

Novel cerium(IV) heteropolyoxotungstate containing two types of lacunary Keggin anions†

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A novel V-shaped polyoxotungstate is formed when Ce^{IV} metal centres bridge monolacunary [PW₁₁O₃₉]⁷⁻ anions to an unusual 1,4-bilacunary [PW₁₀O₃₈]¹¹⁻ anion which appears with an unprecedented bridging structural motif.

Polyoxometalates are a unique family of metal–oxygen clusters¹ which exhibit a myriad of molecular architectures² and find applications in fields ranging from materials science to medicine.³ Lanthanopolyoxometalates have been shown to exhibit catalytic activity^{4,5} and have been used as luminescent centres in functional materials.⁶ The first crystallographic report of a structure containing anions of the [M(XW₁₁O₃₉)₂]ⁿ⁻ type concerned Cs₁₂[U-(GeW₁₁O₃₉)₂]₁₃·14H₂O,⁷ with the corresponding lanthanide undecamolybdophosphate analogues being only recently described by Gaunt *et al.*⁸ which includes the Ce^{III} compound, (NH₄)₁₁[Ce^{III}(PMo₁₁O₃₉)₂]₁₆·H₂O. A handful of reports on monolacunary derived lanthanopolyoxometalates can be found in the literature, namely those by Peacock and Weakley (who have isolated the first Ce^{IV} and Ce^{III} complexes, K₁₀[Ce^{IV}(PW₁₁O₃₉)₂]₂₅·H₂O and K₁₂[Ce^{IV}(SiW₁₁O₃₉)₂]₂₈·H₂O),⁹ Griffith *et al.*,⁵ Haraguchi *et al.* (1 : 2 complexes containing Ce^{III}, such as K₁₃[Ce^{III}(GeW₁₁O₃₉)₂]₂₁·H₂O and K₁₅[Ce^{III}(BW₁₁O₃₉)₂]₂₀·H₂O),¹⁰ Pope and collaborators (who have described the formation of polymeric chains of the 1 : 1 anionic complex [Ce^{III}(α-SiW₁₁O₃₉)(H₂O)₃]⁵⁻),¹¹ Mialane *et al.* (KCs₄[Yb(SiW₁₁O₃₉)(H₂O)₂]₂₄·H₂O and K_{0.5}Nd_{0.5}[Nd₂(SiW₁₁O₃₉)(H₂O)₁₁]₁₇·H₂O),¹² Wang *et al.* (K₁₁H₂[Dy(SiMo₁₁O₃₉)₂]₂₉·H₂O),¹³ Shan *et al.* (K₇H₆[Nd(GeMo₁₁O₃₉)₂]₂₇·H₂O),¹⁴ and Naruke and Yamase (K_{8.5}H_{3.5}[Eu(BW₁₁O₃₉)(W₅O₁₈)]₂₅·H₂O).¹⁵ Following our interest in the synthesis of novel lanthanide complexes containing Keggin-type monolacunary polyoxotungstophosphates or polyoxotungstoborates, which can be used in the preparation of thin films¹⁶ and exhibit catalytic activity for oxidative processes,^{4,5} here we describe the first Ce^{IV} complex with two distinct Keggin-type lacunary polyoxotungstophosphates (monolacunary [PW₁₁O₃₉]⁷⁻ and bilacunary [PW₁₀O₃₈]¹¹⁻ anions, both derived from the α-isomer) H₇K₁₀[Ce₂(PW₁₀O₃₈)(PW₁₁O₃₉)₂]₂₈·H₂O, belonging to a new class of lanthanide-substituted polyoxometalates in which the bilacunary [PW₁₀O₃₈]¹¹⁻ anions show an unprecedented bridging structural motif.† Of the five possible bilacunary α-Keggin anion isomers, the one observed here is the 1,4 isomer (C₂ symmetry), resulting from the removal of two W=O groups in the 1 and 4 positions, respectively, as defined by Pope and Scully.¹⁷

The structure of the title compound was elucidated by single-crystal X-ray diffraction§ and consists of discrete V-shaped anionic complexes, [Ce₂(PW₁₀O₃₈)(PW₁₁O₃₉)₂]¹⁷⁻, ca. 1.5 × 2.2 nm in size and constructed from bilacunary [PW₁₀O₃₈]¹¹⁻ and monolacunary [PW₁₁O₃₉]⁷⁻ polyanions bridged by Ce^{IV} cations (Fig. 1). Highly symmetric and unusual 1,4-bilacunary [PW₁₀O₃₈]¹¹⁻ species are located at the two-fold axis along the *c* direction, leading to two lacunae formed by the O(45), O(47), O(57) and O(58) atoms and

