# Total Synthesis of Bioactive Marine Macrolides<sup>†</sup>

Roger D. Norcross<sup>‡</sup> and Ian Paterson<sup>\*</sup>

Contribution from the University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, UK

Received March 17, 1995 (Revised Manuscript Received June 12, 1995)

#### **Contents**

١.	Introduction	2041
II.	Survey of Syntheses of Marine Macrolides	2043
	A. The Swinholides	2043
	B. The Halichondrins	2052
	C. Aplasmomycin	2065
	D. The Aplyronines	2073
	E. The Scytophycins	2076
	F. The Ulapualides and Halichondramides	2078
	G. The Bryostatins	2082
	H. The Macrolactins	2095
	The Amphidinolides	2098
	J. Tedanolide	2099
	K. The Latrunculins	2100
	L. The Octalactins	2105
III.	Concluding Remarks	2108

#### I. Introduction

Nature has stocked the seas with a seemingly limitless range of diverse and often highly complex secondary metabolites, which exhibit one or more of a variety of biological properties including cytotoxicity, neurotoxicity, antiviral, and antifungal activity. This review focuses on chemical efforts directed toward the total synthesis of a specific subset of biologically active marine natural products—namely those compounds which possess a macrocyclic lactone moiety, *i.e.* the marine macrolides.<sup>2,3</sup> The literature is surveyed from the onset of the subject, in the early 1980s, until the close of 1994.

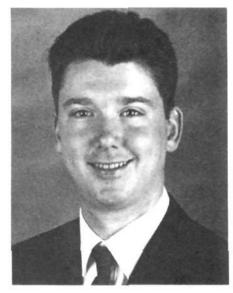
The topical nature of this field of chemical research may be illustrated by reference to the spongistatins, a group of nine extremely potent cytotoxic marine macrolides which have recently been isolated from an Eastern Indian Ocean sponge of the genus Spongia (spongistatins 1-3 (1-3 in Figure 1)), and from the Southwest African marine sponge Spirastrella spinispirulifera (spongistatins 4-9 (4-9)).4 In addition, some presumably identical compounds, the altohyrtins A-C (1, 2, and 10) and 5-desacetylaltohyrtin A (3), have been isolated from the Okinawan marine sponge Hyrtios altum,5 and a compound named cinachyrolide A (assumed to be identical to spongistatin 4) has been isolated from a marine sponge of the genus Cinachyra.<sup>6</sup> The spongistatins represent some of the most potent substances presently known against a subset of highly chemoresistant tumor types in the US NCI panel of 60 human cancer cell lines. They show especially powerful growth inhibitory activity against human melanoma, lung, colon, and brain cancers. Indeed, spongistatins 1 (mean panel  $GI_{50}=(2.5-3.5)\times10^{-11}$  M) and 9 (mean panel  $GI_{50}=4\times10^{-11}$  M) are the most potent members of the spongistatin family and are claimed to be the most cancer cell growth inhibitory antimitotic substances discovered to date.<sup>4</sup>

Synthetic interest in the marine macrolides stems mainly from their biological activities, and, in particular, their potential as chemotherapeutic agents. Most of these natural products are available in only microscopic quantities from their biological source: spongistatin 1 (1), for example, is obtained in only  $3.4 \times 10^{-7}\%$  isolated yield from the whole sponge, and spongistatin 9 (9) is isolated in  $2.2 \times 10^{-7}\%$  yield. Large-scale harvesting of marine organisms, such as sponges, is neither practical nor ecologically acceptable, but total synthesis has, in principle, the potential for supplying sufficient quantities of natural product for biological and pharmaceutical testing. The fact that closely related chemical structures have been found in several disparate marine sources, as in the case of the spongistatins, suggests that these natural products may in fact be produced by symbiotic organisms living in association with the different marine hosts.7 Potentially, in such cases, fermentation cultures of the symbiotic organisms provide a means of obtaining significant quantities of the marine macrolides. To date, however, this appproach has met with only partial success, and total synthesis remains the only viable alternative. In addition, de novo chemical synthesis provides the possibility for preparation of nonnatural analogues, which may be used as probes to determine the mechanism of action of the natural product, as well as being useful for therapeutic evaluation.

