Biosynthesis of Sesquiterpenes

GEOFFREY A. CORDELL

Department of Pharmacognosy and Pharmacology. College of Pharmacy, University of Illinois at the Medical Center. Chicago. Illinois 60612

Received March 11, 1975 (Revised Manuscript Received July 21, 1975)

Contents

Т	Introduction	425
	Biosynthesis of Acyclic Sesquiterpenes	425
	A. Farnesol	425
	B. Juvenile Hormone	428
Ш.	Biosynthesis of Monocyclic Sesquiterpenes	430
	A. Abscisic Acid	430
	B. Fumagillin	431
	C. Germacrene C	431
	D. Dendrolasin	431
	E. trans-y-Monocyclofarnesol	432
IV.	Biosynthesis of Bicyclic Sesquiterpenes	432
	A. Caryophyllene	432
	B. y-Muurolene	432
	C. Petasin	432
	D. Ipomeamarone	433
	E. Carotol	433
	F. Helminthosporal	433
	G. Avocettin	434
	H. Alkaloids of Nuphar	435
	I. Mycophenolic Acid	435
۷.	Biosynthesis of Tricyclic Sesquiterpenes	436
	A. The Illudins, Hirsutic Acid, and the Coriolins	436
	B. Trichothecin and Helicobasidin	437
	C. Picrotoxinin and Dendrobine	441
	D. Sativene	442
	E. Longifolene	443
	F. Santonin	444
	G. Sesquiterpenes of Pogostemon cablin	444
VI.	Biosynthesis of Tetracyclic Sesquiterpenes	444
	A. Gossypol	444
	B. Marasmic Acid	445
VII.	Biosynthesis of Pentacyclic Sesquiterpenes	455
	A. Siccanochromenes	455
	Summary and Conclusions	455
IX.	Addendum	455
Χ.	References	457

J. Introduction

Sesquiterpenes are a diverse group of terpenoid compounds containing a 15-carbon skeleton. Their structure diversification and pharmacologic activity make this group of exceptional interest.

Biosynthetically, their derivation along the lines of the Ruzicka hypothesis^{1,2} is assured. Further elaboration of the primary precursor farnesol is frequent, and this skeletal profusity and the accompanying stereochemical problems have been the subject of substantial speculation,^{3–6} albeit with the minimum of experimental support. Discussion of biogenesis per se is kept to a minimum in this work.

The biosynthetic work in this field has been the subject of only selective review,⁷⁻³⁸ and this contribution attempts to partially fill this void. Organization of this review is based on the number

of carbocyclic rings contained in the sesquiterpene under consideration as summarized above. Farnesol, a precursor of all sesquiterpenes, is therefore treated first.

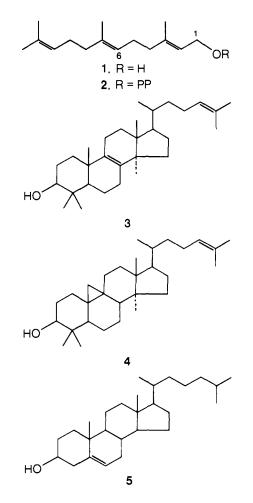
The literature is reviewed to the end of November 1974.

II. Biosynthesis of Acyclic Sesquiterpenes

A. Farnesol (1)

The biosynthesis of farnesol (1) is one of the most elegantly investigated areas in the whole of biosynthesis, and space will permit only a cursory examination of this area.

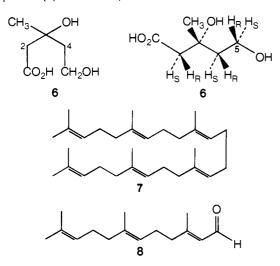
The details of the formation of farnesol (1) are important because of the subsequent tail-to-tail dimerization of farnesyl pyrophosphate (2), leading to the key triterpenes lanosterol (3) (in mammals) and cycloartenol (4) (in plants). Indeed much of the information on farnesol (1) biosynthesis is derived from information of the biosynthesis of cholesterol (5), with the clear implication that the farnesyl units are utilized in the biosynthetic scheme.



In order to both understand and appreciate the subsequent work on the formation of 1 and of the sesquiterpenes from 1, we must reexamine some of the early work on the steroid nucleus.

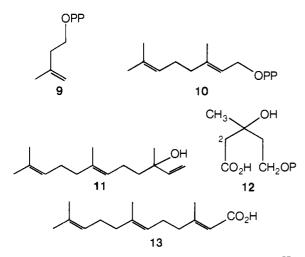
The discovery of mevalonic acid (6) by Folkers³⁹⁻⁴² and Tamura⁴³ and its demonstration as an efficient precursor of cholesterol (5)⁴⁴ in rat liver homogenate gave early experimental support to the biogenetic ideas of Ruzicka and coworkers.² Subsequently, the specific precursor relationship of mevalonic acid (6) to farnesyl pyrophosphate (2) was demonstrated.⁴⁵

At this time there was considerable speculation as to the origin of squalene (7).⁴⁶ However, it was found⁴⁶ that added farnesol



(1) and farnesal (8) considerably inhibited the incorporation of $[1-^{14}C]$ acetate into the unsaponifiable fraction of rat liver homogenate, thus confirming the results of previous workers.^{47,48} These results were followed to their logical conclusion by Lynen⁴⁵ who investigated the formation of squalene (7) from 1. The results confirmed the suspicions of Isler⁴⁶ who regarded the union of two C₁₅ units to be more likely than the union of three C₁₀ units in the biosynthesis of squalene. Others were less successful with C₁₅ precursors of squalene (7).^{49,50}

Lynen also demonstrated⁵¹ the existence, not only of an isopentenyl pyrophosphatase, but also a farnesylpyrophosphate synthetase which successfully converted [1-¹⁴C]isopentenyl pyrophosphate (9)⁵² into geranyl (10) and farnesyl (2) pyrophosphates. From rat liver Popjak^{53,54} obtained an enzyme system which utilized [2-¹⁴C]mevalonate (6) to produce a number of interesting allylic alcohols, including nerolidol (11) and *trans,trans*-farnesol (1) (60% of total activity). When [2-¹⁴C]-5-phosphomevalonic acid (12)⁵⁵ was used, the level of incorporation in 1 and 11 was raised to 88%.^{54,56}



Also isolated (artifact?) from both the rat liver enzyme⁵⁷ and liver alcohol dehydrogenase (LAD)⁵⁸ systems was *trans,trans-*

farnesoic acid (13), suggested^{57,59} and later disproved⁶⁰ as a precursor of squalene (7).

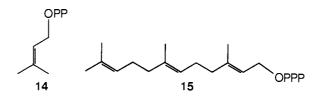
In order to understand the stereochemical formation of 1, we must digress slightly and discuss relevant aspects of the formation of 7. Much of this work was initially due to Bloch.^{61–64}

Later, some of these results were invalidated,^{65–70} and it was conclusively demonstrated that a single hydrogen is stereo-specifically^{71,72} replaced in the central bond of squalene.

Farnesol (1) is therefore produced without oxidation at C-5 of mevalonate (6), and the stereochemical integrity of the 2-carbon of 6 is maintained. Each unit of 6 does, however, lose a hydrogen atom from C-4 on conversion to 1.

In order to investigate the stereospecific formation of the double bonds of squalene (7) (and therefore of *trans,trans*-farnesol (1)), mevalonic acids stereospecifically labeled at the pro-4*R* and pro-4*S* positions were used in the biosynthesis of squalene (7) in rat liver,^{73,74} fungi,^{75,76} and higher plants.⁷⁷⁻⁷⁹ From *pro*-[4*R*-³H]mevalonate (6) essentially all of the tritium was retained in both $1^{76,78}$ and $7,^{74}$ whereas from *pro*-[4*S*-³H]-mevalonate (6) activity was lost. Therefore, in the formation of the trans double bonds, the pro-4*S* proton is specifically lost. Exactly the opposite situation obtains in the formation of rubber,^{78,80,81} where specific retention of the pro-4*S* hydrogen of 6 occurs.

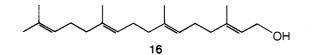
The enzyme isopentenylpyrophosphate isomerase catalyzes the conversion of isopentenyl pyrophosphate (9) to dimethylallyl pyrophosphate (14) (IPP and DMAPP, respectively). Cornforth has recently demonstrated⁸² that the proton added from the medium in this isomerization approaches from the *re* side of the double bond in IPP. This suggests that formation of the double bond in DMAPP takes place by concerted addition–abstraction. Aspects of this and other work by Cornforth and his associates have recently been summarized.⁸³



Further work on the biosynthesis of *trans,trans*-farnesol (1) from mevalonate (6) has concentrated on confirmation of its biosynthetic origin in various organisms and tissues.^{76,78,84–87} In some instances partially purified enzyme systems were obtained which were responsible for the formation of 2 and even for the formation of *trans,trans*-farnesyl triphosphate (15).⁸⁸

Juvenile hormones are probably the best known of the sesquiterpene-type compounds found in insects, and their biosynthesis will be covered in the next section. A number of other sesquiterpenes are also known to occur in insects, and, indeed, these are found to the exclusion of triterpene and steroid metabolites in many cases,^{89,90} apparently because enzymes beyond squalene oxidase are not present or are of extremely low activity.⁹¹ These systems are therefore useful for the isolation of enzyme systems responsible for early reactions in the biosynthetic sequence.

The enzyme responsible for the first step in the biosynthesis of farnesyl pyrophosphate (2) from 6, mevalonic kinase. has been obtained from a number of organisms, $^{92-96}$ including the fly, *Sarcophaga bullata*.⁹⁷ Further work with *S. bullata*⁹⁸ demonstrated that along with *trans,trans*-farnesol (1) and *trans*-nerolidol (11), geranylgeraniol (16) was the most complex metabolite labeled by [2-1⁴C]mevalonate (6). Mevalonic kinase is



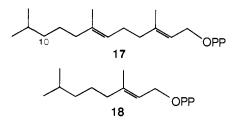
An enzyme system capable of synthesizing **2** was first obtained by Lynen and co-workers.¹⁰¹ This synthetase catalyzed the isomerization of isopentenyl pyrophosphate (**9**) to dimethylallyl pyrophosphate (**14**), the formation of geranyl pyrophosphate (**10**) and the formation of farnesyl pyrophosphate (**2**). From *Micrococcus lysodeikticus*, Bloch's group¹⁰² obtained an enzyme which catalyzed the formation of **2**, **10**, and **16**. An enzyme system from pig liver was found¹⁰³ to catalyze only the synthesis of **2**, indicating that possibly as many as four enzyme systems were involved in these reactions (one isomerase and three synthetases).

Subsequent work¹⁰⁴ demonstrated an inadequacy in the monitoring of geranylpyrophosphate synthetase, and additional work with various systems derived from higher plants and mammalian organs has demonstrated that the same enzyme system (prenyl transferase) is responsible for the synthesis of both **10** and **2**.

Prenyl transferase (farnesylpyrophosphate synthetase) in various degrees of activity has been obtained from a number of sources including pig liver,^{103–105} beef liver,^{85,106} yeast,^{51,107} Andrographis paniculata,¹⁰⁸ Citrus sinensis,¹⁰⁹ Cucurbita pepo,¹¹⁰ Gibberella fujikuroi,¹¹¹ Gossypium hirsutum,¹¹² Iris hollandica,¹¹³ Micrococcus lysodeikticus,^{114–116} Pinus radia-ta,^{109,117} and a Pisum sp.¹¹⁸ The lack of homogeneity of these enzyme systems makes substrate and product specificity highly questionable in this area.

A technique has been described by which many of the intermediates between mevalonate (6) and farnesyl pyrophosphate (1) can be separated chromatographically.¹¹⁹

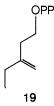
Farnesylpyrophosphate synthetase, as well as catalyzing the reaction of the natural substrates DMAPP (14) and geranyl pyrophosphate (10) with IPP (9). also catalyzes the reaction of a number of other (artificial) substrates. The work of Popjak initiated this important area with the demonstration of the enzymic formation of 10,11-dihydrofarnesyl pyrophosphate (17) from isopentenyl pyrophosphate and 6,7-dihydrogeranyl pyrophosphate (18). Because it appeared that the pyrophosphate unit of



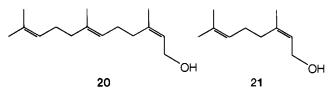
18 was bound to the enzyme, ^{122,123} it was considered that many other similar compounds would act as substrate for the enzyme. The simplest allylic substrate was shown to be DMAPP (**14**), ¹²¹ but several cis and trans higher homologues were also effective. ¹²¹

Independently, a Japanese group,^{124,125} using an enzyme system from pumpkin, confirmed these results and have since extended their work to the higher homologues of isopentenyl pyrophosphate,¹²⁶ geranyl pyrophosphate,¹²⁷ geranylgeranyl pyrophosphate,¹²⁸ and squalene pyrophosphate.¹²⁹ While many changes in the allylic moiety could be made before enzyme specificity limited the reaction, only one other isopentenyl derivative was active as a substrate,¹²⁶ indicating a much narrower specificity for this "half" of the reacting system. Interestingly, from the point of view of juvenile hormone biosynthesis, the second reactive substrate was 3-ethyl-3-butenyl pyrophosphate (**19**).

More recently there has been considerable widespread interest (ref 108, 109, 112, 117, 122, 130, 132) in another isomer of farnesol, namely *cis,trans*-farnesol (20). There had been speculation as to both the origin and subsequent utilization of

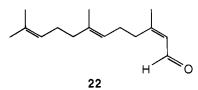


the 2-cis C_{10} and C_{15} pyrophosphates. These isomers are essentially unreactive in further condensation with IPP (**9**),¹¹⁷ yet nerol (**21**) is a key precursor of the cyclic monoterpenes¹³³ and the iridoids,¹³⁴ and 2-*cis*-6-*trans*-farnesol (**20**) is, as we shall see, apparently an important intermediate in the formation of many polycyclic sesquiterpenes.



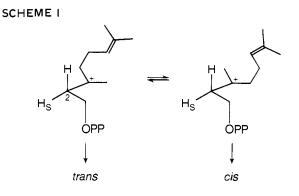
The possible importance of **20** in sesquiterpene biosynthesis was first recognized by Hendrickson³ and extensively elaborated on by Ramage.⁴ The compound was first identified in *Pinus radiata* seedlings,¹¹⁷ and its formation from [1-³H]geranyl pyrophosphate (**10**) was established. Apparently there was no isomerization of [1-³H]-**2**.

In further work, cell-free extracts of orange flavedo¹³⁰ were used to examine the biosynthesis of the farnesols from **10**. The latter compound, derived from $[4-{}^{14}C]$ -IPP (**9**), was a precursor not only of *trans,trans*-farnesol (**1**) but also of *cis,trans*-farnesol (**20**) and the corresponding aldehydes **8** and **22**.¹³⁵ Of the four



products formed, the trans compounds, 1 and 8, were produced more rapidly and in greater quantity than the cis isomers, 20 and 22. Again, isomerization of the farnesyl pyrophosphates was not observed, and a redox scheme was proposed involving the initial formation of both the *cis*- and *trans*-farnesyl pyrophosphates.

More sophisticated work with the farnesylpyrophosphate synthetase system of *Pinus radiata*, ¹⁰⁹ *Citrus sinensis*, ¹⁰⁹ and *Andrographis paniculata*¹⁰⁸ demonstrated that in the formation of both the 2-cis and 2-trans double bonds it was the pro-4*S* proton of **6** which was specifically eliminated.^{108,109} Previous work had shown that the enzyme system has a low specificity with regard to the long-chain part of the allylic substrate, ^{120,121,123,124} and Cori and co-workers suggested that this may permit rotation of any intermediate carbonium ion about the 2,3 bond (Scheme I).



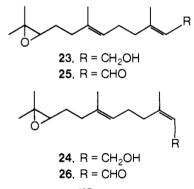
Alternatively, it was suggested ¹⁰⁹ that there may be a specificity in the binding of IPP (**9**) to the enzyme, the double bond being either cis or trans to the alcohol function. Work by Overton, ¹⁰⁸ however, supports a third possibility. When $[5-{}^{3}H_{2},2-{}^{14}C]$ mevalonate (**6**) was used as a precursor, *trans,trans*-farnesol (1) retained all ${}^{3}H$ activity, but *cis,trans*-farnesol (**20**) lost one-sixth of the total ${}^{3}H$ activity, indicating a loss of ${}^{3}H$ activity from C-1 aldehydes as intermediates in the 2,3 isomerization.

Interestingly, in the sequential labeling study of Cori, ¹³⁰ the apparent order of synthesis was *trans,trans*-farnesol (1), *trans,trans*-farnesal (8), *cis,trans*-farnesal (22), and *cis,trans*-farnesol (20).

Overton¹³² has extended this work and determined that the pro-1*S* hydrogen was lost in the oxidation of 1 to 20. It was also demonstrated that a reverse reaction takes place in which 20 is converted to 1, the stereochemistry of this reaction being exactly the opposite of the forward reaction in that the pro-1*R* hydrogen is specifically lost.

trans, trans-farnesol (1)
$$\stackrel{\text{pro-1S lost}}{\underset{\text{pro-1R lost}}{\longleftarrow}} cis, trans-farnesol (20)$$

During an examination of the metabolites of *trans*-10,11epoxyfarnesol (23) in *Helminthosporium sativum*, ¹³⁶ Suzuki and Marumo¹³⁶ found that one of the products was the corresponding cis,trans isomer 24. When $[1-^{2}H_{2}]$ -*trans*,*trans*-10,11-epoxyfarnesol (23) was used as a precursor, 42% of the deuterium was lost at C-1 in 24, indicating the probable intermediacy of the trans (25) and cis (26) aldehydes in the isomerization process. When the trans,trans aldehyde 25 was used as a precursor, it too was incorporated into 24. Similar results were also obtained with $[1-^{2}H_{2}]$ -*trans*,*trans*-farnesol (1).



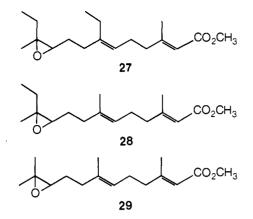
Arigoni and co-workers,¹³⁷ as part of their recent extensive work on sesquiterpene biosynthesis, have demonstrated that alternative routes *not* involving oxidation must also be considered for this isomerization. It was demonstrated, for example, that overall inversion at C-1 of farnesol (1) can occur involving nerolidol (11) as an intermediate. This nonoxidative process affords a *cis,trans*-farnesol (20) with opposite stereochemical labeling at C-1 to that in the original 1. The consequences of this result in terms of evaluating the earlier "stereospecific" losses from C-1 are clear. The factors which influence stereospecific oxidation or an intact inversion at C-1 must be evaluated. Further discussion of the work of Arigoni in this area is deferred to the sections on the tricyclic sesquiterpenes.

Considerable further effort is required in this area, in particular in evaluating and monitoring the enzyme systems involved and their distribution.

B. Juvenile Hormone

Many aspects of the isolation, structure elucidation. and synthesis of the juvenile hormones (JH) have been discussed, ^{138–141} but for the purpose of this review we shall consider only three of these, JH I (27), JH II (28), and JH III (29).

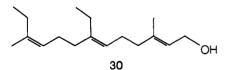
The simplest of these compounds and the one most closely related to farnesol (1) is JH III, 10,11-epoxy methyl farnesoate (29), isolated from organ cultures of the tobacco hornworm moth,



Manduca sexta, ¹⁴² and from the grasshopper, *Schistocera vaga*. ¹⁴³ Preliminary work with simple precursors indicated that methionine labeled only the ester methyl group and that [2-¹⁴C]acetate gave only a low incorporation, the specificity of which could not be determined. ¹⁴³

Similar problems had been reported earlier by Schmialek¹⁴⁴ who found that mevalonate labeled the sesquiterpenes rather than JH I (27) in *Samia cynthia*. Likewise, Roller and co-workers, ¹⁴⁵ when studying the formation of JH I (27) and JH II (28) in *Hyalophora cecropia*, found that whereas methionine provided the ester methyl group for both compounds, it was not utilized in the formation of the carbon skeleton. The remaining carbon atoms were not labeled by farnesol (1), mevalonate (6), or propionate although each was metabolized. Acetate gave only low incorporation into JH I (27) and JH II (28), and the specificity could not be determined. These results led the authors to conclude that either mevalonate was not reaching the site of synthesis or the sesquiterpenoid-like chain of JH I (27) and JH II (28) may be iderived via the fatty acid pathway.

Ajami and Riddiford recently reported¹⁴⁶ the first extensive experimental work on the mode of incorporation of precursors into JH I (27). Acetate, glucose, methionine, isoleucine, and valine were incorporated, but propionate, mevalonate (6), lysine, 4-methyl-*cis*-3-hexenol, 4-methyl-3-pentenol, farnesol (1), and bishomofarnesol (30) were not. On this basis it was suggested¹⁴⁶ that the common pathways of terpenoid or fatty acid biosynthesis were not operating.



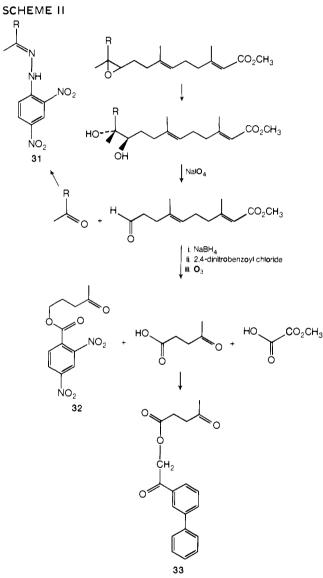
More recently, however, Schooley and co-workers¹⁴⁷ have studied more carefully the biosynthesis of JH II (28) and JH III (29) in *Manduca sexta* and have confirmed their terpenoid origin. An elegant scheme for the degradation of JH II (28) and JH III (29) was evolved whereby each compound was broken down to essentially three fragments. Each of these fragments could be derivatized to a crystalline compound suitable for work with the minute amount of material available. The degradation scheme used by Schooley is shown in Scheme II, and particular attention should be paid to the three fragments obtained as crystalline products, **31**, **32**, and **33**.

Table I summarizes the incorporation and degradation of JH II (28) and JH III (29) after feeding acetate, propionate, and mevalonate (6). These results confirm for the first time the biosynthesis of the juvenile hormone skeleton by the isoprenoid pathway, there being a specific incorporation of mevalonate (6) into JH II (28) and JH III (29).

The results are in agreement with the theory of juvenile hormone biosynthesis propounded by Schooley^{142,147} and mentioned briefly by others.¹⁴⁵ This theory differs drastically from the previous theories which have proposed the addition of one

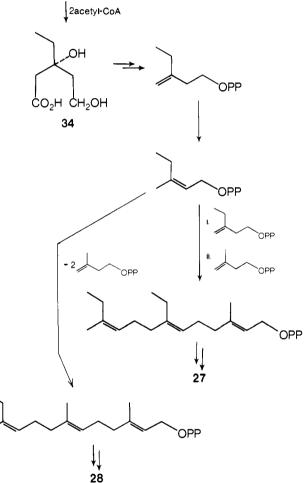
TABLE I. Degradation of JH II (28) and JH III (29)147

Predicted	Fragment					
total no.		31		32		33
atoms	Found	Predicted	Found	Predicted	Found	Predicted
8	0.97	1	3.14	3	2.87	3
2	0.02	0	0.97	1	1.05	1
1	0.98	1	0.05	0	0.05	0
1	0.90	1		0		0
9	1.93	2	3.07	3	3.23	3
3	1.14	1	1.05	1	1.03	1
	labeled atoms 8 2 1 1 1	total no. labeled atoms Found 8 0.97 2 0.02 1 0.98 1 0.90 9 1.93	total no. labeled atoms 31 B Found Predicted 8 0.97 1 2 0.02 0 1 0.98 1 1 0.90 1 9 1.93 2	Babeled atoms 31 Found Found 8 0.97 1 3.14 2 0.02 0 0.97 1 0.98 1 0.05 1 0.90 1 3.07 9 1.93 2 3.07	Babeled atoms 31 32 Babeled atoms Found Predicted Found Predicted 8 0.97 1 3.14 3 2 0.02 0 0.97 1 1 0.98 1 0.05 0 1 0.90 1 0 0 9 1.93 2 3.07 3	Predicted total no. labeled atoms 31 32 Babeled atoms Found Predicted Found Predicted Found 8 0.97 1 3.14 3 2.87 2 0.02 0 0.97 1 1.05 1 0.98 1 0.05 0 0.05 9 1.93 2 3.07 3 3.23

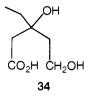


S. CoA

SCHEME III

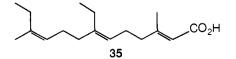


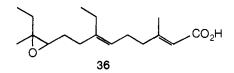
or two single carbon units to a preformed 15-carbon chain. In Schooley's theory JH II (28) is derived from one homoisoprenoid unit and two isoprenoid units, the former unit [homomevalonic acid (34)] being produced from propionate and two acetates rather than from three acetates. The data in Table I indicate the specific incorporation of propionate into the starting unit, and



the specific incorporation of mevalonate (6) only into the units derived from IPP (9). There is therefore no extension of mevalonate (6) by single carbon units. The probable biosynthetic route to JH I (27) and JH II (28) is therefore as shown in Scheme III.

