Enantioselective rhodium-catalyzed addition of arylboronic acids to trifluoromethyl ketones[†]

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The catalytic asymmetric 1,2-addition of a series of arylboronic acids to 2,2,2-trifluoroacetophenones is described with high isolated yields (up to 96%) and good enantioselectivities (up to 83% ee) using a rhodium(I)/phosphoramidite catalyst.

Due to their unique properties and unusual reactivities, fluorinated compounds have found extensive application in the fields of materials, pharmaceuticals, and agrochemistry.¹ As a consequence, the annual number of publications and patents concerning fluorinated products has steadily increased over the last three decades.2 Fluorinated compounds bearing a trifluoromethyl substituent represent an interesting sub-class of this type of structures, often providing unique biological activities.^{1,3} In this context, numerous methods for the trifluoromethylation of carbonyl compounds have been reported.⁴ However, enantioselective trifluoromethylation is difficult to achieve and enantiomeric excesses exceeding 50% are rarely reached, except when the substrate is very hindered. 5

An alternative strategy for the synthesis of trifluoromethyl substituted tertiary alcohols (Fig. 1) would be the addition of carbon nucleophiles to trifluoromethyl ketones. However, the formation of quaternary carbons via the addition of carbon nucleophiles to ketones still constitutes a major challenge in synthetic chemistry.⁶ So far, the use of trifluoromethyl ketones as substrates in enantioselective organometallic addition reactions has been limited.⁷ To the best of our knowledge no catalytic enantioselective arylation of fluorinated ketones has been reported so far.

Most of the enantioselective transformations described for the construction of tertiary alcohols from ketones involve the addition of alkyl, alkenyl and arylzinc reagents.8 The lack of readily available zinc reagents severely limits these methods. We envisioned that, for activated ketones, the introduction of aryl

Fig. 1 Trifluoromethyl substituted diarylmethanols.

moieties by asymmetric rhodium-catalyzed addition of arylboron reagents would be a more convenient method.⁹ Arylboronic acids have received increasing attention as arylation reagents because they are shelf-stable, readily available, and compatible with a large variety of functional groups.¹⁰ Rhodium-catalyzed addition of sp²hybridized carbon nucleophiles has made considerable progress during the last decade.¹¹ Our group has demonstrated that phosphoramidites are excellent ligands for the highly enantioselective rhodium-catalyzed conjugate addition of arylboronic acids to enones.¹² Phosphoramidites¹³ comprise a low-cost class of ligands that are easily tunable and therefore highly suitable for ligand variation.14 Recently, a ligand library approach led to the identification of phosphoramidite ligands that provide high enantioselectivity in the rhodium-catalyzed addition of arylboronic acid to imines¹⁵ and good enantioselectivities in their addition to aldehydes.16 A recent publication of Hayashi describing the asymmetric arylation of isatins with a Rh/MeO-MOP catalyst¹⁷ prompted us to divulge our own results in this area.¹⁸ Herein we report the first rhodium-catalyzed addition of arylboronic acids to 2,2,2-trifluoroacetophenones with enantioselectivities up to 83%.

Initial experiments were performed with 2,2,2-trifluoroacetophenone (1a) and 3 equivalents of para-methoxyphenylboronic acid (2a). Variation of solvents identified methyl tert-butyl ether (MTBE) as the most suitable solvent for this reaction in terms of activity and enantioselectivity, although virtually identical results were obtained in acetone. An array of binol-based phosphoramidite ligands was screened, leading to the identification of phosphoramidite ligand L (Fig. 2) as an efficient ligand for this reaction.

A catalyst was generated in situ from 5 mol% of $[(C_2H_4)_2$ -Rh(acac)] and 12.5 mol% of phosphoramidite L. According to ¹⁹F NMR, 60% conversion was obtained after 16 h in refluxing MTBE. Column chromatography afforded the pure product 3a in 50% yield with a promising enantioselectivity of 68% (Table 1, entry 1). As already observed for the addition of arylboronic acids to aldehydes, $16,19$ the reaction is rather sensitive to electronic effects both in substrate and arylboronic acid. Addition of the less nucleophilic para-chlorophenylboronic acid 2b gave the corresponding tertiary alcohol 3b in even lower yield, but with a slightly

Fig. 2 Phosphoramidite (S)-L.

