www.rsc.org/chemcomm

ChemComm

Chan-Mo Yu,* Seok-Keun Yoon, Young-Taek Hong and Jimin Kim

Department of Chemistry and Lab for Metal-Catalysed Reactions, Sungkyunkwan University, Suwon 440-746, Korea. E-mail: cmyu@chem.skku.ac.kr; Fax: +82 31 290 7075; Tel: +82 31 290 7067

Received (in Cambridge, UK) 29th April 2004, Accepted 23rd June 2004 First published as an Advance Article on the web 23rd July 2004

A novel intramolecular Prins type cyclisation of cyclopropylvinylic aldehydes 1 promoted by TiCl₄ under mild reaction conditions to form *cis*-cyclic products 2 in high yields has been accomplished.

The development of efficient synthetic methods for achieving stereocontrol in the construction of cyclic systems is of considerable current interest in synthetic chemistry.1 During the past decades, substantial progress has been made, and as a result, many stereoselective synthetic routes have been extensively explored.² In this regard, the intramolecular Prins cyclisation has proved to be a useful transformation in the construction of five, six, and even larger membered rings for the synthesis of carbocyclic and heterocyclic biofunctional molecules.³ Notable methods have been developed in recent years for accessing stereocontrol through the Prins type reaction, the vast majority of which involve construction of the oxacyclic ring by carbon-carbon bond formation from an oxocarbonium intermediate via electrophilic cyclisation.⁴ In the course of our research program aimed at finding new synthetic methods for the stereoselective construction of pyran units,5 we became quite interested in the reaction of cyclopropylvinylic aldehyde 1 in the presence of promoter MX to form cyclic compound 2 through the Lewis acid catalyzed Prins procedure as depicted in Scheme 1.

It was envisaged that the cyclisation of **1** with proper promoter MX for the diastereoselective synthesis of 2 through a carbocationic intermediate A could be realized in a predictable fashion. The background to this prediction was the known reaction pathway for the formation of homoallylic halide 4 from cyclopropyl alcohol 3 in acidic media through intermediate B.6 To the best of our knowledge, the use of the cyclopropylvinyl moiety as an electrophile for capturing an incoming nucleophile has not been reported, even though this functionality has been often employed as a substrate for transition metal catalysed cycloadditions.^{7,8} The realization of an efficient method in this reaction should also be valuable because synthetic applications can be foreseen for homoallylic chloride.9 This research led to the discovery of a new type of Prins cyclisation of 1 under mild reaction conditions with remarkably high diastereoselectivity. The starting point of this investigation was the availability of cyclopropylvinylic aldehydes 1: these compounds were readily prepared by the modification of



† Electronic supplementary information (ESI) available: experimental details. See http://www.rsc.org/suppdata/cc/b4/b406362g/

known procedures as described in the literature.^{10,11} The stage was thus set for the Prins cyclisation of cyclopropylvinylic aldehydes **1** mediated by a Lewis acid.

To find optimum conditions a series of experiments was performed with cyclopropylvinylic aldehyde **1a** as a model substrate. Preliminary investigations on the transformation of **1a** indicated that the conversion to the corresponding **2a** could not be achieved satisfactorily with $SnCl_4$ and $InCl_3$ under various conditions mainly due to decomposition of substrate as shown in Table 1.

Fortunately, we found that TiCl₄ was able to promote the reaction process; TiCl₄ was generally superior and was chosen for systematic studies. After surveying numerous conditions for orienting experiments, several key findings emerged: i) 1.1 equivalents of TiCl₄ were required for optimal conditions in terms of chemical yields and reaction rates; ii) reaction performed at -78°C to 0 °C in CH₂Cl₂ resulted in the best chemical yields in comparison with other solvents such as toluene, CH₃CH₂CN, and THF; iii) reaction produced the (E)-vinylic product as a sole isomer according to the analysis of 500 MHz ¹H NMR spectra; iv) only cisisomer 2a was formed as determined by the analysis of 500 MHz ¹H NMR spectra of crude products. Under optimal conditions, the reaction was performed by addition of $TiCl_4$ (1.1 equiv.) to a solution of 1a (1 equiv.) in CH_2Cl_2 at -78 °C. After 2 h at -78 °C, the reaction mixture was allowed to warm to 0 °C and then the reaction was quenched by addition of aqueous NaHCO₃. The crude product was purified by SiO₂ column chromatography to give 2a in 74% yield as a colorless oil. With the notion that this approach might lead to a general and efficient method for the new type of Prins cyclisation, we set out to determine the scope of reaction with several cyclopropylvinylic aldehydes 1 as listed in Table 2. Indeed, the method is successful with a variety of cyclopropylvinylic aldehydes 1 and affords products of high diastereomeric purity as can be seen in Table 2. It is worthy of note that the reaction produced no or only trace amounts of minor products according to the analysis of 500 MHz ¹H NMR spectra of the crude products.

The stereochemical assignments were based on the magnitudes of the vicinal coupling constants of six-membered ring protons for compounds **2a**, **2d**, and **2e**. Small coupling constants for CHOH for all cases clearly indicated that the hydroxy group is located at the axial position. The *cis*-stereochemical relationship for **2a** and **2b**

Table 1 Preliminary investigations^a



then warming to 0 °C. ^{*b*} Yields refer to isolated and purified products.

was also confirmed by NOE experiments: we observed NOE enhancements between stereogenic vicinal protons.

Although the exact mechanistic aspects of this transformation have not been rigorously elucidated, the following pathway could be a probable chemical route on the basis of experimental results. Obtaining the *cis*-products from **1** under TiCl₄ promotion can be accounted for by the intervention of the pseudopericyclic stereochemical models **C** and **D** with minimum steric interactions and optimal orbital interactions, which lead to the major reaction pathway as illustrated in Fig. 1. Thus, we believe that the orientation between olefinic and carbonyl moieties offered by TiCl₄ could be more crucial for stereoselectivity.