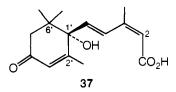
Additional work has demonstrated the possible intermediacy of a number of the acidic metabolites. Bishomofarnesoic acid (**35**) and farnesoic acid (**13**) were incorporated into JH I (**27**) and JH III (**29**). respectively. in *Schistocerca gregarid*.¹⁴⁸ 10-Epoxy-7-ethyl-3,11-dimethyl-2,6-tridecadienoic acid (**36**) was incorporated to the extent of 10% into JH I (**27**).¹⁴⁹





III. Biosynthesis of Monocyclic Sesquiterpenes A. Abscisic Acid

The structure $^{150-158}$ of abscisic acid (**37**), the widely occurring 159 plant growth regulator, suggests two possible routes of formation: (i) oxidative cleavage of a carotenoid derivative or (ii) direct biosynthesis from **1**. The importance of **37** in the regulatory processes in higher plants has given rise to a flurry of activity on biosynthesis and metabolism. This work has been summarized at various stages. $^{160-163}$



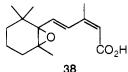
In 1967 Milborrow¹⁵⁹ demonstrated the presence of **37** in a number of ripening fruits. Later, avocado, strawberry, banana, and tomato were shown by Noddle and Robinson¹⁶⁴ to contain **37**, albeit in minute quantities. Each of these fruits was demonstrated¹⁶⁴ to give labeled abscisic acid (**37**) after feeding $[2-^{14}C]$ mevalonate (**6**) to the unripe fruit. The quantity of **37** produced did not permit degradative studies.

These early results were quickly followed by a more discriminating experiment using doubly labeled **8**. Abscisic acid (**37**) contains a terminal cis double bond, and theoretically there are two reasonable pathways for the formation of this group. These pathways may potentially be distinguished by the utility of pro-4*R* and pro-4*S* labeled mevalonates. Thus we may imagine that initial formation of a terminal cis double bond would take place with loss of the pro-4*R* proton, whereas isomerization of an initially formed trans double bond would lead to loss of the pro-4*S* proton and possibly a subsequent loss of the pro-4*R* proton in the isomerization step.

When $[4R-^{3}H]$ - and $[4S-^{3}H]$ mevalonates were separately fed to avocado fruit, ¹⁶⁵ two-thirds and zero tritium labeling were observed, respectively. The terminal cis double bond is therefore produced indirectly at some stage from a trans double bond with retention of tritium during the isomerization. It should be noted that the 4*R* proton of mevalonate (6), retained at C-5 in 1, is lost upon cyclization to 37. Although the stereochemical information is important, exactly the same result would be observed if abscisic acid (37) were produced by degradation of a carotenoid.

When [¹⁴C]phytoene was applied to avocado fruit, label was incorporated into the carotenoids but not into **37**.¹⁶³ Abscisic acid (**37**) is therefore not produced from degradation of a carotenoid.

Compound **38** also acts as a plant growth inhibitor, and Milborrow and co-workers¹⁶⁶ examined the possible intermediacy of **38** in the formation of abscisic acid (**37**). When the 2-cis

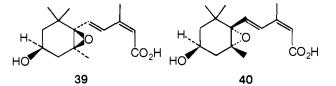


isomer of **38** was used as a precursor of abscisic acid (**37**) in green tomato fruit, a 1.8% incorporation was observed after 48 h. The corresponding 2-trans isomer gave only 0.04% incorporation.

It has been observed that, when wilting is induced in wheat.

there is a very rapid increase (25-fold) in abscisic acid (37) content.¹⁶⁷ This short time period is therefore potentially very useful for the study of abscisic acid (37) biosynthesis, depending upon whether abscisic acid is released from a bound state or if it is synthesized de novo at this time. When [2-3H]mevalonate was used as a precursor, nine times as much activity appeared in the 37 from wilted wheat as in that from normal wheat seedlings, suggesting that much of the abscisic acid (37) produced during wilting is synthesized de novo. When the [2-14C]epoxide (38) was used as a precursor of 37 in wilted wheat, incorporation was observed. If this compound is an intermediate on the route to 37, one of the further steps is formation of a hydroxy group at C-1'. Using the epoxide (38) specifically labeled at the epoxide oxygen with ¹⁸O, Milborrow showed¹⁶⁸ that essentially all (97%) of the heavy isotope label was retained upon conversion to 37; hence only one of the enantiomers of the epoxide is converted to 37.

In addition to **37**, $[2^{-14}C]$ -**38** also gave rise to another labeled product, the structure of which was elucidated to be 1',2'-epi-2-*cis*-xanthoxin acid (**39**), in which the C-1' stereochemistry is opposite to that observed in **37**. Labeled **39** was not a precursor of **37**. However, the opposite 1',2' epimer, (+)-2-*cis*-xanthoxin acid (**40**), was shown to be a precursor of **37** in tomato shoots. ¹⁶⁸

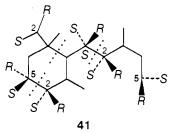


Attempts to show the presence of **40** in avocado fruit failed to give detectable activity by dilution analysis: **40** may therefore not be a normal intermediate in the biosynthesis of **37**, or the reactivity of **40** may prevent a measurable pool size forming.

The nonmetabolite (39) arises because the labeled epoxide (38) originally fed was a synthetic, racemic mixture of 1' epimers. Thus only one of the epimers was carried through to 37.

An important point about both the epoxide **40** and **37** itself is that each compound contains a C-4,5 double bond. The availability of the six stereospecifically labeled mevalonates permits an examination of the double bond forming processes in detail.

Structure **41** indicates the sites of labeling and stereochemistry from 2- and 5-labeled ($\mathbf{6}$) (prior to junction of the isoprene fragments) in the abscisic acid skeleton. Clearly in abscisic acid

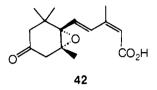


(37) itself, a number of these labels are completely lost, e.g., at C-1 and C-4'; indeed only one of the six C-5 hydrogens of mevalonate (6) should be retained in 37.

The situation regarding the retention of labels from a stereospecifically labeled C-2 in **6** is more complex. It depends not only upon the stereospecificity of elimination at C-4,5 and C-2',3', but also upon the presence of isopentenyl isomerase which tends to isomerize any stereospecifically labeled **6**, resulting in an apparent loss of the 2R-³H and retention of the 2S-³H. The intricacies of the results cannot be discussed, there being sufficient space only to discuss the conclusions from these experiments.

From $[2R^{-3}H]$ mevalonate (6) three tritium atoms were retained in 37,¹⁶⁹ and degradative work indicated that these tritium atoms were specifically retained at the expected positions, in particular at C-4 and C-3'. From $[2S^{-3}H]$ mevalonate (6), only one tritium was retained in **37**. When $[5S^{-3}H]$ mevalonate (6) was used as a precursor, one tritium was specifically retained at C-5 in **37**.¹⁶⁹ There is therefore an overall trans elimination of hydrogen at C-4,5 in the formation of the C-4,5 double bond. This result complements the work of Goodwin and co-workers¹⁷⁰⁻¹⁷² on the formation of the corresponding double bonds in the carotenoids.

Chemical evidence suggests that the 1',2'-cis- and *trans*-diols are moderately unstable, thus possibly precluding their isolation from natural sources. However, Taylor and Burden¹⁵⁴ have shown that 2-*cis*-4'-oxoxanthoxin acid (**42**) is a very labile



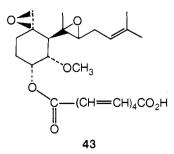
compound, rearranging to **37** under mild conditions, suggesting that possibly dehydrogenation at C-4' occurs prior to ring cleavage of the epoxide.

More recent work by Milborrow¹⁷³ has involved the elaboration of cell-free systems producing **37**. Because the rate of synthesis of **37** in ripening avocado is similar whether the mesocarp slices are turgid or wilted,¹⁷⁴ Milborrow considered that the biosynthesis of **37** from mevalonate (**6**) might be increased by damaging the chloroplasts. Extensive work with lysed chloroplasts, investigating the importance of various cofactors and time, indicated that maximal incorporation of **6** into **37** was observed with a complete medium after **17** h. It was also demonstrated that unlike the situation in wheat leaves,¹⁶³ **37** does not inhibit its own biosynthesis in avocado. Compounds which inhibit carotenoid biosynthesis¹⁷⁵ did not inhibit abscisic acid (**37**) biosynthesis.

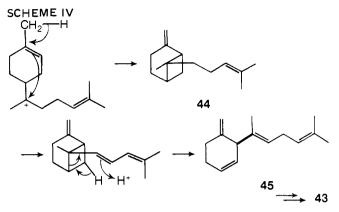
At this point it is not worthwhile considering an overall scheme for the biosynthesis of abscisic acid (**37**), even bearing in mind the stereochemical limitations imposed by the elegant work of Milborrow and associates. It is still not certain that farnesol (**1**) is a true intermediate since there is no record of this experiment having been attempted. Indeed it is true to say that details of the early steps in the scheme are known only very scantily. Thus, still to be investigated are (i) the point at which C-1 oxidation of farnesol occurs; (ii) the intermediacy of a compound having a 1'2' double bond; (iii) the point at which the **4**,5 double bond is introduced; (iv) the mechanism of introduction of the **4**,5 bond; and (v) the mechanism of introduction of the **1**'2' double bond. Clearly there is much further work needed before the intricacies of abscisic acid (**37**) biosynthesis are understood.

B. Fumagillin (43)

The carbon skeleton of fumagillin (**43**), an antibiotic from *Aspergillus fumigatus*, ¹⁷⁶ is not classically isoprenoid, and initial attempts were therefore aimed at establishing the terpenoid nature of this molecule.



 $[1-{}^{14}C]$ - and $[2-{}^{14}C]$ acetates gave good incorporations¹⁷⁷ into fumagillin (43), although, upon hydrolysis of the sebacic acid



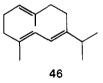
unit, almost half of the activity was lost. The remaining unit was shown by degradation to be derived from acetate in a specific manner, although the complete degradation pattern was not determined.

When mevalonate (6) was used as a precursor, a higher incorporation than with acetate was observed, and degradation indicated that one-third of the activity was found to be specifically in the terminal dimethyl group.¹⁷⁶

A novel mechanism was proposed by Birch¹⁷⁷ to account for the biosynthesis of fumagillin (43) and related compounds (Scheme IV).³⁵ Cyclization of *cis,trans*-farnesol (20) in a manner similar to monoterpene cyclization, and subsequent formation of a four-membered ring, gives 44. Introduction of a second double bond in the side chain and subsequent rearrangement would give rise to 45 which is at the same oxidation state as fumagillin (43).

C. Germacrene C (46)

The thermal instability of many of the simple germacrene derivatives, relative to the elemene derivatives, precluded their isolation for many years. Thus germacrene C (46), one of the

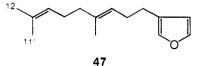


simplest sesquiterpenes, eluded isolation in substantial quantities until 1969.¹⁷⁸ when it was found to be the main constituent of the oil of the seed of *Kadsura japonica*.

Subsequent work on the biosynthesis of germacrene C (46) by the same workers demonstrated its formation as expected from mevalonate (6).¹⁷⁹ A supernatant of the homogenized seeds of *K. japonica* catalyzed the conversion of $[2-^{14}C]$ mevalonate (6) into germacrene C (46) and *trans,trans*-farnesol (1). The latter (as the pyrophosphate 2) was also an effective precursor of 46. No further work has been reported.

D. Dendrolasin

Dendrolasin (47), a simple sesquiterpene derivative containing a furan ring, occurs in the ant, *Lasius filiginosus*, ¹⁸⁰ and a number of higher plants.^{181,182} In 1969, Schmid and co-workers¹⁸³



confirmed the terpenoid derivation of dendrolasin (47). Incorporation into dendrolasin (47) of $[2-^{14}C]$ mevalonate (6) was invariably poor in spite of considerable effort.

Degradation of the labeled dendrolasin (47) after feeding $[2-{}^{14}C]$ mevalonate (6), $[1-{}^{14}C]$ acetate, and $[U-{}^{14}C]$ glucose

TABLE II. Degradation of Dendrolasin (47)

Precursor	% total activity in C-11, -11', and -12
[U-14C] Glucose	20
[1-14C] Acetate	15
[2-'⁴C] Mevalonate	32

indicated that incorporation was specific for the former two precursors and uniform for the labeled glucose (Table II). No further work has been reported, although a discussion of the many related furan derivatives in *Myoporium deserti* from a biogenetic point of view has appeared.¹⁸⁴

E. trans- γ -Monocyclofarnesol (48)

From the disrupted cells of the leaf-spot fungus *Helminthosporium siccans*, a compound was isolated which was demonstrated to be a novel sesquiterpene alcohol, *trans*- γ -monocyclofarnesol (**48**).¹⁸⁵ Further work demonstrated that in a cell-free

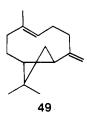
48 CH₂OH

system of *H. siccans*, 25-50% of the total activity from [2-¹⁴C]mevalonate (6) was located in 48.¹⁸⁵ Farnesol (1) was not found in this incubation, and the labeled *trans-* γ -monocyclo-farnesol (48) was not used as a precursor of the more elaborate metabolites of *H. siccans*.

IV. Biosynthesis of Bicyclic Sesquiterpenes

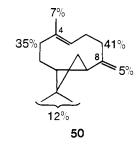
A. Caryophyllene (49)

Caryophyllene (**49**), a bicyclic sesquiterpene, contains a novel 11-membered ring and is the major sesquiterpene of peppermint, *Mentha piperita*. ^{186,187} The earliest work on the formation of caryophyllene (**49**) was, however, carried out using the catnip plant, *Nepeta cataria*. ¹⁸⁸ Waller and co-workers demonstrated the incorporation of $[2-^{14}C]$ mevalonate (**6**), but no degradative results were reported.



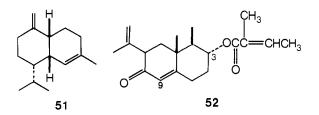
Loomis and Croteau.¹⁸⁷ as part of their work on the physiology of mono- and sesquiterpene biosynthesis,¹⁸⁹ have also investigated the formation of **49**. When [2-¹⁴C] mevalonate (**6**) was fed to peppermint plants through the cut stem, incorporation into **49** was about 0.22%, rising to this level after 6 h and decreasing rapidly, demonstrating a quite rapid turnover of this metabolite. Later work¹⁹⁰ demonstrated that in the presence of added sucrose, the level of mevalonate (**6**) incorporation was doubled after the same period of time, suggesting that sucrose may be an energy source at the site of biosynthesis.

Chemical degradation¹⁸⁷ of the radioactive caryophyllene (**49**) yielded activities at the carbon atoms as shown in **50**. Two interesting observations may be deduced from these results. The first is that the IPP-derived portions of caryophyllene (**49**) incorporate label more extensively than that derived from DMAPP (**14**). The second is that there is some loss of specificity from the label originally at C-2 of **6** into the C-4 methyl and the C-8 methylene, indicating the presence of an IPP isomerase.



B. γ -Muurolene (51)

 γ -Muurolene (**51**) co-occurs with caryophyllene (**49**) in *Mentha piperita*. Concurrently with their work on **49**, Loomis and Croteau¹⁸⁷ also examined the formation of **51** from [2-¹⁴C]mevalonate (**6**). After reaching a maximal incorporation of 0.1% after 4.5 h, the level decreased rapidly, again indicating active metabolism of **51**. With added sucrose the level of incorporation rose to 0.4%. No degradations were reported.



C. Petasin (52)

Petasin (52) and related compounds from *Petasites hybri* dus^{191} are members of the eremophilane series of sesquiterpenes, and again the skeleton cannot be derived by direct cyclization of farnesol (1).

Initial work with $[2-^{14}C]$ mevalonate (6) in this plant succeeded in labeling mainly the sterols and triterpene alcohols,¹⁹² but subsequent work^{193,194} improved the specific incorporation into 52.

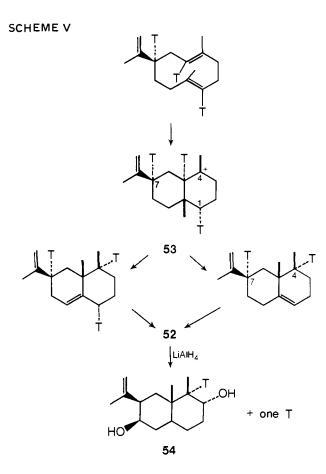
The level of incorporation was raised again when an enzyme inhibitor suppressing triterpene biosynthesis was used.¹⁹³ At a dose level of 100 ppm, the enzyme inhibitor tris(2-diethylam-inoethyl) phosphate trihydrochloride (SK and F 7997-A₃) improved the incorporation into petasin (**57**) fivefold with concomitant reduction of incorporation into the phytosterols.

It was suggested¹⁸⁶ that the eremophilane group of sesquiterpenoids arises from the eudesmane skeleton by shift of the methyl group at C-10 to C-5 with sequential loss of the C-1 α (or C-9 α) proton. Thorough degradation of the product after [2-¹⁴C]mevalonate (**6**)¹⁹⁴ indicated that one-third of the label was in the isopropylidene group, a further third was specifically at C-3, and the final third of the activity was specifically at C-9.

If Scheme V is correct for the formation of the eremophilane skeleton, then $[4R^{-3}H]$ mevalonate (6) should provide the 1α , 5α and 7α hydrogens in an intermediate eudesmane (53), and subsequent rearrangement should lead to an eremophilene labeled at 4α and 7α . Alternatively, if the 9α hydrogen were eliminated, one would expect complete retention of all tritium into petasin (52).

When this experiment was conducted, ¹⁹⁴ it was found that specific labeling of C-4 α occurred, supporting the hypothesis that methyl migration takes place with simultaneous 1,2-hydride shift. Reduction of petasin (52) with LiAlH₄ gave the diol (54) which contained only two tritium atoms. Because positions 1 and 7 of petasin (52) are base labile, however, partial exchange may have occurred.

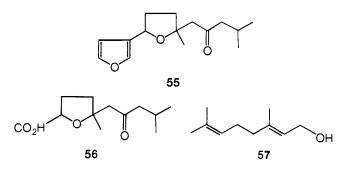
Lawton has proposed¹⁹⁵ an alternative scheme for the biosynthesis of the eremophilane skeleton involving spiro intermediates.



D. Ipomeamarone (55)

When the root of the sweet potato (*Ipomoea batatas*) is infected with the black rot fungus, *Ceratocystis fimbriata*, the former produces a novel sesquiterpene¹⁹⁶ derivative ipomeamarone (**55**),¹⁹⁷ which acts as a potent antifungal agent against the invading fungus.¹⁹⁸

The amount of **55** produced under these conditions is an amazing 2%,¹⁹⁷ after quite short time periods. Preliminary feeding experiments with $[2-^{14}C]$ acetate^{199,200} and $[2-^{14}C]$ -mevalonate (**6**)¹⁹⁹ gave poor incorporations. Subsequently²⁰¹ much higher incorporations were obtained, 11-16% for $[2-^{14}C]$ acetate and 48% for $[2-^{14}C]$ mevalonate (**6**) under slightly different conditions. Degradation to afford ipomeanic acid (**56**) indicated that all of the activity was retained from mevalonate (**6**) and **73**% of the activity from acetate.



Pyruvate and citrate²⁰² were considerably poorer precursors of **55.** In theory **7**/9 of the activity from $[2-^{14}C]$ acetate should be retained in ipomeanic acid (**56**). The slightly (\sim 5%) lower activity observed may indicate some unsymmetrical incorporation of acetate into farnesol (**1**). Further attempts to isolate small fragments failed, but the sesquiterpenoid nature of **55** was clearly established.

Added farnesol (1), nerolidol (11), or geraniol (57) reduced the incorporation of $[2^{-14}C]$ acetate although the level of ipo-

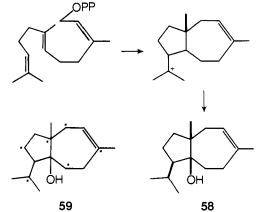
meamarone (55) was markedly increased.¹¹⁹ [2-¹⁴C]Farnesol was evaluated as a precursor of 55 in a time-course study.²⁰³ After 6 h 4.8% of the activity was located in 55, and this maximized to 6.8% after 21 h. The specificity of the label was not determined.

Ethanol is rarely used as a biosynthetic precursor, but when [2-¹⁴C]ethanol was used as a precursor of **55**, it proved to be more readily incorporated than acetate.²⁰⁴ Pyrazole, a potent inhibitor of alcohol dehydrogenase,²⁰⁶ drastically decreased the level of incorporation of ethanol, suggesting that the route of ethanol incorporation was via acetaldehyde and subsequently acetyl-CoA.

E. Carotol (58)

Carotol, one of the principal sesquiterpene alcohols of the seeds of the carrot, *Daucus carota*, ²⁰⁶ has a probable biogenesis from *cis,trans*-farnesol (**20**) as shown in Scheme VI.²⁰⁷ Only very

SCHEME VI

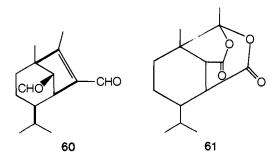


preliminary work has been reported. $[1-^{14}C]$ Acetate should give rise to **58** labeled as in **59**; degradation of acetate-labeled carotol (**58**) afforded the carbon atoms indicated as acetic acid, and these were found to contain 16% of the total activity as expected.²⁰⁷

Thus although the sesquiterpene nature of **58** is established, the mechanism of cyclization and the specificity of any hydride shifts in carotol (**58**) biosynthesis have not been determined.

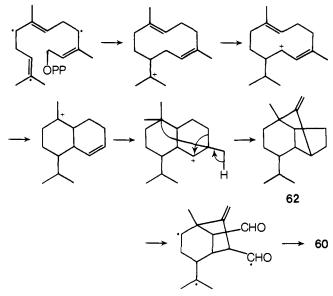
F. Helminthosporal (60)

Helminthosporium sativum is the fungus responsible for the seedling blight, root rot, and leaf spot of a number of cereals but most importantly (economically) of barley. The most important compound obtained from this fungus is the toxin helminthosporal (**60**).²⁰⁸ Although this compound is an artifact of the basic work-up procedure,²⁰⁹ we shall consider it here as a natural product.



Only one experiment has been reported and that some years ago. De Mayo and co-workers,²¹⁰ following up their isolation and structure elucidation studies, also examined the biosynthesis of helminthosporal (**60**) from [2-¹⁴C]mevalonate (**6**). Incorpo-

SCHEME VII

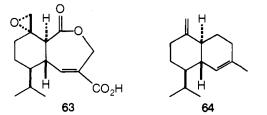


ration was low, but oxidation and subsequent ozonolysis produced a dilactone (61) retaining 62% of the initial activity of 60. In this degradation one carbon atom, the exocyclic carbon of the α , β -unsaturated aldehyde, was specifically lost.

Scheme VII depicts a route for helminthosporal (**60**) biosynthesis proposed by De Mayo et al.²¹⁰ involving a germacrene intermediate which subsequently cyclizes to a tricyclic compound (**62**). This intermediate, sativene, was subsequently isolated from *H. sativum*,²¹¹ thereby adding some credence to its possible biosynthetic utility in the formation of **60**. Recent work by Arigoni on the biosynthesis of sativene (**62**) is discussed later.

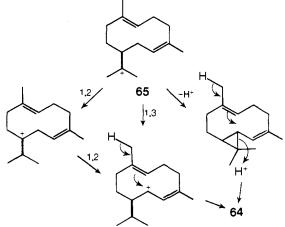
G. Avocettin (63)

Avocettin (63), from Anthostoma avocetta, may be thought of as the end product of extensive oxidative attack on ent- γ cadinene (64). Three possibilities for the formation of 64 from



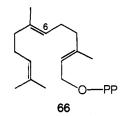
a germacrene cation such as 65 are shown in Scheme VIII. The work on the biosynthesis of avocettin (63), in particular that re-

SCHEME VIII



lating to the conversion of *trans,trans*-farnesol (1) to *ent-* γ -cadinene (64), forms part of an important and elegant study by Arigoni^{*} and co-workers.¹³⁷

[6-³H]Farnesol (1) was used initially as a precursor of avocettin (63). A 0.12% incorporation was observed and the label was at the ring junction position. [2-¹⁴C]Mevalonate (6) was incorporated into avocettin (63), and degradation indicated labeling to be at C-3, C-9, and in the isopropyl side chain. The mode of cyclization of 1 is therefore as illustrated in 66, i.e., from 6-*trans*-farnesol. Distribution of label was fairly even for the active sites [*cf.* caryophyllene (49)].



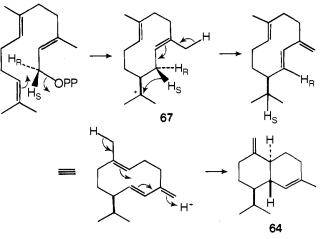
In order to investigate the formation of γ -cadinene (64) more carefully, [4*R*-³H]mevalonate (6) was used in admixture with [2-¹⁴C]-6. Intact incorporation was observed, and degradation indicated that one tritium was located at the carbon bearing the isopropyl group. Any route to 64 involving a 1,2-hydride shift can therefore be discounted.

The mechanism of formation of the isopropyl group of avocettin (63) was then investigated with $[5R-^{3}H]$ - and $[5S-^{3}H]$ mevalonates (6). Degradation of the labeled avocettin (63) indicated that *n*o loss of tritium occurred from C-1 of the farnesyl precursor and that formation of the isopropyl chain involved the stereospecific migration of the pro-5*S* proton of 6.

In the biosynthesis of *cis,trans*-farnesol (20) from 1, one of the hydrogens on C-1 was lost (see earlier discussion). Since this was not observed in the biosynthesis of 63, some alternative mechanism must be postulated to account for the formation of 64, which by its existence contains a cis double bond at the positions derived from C-2,3 of farnesol.

