

3-Hydroxypyrrolidines from epoxysulfonamides and dimethylsulfoxonium methylide†

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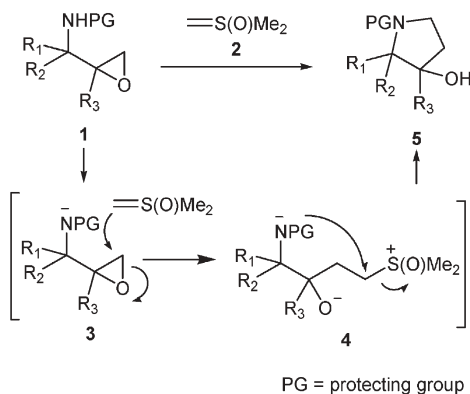
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N-Tosyl-protected 3-hydroxypyrrolidines are prepared by reaction of dimethylsulfoxonium methylide with readily available epoxysulfonamides.

The 3-hydroxypyrrolidine motif **5** is found in a range of naturally occurring bioactive alkaloids,¹ pharmaceuticals² and drug intermediates,³ and a number of synthetic approaches to 3-hydroxypyrrolidines have been developed.^{3,4} In connection with our interest in the reactions of sulfur ylides with three-membered heterocycles,⁵ and stimulated by a report in 2004 by Borhan and co-workers concerning the synthesis of 3-hydroxytetrahydrofurans from epoxy alcohols using dimethylsulfoxonium methylide **2**,⁶ we considered whether 3-hydroxypyrrolidines **5** could be obtained from aminoepoxides **1** using ylide **2** (Scheme 1).

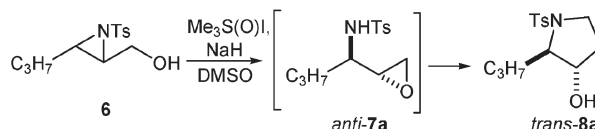


Scheme 1 3-Hydroxypyrrolidines **5** from aminoepoxides **1**.

For the chemistry shown in Scheme 1 to succeed, regioselective ring-opening of the epoxide by the ylide **2** should be followed by 5-*exo-tet* cyclisation in preference to oxetane⁸ formation. It was anticipated that a suitably acidifying *N*-protecting group (e.g. PG = RSO₂) would assist pyrrolidine formation, because the amino functionality would then likely be deprotonated under the reaction conditions⁹ (as shown in intermediates **3** and **4**). Potential complications¹⁰ due to aziridine formation from intermediate **3** by aza-Payne rearrangement should be avoided by using

epoxysulfonamides **1** (PG = RSO₂), since the latter are commonly accessed by base-induced aza-Payne rearrangement of *N*-tosyl 2-aziridinemethanols;¹¹ another synthetically useful route to epoxysulfonamides **1** (PG = RSO₂) proceeds by epoxidation of allylic sulfonamides.¹²

Direct application of Borhan's conditions [Me₃S(O)I (10 equiv.), NaH (10 equiv.), DMSO, 85 °C, 24 h]⁶ to 2-aziridinemethanol **6**¹³ gave the desired 3-hydroxypyrrolidine **8a** (44%), likely by way of (deprotonated) epoxysulfonamide **7a** from *in situ* aza-Payne rearrangement (Scheme 2). However, the potential restriction to using 2-aziridinemethanols as starting materials, together with the requirement for a large excess of the ylide **2**¹⁴ and prolonged reaction time, led us to focus on optimising conditions for 3-hydroxypyrrolidine synthesis from epoxysulfonamides.^{15,16}



Scheme 2 3-Hydroxypyrrolidine **8a** from aziridinemethanol **6**.

Epoxysulfonamide **7a**† (0.1 M in DMSO) was completely consumed within 70 min following reaction with ylide **2** [3 equiv., generated from Me₃S(O)I and NaH] at 80 °C, however 3-hydroxypyrrolidine **8a** was obtained in only 30% yield, with no other products being isolated. Generating ylide **2** from Me₃S(O)I (3 equiv.) and *n*-BuLi (3 equiv.) in THF proved more encouraging, giving 3-hydroxypyrrolidine **7a** in 60% yield after 16 h at rt (Table 1, entry 1); reduced reaction times at rt gave lower yields of **8a**, with starting epoxide **7a** being recovered. Lowering the concentration of epoxide **7a** to 0.02 M slightly reduced the efficiency of the reaction (entry 2). A modest improvement in the yield of **7a** to 65% was obtained by refluxing the reaction mixture for 70 min (entry 3).¹⁷ The yield of **8a** fell when lowering the amount of ylide **2** (to 2 equiv.) either directly (51%, entry 4), or by deprotonating epoxide **7a** with NaH (1 equiv.) first (52%). Using NaHMDS as the base also had a detrimental effect on the yield of **8a**. However, switching solvent to DMPU gave a significant increase in yield of **8a** (87%, entry 5) and, more usefully, this improvement was also observed when using DMPU as an additive (up to 20 equiv.) in THF (entries 6–9). Under the latter conditions, 2 h at reflux was optimal (70 min or 3 h gave slightly reduced yields of **8a**).