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R	CF ₃ 1a: $R = H$ 1 \mathbf{b} : R = Cl		$[(C2H4)2Rh(acac)]$ (3-5 mol%) HO. χ Ar (S) -L $(2.5$ equiv to Rh CF3 3 equiv ArB(OH) $_2$ (2) К MTBE, reflux, 16h $3a-i$			
Boronic Acids:						
	$2a$: 2 _b 2c: $Ar = p$ -tolyl 2d: $Ar = phenyl$ 2e:		$Ar = p$ -methoxyphenyl $Ar = p$ -chlorophenyl $Ar = m$ -methoxyphenyl	$2f$: $Ar = m$ -tolyl $Ar = m$ -chlorophenyl 2g: $2h$: $Ar = o$ -tolyl 2i: $Ar = 2$ -naphthyl		
Entry	Substrate	R	Boronic acid	Product	Yield ^b $(\%)$	ee c, \overline{d}
1	1a	H	2a	3a	50	68
\overline{c}	1a	Н	2 _b	3 _h	28	72
3 ^e	1 _b	C1	2a	3c	96	68
4^e	1 _b	C1	2c	3d	91	83
5	1 _b	C1	2d	3 _b	90	79
6	1b	C1	2e	3e	94	71
7	1b	C1	2f	3f	91	76
8	1b	C1	2g	3g	52	83
9	1b	C1	2 _h	3 _h	40	50
10	1 _b	C1	2i	3i	69	76

Table 1 Enantioselective rhodium/phosphoramidite-catalyzed asymmetric arylation of 1^a

^a Reactions were performed on 0.178 mmol scale in 2.5 mL of MTBE at reflux for 16 h with 3.0 equiv of arylboronic acid (2) with a catalyst generated in situ from 5 mol% $[(C_2H_4)_2Rh(acac)]$ and 12.5 mol% (S)-L. b Isolated yield after column chromatography. c Determined by chiral HPLC. d The absolute configuration of the products is unknown. ^e Conditions could be optimized to 3 mol% of catalyst and 2 equiv of arylboronic acid without affecting the outcome of the reaction.

higher selectivity of 72% ee (entry 2). To our delight, the activated para-chloro-substituted substrate 1b and arylboronic acid 2a could be converted into the corresponding tertiary alcohol with 96% yield and 68% ee (entry 3).

With these results in hand, the scope of the reaction was investigated. Also the addition of phenylboronic acid (2d) proceeded to full conversion and the alcohol 3b could be obtained in 90% yield with 79% ee (entry 5). Electron donating substituents on the aryl-group of the boronic acid 2 increased the rate of the reaction, whereas the presence of electron-withdrawing substituents has a retarding effect (compare entries 1–8). Both para- and meta-substituted aryls could be introduced with good enantioselectivities and high yields (entries 3, 4, 6, 7, and 8). Addition of ortho-substituted aryls did not proceed to full conversion and results in a considerable drop in enantioselectivity (entry 9). Although a good enantioselectivity of 76% was obtained in the addition of the sterically hindered 2-naphthylboronic acid to 1b, the reaction did not proceed to full conversion and the product 3i was obtained in 69% yield (entry 10).

In the addition of *para*-methoxyphenyl and *meta*-methoxyphenyl groups, high yields were obtained with enantioselectivities of 68 and 71%, respectively (entries 3 and 6). meta-Tolyl substituted alcohol 3f was obtained in high yield with 76% ee (entry 7). The best results were achieved with the addition of a para-tolyl group. Tertiary alcohol 3d was obtained in 91% yield with an enantioselectivity of 83% (entry 4). The conditions could be optimized to 3 mol% of catalyst and 2 equivalents of arylboronic acid without affecting the outcome of the reaction (entries 3 and 4).

In summary, catalytic asymmetric synthesis of trifluoromethyl substituted tertiary alcohols has been realized with good enantioselectivities (up to 83%) and high isolated yields (up to 96%) employing a rhodium/phosphoramidite-catalyst. We are currently directing our efforts towards enhancing the scope and enantioselectivity of this methodology.