In addition, any mechanism must also take into account the stereospecific shift of the pro-5*S* of **6**. Scheme IX illustrates the rationalization of these results by Arigoni.¹³⁷

SCHEME IX

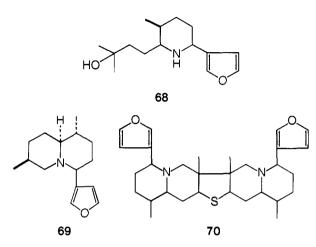


Further work by Arigoni¹³⁷ involved an examination of the stereochemical origin of the methyl groups in the isopropyl side chain. In the biosynthesis of **1**. $[2-^{14}C]$ mevalonate (**6**) labels only the (*Z*)-methyl group in the isopropylidene residue. A brilliant degradation of avocettin (**63**) succeeded in chemically isolating the methyl groups of the isopropyl side chain and demonstrated

 The author is deeply indebted to Professor D. Arigoni, who provided a preprint of his symposium lecture delivered at the 9th IUPAC meeting, Ottawa, Canada, June 1974. the integrity of these groups. Thus the initial ring closure and subsequent hydrogen migration to the isopropyl chain occurs faster than rotation of the side chain in **67**. The explanation of these results is deferred to a later point.

H. Alkaloids of Nuphar

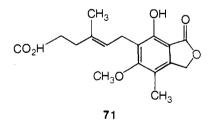
The genus of aquatic plants, *Nuphar*, contains a number of alkaloids containing both a piperidine and furan ring.²¹² Two typical examples are nupharamine (68) and deoxynupharidine (69). In both cases derivation from a linear sesquiterpene can be discerned.



The only reported study of the biosynthesis of this group of compounds is by Schutte and co-workers who made a preliminary study of the formation of the dimeric compound, thiobinupharidine (**70**), in *Nuphar luteum*.²¹³ Cadaverine was not a direct precursor of the piperidine rings, whereas $[2-^{14}C]$ - and $[3,4-^{14}C]$ mevalonates (**6**) were each specifically incorporated. The derivation from a 15-carbon precursor remains to be demonstrated, since at the time of the biosynthetic work the correct structure of thiobinupharidine (**70**) was not known.

I. Mycophenolic Acid (71)

Mycophenolic acid (71) from *Penicillium brevi-compactum*²¹⁴ does not appear to contain any unit derived from a terpenoid precursor. The nucleus has the standard substitution for derivation from acetate, save for an additional nuclear methyl group, and the side chain contains seven, rather than five or ten, carbon atoms.

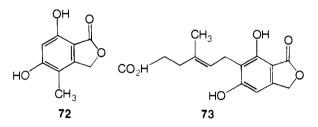


Early work on the biosynthesis of mycophenolic acid (71) demonstrated that methionine was a precursor of both the nuclear and *O*-methyl groups.^{215,216} The methyl group on the side chain is therefore derived from a source other than methionine.

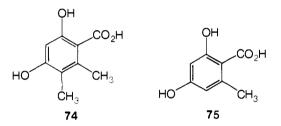
It was Birch²¹⁷⁻²¹⁹ who first demonstrated that the sevencarbon unit was derived from mevalonate (6) rather than acetate. $[1-^{14}C]$ Acetate labeled the side chain *and* the nucleus of $71.^{216-218}$ but $[2-^{14}C]$ mevalonate (6) was incorporated only into the side chain.^{217,218} Although the specificity of labeling within this unit was not determined, it was felt at the time that the origin of each carbon atom in the molecule was defined.

This area remained dormant for many years, but the recent work of Money, and more particularly Canonica and their respective co-workers, has illuminated the steps on the biosynthetic pathway quite elegantly.

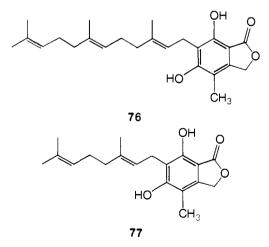
Canonica considered that nuclear methylation was an early step in the sequence and proved this theory in two separate series of experiments. In the first of these, a variety of phthalide derivatives were used as potential precursors. Only one. **72**, was incorporated to any great extent (36.5%).²²⁰ The corresponding compound lacking the 4-methyl group was not incorporated into **71**, but did give rise to a new metabolite, **73**, also lacking the 4-methyl group. When the incubation was interrupted after feeding [¹⁴C] methionine, **72** was isolated radioactivity labeled to the extent of 3% of the total activity.²²¹



Both Money²²² and Canonica²²¹ have investigated more carefully the point at which nuclear methylation occurs. Whereas 4.6-dihydroxy 2,3-dimethylbenzoic acid (**74**) was a precursor, orsellinic acid (**75**), a postulated precursor,²²³ was not. On this basis it was suggested^{221,222} that C-methylation occurs at the stage of an enzyme-bound polyketide.



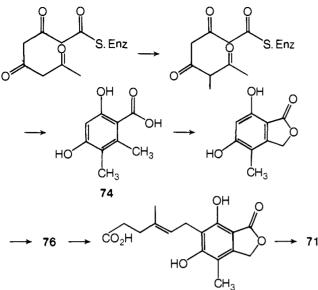
It had been speculated^{224,225} that the side chain of mycophenolic acid was derived from a geranyl unit. However, when the phthalide derivative (72) was added to the medium, a new metabolite (76) containing a C₁₅ side chain was isolated.²²¹ Reintroduction of 76, specifically labeled at the nuclear methyl group, gave 71 containing 33.6% of the total activity.²²¹ The corresponding geranyl derivative (77) was incorporated to a much lesser extent, and labeled geraniol (57) was not a precursor.²²²



The seven-carbon side chain is therefore derived not from a $C_{\rm 10}$ unit, but from a $C_{\rm 15}$ farnesyl, unit. Oxidation of the side chain probably precedes O-methylation.^{226}

A reasonable route for the biosynthesis of mycophenolic acid (71) is shown in Scheme X.

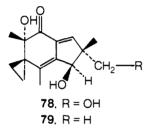
SCHEME X



V. Biosynthesis of Tricyclic Sesquiterpenes

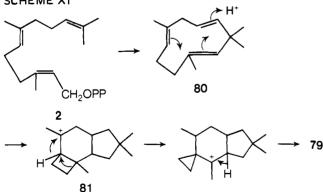
A. The Illudins, Hirsutic Acid, and the Coriolins

The fungal metabolites illudin S $(78)^{227-230}$ and illudin M $(79)^{227-230}$ from the Basidomycetes *Clitocybe illudens*²²⁷ and *Lampteromyces japonicus*, ²²⁸ exhibit antibacterial, ^{231,232} toxic^{231,232} and antitumor²³⁰ activity. Initial biosynthetic work



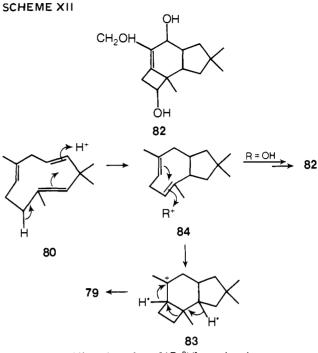
with [2-¹⁴C]mevalonate (6) demonstrated incorporation into **78** and **79**,²³⁰ and it was suggested that the humulene skeleton (80) was an important intermediate as shown in Scheme XI.

SCHEME XI



A compound illudol, with the skeleton **81** was isolated from *C. illudens*, ²³¹ but its structure was not determined to be **82** until 1967.²³³ Preliminary biosynthetic work²³³ indicated that $[2-1^{4}C]$ mevalonate (**6**) was a precursor.

More recently, Hanson and co-workers²³⁴ have investigated the biosynthesis of this group of compounds more carefully. Using $[4R-^{3}H,2-^{14}C]$ mevalonate (6), it was demonstrated that only one tritium was specifically retained in the formation of illudin M (79). confirming the proton losses outlined in Scheme XI. However, degradation of illudin M (79) after feeding $[2-^{3}H_{2},2-^{14}C]$ mevalonate (6) indicated the presence of only three

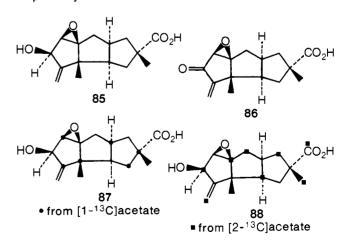


H* protons from [4R-3H]mevalonate

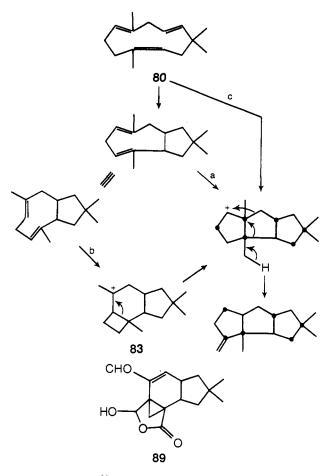
(rather than four) tritium atoms at the indicated positions. The 3 H: 14 C ratio dropped from 9.6:1 in 6 to 4.7:1 in illudin M (79), with a specific incorporation of 0.06%. 234

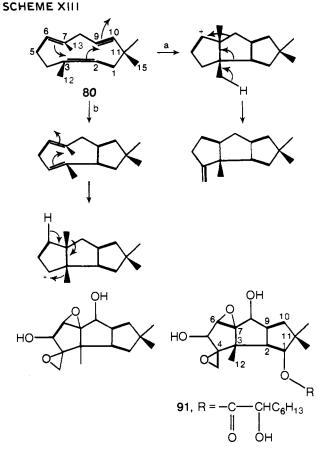
Clearly an alternative mechanism is involved, and it was suggested²³⁴ that the tricyclic intermediate **83** (rearranging as shown) is not produced directly from the humulene skeleton, but that a bicyclic intermediate of the type **84** is involved. The mode of formation of this intermediate would involve loss of a proton derived from C-2 of **6** as shown in Scheme XII. The invocation of a bicyclic intermediate would also allow a rationalization of the formation of illudol (**82**).

CMR spectroscopy has recently been used in a variety of biosynthetic studies,²³⁵ and this technique has been applied to an evaluation of the possible biosynthetic routes to hirsutic acid (**85**) and complicatic acid (**86**) in *Stereum complicatum*.²³⁶ In particular, the utilization of $[1^{-13}C]$ - and $[2^{-13}C]$ acetates as precursors of hirsutic acid (**85**) was studied.²³⁷ From $[1^{-13}C]$ - acetate carbon atoms 1, 3, 4, 6, 9, and 11 were specifically enriched whereas $[2^{-13}C]$ acetate specifically enriched carbon atoms 2, 5, 7, 8, 10, 12, 13, 14, and 15 as shown in **87** and **88**, respectively.



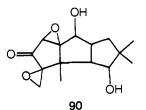
The observed pattern of labeling clearly demonstrated that two 1,2-shifts occurred, and three schemes were postulated to explain the rearrangement from the humulene derivative **80.**²³⁷ One of these (route a) was a modification of a route originally





proposed by Scott,²³⁸ and another involved an intermediate (83) previously deemed important in the formation of illudin M (79), and possibly marasmic acid (89) biosynthesis. It remains to determine which of these routes is the correct one.

The coriolins are a group of novel antitumor antibiotics from *Coriolus consors*, ²³⁹ bearing a close structural relationship to hirsutic acid (**85**) and are typified by **90.** As with the biosynthesis



of hirsutic acid (85) referred to previously, ¹³C NMR spectroscopy was utilized in order to gain additional information concerning the biosynthetic route. The key to the method used in this case is the observation of ¹³C–¹³C coupling constants when an intact [1,2-¹³C₂] acetate moiety is incorporated. If the acetate C–C bond is broken, the coupling is irretrievably lost. In this way the migration of a methyl group, for example, can be easily monitored. As with all ¹³C NMR work, however, the assignment of each carbon resonance in the spectrum must be made.

 $[1,2^{-13}C_2]$ Acetate was therefore used as a precursor of 5dihydrocoriolin C (91) in *Coriolus consors*.²⁴⁰ From the ¹³C NMR spectrum it was deduced that six couplings were present in the labeled 91 out of the 24 possible C–C couplings. On the basis of previous assignments, these ¹³C–¹³C couplings were found to be as follows: 1–2, 3–12, 4–13, 6–7, 9–10 and 11–15.

This single experiment served to distinguish between three possible pathways which the previous ¹³C labeling work with hirsutic acid (85)²³⁷ had not been able to do. The only scheme compatible with these results is route a illustrated in Scheme XIII, from humulene (80), in which intact ¹³C-¹³C couplings are

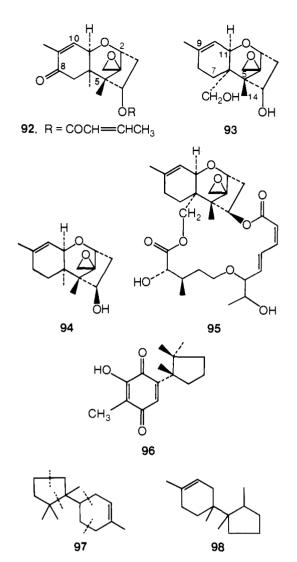
marked with heavy lines (stereochemistry not implied). In order to give some idea of the theory behind this experimment, an alternative scheme (route b) is also included which gives a product with different ${}^{13}C{-}^{13}C$ labeling. Of particular note in this case are the lack of ${}^{13}C{-}^{13}C$ couplings at 3–12 and 4–13 which are lost at the stage of the double methyl migration. Since these couplings are observed in **91**, route b is excluded from consideration.

Also used as precursors were $[1-^{13}C]$ acetate, $[2-^{13}C]$ acetate, and $[2-^{13}C]$ mevalonate. The latter gave specific enrichment at C-5, C-8, and C-14, indicating that the stereospecific integrity of DMAPP (14) was again maintained upon further elaboration to more complex cyclized products.

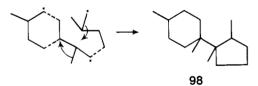
B. Trlchothecin (92) and Helicobasidin (96)

The fungus *Trichothecium roseum* produces a number of metabolites of which trichothecin (92) is the best known.³⁵ The correct structure for trichothecin (92) was first proposed by Godtfredsen and Vangedal,²⁴⁵ and biosynthetically it is one of the best studied of the sesquiterpenes. From a number of related fungi, including *Myrothecium roridum*,^{241–244} *M. verrucaria*,²⁴³ and *Trichoderma viride*,²⁴⁵ closely related compounds such as verrucarol (93),^{241–244,246} trichodermol (94),^{245–247} and roridin (95)²⁴³ were isolated.

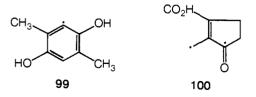
Helicobasidin, one of the two major metabolites of *Helicobasidium mompa*, ²⁴⁸ was shown to have structure **96** by Nishikawa and co-workers.²⁴⁹ Although **92** and **96** have quite different carbon skeleta, their biosyntheses are, as we shall see, quite closely related. If we consider that formation of the quinone ring of helicobasidin (**96**) is a final step, we can imagine a precursor having the carbon skeleton depicted in **97**. If we also break the pyran ring in trichothecin (**92**) and remove all oxygenation, the skeleton **98** is obtained. Of these, only **97** obeys the isoprene rule, the other being related either by a 1,3-methyl shift or two 1,2-methyl shifts. We can naively predict, therefore, that the



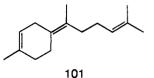
skeleton contained in 97 may well be a precursor of the tricothecane skeleton (98).



Initial work on the biosynthesis of trichothecin (**92**) was reported some years ago by Jones and co-workers.^{250,251} [2-¹⁴C]Mevalonate (**6**) was incorporated to the extent of 0.5% and extensive degradation indicated that one label was located in the six-membered ring fragment **99**, and two labels in the cyclopentenone fragment **100**. [1-¹⁴C]Acetate gave rise to trichothecin (**92**), containing 95% of the label in the crotonate molety.



The crude scheme suggested for trichothecin (92) biosynthesis²⁵¹ was refined by Ruzicka²⁵² and Godtfredsen and Vangedal.²⁴⁵ Each scheme involved a bisabolene derivative such as **101** as the key intermediate, and the latter scheme²⁴⁵ also accounts for the stereochemistry of **92** via a boat conformation of the side chain in the cyclization reaction. There are two possible modes of cyclization of farnesol to give the bisabolene intermediate **101**, depending upon the stereochemical nature of the central double bond of farnesol. At that point there was no information available as to which isomer was the preferred precursor.



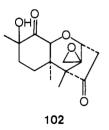
Polish workers have investigated the biosynthesis of verucarol (93) using $[1-^{14}C]$ - and $[2-^{14}C]$ acetates and $[2-^{14}C]$ mevalonate (6).²⁵³ Degradation of the product derived from $[1-^{14}C]$ acetate and $[2-^{14}C]$ mevalonate (6) indicated that the labeling was specific.

Similar work was reported by Bentley and Chen²⁵⁴ and Nishikawa et al.²⁵⁵ on helicobasidin (**96**) in *H. mompa*. [1-¹⁴C]Acetate was incorporated to the extent of 0.9% and [2-¹⁴C]-6 to the extent of 0.04%.²⁵⁴ Degradation of the labeled helicobasidin (**96**) indicated^{254,255} that each precursor was used specifically in accordance with a scheme from **2**, although the various hypotheses^{249,256} could not be distinguished by this method.

In 1967 Hanson published his first results in this area.^{257,258} Farnesyl pyrophosphate (2) was shown to be a good (1.5%) precursor of trichothecin (92) in *T. roseum*. This was the first incorporation of an intact C-15 unit into a cyclic sesquiterpene derivative and was the first of a very successful series of experiments by the Sussex group. Simultaneously, Nozoe and co-workers²⁶⁰ have studied in detail the biosynthesis of helicobasicin (96).

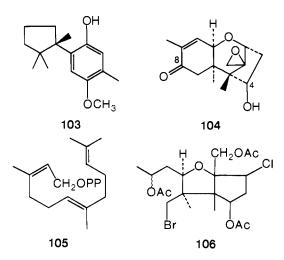
As we have seen previously, it is the pro-4*S* proton of mevalonate (6) which is lost in the formation of *trans,trans*-farnesol (1). Thus from $[4R-^{3}H]$ mevalonate we would expect complete retention of tritium in farnesol (1), loss of one tritium upon incorporation into trichodermol (94) and trichothecin (92), and loss of two tritiums on incorporation into helicobasicin (96). Thus trichothecin (92) and trichodermol (94) should be labeled at C-2 and C-10, and helicobasidin (96) at C-2.

Hanson²⁵⁹ demonstrated that when $[4R^{-3}H]$ mevalonate (6) was used as a precursor of trichothecin (92) in *T. roseum*, and of trichodermol (94) in *Trichoderma* sp., two out of three tritium labels were retained. Degradation of trichodermol (94) indicated that the ketol 102 retained only half of the tritium activity of 94, a result apparently in conflict with the earlier work of Jones and Lowe.²⁵¹



When $[4R^{-3}H]$ mevalonate (6) was used as a precursor of helicobasidin (96) in *H. mompa*,²⁶⁰ an interesting result was observed. Instead of two tritium atoms being lost, two of three were *retained*. Evidence from chemical conversion indicated that the tritium originally of C-8 of farnesol (1) is somehow transferred to the five-membered ring, for **103**, a cometabolite of **96**, retained all tritium activity. Because of this observation, Nozoe²⁶⁰ postulated that a γ -bisabolene derivative such as **101** was *not* involved as a biosynthetic intermediate.

In order to investigate the mode of cyclization of the farnesyl chain, $[1-{}^{3}H_{2}]$ - and $[2-{}^{3}H]$ farnesyl pyrophosphates (2) (each used in admixture with $[2-{}^{14}C]$ -2) were used as precursors of

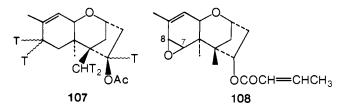


trichothecolone (**104**) in *T. roseum*.^{259,261} With $[1-^{3}H_{2}]$ farnesyl pyrophosphate (**2**) one tritium label was lost in trichothecolone (**104**), but no label was lost from $[2-^{3}H]$ -**2**. On this basis it was suggested²⁵⁹⁻²⁶¹ that *cis,trans*-farnesyl pyrophosphate, cyclizing as in **105**, was the initial intermediate.

Similar work using different precursors has also been reported by Nozoe and Machida.²⁶² [4R-³H,2-¹⁴C]Mevalonate (6) retained two tritium atoms (in agreement with Hanson's work²⁵⁹) upon incorporation into trichothecolone (104). The radioactive trichothecolone (104) was then degraded by a series of steps in which the carbon atom at C-8 was specifically removed. This compound, 106, was shown to have a ³H:¹⁴C atomic ratio of 1.78:2; i.e., the carbon at C-8 was specifically derived from [2-¹⁴C]mevalonate (6), confirming the result of Hanson.^{259,261}

Additional evidence for the farnesol cyclization came from experiments with $[2-{}^{3}H_{2}]$ mevalonate (6). When trichodermin (107) from *Trichoderma sporulosum* and trichothecolone (104) from *T. roseum* were isolated after feeding $[2-{}^{3}H_{2}]$ mevalonate (6), five and four tritium labels were incorporated, respectively.^{261,263} The methyl groups of the three mevalonate units therefore retain their individuality.

When $[2R^{-3}H]$ mevalonate (6) was used as a precursor, the isolated trichodermol (94) contained nearly two tritium labels.^{261,263} This is an expected result if hydroxylation at C-4 proceeds specifically with removal of the pro-2*R* proton (of 6) at this position. This observed specificity was not checked with $[2S^{-3}H]$ mevalonate (6). The retention of four tritium labels into trichothecolone after feeding $[2^{-3}H_2]$ mevalonate (6) is an unexpected result.^{261,263} One label was demonstrated by oxidation to be at C-4. The location of one of the other labels was suggested after the multiply labeled 107 (obtained from $[2R^{-3}H]$ -6) lost only one tritium upon incorporation into trichothecolone (104) and trichothecin (92). A tritium must therefore have migrated from C-8 to C-7 during formation of 92, possibly via an NIH shift type mechanism on the epoxide 108.



When $[5-{}^{3}H_{2},2-{}^{14}C]$ mevalonate (6) was used, 261,264 only four of the expected five tritium labels were observed in tricothecin (92). The assumption (not checked) was that one tritium was lost from C-7 of trichothecin (92) at some stage.

Similar results were obtained by Hanson.²⁶⁵ $[2-{}^{3}H_{2}.2-{}^{14}C]$ -Mevalonate (6) was transformed into helicobasidin (96) in *H. mompa* with retention of four tritium labels, and when [5-

 ${}^{3}H_{2}, 2{-}^{14}C$] mevalonate (6) was used two tritium atoms were retained.

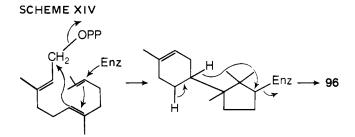
Tamm and co-workers²⁶⁶ investigated the use of $[2-{}^{3}H_{2}]$ mevalonate (**6**) as a precursor of verrucarol (**93**), and one of these was shown (by oxidation) to be at C-4. No tritium was found at C-10 but two tritiums were located at C-8. A bisabolene intermediate was suggested.²⁶⁶ When the individual pro- $[2-{}^{3}H]$ isomers were used as precursors, it was demonstrated that $[2 S-{}^{3}H]$ mevalonate (**6**) gave rise to a tritium at C-4, indicating again that hydroxylation at C-4 was stereospecific.

Tamm and co-workers²⁶⁷ also examined the biosynthesis of verrucarol (**93**) from farnesyl pyrophosphate (**2**). A specific incorporation of 0.22% into verrucarol (**93**) was observed when $[6-{}^{3}\text{H}, 12, 13-{}^{14}\text{C}]$ -*trans,trans*-farnesyl pyrophosphate (**2**) was used. There was no loss of tritium, and degradation indicated that the tritium was probably located at C-2 of verrucarol (**93**). Thus a specific migration of tritium occurs from C-6 to C-2 after formation of the cyclic system. The biosynthesis of verrucarol (**93**), therefore, mirrors that of trichothecin (**92**) in this respect.

Hanson²⁶⁸ followed up Nozoe's²⁶⁰ observation that a hydrogen shift was occurring in the biosynthesis of trichothecin (**92**) and helicobasidin (**96**).

When $[4R^{-3}H]$ mevalonate (6) was used as a precursor of helicobasidin (96) in *H. mompa*, two of the three tritium atoms were retained.^{265–268} In order to investigate whether the tritium from the central prenyl unit was retained, $[2^{-3}H,2^{-14}C]$ geranyl pyrophosphate (10) was used. There was no change in the ³H:¹⁴C ratio and consequently the 6-proton of 1 migrates during the biosynthesis of 96. Similar work^{261,168} demonstrated that this migration also takes place in the biosynthesis of trichothecin (92) and trichodermol (94).

On this basis Hanson^{265,268} suggested that the cyclization of *cis,trans*-farnesyl pyrophosphate (**105**) may be concerted with attack of an enzyme at C-10 initiating cyclization. The hydride from C-6 of farnesol then migrates to C-10 with concomitant displacement of the enzyme as shown in Scheme XIV.

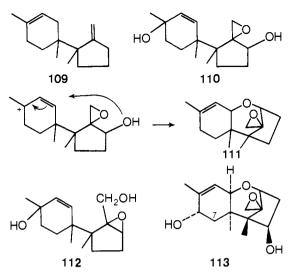


Forrester and Money²⁶⁹ investigated the utility of a number of bisabolene derivatives and a monocyclofarnesol in the biosynthesis of trichothecin (**92**). In each case low incorporations and extensive degradation were observed. The intervention of a bisabolene intermediate in the biosynthesis of trichothecin (**92**) and helicobasidin (**96**) is mitigated both by this work and the work of Hanson and Nozoe.