Table 2 Prins cyclization of 1 promoted by TiCl₄.^a



^{*a*} Reactions were carried out with TiCl₄ (1.1 equiv.) in CH₂Cl₂ at -78 °C for 2 h and then warming to 0 °C. ^{*b*} Yields refer to isolated and purified products.



In summary, a new intramolecular Prins type cyclisation of cyclopropylvinylic aldehydes 1 promoted by TiCl₄ under mild reaction conditions to form *cis*-products 2 has been accomplished in a general and efficient way, which promises to be useful. Further studies including synthetic applications and extension of this method into enantiomeric pathways are in progress.

Generous financial support from grants from the Korea Ministry of Science and Technology through the National Research Laboratory program (NRL) and the Center for Molecular Design and Synthesis (CMDS: KOSEF SRC) at KAIST is gratefully acknowledged.

Notes and references

- Reviews for Lewis acid promoted cyclizations, see: (a) M. Satelli and J.-M. Pons, *Lewis Acids and Selectivity in Organic Synthesis*, CRC Press, Boca Raton, 1995, pp. 91–225; (b) C. Thebtaranonth and Y. Thebtaranonth, *Cyclization Reactions*, CRC Press, Boca Raton, 1994, pp. 5–76.
- Reviews, see: (a) E. Langkopf and D. Schinzer, *Chem. Rev.*, 1995, **95**, 1375–1408; (b) I. Ojima, M. Tzamarioudaki, Z. Li and R. Donovan, *Chem. Rev.*, 1996, **96**, 635–662; (c) M. Lautens, W. Klute and W. Tam, *Chem. Rev.*, 1996, **96**, 49–92; (d) B. M. Trost, F. D. Toste and A. B. Pinkerton, *Chem. Rev.*, 2001, **101**, 2067–2096; (e) C. Aubert, O. Buisine and M. Malacria, *Chem. Rev.*, 2002, **102**, 813–834.
- 3 Reviews, see: (a) D. R. Adams and S. P. Bhatnagar, *Synthesis*, 1977, 661–667; (b) T. A. Blumenkopf and L. E. Overman, *Chem. Rev.*, 1986, 86, 857–873; (c) B. B. Snider, in *Comprehensive Organic Synthesis*, Vol. 2, ed. B. M. Trost, Pergamon Press, New York, 1991, pp. 527–561.
- 4 (a) W. H. Bunnelle, D. W. Seamon, D. L. Mohler, T. F. Ball and D. W. Thompson, *Tetrahedron Lett.*, 1984, 25, 2653–2654; (b) J. Yamada, T. Asano, I. Kadota and Y. Yamamoto, J. Org. Chem., 1990, 55, 6066–6068; (c) Y. Hu, D. J. Skalitzky and S. D. Rychnovsky, *Tetrahedron Lett.*, 1996, 37, 8679–8682; (d) M. J. Cloninger and L. E. Overman, J. Am. Chem. Soc., 1999, 121, 1092–1093; (e) H. Huang and J. S. Panek, Org. Lett., 2001, 3, 1693–1696; (f) X.-F. Yang, J. T. Mague and C.-J. Li, J. Org. Chem., 2001, 66, 739–747; (g) C. E. Davis and R. M. Coates, Angew. Chem., Int. Ed., 2002, 41, 491–493; (h) F. Lopez, L. Castedo and J. L. Mascarenas, J. Am. Chem. Soc., 2002, 124, 4218–4219; (i) Y. S. Cho, H. Y. Kim, J. H. Cha, A. N. Pae, H. Y. Koh, J. H. Choi and M. H. Chang, Org. Lett., 2003, 5, 1499–1502.
- 5 (a) C.-M. Yu, J.-Y. Lee, B. So and J. Hong, Angew. Chem., Int. Ed., 2002, 41, 161–163; (b) C.-M. Yu, J.-M. Kim, M.-S. Shin and M.-O. Yoon, Chem. Commun., 2003, 1744–1745.
- 6 (a) S. F. Brady, M. A. Ilton and W. S. Johnson, J. Am. Chem. Soc., 1968, 90, 2882–2889; (b) J. P. McCormick and D. L. Barton, J. Org. Chem., 1980, 45, 2566–2570; (c) S. Kanemoto, M. Shimizu and H. Yoshioka, Tetrahedron Lett., 1987, 28, 663–666.
- 7 (a) P. A. Wender, H. Takahashi and B. Witulski, J. Am. Chem. Soc., 1995, 117, 4720–4721; (b) P. A. Wender, C. O. Husfield, H. E. Langkopf, J. A. Love and N. Pleuss, *Tetrahedron*, 1998, 54, 7203–7220; (c) P. A. Wender, G. G. Gamber, R. D. Hubbard and L. Zhang, J. Am. Chem. Soc., 2002, 124, 2876–2877.
- 8 (a) B. M. Trost and F. D. Toste, Angew. Chem., Int. Ed., 2001, 40, 1114–1116; (b) B. M. Trost and H. C. Shen, Org. Lett., 2000, 2, 2523–2525.
- 9 (a) M. P. Cooke, Jr., Tetrahedron Lett., 1973, 14, 1581–1584; (b) J. E. McMurry and G. K. Bosch, J. Org. Chem., 1987, 52, 4885–4893.
- 10 P. A. Wender, F. G. Craig, H. E. Langkopf and J. A. Love, J. Am. Chem. Soc., 1999, 121, 5348–5349.
- 11 A. D. Aloise, M. E. Layton and M. D. Shair, J. Am. Chem. Soc., 2000, 122, 12610–12611.