Combining two-directional synthesis and tandem reactions: desymmetrisation by intramolecular cycloaddition/triazoline fragmentation[†]‡

Martin Rejzek,^a Robert A. Stockman,^{*a} Jan H. van Maarseveen^b and David L. Hughes^c

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A tandem azide formation/intramolecular cycloaddition/ triazoline fragmentation/Michael addition, which results in a non-symmetrical quinolizidine from an acyclic symmetrical precursor, is presented.

The strategies of two-directional synthesis¹ and tandem reactions² both offer the potential to substantially reduce the number of operations required to synthesise complex molecules. As part of an on-going programme studying the potential of combining two-directional synthesis and tandem reactions,³ we herein disclose a remarkable self-desymmetrising cascade reaction.

Two-directional synthesis intrinsically generates symmetrical products. By using two-directional synthesis as a means of forming simple symmetrical functionalised chains, then using a tandem reaction to "fold" these chains, creating rings and stereocentres, we have the potential to form complex polycyclic compounds in just a few steps. Our recent syntheses of perhydrohistrionicotoxin⁴ and

^aSchool of Chemical Sciences and Pharmacy, University of East Anglia, Norwich, UK NR4 7TJ. E-mail: r.stockman@uea.ac.uk;
Fax: 44 (0)1603 592005; Tel: +44 (0) 1603 593890
^bInstitute of Molecular Chemistry, University of Amsterdam, Niewe Achteragracht 129, 1018 WS Amsterdam, The Netherlands.
E-mail: jvm@science.uva.nl; Fax: (+31) (0) 20 5255670; Tel: (+31) (0)20 5255671
^cX-Ray Crystallography Centre, School of Chemical Sciences and Pharmacy, University of East Anglia, Norwich, UK NR4 7TJ
† Part 7 in our series concerning two-directional synthesis and tandem

reactions.

‡ This paper is dedicated to the memory of Mark Jones.

hippodamine⁵ exemplify this approach to the synthesis of complex alkaloids.

During an investigation related to our work on hippodamine, we wished to synthesise symmetrical azide 3, such that we could investigate a Staudinger-type azide reduction/tandem Michael cyclisation. In order to access azide 3, we started with the symmetrical ketone 1 (Scheme 1), which is available in 4 steps from 1,3-dithiane, or in 5 steps from ethyl formate.⁶ Reduction of the ketone with sodium borohydride followed by conversion of the resulting pseudo-C2 symmetric alcohol to mesylate 2 proceeded uneventfully in 96% yield over two steps. Reaction of mesylate 2 with sodium azide in DMF at 50 °C was expected to result in the formation of azide 3. TLC showed one clear product, with a further amount of baseline material, which we considered was possibly due to azide overaddition/decomposition. However, upon inspection of the crude ¹H NMR, it was quickly obvious that something far more complicated than azide 3 had been formed in the reaction pot. Indeed, after purification by column chromatography, it was found that the sole isolable product of the reaction was quinolizidine derivative 5 (52%). Key observations which led us to assign the structure of bicyclic diazo compound 5 were the obvious asymmetry present in the ¹H and ¹³C spectra (including two ester carbonyl peaks in the ¹³C NMR spectrum), and the characteristic diazo stretch in the IR at 2150 cm^{-1} . High resolution mass-spectral data confirmed the molecular mass of 338.2075 (M + H). However, in order to further confirm the stereochemistry of our product, we decided it would be best to make a crystalline derivative suitable for X-ray structure



Scheme 1 Two-directional synthesis and tandem reaction of azide 3.



Scheme 2 Proposed mechanism for the formation of 5.



Fig. 1 X-ray structure of hydrazone 6, formed by hydrogenation of 5.

determination. Thus upon exposure to standard hydrogenation conditions (10% Pd/C, 1 atm hydrogen), diazo compound **5** was quantitatively converted to the crystalline hydrazone **6**, whose X-ray structure⁷ is shown in Fig. 1.

Study of the X-ray structure of 6, combined with the isolation of a small amount of triazoline 4 from the reaction mixture when stopped early, has allowed us to postulate a mechanism for this tandem reaction sequence, which is presented in Scheme 2. Thus, [3 + 2] cycloaddition of azide 3 with one of the enoate functionalities results in the thermodynamically most favoured cvcloadduct 4. 1,4-Prototopic shift within the triazoline of 4, followed by fragmentation to yield diazo 2,6-disubstituted piperidine 7 and subsequent Michael-type ring closure is the most satisfactory mechanism. Although the triazoline fragmentation⁸ to a piperidine is precedented under basic conditions,⁹ lending weight to this mechanistic proposal, we were unable to detect quantities of the intermediate piperidine 7. However, we were able to detect these types of piperidine intermediate in our hippodamine synthesis,⁵ which used a similar Michael-type ring closure to a 4,6-disubstituted quinolizidine. It could also be envisaged that the formation of 5 from 4 could be achieved through a concerted enetype mechanism, with the tertiary nitrogen and the proton α to the ester on the triazoline being delivered to either end of the electron deficient alkene, with concomitant triazoline fragmentation.

In conclusion, the power of combining two-directional synthesis and tandem reactions has provided a short and efficient entry into a non-symmetrical 4,6-disubstituted quinolizidine skeleton. Further studies into the use of the tandem azide formation/[3 + 2]/fragmentation reaction are on-going and our results will be reported in due course.

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Base induced equilibration of triazolidine and diazoester was found to be facile by Herdeis.

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