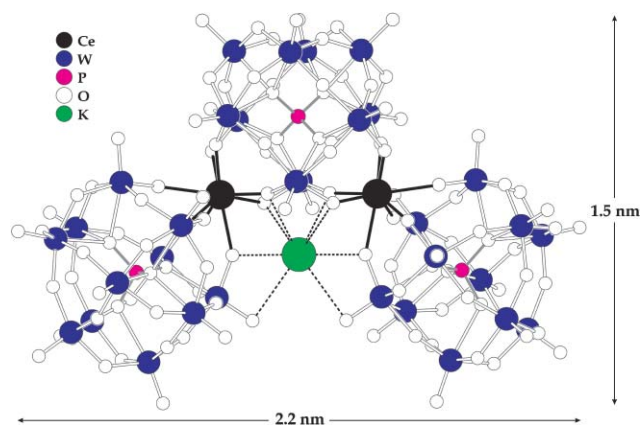


Fig. 1 Molecular structure of H₇K₁₀[Ce₂(PW₁₀O₃₈)(PW₁₁O₃₉)₂]₂₈·H₂O.

occupied by the Ce^{IV} cations with Ce(1)⋯Ce(1)ⁱ of ca. 6.10 Å [symmetry code: (i) -*x*, *y*, 1.5 - *z*] (Fig. 2). The residual available coordination sites of these cations establish a physical connection with a crystallographically independent monolacunary [PW₁₁O₃₉]⁷⁻ polyanion [via the O(19), O(28), O(32) and O(39) atoms], resulting in a {CeO₈} coordination environment, best described as a distorted bicapped trigonal prism. In Fig. 2, Ce–O(19) and Ce–O(39) bonds occupy the capped positions and the Ce–O bond lengths are within the 2.28(3) to 2.44(3) Å range. The PO₄ tetrahedra in the polyanions are almost regular with the P–O and O–P–O bond lengths and angles within the expected ranges [1.53(3)–1.57(3) Å and 1.55(3)–1.57(3) Å, 107.3(16)–111.4(16)° and 108(2)–113(2)° for the mono- and bilacunary polyanions, respectively]. Despite the observed bond ranges for W–O_a [2.31(3)–2.48(3) and 2.36(3)–2.48(3) Å], W–O_{b/c} [1.76(3)–2.02(3) and 1.77(4)–2.05(3) Å]

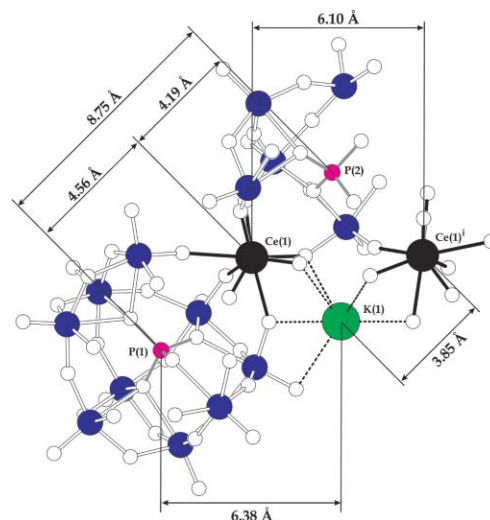


Fig. 2 Geometry of the molecular unit. Colour scheme as in Fig. 1. Distances not shown: K(1)⋯Ce(1) 3.85 Å; P(1)⋯P(1)ⁱ 12.69 Å. Symmetry code used to generate equivalent atoms: (i) -*x*, *y*, 1.5 - *z*.

† Electronic supplementary information (ESI) available: single-crystal X-ray diffraction: details on crystal structure solution and refinement. See <http://www.rsc.org/suppdata/cc/b4/b409034a/>

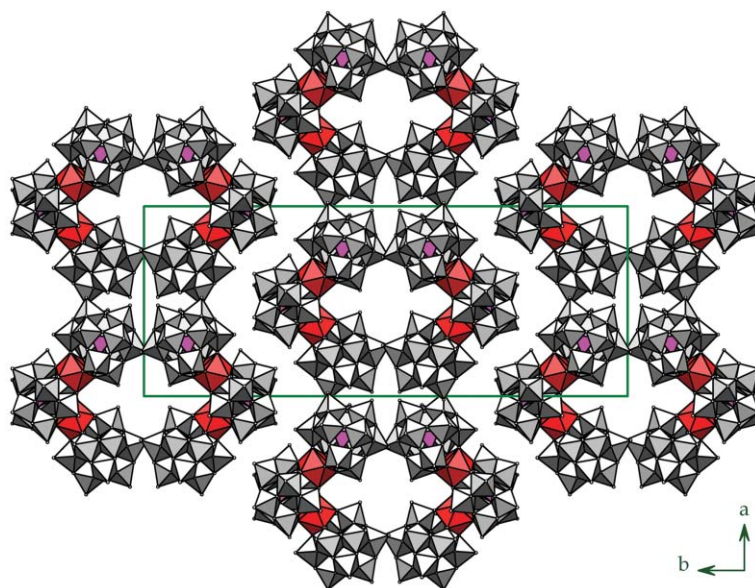


Fig. 3 Crystal packing. WO_6 polyhedra are represented in grey and CeO_8 polyhedra in red.