Substantial progress has also been made in the isolation of various advanced intermediates and their subsequent utilization in the biosynthesis of the major metabolites. In 1970 Nozoe's group isolated from *Trichothecium roseum* a number of new trichothecin-type compounds of considerable biogenetic significance. Among these were trichodiene (109)^{270,271} and trichodiol (110).²⁷¹ The relationship of 109 and 110 to the more elaborate derivatives is clear, for trichodiol (110) needs only to cyclize at C-11 to afford the trichothecin nucleus 111. These precursor relationships have been substantiated by subsequent work.

Tritium-labeled trichodiene (109) was fed to *T. roseum*,²⁷² and a 3.2% incorporation was observed into trichothecolone (104) after base hydrolysis. Also labeled in this study were 12,13-epoxytrichothec-9-ene (111) and trichodiol A (112). Since

trichodiol A (**112**) is an artifact produced²⁷¹ by saponification of trichodiol (**110**), this may be regarded as incorporation (2%) of trichodiene (**109**) into trichodiol (**110**).

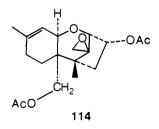


The incorporation of trichodiene (109) into 111 prompted a search for this compound in the culture broth. Although only a small quantity of material was obtained, chemical interrelation with trichodiol (110) confirmed the structure of the new metabolite to be 111.

Thus there remain only two hydroxylations which occur to afford the fully oxygenated trichothecane nucleus, one at C-8 and the other at C-4, the latter having been demonstrated to be stereospecific.^{261,263} The product of these two hydroxylations, trichothecadiol (**113**), was isolated from *T. roseum*.²⁶² Isolation of a biogenetically reasonable metabolite does not prove intermediacy, however, and the results of feeding experiments in this area should be interesting. It should be remembered that Hanson^{261,263} has demonstrated that tritium migrates from C-8 to C-7 in trichothecin (**92**) and has postulated an epoxide intermediate.

However, some of these results are controvertible. When $[2R^{-3}H,2^{-14}C]$ - and $[2S^{-3}H,2^{-14}C]$ mevalonates (6) were used as precursors of trichothecin (92), the results²⁷³ were not as expected from the data based on the work with $[2^{-3}H_2]$ mevalonate. From the 2*R*-labeled mevalonate (6), only one tritium (at C-14) was retained; from the 2*S*-labeled mevalonate (6), two tritium atoms were retained, and one of these was located at C-4 α (the other at C-14?). This is in contrast to Hanson's work^{261,263} which suggested that a tritium migration from C-8 to C-7 had taken place. The need to involve NIH shift in the rearrangement of a 7,8-epoxide to a 7-ketone is therefore questioned. The stereospecificity of hydroxylation at C-4 with removal of a pro-2*R* hydrogen from 6 is, however, confirmed.

Calonectrin (114), which has no oxygenation at either C-4 or C-8, was also investigated with the stereospecifically labeled $[2^{-3}H]$ mevalonates (6). Both pro-2*R* and pro-2*S*-³H mevalonates (6) showed incorporation of three tritium labels, and one of these was established to be at C-8.²⁷⁸



However, calonectrin (114) is oxygenated at C-3, a position which should be labeled by the 5-carbon of 6. When 114 was isolated after incorporation with $[5R-^{3}H]$ mevalonate (6), two

tritium atoms were retained.²⁷³ In the formation of trichothecin, three tritiums were retained, and one of these was at C-7.²⁷³

Several interpretations of these results are possible, but the most straightforward can be presented as follows. In the biosynthesis of the trichothecane skeleton, the 5-carbon of mevalonate (6) should label positions 3, 7, and 11. Carbon 7 which in calonectrin (114) does not have exchangeable hydrogens should retain tritium. Since the tritium at C-7 in trichothecolone (104) was exchanged, C-7 contains a label from pro-5*R*-H 6 in both 92 and 114. The stereochemistry of C-11 is the same in both 92 and 114, and possibly accounts for the location of a second label. That trichothecolone (104) retains the third tritium atom, probably at C-3, indicates that in calonectrin (114) a stereospecific hydroxylation has again occurred, this time with specific loss of the pro-5*R* proton of 6.

An interesting observation can be made on the basis of these experiments. Nozoe's isolation of trichodiol (**110**),²⁷¹ although not demonstrating intermediacy of this precise system, suggests at least that in the formation of the pyran ring attack probably takes place by a 2-hydroxy group on a 10,11-double bond. Since the pro-5*R* proton is apparently retained at C-11, the formation of the 10,11-double bond must proceed stereospecifically with loss of the 5*S* proton. This may occur by formation of an enzyme bound intermediate at C-9 α , which then undergoes stereospecific attack by the hydroxyl group at C-11 in trichodiol (**110**) with displacement of the enzyme. If this were the case, calonectrin (**114**) and trichothecin (**92**) should retain only two tritium labels after incorporation of [5*S*-³H] mevalonate (**6**). This experiment has not yet been reported.

In order to establish conclusively the location of the label from the 2-carbon of mevalonate, $[2-^{13}C]$ mevalonate (6) was used as a precursor of helicobasidin (96) in *H. mompa*.²⁷⁴ Prototropic tautomerization is expected to distribute the label equally between C-8 and C-10, and this was observed. Carbon atoms 4 and 12 were also enriched equally, thus confirming the specificity but giving no information as to the mode of cyclization of farnesol.

Additional experiments with $[1-^{13}C]$ - and $[2-^{13}C]$ acetates gave the expected specific enrichments of carbons 1, 3, 5, 7, 9, and 11 from $[1-^{13}C]$ acetate and of carbons 2, 4, 6, 8, 10, 12, 13, 14, and 15 from $[2-^{13}C]$ acetate.²⁷⁴

Similar work on the biosynthesis of the trichothecane skeleton gave different results.²⁷⁵ Initially the optimum incorporation time was determined using $[2-1^4C]$ mevalonate (6). $[2-1^3C]$ Mevalonate (6) (containing a small quantity of $[2-1^4C]$ mevalonate (6) as tracer) was then used, and after 5 days trichothecin (92) was isolated. The hydrolysis product trichothecolone (104) was examined by ^{13}C NMR. Only three carbons, C-4, C-8, and C-14, showed any enrichment, and all of these were enriched to the extent of 0.35%. That C-8 rather than C-10 is specifically enriched by C-2 of mevalonate (6) confirms the mode of cyclization of farnesyl pyrophosphate (2) determined previously.^{259–261,271}

When trans, trans-farnesyl pyrophosphate (2) was used as

SCHEME XV 2 \rightarrow HO HO OH 109 110 \rightarrow HO HO OH \rightarrow HO HO OH \rightarrow OH

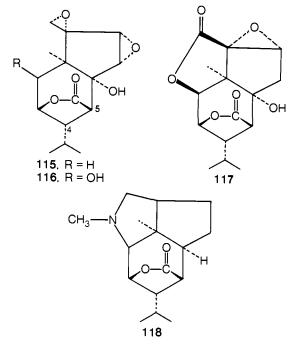
Biosynthesis of Sesquiterpenes

a precursor of trichodiene (109), one label from C-1 of 2 was specifically lost, supporting the possible involvement of *cis,trans*-farnesol (20).

The probable sequence of the late stages in the biosynthesis of **92** is outlined in Scheme XV. Some of the factors and requirements for trichothecin (**92**) biosynthesis in *T. roseum*^{276–280} have recently been established.

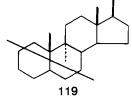
C. Picrotoxinin (117) and Dendrobine (118)

Four closely related sesquiterpenes from widely differing origins are part of a structurally interesting group which has been studied biosynthetically in a number of laboratories. The compounds involved are coriamyrtin (115), tutin (116), picrotoxinin (117), and the alkaloid dendrobine (118). Picrotoxinin (117) is



part of the complex picrotoxin, found in several species of the Menispermaceae,²⁸¹ and has the structure indicated.^{282,283} Coriamyrtin, which occurs in a number of *Coriaria* species,²⁸¹ was shown to have the structure **115**.²⁸⁴ Tutin has the structure^{285,286} and absolute configuration given in **116**²⁸⁷ and has been chemically correlated with coriamyrtin (**115**).²⁸⁸ It cooccurs with coriamyrtin (**115**) in *Corlaria japonica*²⁸¹ and has also been isolated from *Hyenancha globosa* of the Euphorbiaceae.²⁸⁹ Dendrobine, from the Chinese drug "Chiu Shi Hu" (*Dendrobium nobile*),²⁹⁰ was shown to have the structure and absolute configuration **118**.²⁹²

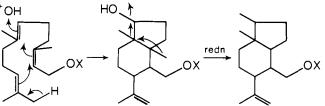
Biogenetically, it was suggested by Conroy^{292,293} that picrotoxinin was derived by the oxidative partial removal of the A and B rings of a 17-methylandrostane derivative (119). Cross,



however. considered²⁹⁴ that these compounds could be derived from farnesol (1) by cyclization and two 1,2-methyl migrations (Scheme XVI).

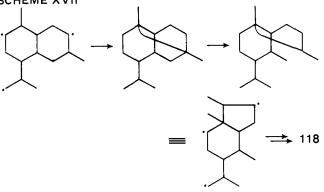
It was not until 1966 that the first biosynthetic results appeared, when Yamazaki and co-workers²⁹⁵ reported preliminary experiments on the biosynthesis of dendrobine (**118**) in *Dendrobium nobile*. A total incorporation from $[2-^{14}C]$ mevalonate (**6**) of 0.012% was observed, and degradation indicated that this



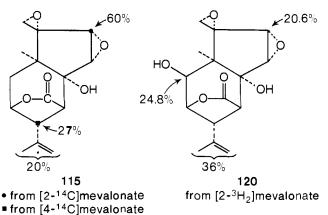


incorporation was specific. On the basis of their results, these workers suggested that dendrobine (118) was derived from a cadalane intermediate in which cyclization and ring cleavage subsequently occurred to give the picrotoxinin skeleton which was then aminated (Scheme XVII).

SCHEME XVII



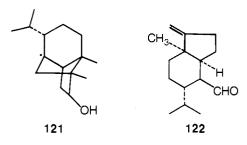
Independent work by the groups of Arigoni²⁹⁶ and Jommi²⁹⁷ afforded considerable information on the biosynthesis of coriamyrtin (115) and tutin (116). $[2-^{14}C]$ - and $[4-^{14}C]$ mevalonates (6) gave only low incorporation into 115 and 116 in *C. japonica*,²⁹⁶ but elegant degradation succeeded in establishing the labeling in 115 as shown. Double-labeling studies of tutin (116) in *C. japonica*,²⁹⁷ with subsequent careful degradation studies, gave a specific incorporation (for ³H) as shown in 120. Two



important points emerge from these results: (i) incorporation of mevalonate (6) into DMAPP (13) is lower than predicted, and (ii) labeling of C-9 and 10 in the isopentenyl unit was essentially equal. The latter observation is rare in terpenoid derivatives other

than the iridoids.¹³⁴ With these results in hand, copaborneol (**121**), the major constituent of *Pinus sylvestris* oil,^{298,299} was implicated in the biosynthesis of tutin (**116**) and related compounds by both Arigoni and Jommi, and when copaborneol (**121**), specifically tritiated as shown, was used as a precursor of tutin (**116**), 90% of the activity was located at C-5 in **116**.³⁰⁰

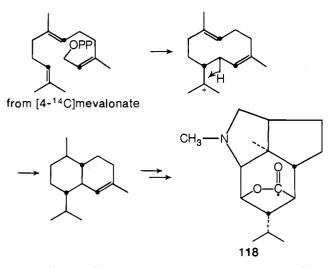
The intermediate nature of copaborneol (**121**) in the biosynthetic scheme confirms the earlier work on the cyclization of farnesol (**1**) and excludes the 1,2-methyl shift hypothesis com-



pletely. One may speculate that as with the ring cleavage of the iridoids to the secoiridoids, ¹³⁴ hydroxylation of the methyl group is the next step followed by fragmentation to give an olefinic aldehyde such as **122**, which is subsequently oxidized. The importance either of **122** or of the ring cleavage reaction in the formation of tutin (**116**) remains to be demonstrated.

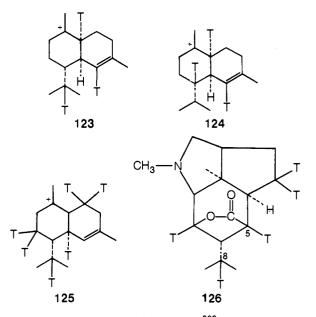
Some details are known of the stereospecificity of the formation of the monocyclic and bicyclic precursors of both tutin (116) and dendrobine (118); this has involved clarification of the steps between *cis,trans*-farnesol (20) and copaborneol (121).

In order to distinguish between the available cyclization schemes, $[4-^{14}C]$ mevalonate (6) was used as a precursor of dendrobine (118).³⁰¹ Degradative work proved to be quite difficult, the aim being to specifically obtain the carbonyl carbon of dendrobine. Eventually this carbon atom, isolated as benzophenone, was shown to contain one-third of the total activity originally present in 118. *cis*, *trans*-Farnesol (20) is therefore a key intermediate in the formation of dendrobine (118).



The initial cyclization of **20** produces a compound with a cation at C-11 which subsequently rearranges to a cation at C-1 of a muurolane derivative.⁴ The use of $[4R^{-3}H]$ mevalonate (6) serves to distinguish some of the possible mechanisms of hydrogen migration involved in this transformation. A double 1,2-hydride shift and cyclization would give a muurolane derivative labeled as in **123**. Alternatively, a 1,3-hydride shift or the intermediate formation of a cyclopropane would leave the tritium originally at C-10 in farnesol (1) intact, giving the muurolane derivative labeled as in **124**. The first mechanism would afford a tutin (**116**) molecule devoid of activity, since C-8 would be labeled and subsequently oxidized. This mechanism was eliminated when tutin (**116**) was found to specifically retain a single tritium at C-4.³⁰²

In order to investigate the possibility of a 1,3-hydride shift, C-1 of farnesol (1) must be labeled, and this carbon atom is derived from C-5 of 6. If the 1,3-hydride shift hypothesis were correct, labeling of the muurolane derivative from $[5-{}^{3}H_{2}]$ mevalonate (6) should be as in 125. Dendrobine would therefore be labeled as shown in 126; i.e., five tritium atoms would be retained including one specifically at C-8. Indeed, it was found that only one-sixth of the tritium was lost in 118.³⁰³ Degradation indicated

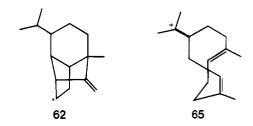


specific tritium labeling at C-8 and at C-5³⁰³ in accordance with the 1,3-hydride shift hypothesis. The remaining activity was specifically located at C-3 and C-11.³⁰⁴ This result also eliminated the possible intermediacy of a cyclopropane derivative.^{305,306}

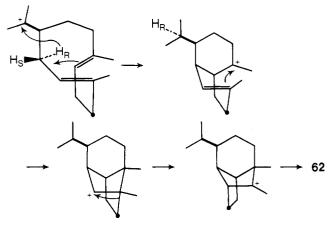
That both hydrogens from C-1 of *trans,trans*-farnesol (1) are retained in dendrobine (118) also indicates a nonoxidative isomerization of 1 to *cis,trans*-farnesol (20). The significance of this result will be discussed subsequently. It is important to note at this point that Arigoni has recently demonstrated that the pro-5*R* proton of **6** specifically migrates from the carbon derived from C-1 of farnesol (1) to C-8 of dendrobine (118).

D. Sativene (62)

In addition to the previously discussed work on avocettin (63) and dendrobine (118) biosynthesis, Arigoni and co-workers have also examined the corresponding stages in the formation of a number of other ring systems. One of these is the compound sativene (62), which, like avocettin (63), may be derived from







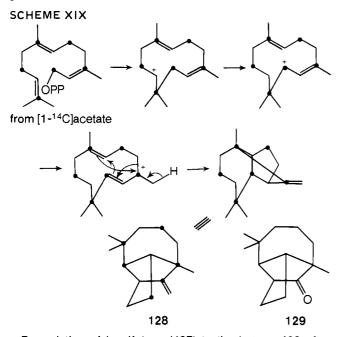
the cation **65.** Having assured the sesquiterpene nature of sativene (**62**), degradation and examination of a number of closely related metabolites established that all of the C-5 protons of mevalonate (**6**) are retained in sativene (**62**). Unlike the situation with avocettin (**63**), but analogous to the data for dendrobine (**118**), it is the pro-5*R* proton which migrates to the isopropyl side chain.¹³⁷ Two further stereochemical points were also established. The isopropyl side chain retains the biogenetic identity of the methyl group (compare tutin (**116**) biosynthesis), and the starred carbon of **62** is also specifically derived from C-2 of **6**.¹³⁷ On the basis of these results Arigoni suggested Scheme XVIII for the biosynthesis of sativene (**62**).

E. Longifolene (127)

Longifolene (**127**), a novel tricyclic sesquiterpene occurring in a number of *Pinus* species,³⁰⁷ was one of the first sesquiterpenes to be studied biosynthetically. Until recently though, only one result^{308,309} had been reported with this compound, and this same 13 years ago.



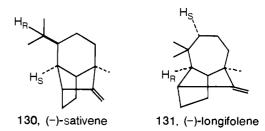
Ourisson³¹⁰ and Hendrickson³ suggested schemes from *cis,trans*-farnesol (**20**) involving a 1,2-shift of the 3,4 bond to 2,4, possibly with simultaneous formation of a 3,7 bond as shown in Scheme XIX. On this basis $[1-^{14}C]$ acetate should label longifolene as shown in **128**.



Degradation of longifolene (**127**) to the ketone **129** after feeding [1-¹⁴C]acetate, and comparison of activities, indicated that none of the original activity had been lost.^{308,309} Specific incorporation of acetate was deduced.

(-)-Sativene (130) and (-)-longifolene (131) co-occur in *Helminthosporium sativum* and *H. victoriae*, ¹³⁷ and it is not difficult to imagine that the latter stages in the biosynthesis of these two metabolites would be extremely similar, the structural differences being due to a difference in the initial cyclization. In the case of longifolene (131), an 11-membered ring is formed, and for sativene (130) a 10-membered ring.

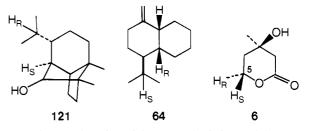
As with sativene (130), there are two possible routes for the



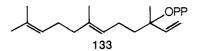
formation of (-)-longifolene (**131**) depending on the enantiomeric nature of the initially formed 11-membered ring compound. Establishment of the specific derivation of the two carbon bridge of (+)-longifolene (**132**) in *Pinus ponderosa*¹³⁷ indicates that formation is direct and does not involve a process which would equilibrate the label in the two-carbon bridges. The stereospecific nature of a 1,3-hydrogen shift to C-3 of (+)-longifolene (**132**) was established.¹³⁷ The migrating hydrogen was shown to be that derived from the pro-5*R* position in mevalonate (**6**), and to be located in an exo orientation.

In the antipodal compound (-)-longifolene (**131**), therefore, we can presume (though it is not established) that the migrating hydrogen is derived from the pro-5*S* position in mevalonate (**6**).

The data for the four compounds investigated by Arigoni and co-workers¹³⁷ are summarized below and have led to a new theory concerning the formation and utilization of germacrene derivatives. The importance (and beauty) of these results cannot be overestimated. They do, however, require considerable explanation, some of it speculative. The facts to be explained may be delineated as follows: (i) although the absolute stereochemistries of (-)-sativene (130) and (-)-longifolene (131) are the same, different hydrogen atoms from C-5 of 6 migrate during ring formation; (ii) copaborneol (121) and (-)-sativene contain isopropyl groups of opposite absolute stereochemistry yet the migrating hydrogen is in each case the same; and (iii) the cadalene derivative (64), however, even though it contains an isopropyl group of the same antipodal nature as that in (-)-sativene (130), is produced by migration of a different hydrogen atom. A common germacrene-type precursor is therefore eliminated in the formation of (-)-sativene (130) and the cadalene derivative (64).



A detailed explanation of these results is beyond the scope of this review. Central, however, is a rationalization of the formation of *cis,trans*-farnesol (20) from *trans,trans*-farnesol (1) *not* involving an oxidation-reduction reaction sequence, but rather nerolidyl pyrophosphate (133), where an inversion of the C-1 protons of farnesol (1) can occur. Previous work had shown that this can be a nonenzymic process.³¹¹



Three parameters were suggested by Arigoni as being important in rationalizing the observed stereochemical preferences for hydrogen migration: (i) the face of attack on the isopropylidene double bond (*re* or *sl*); (ii) the conformation of the cyclizing chain to the olefinic hydrogen of the isopropylidene group (*syn* or *anti*); and (iii) the intermediacy of *cis-* or *trans-*farnesyl pyrophosphates.

The development of this theory precludes an unequivocal

description of the parameters involved in dendrobine (**118**) and avocettin (**63**) biosynthesis, but does permit the conclusion that it is *cis,trans*-farnesylpyrophosphate (**105**) which is involved in the biosynthesis of (–)-sativene (**130**) and (–)-longifolene (**131**). The initial cyclization then takes place on the *re* face of the isopropylidene group from an anti conformation as shown in **134**.



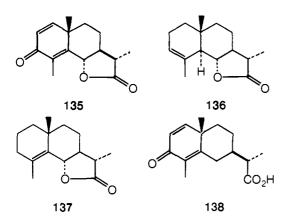
134 re, anti, cis

These results demonstrate that although we may consider apparently similar molecules to be biogenetically derived by common precursors or common routes, Nature almost invariably deludes the organic chemist into a false sense of security, i.e., that we have knowledge of the elaborate processes involved by deductive reasoning without doing the appropriate experimental work.

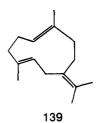
F. Santonin (135)

Santonin (135), one of the classic sesquiterpene lactone derivatives, has been studied only cursorily with respect to its biosynthesis, Barton³¹² having investigated the late stages.

Of the advanced compounds used as precursors, only two were incorporated, **136** and **137**. Compounds such as **138** were not effectively utilized, suggesting that lactone formation precedes oxidation of ring A. However, $[2-^{14}C]$ mevalonate (6) and $[1-^{3}H]$ -*trans*,*trans*-farnesyl pyrophosphate (2) were not incorporated, so that there are clearly some problems in any interpretation of these results.

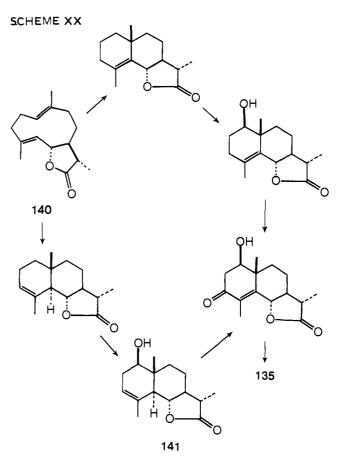


Barton also investigated the stages immediately after the cyclization of farnesol, but with no success.³¹² For example, germacrene B (**139**) was not incorporated. Although incorpo-



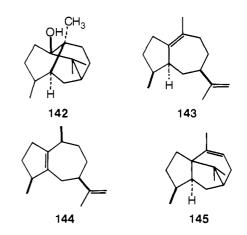
ration data on a number of eudesmenolide derivatives indicated a possible pathway involving 11,12-dihydrocostunolide (140) and 11,12-dihydrosantamarin (141) as key intermediates (see Scheme XX), the low yields and poor incorporation of potentially excellent precursors make these results difficult to interpret.





G. Sesquiterpenes of Pogostemon cablin

Although no extensive data are available, a preliminary study of the formation of some of the sesquiterpenes of *Pogostemon cablin* oil has been carried out.³¹³ Four components of the oil were studied: patchouli alcohol (**142**), α -bulnesene (**143**), α guaiene (**144**), and a mixture of α -patchoulene (**145**) and caryophyllene (**49**). Surprisingly, neither [2-¹⁴C]acetate nor [2-¹⁴C]mevalonate (**6**) was incorporated, whereas [U-¹⁴C]sucrose and glucose were moderately well incorporated into this fraction. Incorporation was rapid as was subsequent metabolism of the labeled materials. Specific terpenoid precursors were apparently unable to reach the site of sesquiterpene synthesis in the leaf disks.



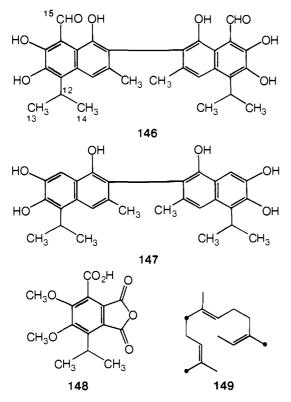
VI. Biosynthesis of Tetracyclic Sesquiterpenes A. Gossypol (146)

The yellow, toxic, phenolic compound gossypol occurs in the resin glands of the seeds,³¹⁴ tap-root bark,³¹⁴ and roots³¹⁵ of the cotton plant *Gossyplum hirsutum*. The structure of gossypol

(146) was first determined by Adams,³¹⁶ but its biosynthesis remained obscure until 1962. Initial work³¹⁵ demonstrated that the excised root tissue produced moderate amounts (1.0%) of gossypol (148) under aseptic conditions.