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Notes and references

- 1 A. M. Thayer, Chem. Eng. News, 2006, 84, 15; B. E. Smart, J. Fluorine Chem., 2001, 109, 3; Organofluorine compounds. Chemistry and Applications, ed. T. Hiyama, Springer, New York, 2000; Biomedical Frontiers of Fluorine Chemistry, ed. I. Ojima, J. R. McCarthy and J. T. Welch, ACS Editions, Washington DC, 1996; Inventory of Industrial Fluoro-Biochemicals, ed. A. Becker, Eyrolles, Paris, 1996; Organofluorine Chemistry, Principles and Commercial Applications, ed. R. E. Banks, B. E. Smart and J. C. Tatlow, Plenum Press, New York, 1994; Organofluorine Compounds in Medicinal and Biomedical Applications, ed. R. Filler, Y. Kobayashi and L. N. Yagupolskii, Elsevier, Amsterdam, 1993; Fluorine in Bioorganic Chemistry, ed. J. T. Welch and S. Eswarakrishnan, Wiley, New York, 1991; Fluorine-Containing Molecules. Structure, Reactivity, Synthesis, and Applications, ed. J. F. Liebman, A. Greenberg and W. R. Dolbier, Jr., VCH, New York, 1988.
- 2 H. Schofield, J. Fluorine Chem., 1999, 100, 7; J. A. Wilkinson, Chem. Rev., 1992, 92, 505; D. Seebach, Angew. Chem., Int. Ed. Engl., 1990, 29, 1320.
- 3 For some examples of pharmacophores containing a trifluoromethyl group, see: B. Jiang and Y.-G. Si, $Angew.$ Chem., Int. Ed., 2004, 43, 216; L. M. Jackson and C. J. Hawkey, Drugs, 2000, 59, 1207; M. L. P. Price and W. J. Jorgensen, J. Am. Chem. Soc., 2000, 122, 9455; L. Tan, C.-Y. Chen, R. D. Tillyer, E. J. J. Grabowski and P. J. Reider, Angew. Chem., Int. Ed., 1999, 38, 711; F. Xu, R. A. Reamer, R. D. Tillyer, J. M. Cummins, E. J. J. Grabowski, P. J. Reider, D. B. Collum and J. C. Huffman, J. Am. Chem. Soc., 2000, 122, 11212.
- 4 For a selection of recent examples of trifluoromethylation of carbonyl compounds, see: Y. Chang and C. Cai, Tetrahedron Lett., 2005, 46, 3161; S. Roussel, T. Billard, B. R. Langlois and L. Saint-Jalmes, Chem. Eur. J., 2005, 11, 939; G. K. S. Prakash, J. Hu and G. A. Olah, J. Org. Chem., 2003, 68, 4457; B. R. Langlois and T. Billard, Synthesis, 2003, 185; J. Joubert, S. Roussel, C. Christophe, T. Billard, B. R. Langlois and T. Vidal, Angew. Chem., Int. Ed., 2003, 42, 3133; G. K. S. Prakash, J. Hu and G. A. Olah, Org. Lett., 2003, 5, 3253; S. Roussel, T. Billard, B. R. Langlois and L. Saint-Jalmes, Synthesis, 2004, 2119.
- 5 For a recent review on nucleophilic trifluoromethylation reagents and their stereoselective aspects, see: B. R. Langlois, T. Billard and S. Roussel, J. Fluorine Chem., 2005, 126, 173.
- 6 For reviews on the formation of quaternary carbon centers, see: Quaternary Stereocenters. Challenges and Solutions for Organic Synthesis, ed. J. Christoffers and A. Baro, Wiley-VCH, Weinheim, 2005; B. M. Trost and C. Jiang, Synthesis, 2006, 369; J. Christoffers and A. Baro, Adv. Synth. Catal., 2005, 247, 1473; D. J. Ramón and M. Yus, Curr. Org. Chem., 2004, 8, 149; I. Denissova and L. Barriault, Tetrahedron, 2003, 59, 10105; E. J. Corey and A. Guzman-Perez, Angew. Chem., Int. Ed., 1998, 37, 388.
- 7 For an enantioselective indium-mediated allylation reaction on trifluoromethyl ketones, see: T.-P. Loh, J.-R. Zhou and X.-R. Li, Tetrahedron Lett., 1999, 40, 9333; for a diastereoselective methyl Grignard addition to trifluoromethyl ketone, see: J. M. Andrés, R. Pedrosa and A. Pérez-Encabo, Eur. J. Org. Chem., 2004, 1558; for the enantioselective alkynylation of trifluoromethyl ketoaniline using metal alkynylides with a stoichiometric amount of chiral mediator, see: B. Jiang and Y. Feng, Tetrahedron Lett., 2002, 43, 2975; M. L. P. Price and W. J. Jorgensen, J. Am. Chem. Soc., 2000, 122, 9455; L. Tan,

C.-Y. Chen, R. D. Tillyer, W. J. J. Grabowski and P. J. Reider, Angew. Chem., Int. Ed., 1999, 38, 711.

