

# Intramolecular C–H insertions adjacent to sulfur for the diastereoselective synthesis of thienofuranones

Paul S. Skerry,<sup>a</sup> Nigel A. Swain,<sup>a</sup> David C. Harrowven,<sup>\*a</sup> Donald Smyth,<sup>b</sup> Gordon Bruton<sup>c</sup> and Richard C. D. Brown<sup>\*a</sup>

<sup>a</sup> Department of Chemistry, University of Southampton, Southampton, UK SO17 1BJ; Fax: +44 (0)23 8059 6805; Tel: +44 (0)23 8059 4108. E-mail: rcb1@soton.ac.uk; dch2@soton.ac.uk

<sup>b</sup> OSI Pharmaceuticals, Watlington Road, Oxford, UK OX4 6LT

<sup>c</sup> GlaxoSmithKline Pharmaceuticals, New Frontiers Science Park, Harlow Essex, UK CM19 5AW

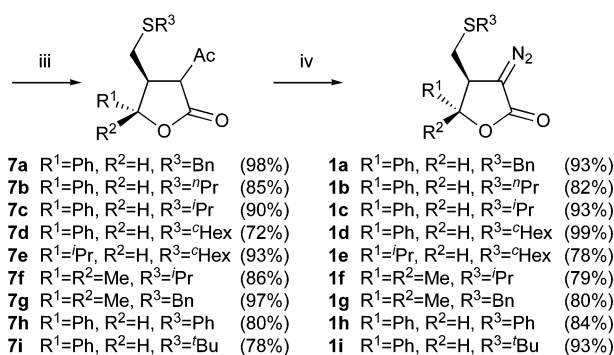
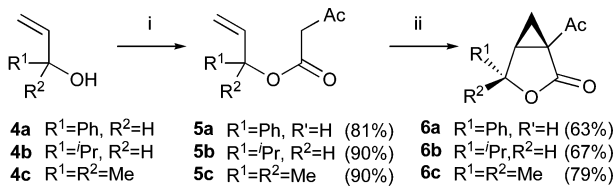
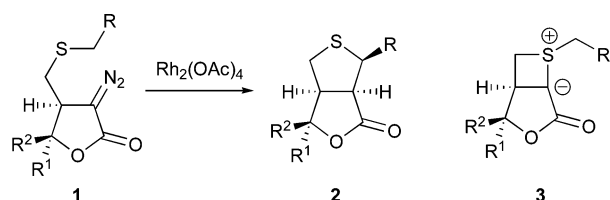
Received (in Cambridge, UK) 20th February 2004, Accepted 26th May 2004

First published as an Advance Article on the web 30th June 2004

A new approach to the diastereoselective synthesis of thienofuranones is described in which an intramolecular 1,5-carbenoid C–H insertion adjacent to sulfur features as a key step.

It is well established that C–H bonds adjacent to oxygen or nitrogen are activated towards insertion by metal carbenoids.<sup>1–3</sup> By contrast, C–H insertions adjacent to sulfur are extremely rare due to the facile capture of the carbenic intermediate by the heteroatom leading to sulfonium ylide formation.<sup>4</sup> Indeed, to the best of our knowledge, the only reported example of C–H insertion adjacent to sulfur was noted as a minor pathway during ylide formation.<sup>5</sup> We reasoned that for substrates such as diazolactone **1**, cyclisation to the strained bicycle **3** was likely to be slow.<sup>6</sup> Consequently, 1,5-insertion to **2** might compete,<sup>7</sup> providing a new route to thienofuranones (Scheme 1).

To test that hypothesis, a series of diazolactones **1a–i** were prepared following the four step sequence outlined in Scheme 2.<sup>8–11</sup> Notably, the key deacylative diazo-transfer reaction, viz. **7**→**1**, was readily accomplished using a one pot procedure involving *in situ* generation of triflyl azide under phase transfer conditions.<sup>11</sup> Each of the diazolactones **1a–i** were then treated with



**Scheme 2** Reagents and Conditions: (i) 2,2,6-trimethyl-4H-1,3-dioxin-4-one, xylenes, 150 °C; (ii) Mn(OAc)<sub>3</sub>, Cu(OAc)<sub>2</sub>, KOAc, AcOH, 75 °C; (iii) R<sup>3</sup>SH, NaHCO<sub>3</sub>, DMSO, 100 °C; (iv) NaN<sub>3</sub>, (Tf)<sub>2</sub>O, <sup>n</sup>Bu<sub>4</sub>NBr, 2 M NaOH–hexane–MeCN (2 : 1 : 1), 0 °C.