Feeding experiments with $[1-{}^{14}C]$ - and $[2-{}^{14}C]$ acetates, 317 and subsequent degradation of the labeled gossypol (**146**), demonstrated specific labeling at carbons 12 (from $[2-{}^{14}C]$ - acetate) and 13(14) and 15 (from $[1-{}^{14}C]$ acetate). This is to be expected on the basis of an isoprenoid rather than an acetate pathway. Confirmation of the isoprenoid pathway came from the incorporation (21.9%) of $[2-{}^{14}C]$ mevalonate (**6**)³¹⁷ into gossypol (**146**).

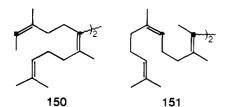
Subsequent work involved the isolation of an enzyme system capable of synthesizing **146** and degradation of the labeled **146**. A key compound in the degradation scheme was apogossypol hexamethyl ether (**147**) in which both C-15 formyl groups have been removed. This compound, **147**, exhibited the same ra-



dioactivity as gossypol (146), thereby confirming the specific incorporation of $[2-^{14}C]$ -6. Similarly, labeled gossic acid (148) retained all the activity of the starting material, thus eliminating a very common mode of cyclization involving *cis,trans*-farnesol (20) as shown in 149, which would have led to a loss of one-third of the label in gossic acid (148). The intact incorporation of a farnesyl unit was demonstrated by this experiment.

The two remaining possible modes of cyclization of **150** and **151** were investigated using geranyl (**57**), neryl (**21**), and farnesyl (**1**) derivatives. The theory behind this simple but elegant series of experiments is shown in **150** and **151**.

If the mode of cyclization were as in 150, [2-14C]geraniol (57)

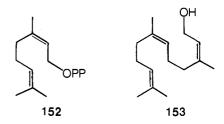


. label from [2-¹⁴C]geraniol (57) or nerol (21)
 . label from [2-¹⁴C]farnesol

should lead to unlabeled gossic acid (148), and $[2^{-14}C]$ farnesol (1) should give labeled gossic acid (148). If cyclization (151) were correct, the exact reverse situation should be observed.

In practice, gossic acid (148) was labeled from $[2^{-14}C]$ geraniol (57) but not from $[2^{-14}C]$ farnesol (1), thereby supporting the mode of cyclization shown in 151.

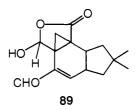
In addition,³¹⁸ all four isomers of farnesol (as the pyrophosphates) were used as precursors. Only the two 2-cis isomers were incorporated and the cis,cis isomer was by far the best precursor. Neryl pyrophosphate (**152**) was a much better precursor than geranyl pyrophosphate (**10**). These results are also in agreement with the mode of cyclization (**151**), which requires a *cis*,*cis*-farnesol (**153**).



Various aspects of the physiological role and dynamics of gossypol (**146**) formation have been studied.^{319,320}

B. Marasmic Acid (89)

The structure of the antibacterial compound marasmic acid from the mold *Marasmus conigenus* was deduced by De Mayo and co-workers to be **89**.³²¹ These workers also investigated the biogenesis of this extremely novel terpenoid derivative. [2-¹⁴C]Mevalonate (**6**) was incorporated into marasmic acid (**89**) to the extent of 0.3% and Kuhn–Roth degradation indicated that marasmic acid (**89**) had one-third of the label at the geminal dimethyl group.



Although no details of the biosynthetic scheme have been elucidated, De Mayo has indicated that one can write a biogenetic scheme for the illudin group of compounds and marasmic acid involving a common precursor as shown in Scheme XXI.³²¹

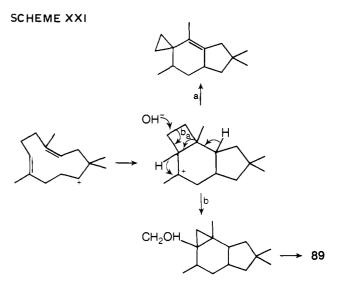


TABLE III. Incorporation Data on the Biosynthesis of Sesquiterpenes

Compound Precursor	Organism	Incorporation	Degradative result	Ref
Abscisic Acid				
[2-'⁴C]Mevalonate	<i>Fragaria</i> cv. T alis m an	NG		164
	Lycopersicon esculentum	0.6		164
	Musa sapientum	NG		164
	Persea gratissima	0.5, 0.4		164,190
[2- ³ H]Mevalonate	Triticum vulgare	0.0013	Nonwilted	166
		0.0012	Wilted, 5 h	166
		0.05	Wilted, 24 h	166
[2- ³ H,]Mevalonate	P. gratissima	0.0022	Intact chloroplast, 2 h	173
	C .	0.005	Lysed chloroplast, 2 h	173
		0.0027	Intact chloroplast, 17 h	173
		0.014	Lysed chloroplast, 17 h	173
		0.022	Intact chloroplast, 17 h	173
			ABA added	
[2 <i>R</i> -³H,2-¹⁴C]Mevalonate	P. gratissima	0.03	Slight ³ H lost	169
[2 <i>R</i> - ³ H,2- ¹⁴ C]Mevalonate	P. gratissima	0.1	Two ³ H lost	169
[4R- ³ H,2- ¹⁴ C]Mevalonate	P. gratissima	0.076	One ³ H lost	165
[4S- ³ H,2- ¹⁴ C]Mevalonate	P. gratissima	0.06	All ³ H lost	165
[5S- ³ H,2- ¹⁴ C]Mevalonate	P. gratissima	0.36	One ³ H retained	169
5-(1,2-Epoxy-2,6,6-tri-	L. esculentum	3.6		166
methylcyclohexyl)-3-	T. vulgare	1.12	None wilted	166
methyl[2-14C]penta-2-		3.42	Wilted, 5 h	166
cis-4-trans-dienoic acid		36	Wilted, 24 h	166
[2-14C,1',2'-18O]Epoxide	T. vulgare	NG		166
5-(1,2-Epoxy-2,6,6-tri-	L. esculentum	0.042		166
methylcyclohexyl)-3-				
methyl[2-14C]penta-2-				
cis,4-trans-dienoic acid				
5-(1,2-Dihydroxy-2,6,6-	L. esculentum	0.017		166
trimethylcyclohexyl)-	L. esculentum	0.017		100
3-methyl[2-14C]penta-2-				
cis,4-trans-dienoic acid				
	7 7 4	0		100
5-(1,2-Dihydroxy-2,6,6-	L. esculentum	0		166
trimethylcyclohexyl)-				
3-methyl[2-14C]penta-2-				
trans-4-trans-dienoic				
acid				
[2-14C]-2-cis-1',2'-epi-	L. esculentum	0		168
Xanthoxin acid				
[2-14C]-2-cis-Xanthoxin	L. esculentum	2.8		168
acid				
[¹⁴C]Phy to ene	Persea gratissima	0		163
Avocettin				
[2-14C]Mevalonate	Anthostoma avocetta	NG	Specific	137
[4 <i>R</i> - ³ H,2- ¹ ⁴ C]Mevalonate	A. avocetta	NG	Incorpn intact	137
[5 <i>R</i> - ³ H,2- ¹⁴ C]Mevalonate	A. avocetta	NG	Incorpn intact	137
[5S-³H,2-1⁴C]Mevalonate	A. avocetta	NG	Incorpn intact, one ³ H	137
			migrates	
[6- ³ H]-trans.trans-	A. avocetta	0.12	Specific	137
Farnesyl pyrophosphate				
Calonectrin				
[2R- ³ H,2- ¹⁴ C]Mevalonate	Calonectria nivalis	1.1	Inc or pn intact	273
[2S-³H,2-'⁴C]Mevalonate	C. nivalis	0.9	Inc or pn intact	273
[5R-³H]Mevalonate	C. niv a lis	NG	Two ³ H retained	273
Carotol				
[1-¹⁴C]Acetate	Daucus carota	NG	Specific	207
Carophyllene				
[2-14C]Mevalonate	Mentha piperata	0.22		187
		0.42	Sucrose added	190
	Nepeta cataria	0.03		188
Coriamyrtin				
[2-14C]Mevalonate	Coriaria japonica	0.004	Specific	296
		NG	Specific	296
Dendrobine				
[2-14C]Mevalonate	Dendrobium nobile	0.012	Specific	295
		NG		301
[4-14C]Mevalonate	D. nobile	0.96	Specific	301
[5-³H₂,2-'⁴C]Mevalonate	D. nobile	1.4	One-sixth ³ H lost	303

Blosynthesis of Sesquiterpenes

Compound Precursor	Organism	Incorporation	Degradative result	Ref
[5 <i>S</i> - ³ H,2- ¹⁴ C]Mevalonate	D. nobile	NG	Two ³ H retained, one ³ H migrates	137
Dendrolasin [U-¹⁴C]GIucose	Lasius fuliginosus	0.15	Lisiferen Ishalina	100
			Uniform labeling	183
[1-14C] Acetate	L. fuliginosus	0.02	Specific	183
[2-14C] Acetate	L. fuliginosus	0.001	Specific	183
Deoxyhelicobasidin [4R-³H,2-'⁴C]-	Helicobasidium mompa	0.7	Two ³ H retained	260
Diacetoxyscirpenol [2-14C]Mevalonate	Fusarium equiseti	1.0		327
Dihydrocoriolin C	- · · ·			
[1-1 ³ C]Acetate	Coriolus consors	NG	Specific	240
[2-1 ³ C] Acetate	C. consors	NG	Specific	240
[1,2- ¹³ C ₂]Acetate	C. consors	NG	Six C-C couplings re- tained	240
[2- ¹³ C]Mevalonate ans.trans-10,11-Epoxy- farnesol	C. consors	NG	Specific	240
[1- ² H]-trans.trans.10,11- Epoxyfarnesal is.trans.10,11-Epoxy-	Helminthosporium sativum	31	Enzyme	136
farnesol $[1-^2H_2]$ -trans.trans-10,11-	H. sativum	NG	E n zyme, 42% of ² H	136
Epoxyfarnesol [1- ² H]-trans.trans-10,11-	H. sativum	17	Enzyme	136
Epoxyfarnesol rans.trans-Farnesal				
[1- ³ H]Geranyl pyro- phosphate + IPP	Citrus sinensis	NG	Enzyme	130
[2-14C] Mevalonate	Samia cynthia	NG		144
$[1,2^{-14}C_2]$ Farnesol s, trans-Farnesal	Tenebrio molitor	NG		328
[1- ³ H]Geranyl pyro- phosphate + IPP	C. sinensis	NG	Enzyme	130
rans.trans-Farnesol [2-14C]Acetate	Hyalophora cecropia	0.004		145
[2-I⁴C] Mevalonate	An duo sumbia a unioutata	0.07		119
	Andrographis paniculata	NG	Enzyme	132
	Beef liver	NG	_	85,106
	Gibberella fujikuroi	NG	Enzyme	111
	Iris hollandica	NG	Enzyme	113
	Nicotiana tabacum	NG		86, 87
	Phytophthora cactorum	NG		76
	Pig liver	NG	Enzyme	329
	Pinus radiata	0.4		117
	Pisum sativum	0.2		118
	Rat liver	NG	Enzyme	53, 54,
	Saccharomyces cereviseae	NG		45, 70
	Samia cynthia	NG		144
	Sarcophaga bullata	0.26		98
	Samia cynthia	NG		144
	Sarcophage bullata	0.26		98
[2-14C] Mevalonate, 5- phospho-	Rat liver	NG		98 54, 56,
[5-3H] Mevalonate	H. cecropia	0.53		145
[2R- ³ H, 2- ¹⁴ C] Mevalonate	Pigliver	NG	Enzyme	82
[2S·3H, 2-14C] Mevalonate	Pig liver	NG	Enzyme	82
$[4R-^{3}H, 2-^{14}C]$ Mevalonate	A. paniculata	NG	Incorph intact	108
	C. sinensis	0.1	Incorph intact	100
	Heve a brasiliensis	NG	Incorph intact	78
	P. cactorum	NG	Incorph intact	76 76
	P. radiata	0.55		
[4S-3H,2-14C] Mevalonate			Enzyme, incorpn intact	109
	A. paniculata	NG 0.07	All ³ H lost	108
	C. sinensis H. huggilizzaia	0.07	All ³ H lost	109
	H. br a siliensis	NG	All ³ H lost	78
	Denstan	NIC .		
	P. cactorum	NG	All ³ H lost	76
(F 20) 0 1401 March 1	P. radiata	1.4	All ³ H lost	109
[5- ² H ₂ ,2- ¹⁴ C] Mevalonate				

448 Chemical Reviews, 1976, Vol. 76, No. 4

Compound Precursor	Organism	Incorporation	Degradative result	Ref
		NG	Enzyme, one ² H lost	66
[5- ³ H ₂ ,2- ¹⁴ C] Mevalonate	A. paniculata	8.5	Enzyme	108
	Pig liver	NG	Enzyme	131
[5-³H,4-'⁴C] Mevalonate	Pig liver	NG	En zyme	69
[1-14C] Isopentenyl pyro-	Gossypium hirsutum	1.56	Enzyme	112
phosphate	G, hirsutum	1.31	Enzyme, iodoacetamide	112
, ,	Micrococcus lysodeikticus		,,	114, 11
	Pig liver	NG		105
	S. cereviseae	Nu		51
(1.140)) (an an a		NO		
[1-14C] Isopentenyl	G. hirsutum	NG	E n zyme	110
pyrophosphate + DMAPP			_	
[1-14C] Isopentenyl		NG	E nz yme	110
pyrophosphate +	G. hirsutum	0.7	Enzyme	112
geranyl pyrophosphate				
[1-14C] Isopentenyl	G. hirsutum	0	Enzyme	112
pyrophosphate +				_
neryl pyrophosphate		NO	F .	100 10
[4-14C] Isopentenyl	Pig Liver	NG	Enzyme	103,12
py r oph o sphate				
[4-14C] Isopentenyl	Pig Liver	NG	Enzyme	104
py r ophosphate +				
DMAPP				
[1,4-14C] Isopentenyl	Pig liver	NG	Enzyme	122, 12
pyrophosphate +			202900	,
geranyl pyrophosphate	– 11		_	
[1- ³ H] Geranyl pyro-	P. radiata		Enzyme	117
ph o sphate				
[1-³H]Geranyl pyro-	C. sinensis	NG	Enzyme	130
phosphate + isopentenyl			-	
pyrophosphate				
	D undiata	0	Enzyme	117
[1- ³ H] Neryl pyrophos-	P. radiata	0	Enzyme	11/
phate		_		
[1- ³ H] Neryl pyrophos-	C. sinensis	0	Enzyme	130
phate + isopentenyl				
pyrophosphate				
[1 <i>R</i> - ³ H,4,8,12- ¹⁴ C ₃]-	A. paniculata	NG	Incorpn intact	132
trans, trans- Farnesol	n puncaura			
[1 <i>R</i> - ³ H,4,8,12- ¹⁴ C ₃]-	A namioulata	NG	All ³ H lost	132
	A. paniculata	NG		152
cis, trans-Farnesol	· · ·			
[1 <i>S</i> - ³ H,4,8,12- ¹⁴ C ₃]-	A. paniculata	NG	Inc or pn intact	132
<i>trans.trans</i> - Farnesol				
[1 <i>S</i> -³H,4,8,12- ¹⁴ C ₃]-	A. paniculata	NG	Incorpn intact	132
cis, trans-Farnesol	·		·	
[1,1,5,5,9,9- ³ H ₆ ,4,8,12-	Trichotecium roseum	2.1	Enzyme, incorpn intact	113
[1,1,3,3,5,5,5]	Tricholecium roseum	2.1	Enzyme, incorph intact	115
⁴ C ₃]-trans.trans-				
Farnesyl pyrophosphate				
<i>trans</i> -Fa rn esol				
[2-'⁴C]Mevalonate	P. radiata	0.1	Enzyme	117
[4 <i>R</i> - ³ H,2- ¹⁴ C]Mevalonate	A. paniculata	NG	Incorph intact	108
	C. sinensis	0.08	Enzyme, incorpn intact	109
	P. radiata	0.2	Enzyme, incorpn intact	109
[4 <i>S</i> -³H,2-'⁴C]Mevalonate	A. paniculata	NG	All ³ H lost	108
	C. sinensis	0	Enzyme	109
	P. radiata	0.27	Enzyme, all ³ H lost	109
[5-³H₂,2-'4C]Mevalonate	A. paniculata	1.5	Enzyme, one-sixth of	108
			³ H label lost	
[1-14C]Isopentenyl pyro-	G. hirsutum	3.2	Enzyme	112
	G. nirsutum	5.2	Enzyme	112
phosphate + geranyl				
pyrophosphate			_	
[1-'⁴C]Isopentenyl pyro-	G. hirsutum	4.4	Enzyme	112
phosphate + neryl pyro-				
phosphate				
[1,4-' ⁴ C ₂] Isopenteny	Pig liver	0		122
pyrophosphate +		~		* ~ ~
		0	F .	1.00
[1,4-14C2]Isopentenyl	Pig liver	0	Enzyme	122
py r ophosphate +				

Biosynthesis of Sesquiterpenes

Compound Precursor	Organism	Incorporation	Degradative result	Ref
[1- ³ H]Geranyl pyrophos- phate	P. radiata			117
[1- ³ H]Geranyl pyrophos- phate + isopentenyl	C. sinensis	NG	Enzyme	130
pyrophosphate [1- ³ H]Neryl pyrophos-	P. radiata	0		117
phate [1- ² H ₂] - <i>trans,trans</i> - Farnesol	H. sativum	NG	Enzyme, 37% of ²H lost	136
[1R- ³ H,4,8,12- ¹⁴ C ₃]- trans.trans-Farnesol	A. paniculata	NG	Incorpn intact	132
$[1R - {}^{3}H, 4, 8, 12 - {}^{14}C_{3}] - cis, trans - Farnesol$	A. paniculata	NG	Incorpn intact	132
[1S- ³ H,4,8,12- ¹⁴ C ₃] - trans,trans-Farnesol	A. paniculata	NG	All ³ H lost	132
[1S- ³ H,4,8,12- ¹⁴ C ₃]- cis.trans-Farnesol	A. paniculata	NG	Incorpn intact	132
[1- ³ H]- <i>trans,trans</i> - Farnesyl pyrophos- phate	P. radiata	0	Enzyme	117
[1,1,5,5,9,9- ³ H ₆ ,4,8,12- ¹⁴ C ₃]- <i>trans.trans</i> - Farnesyl py ro phos- phate	T. roseum	0.4	Enzyme, one-sixth of label lost	168
is,cis-Farnesol [1- ¹⁴ C]Isopentenyl pyro- phosphate + geranyl pyrophosphate	G. hirsutum	0.14	Enzyme	112
[1- ¹⁴ C] Isopentenyl pyro- phosphate + neryl pyrophosphate rans.trans-Farnesoic acid	G. hirsutum	5.2	Enzyme	112
[¹ ⁴ C]HC O ₃	Pseudomonas citronellolis	NG	Enzyme	330
[2-1⁴C]Mevalonate	Rat liver	NG	Enzyme	59,60
trans.trans-Farnesol	T. molitor Rat liver	NG 17	Enzyme	57 60
[1,2-14C ₂]-trans.trans- Farnesol	T. molitor	NG	Liizyiile	328
rans, trans-Nerolidol	N. taka ayaa	NC		87
[2-14C]Mevalonate	N. tabacum P . radiata	NG 0.1	Enzyme	117
	Pisum sativum	0.06	Enzyme	118
	Rat liver	NG	Enzyme	53, 54
	S. bullata	0.06		98
[2-¹⁴C]Mevalonate, 5-phospho-	Rat liver	NG	Enzyme	54, 56
[1- ¹⁴ C]Isopentenyl py ro - phosphate + geranyl pyrophosphate	G. hirsutum	0.8	Enzyme	112
[1-1 ⁴ C]Isopentenyl pyro- phosphate + neryl pyrophosphate	G. hirsutum	0	Enzyme	112
[4- ¹⁴ C] Isopentenyl pyro- phosphate	Pig liver	NG	Enzyme	120
[4- ¹⁴ C] Isopentenyl pyro- phosphate + DMAPP	Pig liver	NG	Enzyme	104
[1,4- ¹⁴ C₂]Isopentenyl pyrophosphate + geranyl pyrophos- phate	Pig liver	NG	Enzyme	122
[1- ³ H]Farnesyl pyro- phosphate Farnesyl-5,7-dihydroxy- 4-methylphthalide	P. radiata	NG	Enzyme	311
[Me- ¹⁴ C]Methionine	Penicillium brevicom- pactum	1.8		221
[7- ¹⁴ C]-5.7-Dihydroxy- 4-methylphthalide	P. brevicompactum	9		221

Compound Precursor	Organism	Incorporation	Degradative result.	Ref
Fumagillin				
[1-14C]Acetate	Aspergillus fumigatus	0.21	Specific	177
[2-14C] Acetate	A. fumigatus	0.2	Specific	177
[2-14C] Mevalonate	A. fumigatus	1.6	Specific	177
Germacrene	Vadauna iamonion	NG		179
[2-14C] Mevalonate	Kadsura japonica K. japonica	NG		179
trans,trans-Farnesyl py ro phosphate	к. јарониса	NG		175
Gossypol				
[1-I4C]Acetate	G. hirsutum	2.1	Specific	317
[2-14C]Acetate	G. hirsutum G. hirsutum	NG	Specific	317
[2-14C] Mevalonate	G. hirsutum	22	Speene	317
		2.2		318
		1.43	Specific	318
[2-14C]Geranyl pyro-	G. hirsutum	1.45	Specific	318
phosphate				
[2- ¹⁴ C]Neryl pyro-	G. hirsutum	2.96	Specific	318
phosphate				
[2-14C] - trans.trans-	G. hirsutum	0.21		318
Farnesyl pyrophos-				
phate				
[2-14C] - cis.trans-	G. hirsutum	0.23		318
Farnesyl py rophos-				
phate				
[2-14C] - trans.cis-	G. hirsutum	1.16	Specific	318
Farnesyl pyrophos-				
phate				
[2-14C] - <i>cis</i> ·cis-	G. hirsutum	5.5		230
Farnesyl py ro ph o s-				
phate				
lelicobasidin				
[1-'⁴C]Acetate	Helicobasidium mompa	0.9	Specific	254
[2-14C]Acetate	H. mompa	0.05	Specific	255
[1- ¹³ C] Acetate	H. mompa	NG	Specific	274
[2-1 ³ C] Acetate	H. mompa	NG	Specific	274
[2-14C]Ethy! malonate	H. mompa	0.07		255
[2-14C]Mevalonate	H. mompa	0.004	Specific	254
		0.11	Specific	255
[2- ¹³ C]Mevalonate	H. mompa	NG	Specific	274
[2-3H2,2-14C]Meva-	Н. тотра	NG	Four ³ H retained	265
lonate	111 1110 111			
[4 <i>R</i> - ³ H,2- ¹⁴ C] -	H. mompa	1.7	Two ³ H retained	260
Mevalonate	11. mompu	NG	Two ³ H retained	265, 26
	11		Two ³ H retained	265
[5- ³ H ₂ ,2- ¹⁴ C]Meva-	H. mompa	NG	Two - H retained	205
Ionate				
[2-³H,2-14C]Geranyl	H. mompa	NG	Incorpn intact	265, 26
py ro phosphate				
lelminthosp or al				
[2-1⁴C]Mevalonate	H. sativum	NG	Specific	210
Hirsutic acid				
[1-' ³ C]Acetate	Stereum complicatum	NG	Specific	237
[2-13C] Acetate	S. complicatum	NG	Specific	237
Iludol M				
[2-14C]Mevalonate	Clitocybe illudens	NG		230
[2- ³ H ₂ ,2- ¹⁴ C]-	C. illudens	0.06, 0.03	Three ³ H lost	234
Mevalonate		,		
[4 <i>R</i> - ³ H,2- ¹⁴ C]-	C. illudens	0.1	Two ³ H l o st	234
Mevalonate	C			
Iludin S				
	C illud			220
[2-14C] Mevalonate	C. illudens	NG		230
lludol				40
[2-1⁴C]Mevalonate	C. illudens	NG		40
pomeamarone				
[2-14C]Acetate	Ipomoea batatas	0.22	Specific	199
		5.45		