- 8 For a review on the enantioselective addition of zinc reagents to ketones, see: D. J. Ramón and M. Yus, Angew. Chem., Int. Ed., 2004, 43, 284; for the first example of the catalytic enantioselective addition of diphenylzinc to ketones, see: P. I. Dosa and G. C. Fu, J. Am. Chem. Soc., 1998, 120, 445; for some recent examples of the enantioselective addition of zinc reagents to ketones, see: H. Li and P. J. Walsh, J. Am. Chem. Soc., 2005, 127, 8355; S.-J. Jeon, H. Li and P. J. Walsh, J. Am. Chem. Soc., 2005, 127, 16416; P. G. Cozzi, J. Rudolph, C. Bolm, P.-O. Norrby and C. Tomasisni, J. Org. Chem., 2005, 16, 3341; M. Ni, R. Wang, Z.-J. Han, B. Mao, C.-S. Da, L. Liu and C. Chen, Adv. Synth. Catal., 2005, 347, 1659; H. Li, C. García and P. J. Walsh, Proc. Natl. Acad. Sci. USA, 2004, 5425; S.-J. Jeon, H. Li, C. García, L. K. LaRochelle and P. J. Walsh, J. Org. Chem., 2005, 70, 448; P. G. Cozzi, Angew. Chem., Int. Ed., 2003, 42, 2895; C. García and P. J. Walsh, Org. Lett., 2003, 5, 3641; for the enantioselective addition of a vinylsilane reagent to an activated ketone, see: D. Tomioka, R. Wada, M. Kanai and M. Shibasaki, J. Am. Chem. Soc., 2005, 127, 4138.
- 9 For a review on catalytic asymmetric approaches toward enantiomerically enriched diarylmethanols and diarylmethylamines, see: F. Schmidt, R. T. Stemmler, J. Rudolph and C. Bolm, Chem. Soc. Rev., 2006, 35, 454.
- 10 N. Miyaura and A. Suzuki, Chem. Rev., 1995, 95, 2457; Boronic Acids: Preparation and Applications in Organic Synthesis and Medicine, ed. D. G. Hall, Wiley-VCH, Weinheim, 2005.
- 11 T. Hayashi, Synlett, 2001, 879; T. Hayashi and K. Yamasaki, Chem. Rev., 2003, 103, 169; Modern Rhodium-Catalyzed Organic Reactions, ed. P. A. Evans, Wiley-VCH Verlag, Weinheim, 2005.
- J.-G. Boiteau, R. Imbos, A. J. Minnaard and B. L. Feringa, Org. Lett., 2003, 5, 681; also see: Org. Lett., 2003, 5, 1385; J.-G. Boiteau, A. J. Minnaard and B. L. Feringa, J. Org. Chem., 2003, 68, 9481; A. Duursma, R. Hoen, J. Schuppan, R. Hulst, A. J. Minnaard and B. L. Feringa, Org. Lett., 2003, 5, 3111; R. B. C. Jagt, J. G. de Vries, B. L. Feringa and A. J. Minnaard, Org. Lett., 2005, 7, 2433.
- 13 B. L. Feringa, Acc. Chem. Res., 2000, 33, 346.
- 14 H. Bernsmann, M. van den Berg, R. Hoen, A. J. Minnaard, G. Mehler, M. T. Reetz, J. G. de Vries and B. L. Feringa, J. Org. Chem., 2005, 70, 943; L. Lefort, J. A. F. Boogers, A. H. M. de Vries and J. G. de Vries, Org. Lett., 2004, 6, 1733; A. Duursma, L. Lefort, J. A. F. Boogers, A. H. M. de Vries, J. G. de Vries, A. J. Minnaard and B. L. Feringa, Org. Biomol. Chem., 2004, 2, 1682.
- 15 R. B. C. Jagt, P. Y. Toullec, D. Geerdink, J. G. de Vries, B. L. Feringa and A. J. Minnaard, Angew. Chem., Int. Ed., 2006, 45, 2789.
- 16 R. B. C. Jagt, P. Y. Toullec, J. G. de Vries, B. L. Feringa and A. J. Minnaard, Org. Biomol. Chem., 2006, 4, 773.
- 17 R. Shintani, M. Inoue and T. Hayashi, Angew. Chem., Int. Ed., 2006, 45, 3353.
- 18 P. Y. Toullec, R. B. C. Jagt, J. G. de Vries, B. L. Feringa and A. J. Minnaard, Org. Lett., 2006, 8, 2715.
- 19 M. Sakai, M. Ueda and N. Miyaura, Angew. Chem., Int. Ed., 1998, 37, 3279; C. Moreau, C. Hague, A. S. Weller and C. G. Frost, Tetrahedron Lett., 2001, 42, 6957.