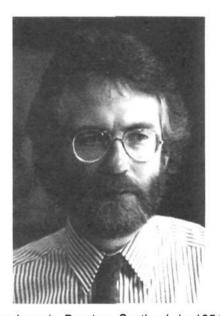
Total synthesis is also a valuable tool for confirming, and in some cases determining, the stereochemistry of marine macrolides, which may not always be possible by spectroscopic or crystallographic means alone. For instance, although a complete absolute configuration for spongistatin 1 ( $\equiv$  altohyrtin A) has been volunteered by Kitagawa and co-workers on the basis of spectroscopic studies on the natural product, 5b this stereochemical assignment is partially in conflict with relative stereochemistry proposed by Pettit et al.  $^{4d,e}$  and by Fusetani et al.  $^6$  Thus, at the time of writing, there is uncertainty about the absolute configuration of portions of the spongistatin stucture, and therefore a need for unambiguous determination of stereochemical configuration via total synthesis. In recent years, the combination of chemical synthesis and conformational analysis using NMR methods and computational molecular modeling has proved

<sup>&</sup>lt;sup>†</sup> We dedicate this review to Professor David A. Evans.

<sup>&</sup>lt;sup>‡</sup> Present address: Corporate Research Units, Ciba-Geigy AG, CH-4002 Basel, Switzerland.



Roger Norcross was born in Manchester, England, in 1967. He studied for a B.A. degree in Natural Sciences at Cambridge University, and it was during this time that his interest in organic synthesis, and asymmetric synthesis in particular, was kindled. He received his Ph.D. degree from Cambridge University in 1992, under the supervision of Dr. lan Paterson, on the subject of new stereoselective aldol methodology and its application in the total synthesis of macrolide antiobiotics. From 1992 to 1994 he was a NATO Postdoctoral Fellow in the laboratories of Professor David A. Evans, at Harvard University. During these immensely stimulating two years he concentrated upon the search for new asymmetric catalysts. Upon returning to Cambridge University in 1994, he reentered the field of total synthesis, focusing upon the development of new stereoselective methods for application in the synthesis of marine macrolides. He is currently working at the Corporate Research Laboratories of Ciba-Geigy in Basel, Switzerland.



lan Paterson was born in Dundee, Scotland, in 1954. He received a B.Sc. degree in Chemistry from St. Andrews University in 1976. In 1979 he obtained his Ph.D. from Cambridge University, working under the supervision of lan Fleming, on the development of new synthetic methods using allylsilanes and silyl enol ethers. After spending a highly rewarding and enjoyable year with Gilbert Stork at Columbia University as a NATO Postdoctoral Fellow, working on the total synthesis of erythromycin A, he joined the faculty at University College, London, in 1980. In 1983 he moved to his present position as a Lecturer at Cambridge University and Fellow of Jesus College. His research interests are centered on the design and development of new synthetic methods for the control of stereochemistry and their application to the total synthesis of a range of biologically active compounds, which currently include several marine macrolides.

valuable as a tool in determining the stereochemistry of marine macrolides, most notably in the case of the halichondrins and aplyronines (vide infra).

The exquisitely complex structures of many of the marine macrolides serve as inspiration for the development of new methodology in organic synthesis, and as an elegant platform for exhibiting the creativity of the modern organic chemist, which seems to be limited only by the structures themselves. The

Figure 1. Structures of the spongistatins.

marine macrolides present a 5-fold synthetic challenge to the organic chemist: (i) The stereochemical challenge provided by target structures possessing, in some cases, in excess of 30 stereogenic centers. Historically, many marine macrolide syntheses have relied on the use of starting materials from the chiral pool, such as carbohydrates, to supply most of the stereogenic centers.8 Increasingly, concomitant with the appearance of new methods for achieving acyclic stereocontrol, alternative strategies have been employed relying on either reagent- or substrate-based asymmetric induction.9

- (ii) Formation of the macrocycle. This has generally been achieved by using macrolactonization reactions, which are increasingly becoming routine procedures in organic synthesis,10 but carbon-carbon bond-forming reactions have occasionally been employed. Thus far, the largest ring constructed has been that of the 44-membered macrodiolide swinholide A (vide infra).
- (iii) The need for efficient processes for coupling complex, often highly oxygenated, fragments, suitable for use in the latter stages of a synthesis.
- (iv) The judicious choice of protecting group arrangements for polyoxygenated structures. In several cases, the success or failure of a synthetic route has depended entirely on the selection of a protecting group for a single hydroxyl-bearing center (vide infra).
- (v) The requirement of many of the marine macrolides for the stereocontrolled formation of di- or trisubstituted double bonds.