2 mol% dirhodium(II) acetate. Pleasingly, for substrates **1a–f** the products of 1,5-insertion **2a–f** were attained in good to high yield (Table 1).<sup>12</sup>

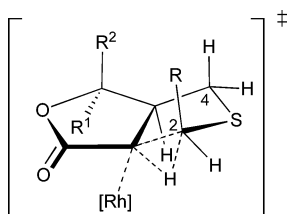
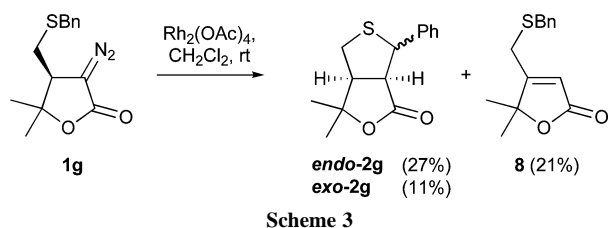
Reactions displayed excellent diastereoselectivity, with **1a** and **1b** yielding only the *endo,exo* products **2a** and **2b** respectively.<sup>13</sup> That preference was significantly reduced in the insertion reaction of diazolactone **1g**, which bears two methyl residues at C5 of the furanone. Indeed, exposure of **1g** to Rh<sub>2</sub>(OAc)<sub>4</sub> gave a complex product mixture from which *endo*-**2g**, *exo*-**2g** and (*5H*)-furanone **8** were each isolated (Scheme 3).

The diminished diastereoselectivity observed for the reaction of **1g**, in comparison to **1a/b**, may be rationalised using the Doyle model for C–H insertion reactions of metal carbenoids.<sup>14</sup> In the transition state leading to *endo* products (see Fig. 1), non-bonding

**Table 1** Rhodium catalysed C–H insertion of diazolactones **1a–f**

Entry	Diazolactone	Thienofuranone	Yield (%) <sup>a</sup>
1			72
2			84
3			75
4			59 (from <b>7d</b> )
5			75
6			51

<sup>a</sup> Isolated yield.

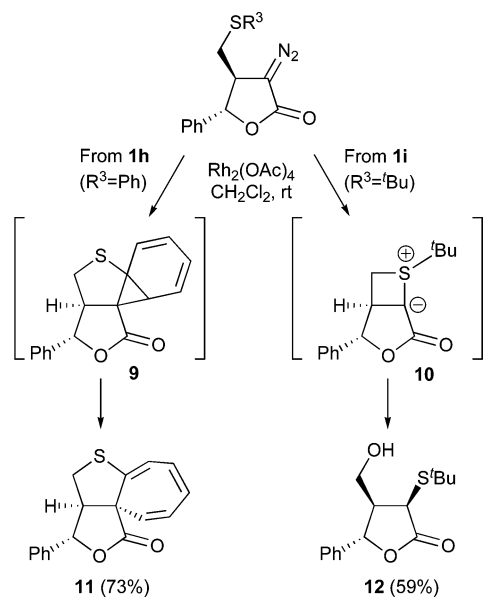


**Fig. 1** Possible transition state model for C–H insertion to afford 2-*endo*-substituted 7-oxa-3-thia-bicyclo[3.3.0]octanes.

interactions between  $R^2$ , the C4  $\beta$ -hydrogen and R increase when  $R^2 = \text{Me}$ . Therefore other pathways, including *exo* C–H insertion and 1,2-insertion, to butenolide **8**, become more important.

To conclude our study we examined two cases where 1,5-insertion was not possible. Thus, exposure of phenyl sulfide **1h** to  $\text{Rh}_2(\text{OAc})_4$  gave cycloheptatriene **11** in high yield via an intramolecular Büchner reaction,<sup>15</sup> while *tert*-butyl sulfide **1i** gave alcohol **12**, presumably via hydrolysis of sulfonium ylide **10** (Scheme 4). No products derived from 1,6-carbenoid C–H insertion were observed in either reaction.

In summary, we have shown that 1,5-carbenoid C–H insertion reactions adjacent to sulfur may proceed efficiently and outpace ylide formation when the latter leads to a strained bicyclic ring system. The method has been used to synthesise a series of



thienofuranones and displays excellent diastereoselectivity. We have also shown that intramolecular Büchner reactions can compete with ylide formation in such cases whereas 1,6-carbenoid C–H insertion reactions do not.

We thank OSI Pharmaceuticals (PSS) and GlaxoSmithKline (NAS) for CASE studentships and the Royal Society for a University Research Fellowship (RCDB). We also acknowledge Dr. M. E. Light and Prof. M. B. Hursthouse for X-ray structural determination.

