

## Facile palladium-catalysed synthesis of 1-aryl-1*H*-indazoles from 2-bromobenzaldehydes and arylhydrazines

Chan Sik Cho,<sup>\*a</sup> Dong Kwon Lim,<sup>b</sup> Nam Ho Heo,<sup>b</sup> Tae-Jeong Kim<sup>b</sup> and Sang Chul Shim<sup>\*b</sup>

<sup>a</sup> Research Institute of Industrial Technology, Kyungpook National University, Daegu 702-701, Korea

<sup>b</sup> Department of Industrial Chemistry, College of Engineering, Kyungpook National University, Daegu 702-701, Korea. E-mail: scshim@knu.ac.kr

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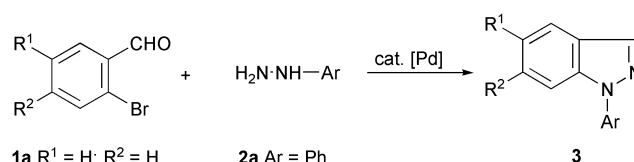
2-Bromobenzaldehydes react with arylhydrazines in toluene at 100 °C in the presence of a catalytic amount of a palladium catalyst and phosphorus chelating ligands such as 1,1'-bis(diphenylphosphino)ferrocene and 1,3-bis(diphenylphosphino)propane along with NaO-*t*-Bu to afford 1-aryl-1*H*-indazoles in good yields.

Palladium-catalysed aromatic carbon–nitrogen bond forming reaction by the cross-coupling of aryl halides (or triflates) and amines has recently had an upsurge as a useful synthetic tool.<sup>1</sup> Various primary, secondary and tertiary amines can be prepared by this protocol and its intramolecular version has been introduced to the synthesis of nitrogen heterocycles as an elegant synthetic alternative.<sup>1</sup> In connection with this report, Song *et al.* have reported a palladium-catalysed intramolecular amination of *N*-aryl-*N'*-(*o*-bromobenzyl)hydrazines leading to 1-aryl-1*H*-indazoles.<sup>2–4</sup> However, the uneasy accessibility of the starting substrates in this method, which are prepared *via* several steps from commercial products,<sup>5</sup> precludes a straightforward approach. This led us to develop an efficient palladium-catalysed synthetic method for 1-aryl-1*H*-indazoles. Herein we report a successful palladium-catalysed synthesis of 1-aryl-1*H*-indazoles from readily available commercial substrates, 2-bromobenzaldehydes and arylhydrazines, *via* an intrinsic intramolecular amination.<sup>6</sup>

The results of several attempted cyclisations of 2-bromobenzaldehyde (**1a**) with phenylhydrazine (**2a**) are listed in Table 1 (Scheme 1). Treatment of equimolar amounts of **1a** and **2a** in toluene in the presence of a catalytic amount of Pd(OAc)<sub>2</sub> (2 mol%) and 1,3-bis(diphenylphosphino)propane (dppp) (3 mol%) along with NaO-*t*-Bu at 100 °C for 15 h afforded 1-phenyl-1*H*-indazole (**3a**) in 84% yield with concomitant formation of hydrazones (< 5%) by condensation between the starting substrates (run 1). The catalytic systems using Pd(OAc)<sub>2</sub> combined with 1,1'-bis(diphenylphosphino)ferrocene (dppf) and PdCl<sub>2</sub> combined with dppp were revealed to be as effective as that using Pd(OAc)<sub>2</sub> combined

with dppp (runs 2, 3). With other phosphorus chelating ligands such as 1,1'-bis(di-*i*-propylphosphino)ferrocene (dipf) and (*S*)-(–)-BINAP, the yield of **3a** was lower than that when dppp and dppf were employed (runs 4, 5), all combined with Pd(OAc)<sub>2</sub>. However, catalytic systems such as Pd(OAc)<sub>2</sub>-PPh<sub>3</sub>, PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> and Pd(PPh<sub>3</sub>)<sub>4</sub> were not effective at all for the present reaction; hydrazone was mainly produced as an identifiable product (runs 6–8). As a result, among the various palladium precursors examined the systems shown in runs 1, 2 and 3 are the choices of preference for the effective formation of **3a**.