In this review, the marine macrolides are presented in order of decreasing ring size (as determined by counting along a contiguous carbon skeleton wherever possible), except that closely related structures are considered in succession in order to aid comparison. After giving a brief description of the natural source and biological properties, a summary of the associated synthetic work then follows. Where total syntheses have been achieved, details for all transformations in the route are to be found in the accompanying diagrams. Discussion in the text concentrates on those key reactions which establish stereochemistry, close rings, or couple complex segments. Wherever possible, an approximate indication of the overall yield and total number of steps is given. As a guide to synthetic efficiency, the approximate number of steps per stereogenic center of the target structure is also provided.

# II. Survey of Syntheses of Marine Macrolides

#### A. The Swinholides

The swinholides (11–17 in Figure 2) are a series of complex macrodiolides, isolated from the marine sponge *Theonella swinhoei*,  $^{11,12}$  which display potent cytotoxicity against a variety of human tumor cell lines.  $^{13d}$  Swinholide A (11) was originally misassigned as a monomeric 22-membered  $^{14}$  macrolide,  $^{11}$  but more recent mass spectroscopic  $^{13a}$  and X-ray crystallographic  $^{13b-d}$  studies have elucidated the true  $C_2$ -symmetrical, 44-membered,  $^{14}$  macrodiolide struc-

ture depicted in Figure 2. Isoswinholide A (18), a minor congener of swinholide A, having an unsymmetrical 46-membered macrodiolide structure, has also been isolated from *Theonella*, <sup>12a</sup> along with the monomeric seco-acid preswinholide A (19), which is believed to be the biosynthetic precursor of swinholide A. 12b,c Other cytotoxic macrodiolides isolated from Theonella sp. include the bistheonellides (20-22), which lack two of the swinholide double bonds and are thus 40-membered macrocycles. 15b-d Note that the structures of the monomeric units of both swinholide A (11) and bistheonellide A (20) (also called misakinolide A<sup>15a,c</sup>) are very similar<sup>13c</sup> to that of scytophycin C (23),16a one of a class of cytotoxic macrolides (23-27) isolated from the terrestial bluegreen alga Scytonema pseudohofmanni (vide infra). 16 This structural homology implies a genetic link between the producing organisms, lending support to the assumption that the swinholides and bistheonellides are actually metabolites of symbiotic microorganisms associated with Theonella sp.7,13c Indeed, the presence of a symbiotic blue-green alga in the marine sponge Theonella swinhoei has been detected using electron microscopy. 13c.

The first total synthesis of swinholide A (and also of its minor congener isoswinholide A) was reported by Paterson *et al.* in 1994, <sup>17g</sup> following syntheses of preswinholide A earlier that year. <sup>17e,f</sup> Two significant segments of swinholide A have been prepared by Nicolaou and co-workers, <sup>18</sup> and Nakata *et al.* have also synthesized a swinholide A segment. <sup>19</sup>

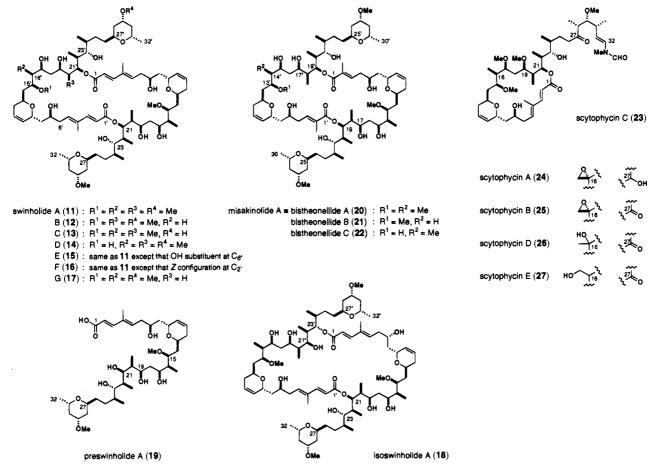


Figure 2. Structures of the swinholides, misakinolides, and scytophycins.

