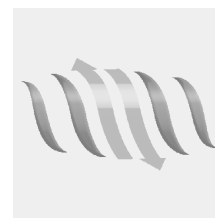


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## Unprecedented formation of an acetamidate-bridged dinuclear platinum(II) terpyridyl complex—correlation of luminescence properties with the crystal forms and dimerization studies in solution

Keith Man-Chung Wong, Nianyong Zhu and Vivian Wing-Wah Yam

*Chem. Commun.*, 2006, 3441–3443 (DOI: 10.1039/b606352g)

The authors inadvertently omitted an important reference. During the course of this study, Rendina and coworkers reported the structure of the same dinuclear acetamido platinum(II) complex that co-crystallized with its mononuclear *N*-acetamido complex. The complex was prepared by a different route through the stepwise reaction of  $[\text{Pt}(\text{terpy})(\text{MeCN})](\text{OTf})_2$  with acetamide, and the dinuclear acetamido platinum(II) complex was found to co-exist in equilibrium with its mononuclear *N*-acetamido complex. The word ‘unprecedented’ should be deleted from the abstract. Nevertheless, the formation of the dinuclear acetamido platinum(II) complex by the slow hydrolysis reaction of  $[\text{Pt}(\text{terpy})(\text{MeCN})](\text{OTf})_2$  in acetone reported in the authors’ work is unexpected and unprecedented. The following communication should be cited:

E. J. Ziolkowski, P. Turner and L. M. Rendina, *Inorg. Chem. Commun.*, 2006, **9**, 53–56.

The authors apologize for the oversight.

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The Royal Society of Chemistry apologises for this error and any consequent inconvenience to authors and readers.

**Additions and corrections can be viewed online by accessing the original article to which they apply.**

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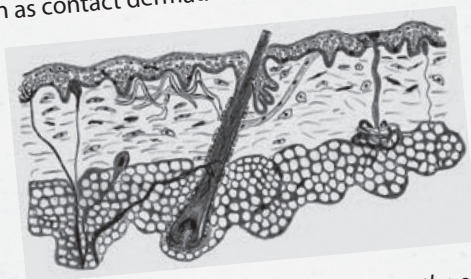
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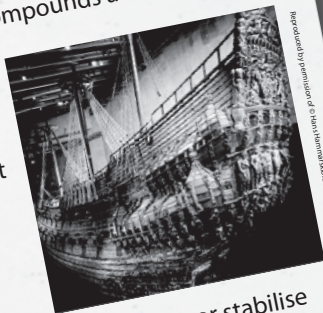
It's our largest body organ, covering about two square metres and weighing around 5 kilogrammes. This review discusses how a deeper understanding of skin biochemistry holds the key to the development of future therapies for skin conditions such as contact dermatitis and skin cancer.



*Biochemistry of human skin—our brain on the outside,*  
D. J. Tobin, *Chem. Soc. Rev.*, 2006, **35**, 52

## Preserving the past

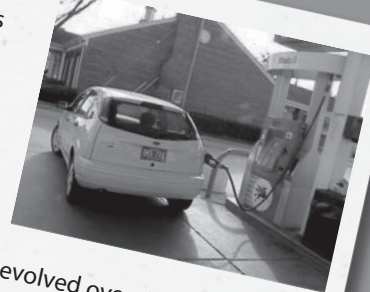
Oxidation of sulfur compounds accumulated in the wood of historical shipwrecks may cause severe acidity in the moist wood, potentially accelerating degradation of recovered archaeological artefacts. Will methods to remove or stabilise sulfur compounds in the wood save the day?



*Sulfur and iron in shipwrecks cause conservation concerns,* Y. Fors and M. Sandström, *Chem. Soc. Rev.*, 2006, **35**, 399

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*Automotive fuels and internal combustion engines: a chemical perspective,* T. J. Wallington *et al.*, *Chem. Soc. Rev.*, 2006, **35**, 335

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