Biosynthesis of Sesquiterpenes

Compound Precursor	Organism	Incorporation	Degradative result	Ref
		11.6	Specific	119
		2.52	10−³ M geraniol added	201
		3.71	10 ⁻³ M nerolidol added	119
		4.41	10 ⁻³ M farnesol added	119
		7.0		119
		6.2		202
		4.6	After 2 h	202
	• • • •			
[2-14C]Ethanol	I. batatas	5.9	After 2 h	204
[3-'⁴C]Pyr u vate	I. batatas	1.7		202
[2,4-14C ₂]Citrate	I. batatas	1.0		202
[2-'⁴C]Mevalonate	I. batatas	0.44	Specific	199
		48	Specific	201
[2-14C]Farnesol	I. batatas	6.8		203
sopetasola				
[2-14C]Mevalonate	Petasites hybridus	0.01	Specific	194
uvenile Hormone I	i clustics hybridus	0.01	opeenie	15.
	II and	0		146
Methionine	H. cecropia	0		146
[³H]Methionine	H. cecropia	0.09		145
[¹⁴C]Methionine	H. cecropia	0.014		145
	Schistocerca vaga	0	In vitro	143
Isoleucine	H. cecropia	0		146
Valine	H. cecropia	0		146
Lysine	H. cecropia	0		146
Glucose	H. cecropia	0		146
	-			146
Acetate	H. cecropia	NG		
[1-'⁴C]Acetate	H. cecropia	0		145
[2-'⁴C] Acetate	H. cecropia	0.001		145
	S. vaga	0	In vit ro	143
P ro pionate	H. cecropia	0		146
[1- ¹⁴ C]P ro pionate	H. cecropia	0		145
Mevalonate	H. cecropia	0		146
	H. cecropia	0		144
[2-14C] Mevalonate	Samia cynthia	0		145
[5-³H]Mevalonate	H. cecropia	0		146
4-Methyl-3-pentenol	H. cecropia	0	146	
4-Methyl <i>-cis</i> -3-	H. cecropia	0		146
hexenol				
Farnesol	H. cecropia	0		146
[2-¹⁴C] Farnesol	H. cecropia	0		145
Bishom o fa r nesol	H. cecropia	0		146
10-Epoxy-7-ethyl-3,11- dimethyl-2.6-tride-	H. cecropia	10		149
cadienoic acid				
luvenile Hormone II				
['⁴C]Methionine	S. vaga	0	In vitro	143
[2-'⁴C]Acetate	Manduca sexta	0.11	In v itro , specific	147
[1-'⁴C]P ro pionate	M. sexta	0.05	In vi tro, specific	147
[2-14C]Propionate	M. sexta	1.23	In vitro, specific	147
[2-14C]Mevalonate	M. sexta	0.16	In vitro, specific	147
uvenile Hormone III				
[¹⁴C]Methionine	Gastrimargus africanus	NG	In vitro	143
	M. sexta	71	In vit ro	142
	S. vaga	NG	In vitro	143
[2-'⁴C]Acetate	G. africanus	NG	In vitro	146
	M. sexta	0.04	In vitro, specific	146
	S. vaga	NG	In vitro	143
[1-'⁴C]Propionate	M. sexta	0	In vitro	146
[2-14C] Propionate	M. sexta	0.21	In vitro, nonspecific	146
[2-14C] Mevalonate	M. sexta	0.51	In vitro, specific	146
+)-Longifolene	-		· · · · · · · · · · · ·	- • -
[1- ¹⁴ C]Acetate	Pinus longifolia	0.07	Specific	308, 30
[2-14C]Mevalonate	Pinus ponderosa	0.1	Specific	137
$[5R - {}^{3}H, 2 - {}^{4}C]$ Meva-	P. ponderosa	NG	Incorpn intact, one ³ H	137
lonate			migrated	207
[5 <i>S</i> - ³ H.2-' ⁴ C] Meva-	P. ponderosa	NG	Incorpn intact	137
F	- · · · · · · · · · · · · · · · · · · ·		·····	

Compound Precursor	Organism	Incorporation	Degradative result	Ref
Marasmic acid [2-14C]Mevalonate	Marasmius conigenus	0.3	Specific	24
trans-γ-Monocyclofarnesol	marasmus conigenus	0.0	Speenie	24
[2-14C] Mevalonate	Helminthosporium siccans	25-50		185
y-Muurolene	neiminosportant siecuns	20 00		100
[2- ¹⁴ C]Mevalonate	Mentha piperata	0.1		187
		0.4	Sucrose added	190
Mycophenolic acid				
[¹⁴ C]Methionine	Pinicillium brevicompactum	77	O- and nuclear methyl	215, 216
	P. brevicompactum	10	?	221
[² H ₃]Methionine	P. brevicompactum	NG		326
[¹⁴ C] Acetic acid	P. brevicompactum	0.4	Random	216-218
[1-'⁴C]-3-Methyl-2-	P. brevicompactum	0.49	Random	218
butenoic acid				
[2-¹⁴C]Mevalonic acid	P. brevicompactum	1.5	Side-chain labeled	218
[1- ³ H ₂]Geraniol	P. brevicompactum	low	N o nspecific	2 22
[1-¹⁴C]Geraniol	P. brevicompactum	low	No n specific	222
[1-³H₂]GeranyI pyr o -	P. brevicompactum	low	Nonspecific	222
phosphate				
[1-'⁴C]-4,6-Dihyd ro xy-	P. brevicompactum	0		221, 222
2-methylbenzoic acid				
[6-'⁴C]-4,6-Dihy dro xy-	P. brevicompactum	0.3		221, 226
2-methylbenzoic acid				
[1-'⁴C]-4,6-Dihy dro xy-	P. brevicompactum	83.5		221, 226
2,3-dimethyIbenzoic				
acid				
[1-'⁴C]-4,6-Dihy dro xy-	P. brevicompactum	11.5	Specific	222
2,3•dimethyIbenzoic				
acid				
[7-'⁴C]-5,7-Dihydroxy-	P. brevicompactum	0.15		220, 226
phthalide				
[7-'⁴C]-5,7-Dihy dro xy -	P. brevicompactum	36.5		220, 226
4-methylphthalide		6.2		221
[7-¹⁴C]-7-Hy dro xy-5-	P. brevicompactum	3.8		220, 226
methoxy-4-methyl-				
phthalide				
[7-14C]-6-GeranyI-5,7-	P. brevicompactum	33.6		220, 226
dihydroxy-4-methyl-				
phthalide	- • •			
[7-14C]-6-Farnesyl-5,7-	P. brevicompactum	33. 6		221, 226
dihydroxy-4-methyl-				
phthalide		70.0		001
[7-I⁴C]-O-Demethyl-	P. brevicompactum	78.3		221
mycophenolic acid				
Nobiline	Devile Liver of Lile	0.71		301
[2-14C]Mevalonate	Dendrobium nobile	0.71		301
Petasin [2-'⁴C]Mevalonate	Petasites hybridus	0.01-0:027	SK and F 7997-A ₃	1 9 3
	Felasiles hybridus	0.01-0.027	added	100
Roridin A				
[2- ³ H ₂ ,2- ¹⁴ C]Meva-	Myrothecium verruc ar ia	3.0	Four ³ H retained	266
Ionate				
[5-14C] Mevalonate	M. verrucaria	2.2	Specific	266
[6- ³ H,12,13- ¹⁴ C ₂]-	M. roridum	0.22		267
Farnesyl pyrophos-				
phate				
Santonin				
[2-¹⁴C]Mevalonate	Artemisia maritima	0		312
[1- ³ H]Farnesol	A. maritima	0.0003		312
[1- ³ H]Farnesyl pyro•	A. maritima	0		312
phosphate				
[8- ³ H]Germacrene	A. maritima	0		312
[11- ³ H]-1,2-Dihydro-	A. maritima	0		312
santonin		0.00		21.0
[11- ³ H]-5a,6β,11β-Eu-	A. maritima	0.03		312
desm-3-en-6,13- ol ide		•		312
C 1 1 31 11 CO 1 1 C -				
[11- ³ H]-6β,11β-E u- desm-4-en-6,13- o lide	A. maritima	0		512

Blosynthesis of Sesquiterpenes

Compound Precursor	Organism	Incorporation	Degradative result	Ref
[2,11- ³ H]-6-Deoxysan-	A. maritima	0		312
toninic acid [11-³H]-1,2-Dihydro-6 -	A. maritima	0		312
deoxysantoninic acid [11,12- ³ H ₂]-11,12-	A. maritima	0.004		312
Dihydrocostunolide [11,12- ³ H ₂]-11,12-	A. maritima	0.006		312
Dihydrosantamarin [11-³H]-1,2,4α,5α-	A. maritima	0		312
Tetrahydrosantonin [11- 3 H]-1,2,4 β ,5 α -	A. maritima	0		312
Tetrahydrosantonin	A. mannina	Ū		512
Sativene [2-14C]Mevalonate	Helminthosporium	NG	Specific	137
[5 <i>R</i> -³H,2-'⁴C]-	victoriae H. victoriae	NG	Incorpn intact, one ³ H	137
Mevalonate	II. vietovice	NG		137
[5 <i>S</i> ⁻³ H,2- ¹⁴ C]- Mevalonate Sesquiterpene fraction of Patchouli oil	H. victori a e	NG	Inc orpn intact	137
[U-14C]Glucose	Pogostemon cablin	0.14	Light	313
[U-14C]Sucrose	P. cablin	0.14	Light, 45 h	313
		0.01	Light, 17 h	313
	D orthin	0 0.01	Dark	313 313
[¹⁴C]HCO₂ [2-¹⁴C]Acetate	P. cablin P. cablin	0.01	Light	313
	r. cubun	0	Dark	313
[2-'⁴C]Mevalonate	P. cablin	õ	Bark	313
[U-¹⁴C]Leucine	P. cablin			
Siccanochromene [Ar-³H]Presiccano- chromene	Helminthosporium sicc a ns	25		325
[Ar- ³ H]Epoxypresic-	H. siccans	7 0		325
canochromene Siccanochromenic acid fraction				
[2-14C]Mevalonate Thiobinupharidine	H. siccans	4		324
[2-14C]Mevalonate	Nuphar luteum	0.0 0 7	Specific	213
[3,4-14C ₂]Mevalonate	N. luteum	0.084	Specific	213
[1,5-'⁴C₂]Cadaverine Trichodermol	N. luteum	0.061	Random	213
[2- ³ H₂,2- ¹⁴ C]- Mevalonate	Trichoderma sporulosum		Five ³ H retained	263
[2 <i>R</i> - ³H,2-' ⁴C] - Mevalonate	T. sporulosum	0.05	Two ³ H retained	263, 27
[4 <i>R</i> - ³ H,2- ¹⁴ C]-	Trichoderma sp.	NG	Two ³ H retained	259
Mevalonate		6.9	Two ³ H retained	261
[2-³H,2-'⁴C] -	T. polysporum	NG	Inc orp n intact	268
Mevalonate	T. sporulosum	NG	Incorpn intact	261, 26
[1- ³ H ₂ ,2- ¹⁴ C]- Mevalonate	T. sporulosum	NG	One ³ H lost	261
Trichodiene $[1,1,5,5,9,9,-^{3}H_{6},4,$ $8,12^{-14}C_{3}]$ -trans.trans-	Trichothecium roseum	5.37	One-sixth tritium lost	131
Farnesyl pyrophosphate Trichodiol				
[4R- ³ H,2- ¹⁴ C]- Mevalonate	Trichothecium roseum	0.04		262
[9- ³ H] Trichodiene Trichothecin	T. roseum	2.0		272
[1-14C]Acetate	Trichothecium roseum	0.2	95% act. in c ro tonate	250, 25
[2-¹⁴C]Mevalonate	T. roseum	0.5	Specific	250, 25
[2- ³ H ₂ ,2- ¹⁴ C]	T. roseum	NG, 0.24	Four ³ H retained	261, 26
Mevalonate [2 <i>R-</i> ³H,2-'4C]-	T. roseum	0.55	One ³ H retained	070
Mevalonate	1. 10seunt	0.00		273

TABLE I	11 -	(Continued)
---------	------	-------------

Compound Precursor	Organism	Incorporation	Degradative result	Ref
[2 <i>R</i> -³H,2-¹⁴C]- Mevalonate	T. roseum	0.55	Two ³ H retained	273
[4R - ³ H,2- ¹⁴ C] - Mevalonate	T. roseum	NG	Two ³ H retained	259
[5- ³ H ₂ ,2- ¹⁴ C]-	T. roseum	0.8	Four ³ H retained	261
Mevalonate [5 <i>R</i> - ³ H,2- ¹⁴ C]-	T. roseum	NG	Three ³ H retained	273
Mevalonate [2-³H,2- ¹⁴ C]Geranyl	T. roseum	NG	Inc or pn intact	261, 268
pyrophosphate [1- ¹⁴ C]Farnesyl pyro-	T. roseum	1.51	Specific	257, 258
phosphate [1-³H₂,2-¹⁴C]Farnesyl	T. roseum	NG	One ³ H retained	259
py ro phosphate [2-³H,2- ¹⁴ C]Farnesyl	T. roseum	NG	Inc or pn intact	259
pyrophosphate	-	0.000		• • •
[9- ¹⁴ C] -α-Bisabolol	T. roseum	0.008	Random	269
$[9^{-14}C]\beta$ and γ - bisabolene	T. roseum	0.003	Random	269
[4α,8α,8β,14α,14β- ³ H _s]Trichodermol richothecolone	T. roseum	NG	Three ³ H retained	263
[2- ³ H ₂ ,2- ¹⁴ C] - Mevalonate	T. roseum	0.9	Four ³ H retained	261
[4 <i>R</i> - ³ H,2- ¹⁴ C] -	T. roseum	1.6	Two ³ H retained	261
Mevalonate	1. 1000000	1.2	Two ³ H retained	262
[1- ³ H ₂ ,2- ¹⁴ C]-	T. roseum	NG	One ³ H retained	261
Farnesyl pyrophos- phate	1. / 0.5cum		One micrailed	201
[2- ³ H,2- ¹⁴ C]- Farnesyl pyrophos- phate	T. roseum	NG	Incorpn intact	261
2,13-Epoxytrichothec-9- ene				
[9- ³ H]Trichodiene utin	T. roseum	0.6		272
[2-14C]Mevalonate	Coriaria japonica	0.0004 0.03	Specific Specific	296 297
[4 <i>R</i> -³H,2-'⁴C] - Mevalonate	C. japonica	0.015	Two ³ H lost	302
	C. imperies	0.001	Coordine out twin	200
[- ³ H] Copaborneol	C. japonica	0.001 0.1	Specific, cut twig Specific, plant	300 300
nnamed from <i>H. mompa</i>		0.6	the second	0.00
[4R- ³ H,2- ¹⁴ C]- Mevalonate	H. mompa	0.6	Incorpn tract	260
errucarol	M. want down	NG	Constitu	252
[1- ¹⁴ C]Acetate [2- ¹⁴ C]Acetate	M. roridum M. roridum	NG NG	Specific Ran do m	253 253
[2-14C]Acetate [2-14C]Mevalonate	M. roridum M. roridum	1.0	nanuum	255 244
	m. ronaum	1.0 NG	Specific	244 253
[2- ³ H ₂ ,2- ¹⁴ C]-	M norrhogria	0.45	Approx, five ³ H re-	253
Mevalonate	M. verrucaria		tained	
[2R- ³ H,2- ¹⁴ C]- Mevalonate	M. verrucaria	0.02	Two ³ H retained	266
[2S-³H,2-' ⁴ C]- Mevalonate	M. verruc a ria	0.33	Three ³ H retained	266
[5-14C]Mevalonate	M. verrucari a	1.5	Specific	266
[6- ³ H,12,13- ¹⁴ C ₂]- Farnesyl pyrophos-	M. roridum	0.22	Incorpn intact	267
phate - <i>cis</i> -Xanthoxin acid [2- ¹⁴ C]Mevalonate	P. gratissima	0		168

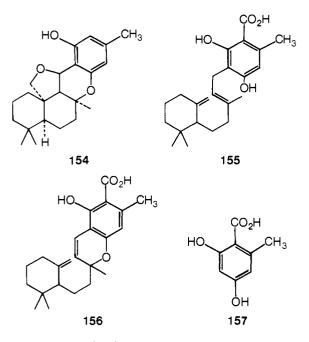
^a Hydrolysis of petasin and isopetasin esters.

VII. Biosynthesis of Pentacyclic Sesquiterpenes

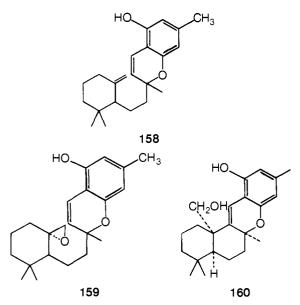
A. Siccanochromenes

As well as *trans*- γ -monocyclofarnesol (**48**) mentioned previously, *Helminthosporium siccans* also produces a novel group of chromene derivatives, the siccanochromenes³²² and the chroman derivative siccanin (**154**).³²³

Two intermediates were isolated (as their methyl esters) after $[2-^{14}C]$ mevalonate (6) indicated that 4% of the total activity was in the carboxylic acid fraction.³²⁴ These intermediates were shown to be presiccanochromenic acid (155) and siccanochromenic acid (156).



The formation of **156** apparently occurs by alkylation of orsellinic acid (**157**) with *trans-* γ -monocyclofarnesol (**48**). Evidence for the reaction sequence after **155** comes from the isolation³²² and feeding of the compounds **158** and **159**.³²⁵ Tritiated **158** was incorporated into siccanin (**154**) to the extent of 25 %, whereas 70 % of the activity was in **154** when labeled **159** was a precursor.³²⁵



An additional intermediate (160) was also isolated,³²⁴ so that we may formulate the sequence as orsellinic acid (157) + *trans*- γ -monocyclofarnesol (48) \rightarrow 155 \rightarrow 156 \rightarrow 158 \rightarrow 159 \rightarrow 160 \rightarrow siccanin (154).

VIII. Summary and Conclusions

Biosynthetic data for the sesquiterpenes are summarized in Table III.

The biosynthesis of sesquiterpenes, as with all biosynthetic work, is at an extremely exciting stage of development. The work of Arigoni¹³⁷ and the increasing use of ¹³C NMR indicate that the challenges of deriving sophisticated biosynthetic schemes can be met with adequate effort and intellect. The wide range of pharmacologic activity exhibited by this group of compounds merits increasing attention to their precise biosynthetic derivation.

IX. Addendum

Since the completion of the main body of this review, a number of articles have appeared which bear directly on the biosynthesis of sesquiterpenes.

II.B. Juvenile Hormone

A recent review³³¹ has summarized the biosynthesis, secretion, and action of Cecropia juvenile hormone, and a full paper on the previously discussed work of Ajami and Riddiford has appeared.³³²

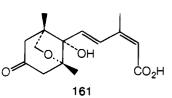
Peter and Dahm³³³ have studied the biosynthesis of JH I (27) and JH II (28) using variously labeled propionates. [1-¹⁴C]Propionate was specifically incorporated into only C-7 and C-11 of 27. Propionate thus serves as a precursor of homomevalonate in agreement with the previous work of Schooley and coworkers.¹⁴⁷ Both [2-¹⁴C]- and [3-¹⁴C]propionates gave rise to extensive randomization of label.

As an extension of their original work, ¹⁴⁷ Schooley and coworkers³³⁴ have examined the importance of $[5-{}^{3}H_{2}]ho$ momevalonate as a precursor of JH I (27) and JH II (28). Degradation of the labeled JH II (28) as discussed previously indicated that the only site of the label was at the predicted position C-9.

III.A. Abscisic Acid

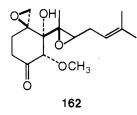
Milborrow, ³³⁵ continuing his work on the biosynthesis of abscisic acid (**37**), has elucidated the specificity of mevalonate in the 6'-methyl groups of abscisic acid (**37**) and phaseic acid (**161**). The latter compound occurs as a single isomer and is therefore admirable for such a study.

Degradation of **161** after feeding abscisic acid (**37**) biosynthesized from $[2^{-14}C, 2^{-3}H_2]$ mevalonate (**6**) demonstrated the *pro*-6^{*t*}*R*-methyl group was specifically derived from C-2 of mevalonate.³³⁵



As a double check on the above work, the derivation of abscisic acid (**37**) from mevalonic acid specifically labeled at the methyl group was investigated. If C-2 and C-3' of mevalonate retain their integrity throughout the biosynthetic process, the carbon atoms derived from C-3' of mevalonate should appear as the pro-6'*S*, the C-2', and the C-3 methyl groups of **37**. (+)-Abscisic acid (**37**) derived from $[3'-{}^{14}C,3'-{}^{3}H]$ mevalonic acid in avocado showed no loss of label.³³⁶ Treatment with base, however, increased the original ${}^{14}C:{}^{3}H$ ratio from 3:3 to 3:2. The exchange of ${}^{3}H$ at the C-2 methyl group was confirmed by modified Kuhn–Roth oxidation. Consequently, C-2 and C-3' of mevalonate do retain their integrity in the formation of **37** and at no point become equivalent. III.B. Ovalicin

Two groups have investigated the biosynthesis of ovalicin (162), a close relative of fumagillin (43).



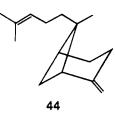
In the first study, conducted by Tanabe and Suzuki³³⁷ [1,2- $^{13}C_2$] acetate was used as a precursor. Six intact acetate units were incorporated, thereby demonstrating that no methyl group migrations were involved in the biosynthesis.

Cane and Levin,³³⁸ using $[4-^{13}C]$ mevalonate (6) as a precursor showed that enrichment was specific at C-1, C-3, and C-10.

Very recently Cane and Levin³³⁹ have extended their work on ovalicin (**162**) with the use of $[3,4-^{13}C_2]$ mevalonate. [The author is very grateful to Dr. David Cane, Brown University, for a copy of a preprint of this paper prior to publication.] Carbons 1 and 6 of ovalicin (**162**), derived from C-4 and C-3, respectively, of the same mevalonate molecule, gave rise to a pair of doublets (J = 53 Hz). Carbon 7 derived from C-3 of mevalonate appears as an enhanced singlet uncoupled to the adjacent C-1.

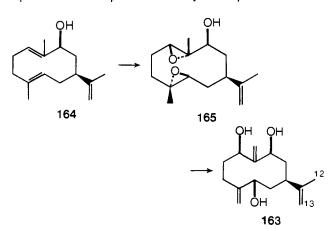
Similarly C-10 and C-11, derived from carbon atoms 4 and 3 of mevalonate, respectively, appeared as doublets (J = 74 Hz) and must therefore be derived from the same molecule of mevalonate.

The data of Cane and Levin^{338,339} complement those of Tanabe³³⁷ and are additional evidence for the intermediacy of β -bergamotene (44) in the biosynthetic scheme, as originally suggested by Birch¹⁷⁷ for fumagillin (43).





The monocyclic sesquiterpenoid ageratriol (**163**) is the principal sesquiterpene of *Achillea ageratum*³⁴⁰ and may be formulated as being derived from agerol ((+)-9 β -hydroxygermacrene A) (**164**) which co-occurs with ageratriol.³⁴¹ [9-³H]-Agerol (**164**) was administered³⁴² and a small specific incorporation was observed into both ageratriol (**163**) and agerol diepoxide (**165**), a postulated biosynthetic precursor. The latter compound labeled at C-13 was incorporated into ageratriol (**163**). Degradation of the labeled ageratriol (**163**) after feeding [2-¹⁴C]-and [2-³H]mevalonate indicated³⁴² that C-12 and C-13 become equivalent at some point in the biosynthetic process.

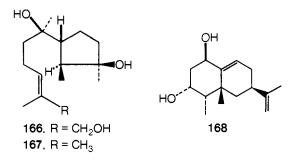


III.F. Cyclonerodiol and Cyclonerotriol

Cyclonerotriol (**166**) and the closely related cyclonerodiol (**167**) from *Fusarium culmorum*³⁴³ have recently been the subject of study by Hanson and co-workers.³⁴⁴

Using $[4,5^{-13}C_2]$ mevalonate the folding of the farnesyl chain was established since all three $^{13}C^{-13}C$ couplings remained. Three other precursors were also used. With $[2^{-3}H_2,2^{-14}C]$ -mevalonate considerable tritium was removed by the action of a prenyl isomerase, but $[4R^{-3}H,2^{-14}C]$ - and $[5^{-3}H_2,2^{-14}C]$ -mevalonates were incorporated without loss of label.

all-trans-Farnesyl pyrophosphate (2) was a precursor of cyclonerodiol (167), which itself was a precursor (15% incorporation) of the triol. Nerolidol was not a precursor, indicating that an enzyme bound intermediate of some type is probably involved.



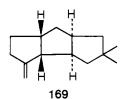
IV.C. Capsidiol

An additional member of the erimophilane group of sesquiterpenoids has been investigated in order to distinguish between the classical¹⁸⁶ and Lawtons spiromechanism for the valencane skeleton.¹⁹⁵ Capsidol (**168**) (a 4-epieremophilane) from *Capsicum frutescens* was labeled by $[2^{-14}C]$ mevalonate and $[2^{-14}C]$ acetate to the extent of 1-3%. In order to evaluate the possibility of a migration of a methyl group from C-10 to C-5, $[1,2^{-13}C_2]$ acetate was used as a precursor.³⁴⁵ Since no $^{13}C^{-13}C$ coupling was observed between C-5 and C-15, these carbon atoms must be derived from different acetate units; instead C-5 was coupled to C-6. This result is in agreement with a mechanism involving a methyl migration. If a spiro intermediate were involved, the C-5 and C-15 carbons would have remained paired.

V.A. Illudins and Coriolins

Matsumoto and co-workers³⁴⁶ have synthesized equivalents of the cation **83**, a probable intermediate in the biosynthesis of the illudins.

From *Coriolus consors* Shibata and co-workers³⁴⁷ have isolated hirsutene (**169**) (pre **91** in Scheme XIII), a postulated precursor of hirsutic acid, complicatic acid, and the coriolins. The structure of hirsutene was confirmed by synthesis.



V.B. Trichothecin and Verrucarin

Tamm has published a series of papers on his extensive work on the biosynthesis of verrucarin and roridin.^{348–351}

Verrucarin A and roridin A differ in the structure of their diacid components, but the sesquiterpene in both cases is verrucarol (93).

Using labeled $[1-^{14}C]$ - and $[2-^{14}C]$ acetate and $[2-^{14}C,2-^{3}H_2]$ mevalonate, it was shown that the *cis,trans*-muconic acid unit derives from mevalonate.³⁴⁸

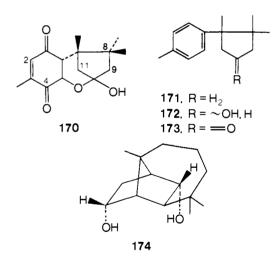
Previous work^{261,263,273} had demonstrated that hydroxylation at C-4 of the trichothecane skeleton was stereospecific with an overall retention of configuration. This has now been confirmed for verrucanol using mevalonate stereospecifically tritiated at C-2. This experiment also confirmed the derivation of C-8 of the trichothecane skeleton from C-2 of mevalonate.