#### Scheme 1

## 1. Paterson Total Synthesis<sup>17</sup>

The synthesis of swinholide A (11) by Paterson *et al.* was based on the selective deprotection and regiocontrolled dimerization of **28**, a fully protected version of the monomeric seco-acid preswinholide A (Scheme 1). The monomeric unit **28** was constructed by the union of the  $C_1-C_{15}$  aldehyde segment **29** and the  $C_{19}-C_{32}$  aldehyde segment **30** employing a butanone synthon (**31**) as a linking unit.<sup>20</sup> By varying the order of the  $C_{15}-C_{16}$  and  $C_{18}-C_{19}$  aldol bond constructions, the stereochemically most efficient sequence for this key segment coupling was determined. The stereocontrolled syntheses of **29** and **30**, in turn, relied heavily on various types of asymmetric aldol reactions, which were used to form the  $C_6-C_7$ ,  $C_{12}-C_{13}$ , and  $C_{22}-C_{23}$  bonds.

a.  $C_1$ - $C_{15}$  Segment Synthesis. 17b,c Preparation of the  $C_1-C_{15}$  segment 29 began by employing methodology developed by Paterson for the enantioselective synthesis of dihydropyrones.<sup>21</sup> Thus, reagentcontrolled asymmetric aldol reaction of aldehyde 32 with the bis(isopinocampheyl) enol borinate 33, derived from enolization of ketone 3422 by (+)-bis-(isopinocampheyl)boron chloride 35, gave the aldol adduct 36 in 80% ee (Scheme 2). Cyclization of 36 to the dihydropyrone 37 was then accomplished by treatment with TMSOTf/iPr2NEt and recrystallization provided 37 in enantiomerically pure form. Stereoselective Luche reduction<sup>23</sup> of **37** to the corresponding allylic alcohol (α-face attack), followed by acetylation, then supplied the glycal 38. Stereoselective introduction of the aldehydic side chain at C<sub>9</sub>, with concomitant allylic transposition, was achieved by employing a variant of the Ferrier rearrangement.24 Thus, Ti(O'Pr)2Cl2-mediated reaction of 38 with silyl enol ether 3925 afforded aldehyde 40 with 97% diastereoselectivity (ds). Chain extension at C7 was effected by means of a novel vinylogous Mukaiyama aldol reaction between 40 and the silvl dienol ether 4126 promoted by BF3 OEt2, which provided 42 with 81% ds in favor of the required stereochemistry at  $C_7$  and with the correct (E)-enal terminus. Note that this reaction results in exclusive γ-attack on 41 and proceeds without chelate participation from the dihydropyran oxygen of 40. The second (E)-double bond of the diene ester moiety was cleanly introduced by Horner-Emmons olefination of 42 to afford 43. Protection of the  $C_7$  hydroxyl, followed by deprotection and Dess-Martin oxidation<sup>27</sup> at  $C_{15}$ , then gave the  $C_1-C_{15}$  segment 29 of swinholide A. Thus, in this segment synthesis, introduction of the C<sub>13</sub> stereogenic center was achieved by a reagent-controlled boron aldol reaction, 32 + 33→ 36, while the remaining two stereogenic centers were set up by the sequence  $38 \rightarrow 40 \rightarrow 42$  using substrate control. Note that addition of ethylmagnesium bromide to aldehyde 29, followed by oxidation, gave 44, a C<sub>1</sub>-C<sub>16</sub> segment of scytophycin C.  $[C_1-C_{15} \text{ segment } \mathbf{29}: 11\% \text{ overall yield from } \mathbf{34}; 10$ steps; ~3 steps per stereogenic center.]