and $\text{W}-\text{O}_i$ [1.65(3)–1.76(3) and 1.70(3)–1.75(3) Å for the mono- and bilacunary polyanions, respectively] of the $\{\text{WO}_6\}$ octahedra being approximately identical to the expected values, the heterogeneity of bond lengths seems to indicate a generalised distortion of the $[\text{Ce}_2(\text{PW}_{10}\text{O}_{38})(\text{PW}_{11}\text{O}_{39})_2]^{17-}$ anionic complex which is, consequently, reflected in the previously described coordination environment of the Ce^{IV} metal centres. Interestingly, the complex has an internal molecular cavity occupied by a K^+ cation with $\text{K}(1)\cdots\text{O}$ interactions ranging from 2.59(3) to 3.24(3) Å. The cation can simultaneously stabilise the complex and contribute to the observed distortions. Further geometric details of the complex are given in Fig. 2.

Individual complexes close-pack in the ab plane, with the steric hindrance imposed by the unusual V-shape leading to several types of channels filled with crystallisation water molecules (Fig. 3). These are, in turn, involved in a complex and extended hydrogen-bonding network (not shown).

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Notes and references

‡ *Synthesis*: Reagents were readily available from commercial sources and were used as received without further purification. Phosphotungstic acid hydrate (4.3 g, 1.5 mmol) was dissolved in hot water (*ca.* 5 cm^3) followed by the addition of an aqueous solution of $(\text{NH}_4)_2[\text{Ce}(\text{NO}_3)_6]$ (0.41 g, 0.75 mmol in *ca.* 2 cm^3 of water). A concentrated solution of potassium acetate (5.0 g, 50 mmol in *ca.* 5 cm^3 of water; pH adjusted to *ca.* 7 with acetic acid) was added dropwise with vigorous magnetic stirring. The mixture was filtered and the resulting aqueous solution refrigerated. The precipitate formed after one week was manually separated and identified as $\text{K}_{10}[\text{Ce}^{\text{IV}}(\text{PW}_{11}\text{O}_{39})_2]\cdot 8\text{H}_2\text{O}$ (1.87 g; 41% yield) according to Haraguchi *et al.*¹⁰ [Elemental composition (in %) calculated: K 6.48, P 1.03, W 67.08, Ce 2.32. Found (in %): K 6.97, P 1.01, W 66.40, Ce 2.56. Selected FT-IR data (in cm^{-1}): $\nu(\text{W}-\text{O})$ at 956, $\nu(\text{W}-\text{O}-\text{W})$ at 892, 813 and 746, and $\nu(\text{P}-\text{O})$ at 1105 and 1051.] Yellow crystals of the title compound suitable for X-ray measurements were obtained by slow evaporation of the aqueous solution over several weeks (0.35 g, 10% yield). Elemental composition (in %) calculated (based on crystallographic data): K 4.34, P 1.03, W 65.26, Ce 3.11. Found (in %): K 4.43, P 0.5, W 65.00, Ce 2.73. Selected FT-IR data (from KBr disk; Aldrich 99%+): $\nu(\text{W}-\text{O})$ at 953, $\nu(\text{W}-\text{O}-\text{W})$ at 889, 827, 764 and 721 cm^{-1} , and $\nu(\text{P}-\text{O})$ at 1105, 1090, 1051 and 1022 cm^{-1} .

§ *Crystal data*: $\text{H}_{63}\text{Ce}_2\text{K}_{10}\text{O}_{144}\text{P}_3\text{W}_{32}$ (including the 63 H-atoms from the charge balancing H^+ and crystallisation water molecules), $M = 9014.85$, monoclinic, space group $C2/c$, $Z = 4$, $a = 21.035(4)$ Å, $b = 53.449(16)$ Å, $c = 12.472(3)$ Å, $\beta = 91.67(3)^\circ$, $V = 14016(6)$ Å³, $\mu(\text{Mo}-\text{K}\alpha) = 27.221$ mm^{-1} , $D_c = 4.272$ g cm^{-3} . Of a total of 21642 reflections collected, 6872 were independent ($R_{\text{int}} = 0.0738$). Final $R1 = 0.0752$ [$I > 2\sigma(I)$] and

$wR2 = 0.1723$ (all data). Data completeness to $\theta = 20.58^\circ$, 96.5%. Details for crystal structure solution and refinement are given as Supporting Information.† CCDC 243257. See <http://www.rsc.org/suppdata/cc/b4/b409034a/> for crystallographic data in .cif or other electronic format.

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