A series of epoxysulfonamides **7**, prepared in two steps from the corresponding allylic alcohols by Sharpless aziridination¹³ followed by aza-Payne rearrangement¹¹ [KH (4 equiv.), THF, –78 °C to 0 °C, 2 h],† were then subjected to the optimised conditions

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† Electronic supplementary information (ESI) available: Experimental procedures and characterisation data for all new epoxysulfonamides and 3-hydroxypyrrolidines, and ¹H and ¹³C NMR spectra for 3-hydroxypyrrolidines. See DOI: 10.1039/b606583j

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Table 1 Optimisation of 3-hydroxypyrrolidine **8a** synthesis from epoxysulfonamide **7a**

Entry ^a	DMPU (equiv.)	Temperature/°C	Time/h	Yield (%)
1	—	rt	16	60
2 ^b	—	rt	16	50
3	—	Reflux	1.2	65
4 ^c	—	Reflux	1.2	51
5	Neat	80 °C	2	87
6	5	Reflux	2	69
7	10	Reflux	2	75
8	15	Reflux	2	83
9	20	Reflux	2	86

^a 0.1 M in epoxide **7a** with *n*-BuLi (3.3 equiv.) and Me₃S(O)I (3 equiv.) used unless indicated otherwise. ^b 0.02 M in epoxide **7a**. ^c *n*-BuLi (2.3 equiv.) and Me₃S(O)I (2 equiv.) used.

developed above (Table 1, entry 9) to give the corresponding 3-hydroxypyrrolidines **8** in 72–88% yield (Table 2).[‡]

Either *trans*- or *cis*-2-substituted-3-hydroxypyrrolidines **8** could be prepared in good yield, starting from the corresponding *anti*- or *syn*-epoxysulfonamides **7** (entries 1–4). The simple 3-hydroxypyrrolidine **8c** (entry 5), 2-aryl-3-hydroxypyrrolidines **8d** and **8e** (entries 6 and 7), tertiary alcohol-containing pyrrolidine **8f** (entry 8) and 2,2-disubstituted-3-hydroxypyrrolidine **8g** (entry 9) were all accessible using this methodology. The relative stereochemistries of 3-hydroxypyrrolidines **8** were generally determined by NOE experiments. In the case of 2-aryl-3-hydroxypyrrolidine **8e**, the structure was supported by X-ray crystallographic analysis (Fig. 1).¹⁸

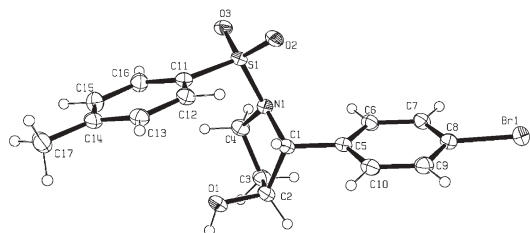


Fig. 1 X-Ray structure of 3-hydroxypyrrolidine **8e** with thermal ellipsoids at the 40% probability level.

Both spiro- and *cis*-fused hydroxypyrrolidines **10a** and *cis*-**10b** could be made using this chemistry (Table 3, entries 1 and 2). Interestingly, the more strained *trans*-fused [4.3.0] system *trans*-**10b** was also successfully generated (entry 3). Epoxysulfonamides **9a** and *syn*-**9b** were prepared by epoxidation (the latter in a highly diastereoselective manner)¹² of the corresponding allylic sulfonamides.†

It has also been found possible to extend the methodology to a 2,3-disubstituted epoxide, when one of the substituents supports the ring-opening process. Thus, 2,3,4-trisubstituted pyrrolidine **12** was formed in 77% yield from epoxysulfonamide **11**† (Scheme 3). The relative stereochemistry was assigned from NOE experiments.

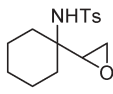
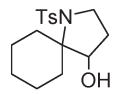
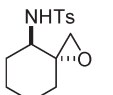
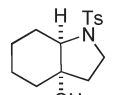
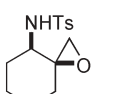
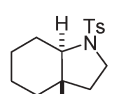
Table 2 3-Hydroxypyrrolidines **8** from epoxysulfonamides **7**

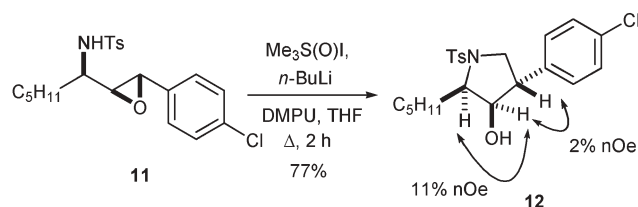
Entry	Epoxide 7	Pyrrolidine 8	Yield (%)
1			86
2			83
3			74
4			76
5			72
6			88
7			82
8			80
9			72

In summary, we have established a process of useful generality for the conversion of epoxysulfonamides to stereodefined 3-hydroxypyrrolidines. The method uses readily available reagents and occurs under experimentally straightforward conditions. Additional studies in the area of epoxysulfonamides and sulfonium ylides are currently underway.