b.  $C_{19}$ - $C_{32}$  Segment Synthesis. 17a Synthesis of the C<sub>19</sub>-C<sub>32</sub> segment 30 began with kinetic resolution of racemic allylic alcohol 45 using Sharpless asymmetric epoxidation, 28a,b which provided epoxide 46 with high diastereo- and enantiomeric purity (95% ds, 96% ee, Scheme 2).28c Directed reductive opening<sup>29</sup> of **46** then afforded diol **47**, with the correct configuration at both  $C_{29}$  and  $C_{31}$ . Ozonolysis of 47 followed by acidic workup and O-methylation furnished the acetal 48 as a mixture of anomers; and highly diastereoselective TMSOTf-catalyzed allylsilane addition<sup>30</sup> to **48** then gave the tetrahydropyran **49** with the correct configuration at  $C_{27}$ , via kinetically controlled axial attack on the oxonium ion 50. After ozonolysis of **49** to give the  $C_{25}$  aldehyde, stereoselective Wittig homologation<sup>31</sup> afforded the (E)-enal **51**. Alternatively, **51** could be obtained from acetal 48 in a single transformation, viz. TMSOTfcatalyzed addition of the silyl dienol ether 41. Construction of the  $C_{22}\text{--}C_{23}$  bond by a highly diastereoselective anti aldol reaction<sup>32</sup> between aldehyde 51 and the (E)-dicyclohexyl enol borinate 52, derived<sup>33</sup> from ketone 53 which was in turn prepared from (S)methyl 3-hydroxy-2-methylpropionate (vide infra),34 then gave the  $\beta$ -hydroxy ketone **54** with the required configuration at  $C_{22}$  and  $C_{23}$ . Stereoselective reduction of 54 using the Saksena-Evans reagent<sup>35</sup> provided the  $C_{21}$ ,  $C_{23}$ -anti diol **55**, which was protected as its di-tert-butylsilylene derivative 56. Note that Paterson et al. did not opt for differential protection of the C21 and C23 hydroxyls, and thus a regioselective lactonization would be required later in the synthesis (vide infra). The remaining stereogenic center at  $C_{24}$ was installed by a substrate-controlled<sup>36</sup> hydroboration of **56**, to give alcohol **57** with  $\geq$  97% ds, followed by Barton deoxygenation<sup>37</sup> of the surplus hydroxyl at C25 to afford 58. Finally, deprotection at C19 and subsequent Swern oxidation38 then supplied the C19-C<sub>32</sub> segment **30** of swinholide A. Note that this also corresponds to the  $C_{17}$ – $C_{30}$  segment of bistheonellide A. Thus, in the synthesis of segment 30, introduction of the  $C_{29}$  and  $C_{31}$  stereogenic centers was achieved by reagent-control (Sharpless epoxidation of 45 →

# Scheme 2. Paterson Preswinholide A Synthesis 17a-c,e a

 $^a$  (a) 34, (+)-Ipc<sub>2</sub>BCl,  $^iPr_2NEt;$  32;  $H_2O_2;$  (b) TMSOTf,  $^iPr_2NEt;$  (c) recrystallize; (d) NaBH<sub>4</sub>, CeCl<sub>3</sub>; (e) Ac<sub>2</sub>O,  $^iPr_2NEt;$  (f) 38 + 39, Ti(O<sup>i</sup>Pr)<sub>2</sub>Cl<sub>2</sub>; (g) 40 + 41, BF<sub>3</sub>·OEt<sub>2</sub>; (h) (MeO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Me,  $^nBuLi;$  (i) TBSOTf, 2,6-lutidine; (j) K<sub>2</sub>CO<sub>3</sub>, MeOH; (k) Dess-Martin periodinane; (l) EtMgBr; (m) Dess-Martin periodinane; (a') (+)-DIPT, Ti(O<sup>i</sup>Pr)<sub>4</sub>,  $^iBuOOH;$  (b') Red-Al; (c') O<sub>3</sub>, MeOH; Me<sub>2</sub>S; 1 M HCl; (d') NaH, MeI; (e') 48 + 41, TMSOTf; (f') H<sub>2</sub>C=CHCH<sub>2</sub>TMS, TMSOTf; (g') O<sub>3</sub>; Me<sub>2</sub>S; (h') Ph<sub>3</sub>P=C(Me)CHO; (i') 53, (°Hex)<sub>2</sub>BCl, Et<sub>3</sub>N; 51; H<sub>2</sub>O<sub>2</sub>; (j') Me<sub>4</sub>NBH(OAc)<sub>3</sub>; (k')  $^iBu_2$ Si(OTf)<sub>2</sub>, 2,6-lutidine; (l') thexylborane; H<sub>2</sub>O<sub>2</sub>, NaOH; (m') (imid)<sub>2</sub>C=S; (n')  $^nBu_3$ SnH; (o') H<sub>2</sub>, 10% Pd-C; (p') Swern oxidation; (q') 30 + 60, TiCl<sub>4</sub>; (r') O<sub>3</sub>; Me<sub>2</sub>S; (s') Cl<sub>3</sub>CC(=NH)OPMB, TfOH; (t') 59, (°Hex)<sub>2</sub>BCl, Et<sub>3</sub>N; 29; H<sub>2</sub>O<sub>2</sub>; (u') catecholborane; (v') DDQ; (w') Dess-Martin periodinane; (x') LiAl(O<sup>i</sup>Bu)<sub>3</sub>H; (y') MeOTf, 2,6-di-tert-butylpyridine; (z') HF; (a'') NaOH.

