** in toluene at  $-78^\circ C$  giving **4a**

Entry	<b>1</b>	$R^1$	time (h)	Yield (%)	Ee (%)
1	<b>a</b>	H	0.5	64	5
2	<b>b</b>	TMS	0.5	94	69
3	<b>c</b>	TES	1	91	83
4	<b>d</b>	TBS	1	95	86
5 <sup>a</sup>	<b>d</b>	TBS	1.5	95	89
6	<b>e</b>	TIPS	5	0	0

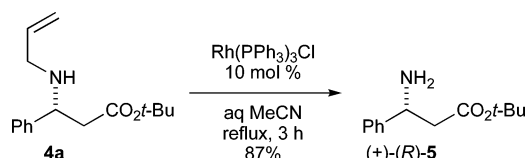
<sup>a</sup> Performed at  $-95^\circ C$ .

† Electronic supplementary information (ESI) available: general procedure for addition reaction, deallylation, silylation of allylamine and data for compounds. See <http://www.rsc.org/suppdata/cc/b4/b405347h/>

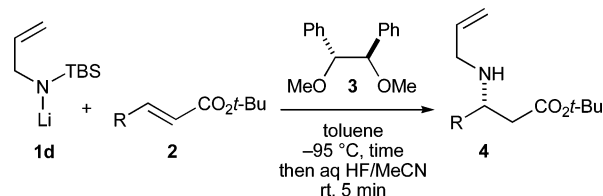
established absolute configuration as shown below. Dramatic improvement of enantioselectivity was observed by changing a TMS group to TES (triethylsilyl) (**1c**:  $R^1 = TES$ ) giving **4a** with 83% ee in 91% yield (entry 3). A TES group was not removed on silica gel column chromatography, but easily protodesilylated with aqueous HF in acetonitrile at room temperature for 5 min to afford **4a**.<sup>7</sup> TBS (*tert*-butyldimethylsilyl) amide **1d** ( $R^1 = TBS$ )<sup>8</sup> was much more effective to give **4a** with 86% ee in 95% yield (entry 4). The enantioselectivity was improved to 89% ee by conducting the reaction at  $-95^\circ C$  (entry 5). TIPS (triisopropylsilyl) amide **1e** ( $R^1 = TIPS$ ) was a poor nucleophile not to give a conjugate addition product probably because of a too much steric hindrance (entry 6).

The allyl group of **4a** was easily removed with rhodium chloride<sup>9</sup> in refluxing aqueous acetonitrile for 3 h to give (+)-(*R*)-**5**<sup>10</sup> in 87% isolated yield without any racemisation (Scheme 2). The stereochemistry of the newly created stereogenic center was thus established to be (*R*).

The asymmetric addition reactions of **1d** with other acyclic and cyclic alkenoates **2** were also controlled by **3** at  $-78^\circ C$  or  $-95^\circ C$  in toluene to afford **4** with up to 94% ee (Scheme 3). Crotonate **2b** ( $R = Me$ ) was converted to **4b** in 90% yield (Table 2, entry 1). The ee of **4b** was determined to be 90% by <sup>1</sup>H NMR using (*S*)-(-)-1,1'-bi-2,2'-naphthol as a chiral shift reagent.<sup>11</sup> The reaction of



**Scheme 2** Removal of the allyl group of **4a** to give **5** with the established absolute configuration.



**Scheme 3** Asymmetric addition of **1d** to **2** by the mediation of a chiral ligand **3**.

**Table 2** Asymmetric addition of *N*-TBS-allylamide **1d** to **2** in toluene at  $-95^\circ C$  giving **4**†

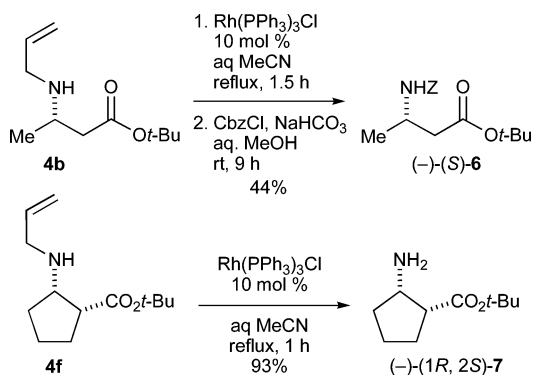
Entry	<b>2</b>	R	time (h)	<b>4</b>	Yield (%)	Ee (%)
1	<b>b</b>	Me	1	<b>b</b>	90	90
2 <sup>a</sup>	<b>c</b>	<i>i</i> -Pr	5	<b>c</b>	84	76
3 <sup>a</sup>	<b>d</b>	( <i>E</i> )-MeCH=CH	18	<b>d</b>	88	75
4	<b>e</b>	1-Naph	3	<b>e</b>	89	94
5	<b>f</b>		1.5	<i>cis</i> - <b>f</b>	82	82
				<i>trans</i> - <b>f</b>	11	84

<sup>a</sup> Performed at  $-78^\circ C$ .