Given these results, the reactions of various 2-bromobenzaldehydes **1** and arylhydrazines **2** were screened using three sets of reaction conditions, Pd(OAc)<sub>2</sub>-dppp (condition A), Pd(OAc)<sub>2</sub>-dppf (condition B) and PdCl<sub>2</sub>-dppp (condition C), and several representative results are summarised in Table 2.<sup>†</sup> The reaction of **1b–1d** with **2a** under conditions A–C gave the corresponding 1-phenyl-1*H*-indazoles (**3b–3d**) in 47–79% yield with minimal formation of hydrazones as identifiable side products (runs 1–5). The product yield was not significantly affected by the electronic



- 1a** R<sup>1</sup> = H; R<sup>2</sup> = H      **2a** Ar = Ph  
**1b** R<sup>1</sup> = OMe; R<sup>2</sup> = H      **2b** Ar = 3-MeC<sub>6</sub>H<sub>4</sub>  
**1c** R<sup>1</sup> = OMe; R<sup>2</sup> = OMe      **2c** Ar = 4-MeC<sub>6</sub>H<sub>4</sub>  
**1d** R<sup>1</sup> = F; R<sup>2</sup> = H      **2d** Ar = 4-MeOC<sub>6</sub>H<sub>4</sub>  
**2e** Ar = 2-ClC<sub>6</sub>H<sub>4</sub>  
**2f** Ar = 4-FC<sub>6</sub>H<sub>4</sub>  
**2g** Ar = 2,4-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>  
**2h** Ar = 3-Cl-4-MeC<sub>6</sub>H<sub>3</sub>

Scheme 1

**Table 1** Palladium-catalysed reaction of **1a** with **2a** under several catalytic systems<sup>a</sup>

Run	Palladium catalysts	Ligands (mol%)	Isolated yield (%)
1	Pd(OAc) <sub>2</sub>	dppp (3)	84
2	Pd(OAc) <sub>2</sub>	dppf (3)	84
3	PdCl <sub>2</sub>	dppp (3)	83
4	Pd(OAc) <sub>2</sub>	dipf (3)	65
5	Pd(OAc) <sub>2</sub>	( <i>S</i> )-(–)-BINAP (3)	69
6	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub> (6)	trace
7	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	—	trace
8	Pd(PPh <sub>3</sub> ) <sub>4</sub>	—	trace

<sup>a</sup> Reaction conditions: **1a** (1 mmol), **2a** (1 mmol), palladium catalyst (2 mol%), NaO-*t*-Bu (2 mmol), toluene (10 ml), 100 °C, for 15 h, under argon.