46), while the remaining stereogenic centers were set up by a series of substrate-controlled reactions: 48

 $\rightarrow$  **49**, **51** + **52**  $\rightarrow$  **54**  $\rightarrow$  **55**, and **56**  $\rightarrow$  **57**. [C<sub>19</sub>-C<sub>32</sub> segment **30**: 5.7% overall yield from **45**; 15 steps

longest linear sequence; 18 steps total;  $\sim$ 2 steps per stereogenic center.]

c. The Synthesis of Preswinholide  $A.^{17e,f}$  By varying the order of the  $C_{15}-C_{16}$  and  $C_{18}-C_{19}$  bond constructions in the segment coupling sequence, two different syntheses of the fully protected monomeric seco-acid preswinholide A (28) were achieved by Paterson et al. The first route (Scheme 2) was based on formation of the  $C_{18}-C_{19}$  bond before the  $C_{15}-C_{16}$  bond. Thus, addition of a butanone equivalent to the  $C_{19}-C_{32}$  aldehyde 30 gave the ethyl ketone 59 which was then aldol coupled to the  $C_{1}-C_{15}$  aldehyde 29.

TiCl<sub>4</sub>-promoted addition of allylsilane 60<sup>39</sup> to aldehyde 30, under Felkin-Anh control, gave the adduct 61 with 95% ds in favor of the desired configuration at C<sub>19</sub>. Ozonolysis of the alkene and protection of the  $C_{19}$  hydroxyl then gave **59**. Note that model studies<sup>17d</sup> indicated that, in contrast, enol borinate addition to **30** would give the undesired anti-Felkin epimer 19epi-59. The model studies also indicated that for a syn aldol coupling of aldehyde 29 with the (Z)-enol borinate derived from ketone 59, the intrinsic diastereofacial selectivities of the two chiral components were matched in the wrong stereochemical sense for swinholide A. Unfortunately, attempts to confer reagent control in this reaction by using isopinocampheyl enol borinates<sup>40</sup> proved unsuccessful. Hence, a boron-mediated anti aldol reaction was used instead for the coupling of 29 and 59, which meant that a stereochemical inversion was required later in the synthesis. Thus, substrate-controlled aldol reaction of aldehyde **29** with the (E)-dicyclohexyl enol borinate **62** derived<sup>33</sup> from ketone **59** gave the two *anti* aldol isomers 63 and 64 in 60:40 ratio, together with a small amount of syn aldol isomers (anti/syn = 87:13). Compound **63** had the correct configuration of the  $C_{16}$ methyl, but required inversion of the hydroxyl at  $C_{15}$ . Catecholborane reduction<sup>41</sup> of **63** gave the corresponding  $C_{17}$ ,  $C_{19}$ -syn diol with >95% ds, and treatment with DDQ then induced cyclization of the C<sub>19</sub> PMB ether<sup>42a</sup> onto the C<sub>17</sub> hydroxyl to afford acetal **65**. Inversion of configuration at  $C_{15}$  was then accomplished by oxidation, followed by selective reduction, to supply 15-epi-65 with 83% ds. Note that the aldol adduct 64 could also be used productively in the synthesis of preswinholide A by conversion into the ketone precursor of 15-epi-65 by a four-step sequence involving epimerization at  $C_{16}$ . With all the stereogenic centers required for preswinholide A installed, methylation of the C<sub>15</sub> hydroxyl of 15-epi-65 gave the fully protected seco-acid 28 in 10% overall yield over the nine steps from 30. Total deprotection of 28 afforded preswinholide A (19), which served to confirm the complete stereostructure.