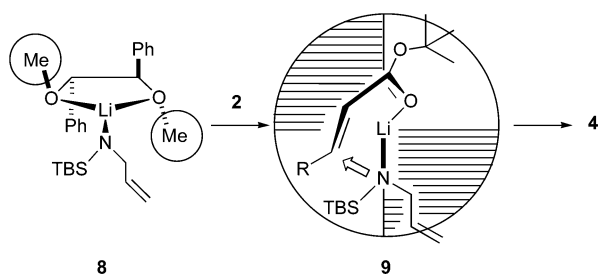
alkenoate **2c** having an isopropyl terminal group (R = *i*-Pr) gave **4c** with 76% ee in 84% yield (entry 2). Sorbate **2d** (R = (*E*)-MeCH=CH) was regioselectively converted to 1,4-addition product **4d** with 75% ee in 88% yield (entry 3). The reaction of **2e** (R = 1-naphthyl) gave **4e** with 94% ee in 89% yield (entry 4). Cyclopentencarboxylate **2f** was also applicable in the reaction to give diastereoselectively *cis*-**4f** with 82% ee as a major product in 82% yield (entry 5). The minor *trans*-**4f** with 84% ee was also produced in 11% yield. It is important to note that chiral ligand **3** was recovered in high yield and was re-used in the asymmetric reaction without loss of selectivity.

Deallylation of **4b** and *cis*-**4f** was achieved by treatment with Wilkinson's catalyst in refluxing aqueous acetonitrile to give (–)-(*S*)-**6** in 44% overall yield after conversion to a Cbz derivative,<sup>12</sup> and (–)-(1*R*,2*S*)-**7**<sup>13</sup> in 93% yield without any racemisation. It is important to note that allylamine attacks to the bottom face of linear and cyclic alkenoates shown as **2** giving **4**.

The stereocontrolled formation of chiral **4** is predictable by using a model (Scheme 5). Coordination of the carbonyl oxygen atom of **2** to a lithium atom in a chelate **8** may be the first event for the reaction. Two etheral methyl groups are fixed in *trans* relationship to the adjacent two phenyl groups. It is reasonable to speculate that coordination of **2** takes place in the less hindered region avoiding steric repulsion by the methyl groups of **3** as shown in **9**.<sup>14</sup> The intra-complex attack of the nitrogen atom to the *si*-face (for example R = Me) at the 3-position of *s-cis*-**2** provides chiral **4** with the observed absolute configuration.



**Scheme 4** Removal of the allyl groups of **4** giving **6** and **7** with the established absolute configurations.



**Scheme 5** Plausible stereoselection for the production of **4** from **9**.

In summary, an external chiral ligand-controlled asymmetric conjugate addition of allylamine to alkenoates was developed. Since both of enantiomers **3** are available, either enantiomer of **4** is accessible. Synthetic utility of the products is the next target.

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## Notes and references

‡ *Procedure* (Table 2, entry 1): *n*-BuLi (3.0 mmol) was added to **1d** (3.0 mmol) in toluene (8 mL) at  $-78^{\circ}\text{C}$  over 5 min. After 0.5 h, **3** (3.6 mmol) in toluene (6 mL) was added. The mixture was stirred for 0.5 h at  $-78^{\circ}\text{C}$ , then cooled to  $-95^{\circ}\text{C}$ . A toluene (2 mL) solution of **2b** (1.0 mmol) was added over 5 min. The mixture was stirred at  $-95^{\circ}\text{C}$  for 1.5 h and quenched with satd.  $\text{NH}_4\text{Cl}$  (3 mL). After addition of satd.  $\text{NaHCO}_3$  (4.5 mL), the whole was extracted with AcOEt. The organic layer was washed with brine, dried, and concentrated. The crude product in MeCN (30 mL) was treated with 46% HF (3 mL) at rt for 5 min. After addition of satd.  $\text{NaHCO}_3$ , the mixture was extracted with AcOEt. The organic layer was washed with brine and dried. Concentration and chromatography (AcOEt/hexane = 1/1) gave **3** (quant. recovery) and **4b** with 90% ee (determined by  $^1\text{H}$  NMR using (*S*)-(–)-1,1'-bi-2,2'-naphthol as a chiral shift reagent).

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