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Table 3 Spiro- and fused-hydroxypyrrolidines **10** from epoxysulfonamides **9**

Entry	Epoxide 9	Pyrrolidine 10	Yield (%)
1			81
2			69
3			66



Scheme 3 Preparation and determination of stereochemistry of trisubstituted hydroxypyrrolidine **12**.

Notes and references

§ *Typical procedure for synthesis of 3-hydroxypyrrolidines from epoxysulfonamides:* *n*-BuLi (1.6 M in hexanes 0.38 mL, 0.61 mmol) was added dropwise to a stirred suspension of Me₃S(O)I (123 mg, 0.56 mmol) in THF (1.4 mL) at $-78\text{ }^{\circ}\text{C}$ and stirred at this temperature for 15 min, and then at $0\text{ }^{\circ}\text{C}$ for 15 min. The mixture was re-cooled to $-78\text{ }^{\circ}\text{C}$ and a solution of *anti*-**7a** (0.19 mmol) in THF (0.5 mL) was added dropwise, followed by DMPU (0.45 mL, 3.74 mmol) and the reaction then warmed to rt over 5 min and heated to reflux. After 2 h, 5% aq. NH₄Cl (10 mL) and EtOAc (10 mL) were added and the layers separated. The aqueous layer was extracted with EtOAc (3 \times 20 mL) and the combined organic layers were dried (MgSO₄) and evaporated under reduced pressure. The residue was purified by column chromatography (60% Et₂O in petrol) to give the corresponding *trans*-**8a** (45 mg, 86%) as a colourless oil; *R*_f 0.18 (70% Et₂O in petrol); IR (neat)/cm⁻¹ 3510br, 2960s, 1599s, 1494s, 1336s, 1156s; ¹H NMR (400 MHz) δ 7.74 (d, *J* = 8 Hz, 2H), 7.31 (d, *J* = 8 Hz, 2H), 4.05 (d, *J* = 3 Hz, 1H), 3.49–3.44 (m, 2H), 3.24 (ddd, *J* = 10.5, 9.5, 7 Hz, 1H), 2.41 (s, 3H), 2.06–1.97 (m, 1H), 1.77–1.67 (m, 3H), 1.50–1.33 (m, 2H), 1.26 (br, 1H), 0.94 (t, *J* = 7 Hz, 3H); ¹³C NMR (100 MHz) δ 143.4, 134.2, 129.5, 127.7, 74.8, 69.1, 46.2, 37.3, 32.4, 21.5, 19.5, 14.0; MS *m/z* (CI) 301 (*M* + NH₄⁺, 100), 284 (68), 130 (50), 48 (33), 86 (29), 72 (30); HRMS calcd for C₁₄H₂₅N₂O₃S (*M* + NH₄⁺) 301.1586, found 301.1577.

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- Use of 5 equivalents of ylide **2** gave 3-hydroxypyrrolidine **8a** in 13% yield.
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- During the preparation of this manuscript, the *in situ* process was reported: R. A. Jones, J. M. Schomaker and B. Borhan, Ylide-mediated expansion of 2,3-aziridin-1-ols to substituted pyrrolidines: Application to the syntheses of pyrrolizidine alkaloids, *Abstracts of Papers*, 231st ACS National Meeting, Atlanta, March 26–30, 2006, CHED-568.
- Using Me₃S(O)Cl as a source of ylide **2** under these conditions gave an identical yield of **8a**.
- Crystallographic data for **8e**: C₁₇H₁₈BrNO₃S, *M_r* = 396.30, crystal size 0.04 \times 0.04 \times 0.20 mm, colourless needles, crystal system triclinic, *a* = 7.4856(4), *b* = 9.7383(5), *c* = 11.5200(6) Å, α = 95.766(2), β = 96.244(2), γ = 92.306(3) $^{\circ}$, *V* = 829.49(8) Å³, *Z* = 2, *D_c* = 1.587 mg m⁻³, *F*₀₀₀ = 404, *T* = 150 K, space group *P* $\bar{1}$, *Z* = 2, μ = 2.617 mm⁻¹, 10663 reflections were measured, *R* = 0.0423, *wR* = 0.0507. CCDC 602881. For crystallographic data in CIF or other electronic format see DOI: 10.1039/b606583j.