The second, and more efficient, route to preswinholide A developed by Paterson *et al.* (Scheme 3) was based on the opposite order for the segment coupling sequence: *i.e.*  $C_{15}-C_{16}$  bond construction prior to  $C_{18}-C_{19}$  bond construction.<sup>17f</sup> Model studies<sup>17d</sup> had indicated that the greatest stereochemical efficiency would be expected from reagent-controlled addition of a butanone equivalent to the  $C_1-C_{15}$  aldehyde **29** to give the methyl ketone **66**, followed by substrate-

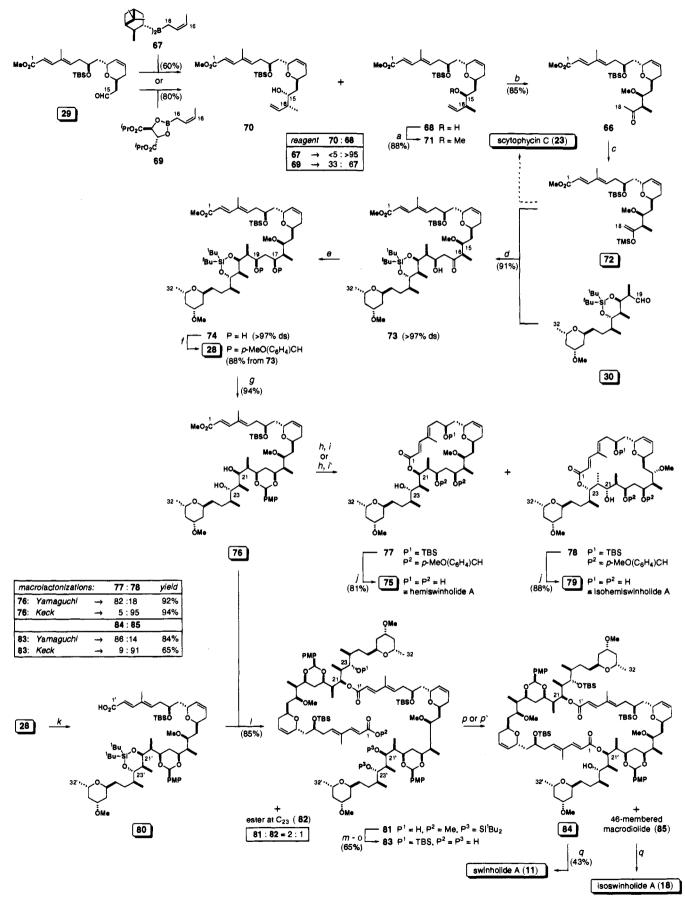
controlled Mukaiyama aldol coupling<sup>25</sup> with the  $C_{19}$ – $C_{32}$  aldehyde **30**.

Thus,  $C_{15}-C_{16}$  bond construction by syn crotylboration of 29 using the Brown chiral crotylboron reagent 6743 furnished the desired homoallylic alcohol 68 with >95% ds. The corresponding Roush reagent 6944 proved less selective in this mismatched situation, generating a 2:1 ratio of 68 and 70. After O-methylation of **68** to give **71**, Wacker oxidation<sup>45</sup> of the terminal alkene supplied the methyl ketone **66**.  $C_{18}-C_{19}$  bond construction by Mukaiyama aldol coupling of aldehyde 30 with the silyl enol ether 72, derived from kinetic enolization of ketone 66, then afforded the Felkin-Anh aldol adduct 73 as the sole product. Note that the silyl enol ether 72 is also a  $C_1-C_{18}$  segment for scytophycin C. A modified Narasaka-Prasad<sup>46</sup> syn reduction of  $\beta$ -hydroxy ketone 73, via the preformed boron chelate, then gave the  $C_{17}$ ,  $C_{19}$ -syn diol 74. Thus, the four stereogenic centers spanning C<sub>15</sub>-C<sub>19</sub> of preswinholide A had been introduced with an overall diastereoselectivity of approximately 95%. Protection of diol 74 as its p-methoxybenzylidene acetal then supplied the fully protected derivative 28 of preswinholide A in 36% overall yield over the seven steps from 29. [Preswinholide A (19): first route—0.2% overall yield from 45; 26 steps longest linear sequence; 39 steps total;  $\sim 2-3$ steps per stereogenic center; second route-1.6% overall yield from 34; 20 steps longest linear sequence; 37 steps total; ~2-3 steps per stereogenic center.]

d. Completion of the Total Syntheses of Hemiswinholide A, Isohemiswinholide A, Swinholide A, and Isoswinholide A. <sup>17g</sup> In order to complete a synthesis of swinholide A, Paterson et al. required selective deprotection and regiocontrolled dimerization of 28. Note that by using the cyclic di-tert-butylsilylene group, these researchers had forgone the opportunity for selective protection of the C<sub>21</sub> and C<sub>23</sub> hydroxyls. This strategy was bold, but ultimately proved to be successful.