**Table 2** Palladium-catalysed synthesis of **3** from **1** and **2a**<sup>a</sup>

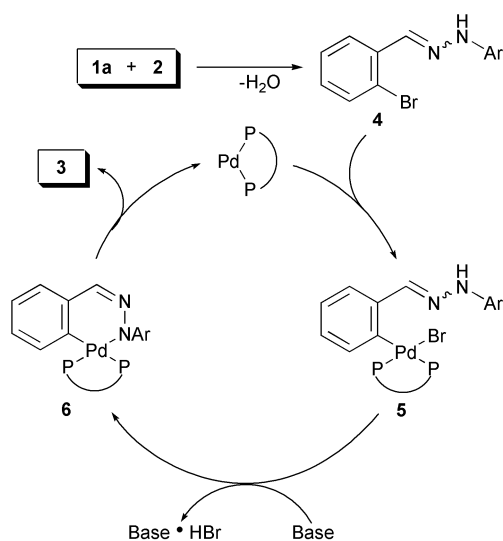
Run	<b>1</b>	<b>2</b>	Conditions <sup>b</sup>	Products	Yield (%)
1	<b>1b</b>	<b>2a</b>	B	<b>3b</b>	79
2	<b>1b</b>	<b>2a</b>	C	<b>3b</b>	47
3	<b>1c</b>	<b>2a</b>	A	<b>3c</b>	65
4	<b>1d</b>	<b>2a</b>	B	<b>3d</b>	77
5	<b>1d</b>	<b>2a</b>	C	<b>3d</b>	51
6	<b>1a</b>	<b>2b</b>	B	<b>3e</b>	79
7	<b>1a</b>	<b>2c</b>	B	<b>3f</b>	55
8	<b>1a</b>	<b>2d</b>	B	<b>3g</b>	61
9	<b>1a</b>	<b>2e</b>	A	<b>3h</b>	78
10	<b>1a</b>	<b>2f</b>	A	<b>3i</b>	73
11	<b>1a</b>	<b>2g</b>	B	<b>3j</b>	86
12	<b>1a</b>	<b>2g</b>	C	<b>3j</b>	60
13	<b>1a</b>	<b>2h</b>	A	<b>3k</b>	86

<sup>a</sup> Reaction conditions: **1** (1 mmol), **2** (1 mmol), palladium catalyst (2 mol%), chelating ligand (3 mol%), NaO-*t*-Bu (2 mmol), toluene (10 ml), 100 °C, for 15 h, under argon. <sup>b</sup> A: Pd(OAc)<sub>2</sub>-dppp; B: Pd(OAc)<sub>2</sub>-dppf; C: PdCl<sub>2</sub>-dppp.

nature of the substituent on **1**. In contrast to the reaction between **1a** and **2a**, the product yield varied according to the employed reaction conditions (runs 1,2 and 4,5). From the reactions between **1a** and various arylhydrazines (**2b–2h**), the corresponding 1-aryl-1*H*-indazoles (**3e–3k**) were also produced in good yields irrespective of the examined functional groups on the aromatic ring of **2** (runs 6–13). Comparing the result of runs 11 and 12 with that of two sets of reactions shown in runs (1 and 2) and (4 and 5), condition B was generally superior to condition C for the formation of 1-aryl-1*H*-indazoles.

A plausible reaction pathway is depicted in Scheme 2. Oxidative addition of the carbon–bromide bond of hydrazone **4**, initially formed *in situ* between **1a** and **2**, to Pd(0) produces an arylpalladium(II) complex **5**. This is followed by intramolecular closure to give a palladacycle **6** which can reductively eliminate to afford 1-aryl-1*H*-indazole **3**.<sup>7,8</sup>

In summary, we have demonstrated that 2-bromobenzaldehydes are cyclised with various arylhydrazines in the presence of a catalytic amount of a palladium catalyst and a phosphorus chelating ligand along with NaO-*t*-Bu to give 1-aryl-1*H*-indazoles in good yields. The present reaction is a straightforward methodology for the synthesis of 1-aryl-1*H*-indazoles from readily available 2-bromobenzaldehydes and arylhydrazines. Similar reaction of alkyl



Scheme 2

2-bromoaryl ketones with arylhydrazines leading to 1,3-disubstituted indazoles is currently under investigation.

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## Notes and references

† *Typical experimental procedure*: a mixture of **1a** (1 mmol), **2a** (1 mmol), Pd(OAc)<sub>2</sub> (0.02 mmol), dppp (0.03 mmol) and NaO-*t*-Bu (2 mmol) in dry toluene (10 ml) was placed in a pressure vessel. The system was flushed with argon and allowed to react at 100 °C for 15 h. The reaction mixture was filtered through a short silica gel column (ethyl acetate) to eliminate inorganic salts. Removal of the solvent left a crude mixture, which was separated by column chromatography (silica gel, ethyl acetate–hexane = 1 : 5) to give **3a** (84%).

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