## Notes and references

- For examples of oxygen activated C–H insertions, see: J. Adams and R. Frenette, *Tetrahedron Lett.*, 1987, **28**, 4773; J. Adams, M. A. Poupart, L. Grenier, C. Schaller, N. Ouimet and R. Frenette, *Tetrahedron Lett.*, 1989, **30**, 1749.
- For examples of nitrogen activated C–H insertions, see: T. C. Smale, *Tetrahedron Lett.*, 1984, **25**, 2913; P. Brown and R. Southgate, *Tetrahedron Lett.*, 1986, **27**, 247.
- For reviews of diazo/carbenoid reactivity, see: M. P. Doyle, *Chem. Rev.*, 1986, **86**, 919; J. Adams and D. M. Spero, *Tetrahedron*, 1991, **47**, 1765–1808; A. Padwa and K. E. Krumpke, *Tetrahedron*, 1992, **48**, 5385; T. Ye and M. A. McKerver, *Chem. Rev.*, 1994, **94**, 1091; M. P. Doyle and D. C. Forbes, *Chem. Rev.*, 1998, **98**, 911.
- A. Padwa and S. F. Hornbuckle, *Chem. Rev.*, 1991, **91**, 263.
- A. Padwa, S. F. Hornbuckle, G. E. Fryxell and P. D. Stull, *J. Org. Chem.*, 1989, **54**, 817.
- For formation of a four-membered cyclic sulfonium ylide, see: H. M. L. Davies and L. V. Crisco, *Tetrahedron Lett.*, 1987, **28**, 371.
- Other examples of cyclisation reactions being biased towards an abnormal course through the inclusion of a five membered ring in the tethering chain include: D. C. Harrowven, N. L'Helias, J. D. Moseley, N. J. Blumire and S. R. Flanagan, *Chem. Commun.*, 2003, 2658; J. W. Dankwardt and L. A. Flippin, *J. Org. Chem.*, 1995, **60**, 2312.
- R. J. Clemens and J. A. Hyatt, *J. Org. Chem.*, 1985, **50**, 2431.
- B. B. Snider and B. A. McCarthy, *Tetrahedron*, 1993, **49**, 9447; M. P. Bertrand, H. O. Mahamat, C. Moustrou and J. M. Surzur, *J. Org. Chem.*, 1989, **54**, 5684.
- C. S. Lee, K. I. Lee and A. D. Hamilton, *Tetrahedron Lett.*, 2001, **42**, 211; C. S. Lee, K. I. Lee and A. D. Hamilton, *Tetrahedron Lett.*, 2001, **42**, 2929.
- N. A. Swain, R. C. D. Brown and G. Bruton, *J. Org. Chem.*, 2004, **69**, 122; R. C. D. Brown, C. J. R. Bataille, G. Bruton, J. D. Hinks and N. A. Swain, *J. Org. Chem.*, 2001, **66**, 6719.
- Procedure for the rhodium-catalysed C–H insertion of compound **1a**: To a solution of diazoactone **1a** (58 mg, 0.18 mmol) in  $\text{CH}_2\text{Cl}_2$  (4 mL) at room temperature was added  $\text{Rh}_2(\text{OAc})_4$  (2 mg, 2 mol%) and the resulting pink, slowly effervescent ( $\text{N}_2$ ) reaction mixture was stirred for 24 h. The reaction was concentrated *in vacuo* to afford crude furofuranone as a pink gum (61 mg). Purification ( $\text{SiO}_2$ ) eluting with 10–20% EtOAc in hexane gave compound **2a** as a white crystalline solid (38 mg, 0.13 mmol, 72%); mp 136–138 °C (EtOAc/hexane);  $\nu_{\text{max}}$  (neat)/ $\text{cm}^{-1}$  1761;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 7.43–7.30 (10H, m, PhH), 5.43 (1H, d,  $J$  6.9, PhCHO–), 4.87 (1H, d,  $J$  9.0, PhCHS–), 3.58 (1H, t,  $J$  9.0,  $-\text{O}_2\text{CCH}-$ ), 3.41 (1H, dddd,  $J$  9.0, 6.9, 5.8, 1.0,  $-\text{SCH}_2\text{CH}-$ ), 3.27 (1H, dd,  $J$  12.6, 5.8,  $-\text{SCHH}-$ ), 3.08 (1H, d,  $J$  12.6,  $-\text{SCHH}-$ );  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 174.5, 139.5, 135.4, 129.1, 128.9, 128.9, 128.7, 128.5, 125.8, 85.2, 56.8, 54.6, 53.9, 37.0;  $m/z$  (CI) 314 ( $[\text{MNH}_4]^+$ , 100%), 297 ( $[\text{MH}]^+$ , 96).
- The stereochemistry of **2a** was determined by X-ray crystallography; details to be published elsewhere.
- M. P. Doyle, L. J. Westrum, W. N. E. Wolthuis, M. M. See, W. P. Boone, V. Bagheri and M. M. Pearson, *J. Am. Chem. Soc.*, 1993, **115**, 958.
- E. Büchner and T. Curtius, *Chem. Ber.*, 1885, **18**, 2374.