Tamm also investigated the mode of cyclization of trichodiene $(109)^{351}$ in the same way as Hanson.²⁷³ Using $[5R-^{3}H]$ meva-Ionate it was conclusively shown that verrucarol (93) retained three ³H labels and that these were located at C-11, C-3, and probably C-7. That ³H is retained at C-11 indicated a stereospecific loss of the pro-1S proton of farnesol in the formation of trichodiene (109). An entirely analogous result was obtained by Hanson and Evans³⁵² using a cell-free system from Trichothecium roseum. The precise mechanism of the formation of trichodiene remains to be determined particularly whether an oxidation-reduction is involved in the cis-trans isomerization of C-2,3.

Bisabolene derivatives were not incorporated into verrucarol (93).³⁵¹

V.B. Lagopodin

Lagopodin B (170), a guinone antibiotic from Coprinus lagopus,³⁵³ has also been the subject of study by Arigoni's group. The author thanks Dr. David Cane, Brown University, for provision of relevant manuscripts and data prior to publication.] [2-14C]Mevalonate was specifically incorporated into 170,354 degradation localizing a third of the activity at C-11, a further third at the C-8 α -methyl, and one-sixth each at C-2 and C-4. On this basis a scheme involving cuparene (171) as an intermediate was proposed.³⁵⁴ Subsequently, a number of intermediates including cuparene (171), cuparenol (172), and cuparenone (173) were shown to be efficient precursors of 170.355 With the overall



scheme evaluated, attention was turned to the mechanism of the second cyclization from a bisabolene-type compound to eventually give cuparene (171). [4R-3H,2-14C]Mevalonate was specifically incorporated into 170. Degradative work indicated that a half tritium equivalent was at C-2, and that the remaining two equivalents of tritium were localized at C-9. In the cyclization, therefore, a 1,4-hydride shift from C-6 to C-9 occurs and a bisabolene intermediate ($\Delta^{6.7}$) is eliminated. The results are in agreement with those of previous workers 168,261,265-268 on the biosynthesis of helicobasidin (96) and trichodermin (107). V.C. Dendrobine

Jommi and co-workers³⁵⁶ have published additional details of the biosynthesis of dendrobine (118). Incorporation of [1-³H₂]-2-trans-6-trans-farnesol (1) into dendrobine (118) in Dendrobium nobile and degradation indicated that 52% of the label was located at C-5 and 47 % at C-8. The 2-cis isomer was not incorporated. [1S-3H]-2-trans-6-trans-Farnesol (1) was incorporated into dendrobine (118), and 88% of the label was located at C-5. The hydride shift from C-1 of farnesol is therefore highly stereospecific with the 1-pro-R hydrogen being transferred.

Farnesal is therefore not an intermediate in the cis-trans isomerization of farnesol. These results are analogous to those of Arigoni³⁵⁷ who used stereospecifically labeled mevalonate.

V.E. Culmorin

Another tricyclic sesquiterpene which has recently been investigated is culmorin (174), from Fusarium culmorum.358 Hanson and Nyfeler³⁵⁹ using [2-³H₂]-, [4R-³H]-, and [5-³H₂]mevalonate (administered with [2-14C]mevalonate) incorporated them into culmorin (174). By degradation, the labeling pattern associated with C-1, C-8, C-10, and C-11 was formed to be in accord with a formation from farnesyl pyrophosphate. From [2S-3H] mevalonate it was demonstrated³⁶⁰ that the endo-C-10 proton of culmorin (174) originates from the 2S position of mevalonate. The total number of hydrogens derived from [5-³H₂]mevalonate (five of six), however, indicated that a migration of [5-3H] label had occurred, possibly from C-7 to C-5.359

Degradative work placed a pro-5S-mevalonoid hydrogen at C-11 and a pro-5*R*-mevalonoid hydrogen at C-7. It is C-7 which originates from C-1 of farnesol and consequently during the course of cyclization the pro-1S hydrogen of farnesyl pyrophosphate has migrated (probably to C-5).

This is in agreement with the work of Arigoni and co-workers^{357,361} on the biosynthesis of sativene and longifolene. The stereochemistry of the loss of hydrogen from C-1 is therefore dependent upon the size of the initial ring formed. When C-1 of farnesyl pyrophosphate attacks the distal double bond, a cyclopropyl ion is formed, which subsequently collapses in one of two ways: (i) to a 10-membered ring with migration of the pro-1R hydrogen, e.g. dendrobine; (ii) to an 11-membered ring with migration of the pro-1S hydrogen, e.g., culmorin, sativene, longifolene. In either instance secondary reactions subsequently take place by reaction of the carbonium ion at C-1 of the farnesyl system.

X. References

- (1) L. Ruzicka, A. Eschenmoser, and H. Heuser, Experientia, 9, 357 (1953).
- (2) A. Eschenmoser, L. Ruzicka, O. Jeger, and D. Arigoni, Helv. Chim. Acta, 38, 1890 (1955).
- J. B. Hendrickson, Tetrahedron, 7, 82 (1959).
- W. Parker, J. S. Roberts, and R. Ramage, Q. Rev., Chem. Soc., 21, 331 (4)
- (1967).
 (5) T. K. Devon and A. I. Scott, "Handbook of Naturally Occurring Compounds", Vol. II, "Terpenes", Academic Press, New York, N.Y., 1972. N. H. Andersen and D. D. Syrdal, Tetrahedron Lett., 899 (1972).
- D. H. R. Barton and P. De Mayo, Q. Rev., Chem. Soc., 11, 189 (1957). (8) A. J. Birch in Ciba Found. Symp. Quinones Electron Transp., 1961, 233
- (1961).
- (9) A. J. Birch, Proc. Chem. Soc., London, 3 (1962). (10) T. G. Halsall and D. W. Theobald, *Q. Rev., Chem. Soc.*, 16, 101
- (1962).
- O. Karlstrom, Sven. Kem. Tidskr., 74, 523 (1962).
- (12) H. J. Nicholas in "Biogenesis of Natural Compounds", 2nd ed, P. Bernfeld, Ed., Pergamon Press, New York, N.Y., 1963, p 641.
- (13) L. Ruzicka, Pure Appl. Chem., 6, 493 (1963).
- W. B. Whalley in ref 12, p 797.
- (15) J. H. Richards and J. B. Hendrickson, "Biosynthesis of Terpenes, Steroids, and Acetogenins", W. A. Benjamin, New York, N.Y., 1964, p 225. J. B. Hendrickson, "The Molecules of Nature", W. A. Benjamin, New York, (16)
- N.Y., 1965, p 110.
- (17) R. B. Clayton, *Q. Rev., Chem. Soc.*, 19, 201 (1965).
 (18) G. Popjak and J. W. Cornforth, *Biochem. J.*, 101, 553 (1966).
- (19) H. V. Daeniker, Givaudanian, 3 (1966).
- (20) A. J. Birch, Science, 156, 202 (1967).
 (21) F. W. Hemming in "Terpenoids in Plants", J. B. Pridham, Ed., Academic Press, London, 1967.
- (22) P. Teisseire, Recherches, 18, 77 (1969).

- (22) G. Popjak in Methods Enzymol., 15, 393 (1969).
 (24) R. Ramage, Flavour Ind., 1, 323 (1970).
 (25) S. Nozoe, Kagaku To Seibutsu, 8, 332 (1970).
 (26) J. W. Cornforth and R. H. Cornforth in "Natural Substances Formed Biologically From Mevalonic Acid", T. W. Goodwin, Ed., Academic Press, London, 1970, p 5.
- (27) G. Popjak in ref 26, p 17.
- V. Herout in "Aspects of Terpenoid Chemistry and Biochemistry", T. W. Goodwin, Ed., Academic Press, London, 1971, p 53. (28)
- (29) G. P. Moss, J. Soc. Cosmet. Chem., 22, 231 (1971).
- (30)R. Rudnicki, Postepy Biochem., 17, 43 (1971 (31) G. P. Moss in Terpenoids Steroids, 1, 221 (1971).
- (32)
- J. S. Roberts in "Chemistry of Terpenes and Terpenoids", A. A. Newman, Ed., Academic Press, London, 1972, p 88.

- (33) D. V. Banthorpe and B. V. Charlwood In ref 32, p 337.
 (34) G. Jommi, *Cron. Chim.*, No. 35, 20 (1972).
 (35) W. B. Turner, "Fungal Metabolites", Academic Press, New York, N.Y., 1972.
- (36) M. Luckner, "Secondary Metabolism in Plants and Animals", Academic Press, New York, N.Y., 1972, p 129.

- Press, New York, N.Y., 1972, p 129.
 (37) G. P. Moss in *Terpenoids Steroids*, 2, 197 (1972).
 (38) H. J. Nicholas in "Phytochemistry", Vol. II, L. P. Miller, Ed., Van Nostrand-Reinhold, New York, N.Y., 1973, p 254.
 (39) D. E. Wolf, C. H. Hoffman, P. E. Aldrich, H. R. Skeggs, L. D. Wright, and K. Folkers, *J. Am. Chem. Soc.*, 78, 4499 (1956).
 (40) L. D. Wright, E. L. Cressow, H. R. Skeggs, G. D. E. MacRae, C. H. Hoffmann, D. E. Wolf, and K. Folkers, *J. Am. Chem. Soc.*, 78, 5273 (1956).
 (41) D. E. Wolf, C. H. Hoffman, P. E. Aldrich, H. R. Skeggs, L. D. Wright and K. Folkers, *J. Am. Chem. Soc.*, 79, 1486 (1957).
 (42) K. Folkers, C. H. Shunk, B. O. Linn, F. M. Robinson, P. E. Wiltreich, J. W. Huff, J. L. Gilfillian, and H. R. Skeggs in *Ciba Found. Symp. Biosynth. Terpenes Sterols*, 1956, 135 (1959). Terpenes Sterols, 1958, 135 (1959).
- (43) G. Tamura, J. Gen. Microbiol., 2, 431 (1956).
- (44) P. A. Tavormina, M. H. Gibbs, and J. W. Huff, J. Am. Chem. Soc., 78, 4498 (1956). (45) F. Lynen, H. Eggerer, U. Henning, and I. Kessel, Angew. Chem., 70, 738
- (1958).
- (46) O. Isler, R. Ruegg, G. Saucy, J. Wursch, K. F. Gey, and A. Pletscher in *Ciba Found. Symp. Biosynth. Terpenes Sterols*, 1958, 135 (1959).
 (47) L. D. Wright and M. Cleland, *Proc. Soc. Exp. Biol. Med.*, 96, 219 (1957).
- (48) F. Dituri, J. L. Rabinowitz, R. P. Hullin, and S. Gurin, J. Biol. Chem., 229, 825 (1957).
- (49) W. Sandermann and H. Stockmann, Naturwissenschaften, 43, 581 (1956), (50) H. Rilling and K. Bloch, unpublished results, quoted in ref 62
- (51) F. Lynen, B. W. Agranoff, H. Eggerer, U. Henning, and E. M. Moslein, Angew. Chem., 71, 657 (1959).
 (52) K. Bloch, S. Chaykin, A. H. Phillips, and A. de Waard, J. Biol. Chem., 234,
- 2595 (1959).
- G. Popjak, Tetrahedron Lett., 19 (1959). (53)
- (54) DeW. S. Goodman and G. Popjak, *J. Lipid Res.*, 1, 286 (1960).
 (55) U. Henning, E. M. Moslein, and F. Lynen, *Arch. Biochem. Biophys.*, 83, 259 (1959).
- (56) DeW. S. Goodman and G. Popjak, *Biochem. J.*, 74, 35 (1960).
 (57) F. Dituri, J. L. Rabinowitz, R. P. Hullin, and S. Gurin, *J. Biol. Chem.*, 229, 825 (1957).
- J. Christophe and G. Popjak, J. Lipid Res., 2, 244 (1961). (58)
- G. Popjak in Ciba Found. Symp. Biosynth. Terpenes Sterols, 1958, 148 (59)
- (1959) (60) G. Popjak, M. Horning, H. L. R. Bucher, and R. H. Cornforth, Blochem. J., 72.34 (1959).
- (61) B. H. Amdur, H. Rilling, and K. Bloch, J. Am. Chem. Soc., 79, 2647 (1957).
 (62) H. Rilling, T. T. Tchen, and K. Bloch, Proc. Natl. Acad. Sci., U.S.A., 44,
- 167 (1958).
- (63) K. Bioch in Ciba Found. Symp. Biosynth. Terpenes Sterols, 1958, 4 (1959).
 (64) H. C. Rilling and K. Bloch, J. Biol. Chem., 234, 1424 (1959).
 (65) DeW. S. Goodman and G. Popjak, Proc. Int. Conf. Biochem. Lipids, 6th.
- 1960, 278 (1960); Chem. Abstr., 58, 128137 (1963).
 (66) G. Popjak, DeW. S. Goodman, J. W. Cornforth, R. H. Cornforth, and R. Ryhage, Biochem. Biophys. Res. Commun., 4, 138 (1961).
- (67) G. Popjak, DeW. S. Goodman, J. W. Cornforth, R. H. Cornforth, and R.
- Ryhage, *J. Biol. Chem.*, **236**, 1934 (1961). (68) G. Popjak, J. W. Cornforth, R. H. Cornforth, R. Ryhage, and DeW. S. Goodman, *J. Biol. Chem.*, **237**, 56 (1962).
- (69) C. Donninger and G. Popjak, *Biochem. J.*, **91**, 10P (1964).
 (70) C. R. Childs and K. Bloch, *J. Biol. Chem.*, **23**7, 62 (1962).
- (71) G. Popjak, G. Schroepfer, and J. W. Cornforth, Biochem. Biophys. Res. Commun., 6, 438 (1961).
- (72) B. Samuelsson and DeW. S. Goodman, Biochem. Biophys. Res. Commun., 11, 125 (1963).
- (73) G. Popjak, *Biochem. J.*, 96, 1P (1965).
 (74) J. W. Cornforth, R. H. Cornforth, C. Donninger, and G. Popjak, *Proc. R.* (74) (14) J. W. Guindin, I. J. Schman, et al. (1966).
 (75) K. J. Stone and F. W. Hemming, *Biochem. J.*, **104**, 43 (1968).
 (76) J. B. Richards and F. W. Hemming, *Biochem. J.*, **128**, 1345 (1972).
 (76) J. B. Richards and F. W. Hemming, *Biochem. J.*, **128**, 1345 (1972).
- (77) T. W. Goodwin and R. J. H. Williams, Proc. R. Soc. London, Ser. B, 163, 515 (1966).
- (78) B. L. Archer, D. Barnard, E. G. Cockbain, J. W. Cornforth, R. H. Cornforth, and G. Popjak, *Proc. R. Soc. London, Ser. B*, **163**, 519 (1966).
 (79) A. R. Wellburn, K. J. Stone, and F. W. Hemming, *Biochem. J.*, **100**, 230
- (1966).
- (80) B. L. Archer and D. Barnard, *Biochem. J.*, 96, 1P (1965).
 (81) B. L. Archer and E. G. Cockbain in *Methods Enzymol.*, 15, 476 (1969).
- (82) K. Clifford, J. W. Cornforth, R. Mallaby, and G. T. Phillips, Chem. Commun., 1599 (1971).
- (83) J. W. Cornforth in "Biosynthesis and its Control in Plants", B. V. Milborrow, Ed., Academic Press, London, 1973, p 178.
 (84) D. A. Beeler, D. G. Anderson, and J. W. Porter, Arch. Biochem. Biophys., 2017 (2017)
- 102, 26 (1963).
- (85) L. Gosselin and J. Duvivier, Arch. Int. Physiol. Biochim., 73, 367 (1965).
- (86) P. Benveniste, L. Hirth, and G. Ourisson, C. R. Acad. Sci., Ser. D. 265, 1749 (1967).
- P. Benveniste, G. Ourisson, and L. Hirth, Phytochemistry, 9, 1073 (1970).
- (a) P. Berveniste, G. Odrisson, and L. nirdt, Phylochemistry, 9, 1073 (1970).
 (88) I. Shechter, Biochim. Biophys. Acta, 316, 222 (1973).
 (89) A. J. Clark and K. Bloch, J. Biol. Chem., 234, 2578 (1959).
 (90) L. I. Gilbert and R. D. Goodfellow, Zool. Jahib., Abt. Alig. Zool. Physiol. Tiere, 71, 718 (1965), and references cited therein.
 (91) R. D. Goodfellow and G. C. K. Liv, J. Insect Physiol., 19, 95 (1972).
 (92) T. T. Tchen, J. Biol. Chem., 233, 1100 (1958).
 (93) H. B. Lawy and G. Ropick, Biochem. 4, 72, 359 (1950).

- (93) H. R. Levy and G. Popjak, Biochem. J., 72, 35P (1959).

- (94) H. R. Levy and G. Popjak, Biochem. J., 75, 417 (1960).
- (95) K. Markely and E. Smallman, Biochim. Biophys. Acta, 47, 327 (1961) (96) L. J. Rogers, S. P. J. Shah, and T. W. Goodwin, Biochem. J., 99, 381 (1966).

Geoffrey A. Cordell

- (97) R. D. Goodfellow and F. J. Barnes, Insect Biochem., 1, 271 (1971).
- (98) R. D. Goodfellow, Y.-S. Huang, and H. E. Radtke, Insect Biochem., 2, 467 (1972)
- (99) J. K. Dorsey and J. W. Porter, J. Biol. Chem., 243, 4667 (1968).
- (100) A. P. F. Flint, Biochem. J., 120, 145 (1970).
- (101) F. Lynen in Ciba Found., Symp. Quinones Electron Transp., 1961, 244 (1961).
- (102) A. A. Kandutsch, H. Paulis, E. Levin, and K. Bloch, J. Biol. Chem., 239, 2504 (1964).
- (103) C. R. Benedict, J. Kett, and J. W. Porter, Arch. Biochem. Biophys., 110, 611 (1965).
- (104) J. K. Dorsey, J. A. Dorsey, and J. W. Porter, J. Biol. Chem., 241, 5353 (1966).
- (105) P. W. Holloway and G. Popjak, Biochem. J., 100, 61P (1966).
- (106) L. Gosselin and J. Duvlvier, *Bull. Soc. Chim. Biol.*, 47, 359 (1965).
 (107) K. Kirschner, Ph.D. Thesis, Ludwig Maximilians-Universitat, Munich. Germany, 1961.
- (108) K. H. Overton and F. M. Roberts, J. Chem. Soc., Chem. Commun., 378 (1973).
- (109) E. Jedlicki, G. Jacob, F. Faini, and O. Cori, Arch. Biochem, Biophys., 152, 590 (1972). (110) K. Ogura, T. Nishino, and S. Seto, *J. Biochem. (Tokyo)*, **64**, 197 (1968).

- (111) I. Shechter and C. A. West, *J. Biol. Chem.*, 244, 3200 (1969).
 (112) S. R. Adams and P. F. Heinstein, *Phytochemistry*, 12, 2167 (1973).
 (113) G. L. Staby, W. P. Hackett, and A. A. De Hertogh, *Plant Physiol.*, 52, 416 (1973).
- A. A. Kandutsch, H. Paulus, E. Levin, and K. Bloch, J. Biol. Chem., 239, (114)
- 2507 (1964). (115) C. M. Allen, W. Alworth, A. Macrae, and K. Bloch, J. Biol. Chem., 242, 1895 (1967).
- A. A. Kandutsch in Methods Enzymol., 15, 490 (1969).
- (117) G. Jacob, E. Cardemil, L. Chayet, R. Tellez, R. Pont-Lezica, and O. Cori, *Phytochemistry*, **11**, 1683 (1972).
 (118) C. J. Pollar, J. Bonner, A. J. Haagen-Smit, and C. C. Nimmo, *Plant Physiol.*,
- 41, 66 (1966).
- (119) I. Oguni, K. Oshima, H. Imaseki, and I. Untani, Agric. Biol. Chem., 33, 50 (1969).
- (120) G. Popjak, P. W. Holloway, and M. Barron, Biochem. J., 111, 325 (1969).
- (121) G. Popjak, J. L. Rabinowitz, and J. M. Baron, Biochem. J., 113, 861 (1969).
- (122) P. W. Holloway and G. Popjak, *Blochem. J.*, **104**, 57 (1967).
 (123) G. Popjak, P. W. Holloway, R. P. M. Bond, and M. Roberts, *Blochem. J.*.
- 111, 333 (1969). (124) K. Ogura, T. Nishino, T. Koyama, and S. Seto, J. Am. Chem. Soc., 92, 6036
- (1970). (125) T. Nishino, K. Ogura, and S. Seto, Biochim. Biophys. Acta, 235, 322
- (1971).(126) K. Ogura, T. Koyama, and S. Seto, J. Chem. Soc., Chem. Commun., 881
- (1972).
- (127) T. Nishino, K. Ogura, and S. Seto, J. Am. Chem. Soc., 94, 6849 (1972).
- (128) T. Nishino, K. Ogura, and S. Seto, J. Am. Chem. Soc., 93, 794 (1971).
 (129) K. Ogura, T. Koyama, and S. Seto, J. Am. Chem. Soc., 94, 307 (1972).
- (130) L. Chayet, R. Pont-Lezica, C. George-Nascimento, and O. Cori, Phyto-
- chemistry, 12, 95 (1973). (131) R. Evans, A. M. Holtom, and J. R. Hanson, J. Chem. Soc., Chem. Commun.,
- 465 (1973). (132) K. H. Overton and F. M. Roberts, J. Chem. Soc., Chem. Commun., 385
- (1974).
- (133) D. V. Banthorpe, B. V. Charlwood, and M. J. O. Francis, Chem. Rev., 72, 115 (1972).

(141) J. J. Menn and M. Beroza, Ed., "Insect Juvenile Hormones", Academic

(142) K. J. Judy, D. A. Schooley, L. L. Dunham, M. S. Hall, B. J. Bergot, and J. B. Siddail, *Proc. Natl. Acad. Sci. U.S.A.*, 70, 1509 (1973).
 (143) K. J. Judy, D. A. Schooley, M. S. Hall, B. J. Bergot and J. B. Siddail, *Life Sci.*, 13, 1511 (1973).

(144) P. Schmialek, Z. Naturforsch. B, 18, 462 (1963).
 (145) M. Metzler, K. Hdahm, D. Meyer, and H. Roller, Z. Naturforsch. B, 26, 1270

(146) A. M. Ajami and L. M. Riddiford, Fed. Proc., Fed. Am. Soc. Exp. Biol., 32, 1382 (1973).

(147) D. A. Schooley, K. J. Judy, B. J. Bergot, M. S. Hall, and J. B. Siddall, *Proc. Natl. Acad. Sci. U.S.A.*, 70, 2921 (1973).
(148) G. E. Pratt and S. S. Tobe, unpublished results, quoted in ref 143.
(149) M. Metzler, D. Meyer, K. H. Dahm, H. Roller, and J. B. Siddall, *Z. Natur-*Ocyclement (1997).

(150) R. S. Burden and H. F. Taylor, *Tetrahedron Lett.*, 4071 (1970). (151) S. Isoe, S. B. Hyeon, S. Katsumura, and T. Sakan, *Tetrahedron Lett.*, 2517

(151) S. 1506, S. D. Hyeon, S. Ratsumara, and T. Carana, (1972).
(152) T. Oritani and K. Yamashita, *Tetrahedron Lett.*, 2521 (1972).
(153) G. Ryback, *J. Chem. Soc., Chem. Commun.*, 1190 (1972).
(154) H. F. Taylor and R. S. Burden, *Proc. R. Soc. London, Ser. B*, 180, 317 (1972).

T. Oritani and K. Yamashita, Agric. Biol. Chem., 37, 1115 (1973). M. Koreeda, G. Weiss, and K. Nakanishi, J. Am. Chem. Soc., 95, 239

(134) G. A. Cordell, Lloydia, 37, 219 (1974).

Press, New York, N.Y., 1972.

forsch. B. 27, 321 (1972).

(1971).

(1972)

(1973)

(155) (156)

- (135) C. George-Nascimento and O. Cori, *Phytochemistry*, **10**, 1803 (1971).
 (136) Y. Suzuki and S. Marumo, *Tetrahedron Lett.*, 5101 (1972).
 (137) D. Arigoni, *Pure Appl. Chem.*, **41**, 219 (1975).
 (138) B. M. Trost, *Acc. Chem. Res.*, **3**, 120 (1970).

- (139) K. Eiter, *Prog. Chem. Org. Nat. Prod.*, **28**, 204 (1970). (140) A. Pfiffner in ref 28, p 95.

- (157) N. Harada, *J. Am. Chem. Soc.*, **95**, 240 (1973). (158) K. Mori, *Tetrahedron*, **30**, 1065 (1974).
- (159) B. V. Milborrow, Planta, 76, 96 (1967).

- (159) B. V. Milborrow, *Planta, 16, 36* (1957).
 (160) B. V. Milborrow, *Biochem. J.*, **11**, 1P (1969).
 (161) B. V. Milborrow, *J. Exp. Bot.*, **21**, 17 (1970).
 (162) B. V. Milborrow in ref 28, p 137.
 (163) B. V. Milborrow in *Plant Growth Subst.*, *Proc. Int. Conf.*, 7th, 1970, 281 (1972)
- (164) R. C. Noddle and D. R. Robinson, *Biochem. J.*, **112**, 547 (1969).
 (165) D. R. Robinson and G. Ryback, *Biochem. J.*, **113**, 895 (1969).
 (166) B. V. Milborrow and R. C. Noddle, *Biochem. J.*, **119**, 727 (1970).

- (167) S. T. C. Wright and R. W. P. Hiron. *Nature (London)*, 224, 719 (1869).
 (168) B. V. Milborrow and M. Garmston. *Phytochemistry*, 12, 1597 (1973).
 (169) B. V. Milborrow, *Biochem. J.*, 128, 1135 (1972).
 (170) T. W. Goodwin and R. J. H. Williams, *Biochem. J.*, 94, 5C (1965).

- (171) R. J. H. Williams, G. Britton, and T. W. Goodwin, Biochem. J., 105, 99 (1967).

- (1367).
 (172) T. J. Walton, G. Britton, and T. W. Goodwin, *Biochem. J.*, **112**, 383 (1969).
 (173) B. V. Milborrow, *Phytochemistry*, **13**, 131 (1974).
 (174) B. V. Milborrow and D. R. Robinson, *J. Exp. Bot.*, **24**, 537 (1973).
 (175) P. G. Bartels and C. McCullough, *Biochem. Biophys. Res. Commun.*, **48**, 61 (1972). 16 (1972).
- (176) D. S. Tarbell, R. M. Carman, D. D. Chapman, S. E. Cremer, A. D. Cross, K. R. Huffman, M. Kunstmann, N. J. McCorkindale, J. G. McNally, A. Ro-sowsky, F. H. L. Varino, and R. L. West, *J. Am. Chem. Soc.*, 83, 3096 (1961)
- A. J. Birch and S. F. Hussain, J. Chem. Soc. C, 1473 (1969) (177)
- (178) K. Morikawa and Y. Hirose, Tetrahedron Lett., 1799 (1969).

- (179) K. Morikawa and T. Hinose, *retrahedron Lett.*, 1139 (1907).
 (179) K. Morikawa, Y. Hirose, and S. Nozoe, *Tetrahedron Lett.*, 1131 (1971).
 (180) A. Quilico, F. Piozzi, and M. Pavan, *Tetrahedron*, 1, 177 (1957).
 (181) Y. Hirose, M. Abu, and Y. Sekiya, *J. Chem. Soc. Jpn*, 82, 725 (1961).
 (182) T. Sakai, K. Nishimura, and Y. Hirose, *Tetrahedron Lett.*, 1171 (1963).
 (182) E. Waltara, C. Schlutter, and H. Schward, *J. Chem. Acta*, 52, 15
- (183) E. E. Waldner, C. Schlatter, and H. Schmid, Helv. Chim. Acta, 52, 15 (1969).
- (184) M. D. Sutherland and R. J. Park in ref 21, p 147.
- (185) K. T. Suzuki, N. Suzuki, and S. Nozoe, *Chem. Commun.*, 527 (1971).
 (186) L. H. Zalkow, F. X. Markley, and C. Djerassi, *J. Am. Chem. Soc.*, 82, 6354 (1960).
- (187)R. Croteau and W. D. Loomis, Phytochemistry, 11, 1055 (1972).
- (188) F. E. Regnier, E. J. Eisenbraun, E. R. Waller, and H. Auda, Abstracts, 150th National Meeting of the American Chemical Society, Atlantic City, N.J., Sept 1965, p 116C.
- W. D. Loomis and R. Croteau, Recent Adv. Phytochem., 6, 1 (1973)
- (190) R. Croteau, A. J. Burbott, and W. D. Loomis, Phytochemistry, 11, 2937 (1972)
- (191) A. Aebi and C. Djerassi, Helv. Chim. Acta, 42, 1785 (1959), and references cited therein.
- (192) C. J. W. Brooks, J. A. Zabkiewicz, R. A. B. Keates, and A. M. M. Berrie, unpublished results. J. A. Zabkiewicz, R. A. B. Keates, and C. J. W. Brooks, *Phytochemistry*,
- (193) 8, 2087 (1969).
- (194) C. J. W. Brooks and R. A. B. Keates, *Phytochemistry*, **11**, 3235 (1972).
 (195) D. J. Dunham and R. G. Lawton, *J. Am. Chem. Soc.*, **93**, 2075 (1971).
- (196) I. Uritani, M. Uritani, and H. Yamada, *Phytopathology*, **50**, 30 (1961).
 (197) T. Akazawa and K. Wada, *Plant Physiol.*, **36**, 139 (1961).
 (198) I. Uritani and T. Akazawa, *Science*, **121**, 216 (1955).

- T. Akazawa, I. Uritani, and Y. Akazawa, Arch. Biochem. Biophys., 99, (199)52 (1962).
- (200)T. Akazawa and I. Uritani, Agric. Biol. Chem., 26, 131 (1962).

- (201) T. Akazawa and I. Ortani, Agric. Biol. Crieffi, 20, 151 (1964).
 (202) K. Oba, H. Shibata, and I. Uritani, Plant Cell Physiol. 11, 507 (1970).
 (203) I. Oguni and I. Uritani, Agric. Biol. Chem., 34, 156 (1970).
 (204) I. Oguni and I. Uritani, Agric. Biol. Chem., 35, 357 (1971).
- (205) H. Theorell and T. Yonetani, Acta Chem. Scand., 23, 255 (1969)
- (206) V. Kykora, L. Novotny, M. Holub, V. Herout, and F. Sorm, Collect. Czech. Chem. Commun., 26, 788 (1961).
- (207) M. Soucek, Collect. Czech. Chem. Commun., 27, 2929 (1962).
- (208) P. De Mayo, E. Y. Spencer, and R. W. White, Can. J. Chem., 41, 2996 (1963)
- (209) P. De Mayo, R. E. Williams, and E. Y. Spencer, Can. J. Chem., 43, 1357 (1965).
- (210) P. De Mayo, J. R. Robinson, E. Y. Spencer, and R. W. White, Experientia. 18. 359 (1962).
- P. De Mayo and R. E. Williams, J. Am. Chem. Soc., 87, 3275 (1965).
- (211) P. De Mayo and R. E. Williams, J. Am. Chem. Soc., 61, 327, (1963).
 (212) O. E. Edwards in "Cyclopentanoid Terpene Derivatives", W. I. Taylor and A. R. Battersby, Ed., Marcel Dekker, New York, N.Y., 1969, p 357.
 (213) H. R. Schutte and J. Lehfeldt, Arch. Pharm., 298, 461 (1965).
 (214) J. H. Birkinshaw, H. Baistrick, and D. J. Ross, Biochem. J., 50, 630 (1952).

- (215) A. J. Birch, R. J. English, R. A. Massy-Westropp, M. Slaytor, and H. Smith, Proc. Chem. Soc., London, 204 (1957).
- (216) A. J. Birch, R. J. English, R. A. Massy-Westropp, M. Slaytor, and H. Smith, J. Chem. Soc., 365 (1958).
- (217) A. J. Birch, R. J. English, R. A. Massy-Westropp, and H. Smith, *Proc. Chem. Soc., London*, 233 (1957).
 (218) A. J. Birch, R. J. English, R. A. Massy-Westropp, and H. Smith, *J. Chem.* 260 (1959).
- Soc., 369 (1958)
- (219) A. J. Birch, R. J. English, and H. Smith, unpublished results, quoted in ref 65
- (220) L. Canonica, W. Kroszczynski, B. M. Ranzi, B. Rindone, and C. Scolastico,
- (220) L. Ganorica, W. Kroszczynski, B. M. Ranzi, B. Hildolfe, and C. Scolastico, Chem. Commun., 1357 (1970).
 (221) L. Canonica, W. Kroszczynski, B. M. Ranzi, B. Rindone, and C. Scolastico, Chem. Commun., 257 (1971).
 (222) C. T. Bedford, J. C. Fairlie, P. Knittel, T. Money, and G. T. Phillips, Chem. Commun. 200 (2021).
- Commun., 323 (1971).
- A. J. Birch, Annu. Rev. Plant Physiol., 19, 321 (1968).
- (224) A. J. Birch in "Biogenesi Delle Sostanze Naturali", Accademia Nazionale

- dei Lincei, Rome, 1964, p 57.
- (225) A. J. Birch and J. J. Wright, Aust. J. Chem., 22, 2635 (1969).
- (226) L. Canonica, W. Kroszczynski, B. M. Ranzi, B. Rindone, E. Santaniello,

Chemical Reviews, 1976, Vol. 76, No. 4 459

- and C. Scolastico, J. Chem. Soc., Perkin Trans. 1, 2639 (1972). (227) T. McMorris and M. Anchel, J. Am. Chem. Soc., 85, 831 (1963).
- (228) M. Tada, Y. Yamada, N. S. Bhacca, K. Nakanishi, and M. Ohashi, Chem.
- (226) M. Fada, Y. Famada, N. S. Bhacca, K. Nakanishi, and M. Ohashi, *Chem. Pharm. Bull.*, **12**, 853 (1964).
 (229) M. Tada, Y. Yamada, N. S. Bhacca, K. Nakanishi, and M. Ohashi, *Chem. Pharm. Bull.*, **12**, 856 (1964).
 (230) T. C. McMorris and M. Anchel, *J. Am. Chem. Soc.*, **87**, 1594 (1965).
- (231) M. Anchel, A. Hervey, and W. J. Robbins, Proc. Natl. Acad. Sci., U.S.A., 36, 300 (1950).
- M. Anchel, A. Hervey, and W. J. Robbins, Proc. Natl. Acad. Sci. U.S.A., (232)38, 927 (1952).
- (233) T. C. McMorris, M. S. R. Nair, and M. Anchel, J. Am. Chem. Soc., 89, 4562 (1967).
- (234) J. R. Hanson and T. Marten, J. Chem. Soc., Chem. Commun., 171 (1973). (235) A. I. Scott, Science, 184, 760 (1974).
 (236) G. Mellows, P. G. Mantle, T. C. Feline, and D. J. Williams, *Phytochemistry*.
- 12, 2717 (1973)
- (237) T. C. Feline, G. Mellows, R. B. Jones, and L. Phillips, J. Chem. Soc., Chem. Commun., 63 (1974).
- (238) F. W. Comer, F. McCapra, I. H. Qureshi, and A. I. Scott, Tetrahedron, 23, 4761 (1967)
- (239) S. Takahashi, H. Naganawa, H. Jinuma, T. Takita, K. Maeda, and H. Umezawa, Tetrahedron Lett., 1955 (1971).
- (240) M. Tanabe, K. T. Suzuki, and W. C. Jankowski, Tetrahedron Lett., 2271 (1974)
- (241) M. Kocor, A. Nespiak, and A. Siewinski, Bull. Acad. Pol. Sci., Ser. Sci. Chim., 9, 207 (1961).
- (242) M. Kocor, A. Nespiak, and A. Siewinski, *Nature (London)*, 112, 138 (1961).
 (243) E. Harri, W. Loeffler, H. P. Sigg, H. Stahelin, C. Stoll, C. Tamm, and D. Wiesinger, *Helv. Chim. Acta*, 45, 839 (1962).
- (244) J. Gutzwiller and C. Tamm, *Helv. Chim. Acta*, 56, 1786 (1963).
 (245) W. O. Godtfredsen and S. Vangedal, *Acta Chem. Scand.*, 19, 1088 (1965).
- (246) J. Gutzwiller, R. Mauli, H. P. Sieg, and C. Tamm, Helv. Chim. Acta, 47, 2234 (1964).
- (247) S. Abrahamsson and B. Nilsson, Proc. Chem. Soc., London, 188 (1964).
- (248) H. Nishikawa, Agric. Biol. Chem., 26, 696 (1962). (249) S. Natori, H. Ogawa, and H. Nishikawa, Chem. Pharm. Bull., 12, 236 (1964).
- (250) J. Fishman, E. R. H. Jones, G. Lowe, and M. C. Whiting, Proc. Chem. Soc., London, 127 (1959)
- (251) E. R. H. Jones and G. Lowe, J. Chem. Soc., 3959 (1960).
 (252) L. Ruzicka, Pure Appl. Chem., 6, 512 (1963).

(1970).

263.

(1973)

857 (1974)

Chem. Abstr., 66, 64343 / (1967).

(285) B. M. Craven, Nature (London), 168, 1193 (1963).

(1973).

(275)

(262)

Trans. 1, 1425 (1972).

- (253) M. Kocor and A. Siewinski, Buli. Acad. Pol. Scl., Ser. Sci. Chim., 14, 341 (1966).
- (254) R. Bentley and D. Chen, *Phytochemistry*, 8, 2171 (1969).
 (255) S. Natori, Y. Inouye, and H. Nishikawa, *Chem. Pharm. Bull.*, 15, 380 (1967).

(260) S. Nozoe, M. Morisaki, and H. Matsumoto, *Chem. Commun.*, 926 (1970). (261) B. A. Achiladelis, P. M. Adams, and J. R. Hanson, *J. Chem. Soc., Perkin*

(264) B. Achilladelis and J. R. Hanson, unpublished results, quoted in ref

(265) P. M. Adams and J. R. Hanson, J. Chem. Soc., Perkin Trans. 1, 586 (1972). (266) R. Achini, B. Muller, and C. Tamm, *Chem. Commun.*, 404 (1971). (267) D. Arigoni, D. E. Cane, B. Muller, and C. Tamm, *Helv. Chim. Acta*, 56, 2946

- (256) W. G. Dauben and P. Oberhansli, J. Org. Chem., 31, 315 (1966).
- (257) J. R. Hanson and B. Achiladelis, *Chem. Ind.* (London), 1643 (1967).
 (258) B. Achiladelis and J. R. Hanson, *Phytochemistry*, 7, 589 (1968).
 (259) B. Achiladelis, P. M. Adams, and J. R. Hanson, *Chem. Commun.*, 511

Y. Machida and S. Nozoe, Tetrahedron, 28, 5113 (1972)

(263) P. M. Adams and J. R. Hanson, Chem. Commun., 1569 (1970)

(268) P. M. Adams and J. R. Hanson, Chem. Commun., 1414 (1971) (269) J. M. Forrester and T. Money, *Can. J. Chem.*, **50**, 3310 (1972).
 (270) S. Nozoe and Y. Machida, *Tetrahedron Lett.*, 2671 (1970).
 (271) S. Nozoe and Y. Machida, *Tetrahedron*, **28**, 5105 (1972).

Y. Machida and S. Nozoe, Tetrahedron Lett., 1969 (1972).

(273) R. Evans, J. R. Hanson, and T. Marten, J. Chem. Soc., Perkin Trans. 1,

(274) M. Tanabe, K. T. Suzuki, and W. C. Jankowski, Tetrahedron Lett., 4723

(277) R. A. Maksimova, B. G. Khuratova, and A. B. Silaev, *Biol. Nauki (Moscow)*, 97 (1969); *Chem. Abstr.*, 71, 59590*m* (1970).
 (278) R. A. Maksimova, N. P. Pal'mova, and A. A. Alekseeva, *Antibiotiki (Moscow)*, 15, 229 (1970); *Chem. Abstr.*, 73, 11583*a* (1971).
 (270) R. A. Maksimova, C. Wurster, and A. D. Dilson (2011) (Article 14, 2020).

(Moscow), 15, 229 (1970); Chem. Abstr., 73, 11583a (1971).
(279) R. A. Maksimova, B. G. Khuratova, and A. B. Silaev, *Biol. Nauki*, 14, 138 (1971); Chem. Abstr., 75, 47462r (1972).
(280) A. N. Akhmedova, L. L. Velikanov, and I. I. Sidorova, Vestn. Mosk. Univ., *Biol., Pochvoved.*, 26, 49 (1971); Chem. Abstr., 75, 150219r (1972).
(281) L. A. Porter, Chem. Rev., 67, 441 (1967).
(282) H. Conroy, J. Am. Chem. Soc., 73, 1889 (1951).
(283) H. Conroy, J. Am. Chem. Soc., 79, 5550 (1957).
(284) T. Okuda and T. Yoshida, Tetrahedron Lett., 694 (1964).
(285) B. M. Craven. Nature (London) 168, 1193 (1963).

(286) M. Graveri, *Nature (London)*, **166**, 1133 (1963).
(286) M. F. McKey and A. M. Mathieson, *Tetrahedron Lett.*, 1399 (1963).
(287) T. Okuda and T. Yoshida, *Tetrahedron Lett.*, 2137 (1965).
(288) T. Okuda and T. Yoshida, *Tetrahedron Lett.*, 4191 (1965).
(289) G. Jommi, P. Manitto, and C. Scolastico, *Chim. Ind. (Milan)*, **47**, 407 (1965);

J. R. Hanson, T. Marten, and M. Siverns, J. Chem. Soc., Perkin Trans. 1, 1033 (1974). (276) A. B. Silaev, Z. E. Bekker, R. A. Maksimova, T. M. Gus'kova, M. V. Karnaukhova, R. N. Zelengya, and N. Y. Paramonov, S'kh. Biol., 1, 627 (1966);

- Chem. Abstr.. 63, 14917 e (1965). (290) S. Yamamura and Y. Hirata, *Tetrahedron Lett.*, 79 (1964)
- (291) Y. Sasaki, Y. Tsuda, B. Yasui, T. Konita, J. Matsumoto, E. Katarao, and J. Nakano, Tetrahedron, 20, 2007 (1964).
- (292) H. Conroy, J. Am. Chem. Soc., 74, 3046 (1952).
 (293) H. Conroy, J. Am. Chem. Soc., 74, 3050 (1952).
 (294) A. D. Cross, Q. Rev., Chem. Soc., 14, 317 (1960).
- (295) M. Yamazaki, M. Matsuo, and K. Arai, Chem. Pharm. Bull., 14, 1058
- (1966). (296) M. Biollaz and D. Arigoni, *Chem. Commun.*, 633 (1969).
- (297) A. Corella, P. Garlboldi, G. Jommi, and C. Scolastico, Chem. Commun.,
- 634 (1969).
- (298) M. Kolbe and L. Westfelt, Acta Chem. Scand., 21, 585 (1967)
- (299) M. Kolbe-Haugwitz and L. Westfelt, Acta Chem. Scand., 24, 1623 (1970). (300) K. W. Turnbull, W. Acklin, D. Arigoni, A. Corbella, P. Gariboldi, and G.
- Jommi, J. Chem. Soc., Chem. Commun., 598 (1972). (301) O. E. Edwards, J. L. Douglas, and B. Mootoo, Can. J. Chem., 48, 2517 (1970).
- (302) A. Corbella, P. Gariboldi, and G. Jommi, J. Chem. Soc., Chem. Commun., 600 (1972).
- (303) A. Corbella, P. Gariboldi, and G. Jommi, J. Chem. Soc., Chem. Commun., 7729 (1973).
- (304) A. Corbella, P. Gariboldi, and G. Jommi, unpublished results, quoted in ref 303.
- (305) P. Bollinger, Dissertation No. 3595, ETH, Zurich, 1965.
- (306) B. Muller, Dissertation No. 4000, ETH, Zurich, 1967.
 (307) J. L. Simonsen and D. H. R. Barton, Ed., "The Terpenes", Vol. III. Cam-
- bridge University Press, Cambridge, 1952, p 92.
 (308) W. Sandermann and K. Bruns, *Chem. Ber.*, **95**, 1863 (1962).
 (309) W. Sandermann and K. Bruns, *Tetrahedron Lett.*, 261 (1962).
- (310) G. Ourisson, Bull. Soc. Chim. Fr., 895 (1955).
- (311) C. George-Nascimento, R. Pont-Lezica, and O. Cori, Biochem. Biophys. Res. Commun., 45, 119 (1971).
- (312) D. H. R. Barton, G. P. Moss, and J. A. Whittle, J. Chem. Soc. C, 1813 (1968).
- (313) M. J. O. Francis, Planta Med., 22, 201 (1972)
- (314) H. D. Boyce, J. R. Harrison, and E. R. Hahn, Oil Soap (Chicago). 18, 27 (1941).
- (315) F. H. Śmith, Nature (London), 192, 888 (1961).
- (316) R. Adams, R. C. Morris, T. A. Geissman, D. J. Butterbaugh, and E. C. Kirkpatrick, *J. Am. Chem. Soc.*, **60**, 2193 (1938).
 (317) P. F. Heinstein, F. H. Smith, and S. B. Tove, *J. Biol. Chem.*, **23**7, 2643
- (1962).
- (318) P. F. Heinstein, D. L. Herman, S. B. Tove, and F. H. Smith, *J. Biol. Chem.*, 245, 4658 (1970).
 (319) A. S. Sadykov, A. I. Ismailov, and D. Iskandarova, *Nauchn. Tr., Tashk. Gos.*
- Univ., No. 263, 109 (1964); Chem. Abstr., 62, 9026 (1965).
- (320) A. S. Sadykov, A. I. Ismailov and Yu. U. Mavlyanova, Nauchn. Tr., Tashk. Gos. Univ., No. 263, 104 (1964); Chem. Abstr., 62, 16632b (1965). (321) J. J. Dugan, P. De Mayo, M. Nisbet, J. R. Robinson, and M. Anchel, J. Am.
- Chem. Soc., 88, 2838 (1966).
- (322) S. Nozoe, K. T. Suzuki, and S. Okuda, Tetrahedron Lett., 3643 (1968).

- (323) K. Hirai, S. Nozoe, K. Tsuda, Y. Jitaka, K. Ishibashi, and M. Shirasawa, Tetrahedron Lett., 2177 (1967). (324) K. T. Suzuki and S. Nozoe, Tetrahedron Lett., 2457 (1969). (325) K. T. Suzuki and S. Nozoe, Chem. Commun., 527 (1971).

- (326) G. Jaureguiberry, G. Farrugia-Fougerouse, H. Audier, and E. Lederer, C. R. Acad. Scl., Ser. D, 3108 (1964).
 (327) A. W. Dawkins, J. Chem. Soc. C, 116 (1966).
- (328) H. Emmerich, G. Drews, K. Trautmann, and P. Schmialek, Z. Naturforsch. B. 20, 211 (1965).
- (329) R. E. Dugan, E. Rasson, and J. W. Porter, Anal. Biochem., 22, 249 (1968) (330) W. Sewbert, E. Fass, and U. Remberger, Biochem. Z. 338, 265 (1963)
- (331) T. Ohtaki and S. Sakurai, Tampakushitsu Kakusan Koso, 20, 529 (1975).
- (332) A. M. Ajami, J. Insect Physiol., 20, 2497 (1974).
 (333) M. G. Peter and K. H. Dahm, Helv. Chim. Acta, 58, 1037 (1975).
- (334) R. C. Jennings, K. J. Judy, and D. A. Schooley, J. Chem. Soc., Chem.
- Commun., 21 (1975).
- (335) B. V. Milborrow. *Phytochemistry*, **14**, 123 (1975). (336) B. V. Milborrow. *Phytochemistry*, **14**, 2403 (1975).
- (337) M. Tanabe and K. T. Suzuki, Tetrahedron Lett., 4417 (1974)

- (338) D. E. Cane and R. H. Levin, J. Am. Chem. Soc., 97, 1282 (1975).
 (339) D. E. Cane and R. H. Levin, J. Amer. Chem. Soc., 98, 1183 (1976).
 (340) R. Grandi, A. Marchesini, U. M. Pagnoni, R. Trave, and L. Garanti, Tetrahedron. 30, 3821 (1974).
- (341) R. Grandl, A. Marchesini, U. M. Pagnoni, and R. Trave, Tetrahedron Lett., 1765 (1973).
- (342) F. Bellesla, R. Grandi, A. Marchesini, U. M. Pagnoni, and R. Trave, Phytochemistry, 14, 1737 (1975). (343) J. R. Hanson, P. B. Hitchcock, and R. Nyfeler, J. Chem. Soc., Perkin Trans.
- 1, 1586 (1975),
- (344) R. Evans, J. R. Hanson, and R. Nyfeler, J. Chem. Soc., Chem. Commun., 814 (1975).
- (345) F. C. Baker, C. J. W. Brooks, and S. A. Hutchlson, J. Chem. Soc., Chem. Commun., 293 (1975).
- (346) Y. Ohfune, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., 4377 (1975).
- (347) S. Nozoe, J. Furukawa, U. Sankawa, and S. Shibata, Tetrahedron Lett., 195 (1976).
- (348) B. Muller, R. Achlni, and C. Tamm, *Helv. Chim. Acta*. **58**, 453 (1975). (349) B. Muller, R. Achini, and C. Tamm, *Helv. Chim. Acta*. **58**, 471 (1975).
- (350) B. Muller and C. Tamm, Helv. Chim. Acta, 58, 483 (1975).
 (351) W. Knoll and C. Tamm, Helv. Chim. Acta, 58, 483 (1975).
- (351) W. Khoi and C. Tamm, Heiv. Chim. Acta. 56, 1162 (1975).
 (352) R. Evans and J. R. Hanson, J. Chem. Soc., Chem. Commun., 231 (1975).
 (353) P. Bollinger, Dissertation No. 3595, ETH, Zurich, 1965.
 (354) D. Arigoni and M. Buzzolini, unpublished data provided by D. E. Cane.
 (365) D. Arigoni, D. E. Cane, and S. Gotfredsen, unpublished results.

- (356) A. Corbella, P. Gariboldi, C. Jomml, and M. Sisti, J. Chem. Soc., Chem.
- Commun., 288 (1975). (357) D. Arigoni, Pure Appl. Chem., **41**, 219 (1975).
- (358) D. H. R. Barton and N. H. Werstiuk, J. Chem. Soc. C. 148 (1968).
- (359) J. R. Hanson and R. Nyfeler, J. Chem. Soc., Chem. Commun., 171 (1975).
- (360) J. R. Hanson and R. Nyfeler, J. Chem. Soc., Chem. Commun., 824 (1975).
- (361) F. Dorn, P. Bernasconi, and D. Arigoni, Chimia, 29, 24 (1975).