The synthesis of the 22-membered<sup>14</sup> macrolide **75**, designated hemiswinholide A, corresponding to the erroneous monomeric structure initially proposed for swinholide A,11 is outlined in Scheme 3. Selective removal of the silylene group of 28 gave the C21, C23 diol 76. After base-catalyzed hydrolysis of the terminal methyl ester of 76, macrolactonization was attempted. By employing the Yamaguchi protocol (formation of the mixed anhydride by treatment with 2,4,6-Cl<sub>3</sub>(C<sub>6</sub>H<sub>2</sub>)COCl/Et<sub>3</sub>N, followed by DMAP-promoted cyclization in toluene),47 an 82:18 mixture of the 22- and 24-membered macrolides 77 and 78 was obtained. In contrast, use of the Keck conditions (DCC, DMAP, DMAP·HCl in chloroform)<sup>48</sup> led to a reversal of selectivity, furnishing a 5:95 ratio of 77 and 78. Note that performing the Keck macrolactonization in toluene, as in the Yamaguchi procedure. gave a 40:60 mixture of 77 and 78. Thus the regioselectivity of macrolactonization appears to be sensitive to solvent polarity, which presumably alters the conformational preferences of the activated secoacid. Note also that only monomeric lactones were obtained. Deprotection of the acetal and silyl protecting groups of 77 supplied hemiswinholide A (75).

### Scheme 3. Paterson Swinholide A and Isoswinholide A Synthesis 17f,g a



 $^{a} \text{ (a) MeOTf, 2,6-di-} \textit{tert-} \textbf{butylpyridine; (b) PdCl}_{2}, \textbf{ CuCl, O}_{2}; \textbf{ (c) LiHMDS, TMSCl, Et}_{3}N; \textbf{ (d) 30 + 72}, \textbf{ BF}_{3}\textbf{\cdot}\textbf{OEt}_{2}; \textbf{ (e) } ^{n}\textbf{Bu}_{2}\textbf{BOMe}; \textbf{ LiBH}_{4}; \textbf{ H}_{2}\textbf{O}_{2}; \textbf{ (f) } p\textbf{-MeO(C}_{6}\textbf{H}_{4})\textbf{CH(OMe)}_{2}, \textbf{CSA}; \textbf{ (g) HF} \textbf{\cdot}\textbf{py}; \textbf{ (h) NaOH; (i) 2,4,6-Cl}_{3}(\textbf{C}_{6}\textbf{H}_{2})\textbf{COCl, Et}_{3}N; \textbf{DMAP}; \textbf{ (i') DCC, DMAP, DMAP+HCl; (j) HF}; \textbf{ (k) NaOH; (l) 80, 2,4,6-Cl}_{3}(\textbf{C}_{6}\textbf{H}_{2})\textbf{COCl, Et}_{3}N; \textbf{76, DMAP; (m) TBSCl, Et}_{3}N, \textbf{DMAP; (n) HF} \textbf{\cdot}\textbf{py}; \textbf{ (o) Ba(OH)}_{2}; \textbf{ (p) 2,4,6-Cl}_{3}(\textbf{C}_{6}\textbf{H}_{2})\textbf{COCl, Et}_{3}N; \textbf{DMAP; (p') DCC, DMAP, DMAP+HCl; (q) HF}.$ 

Similarly, **78** was converted to isohemiswinholide A (**79**).

The synthesis of swinholide A itself exploited the differentiation of the  $C_{21}$  and  $C_{23}$  hydroxyls uncovered above. Thus, hydrolysis of the C1 ester in 28 gave the acid 80, which was used to selectively esterify the  $C_{21}$  hydroxyl of diol **76** (Scheme 3). Activation of 80 using the Yamaguchi conditions<sup>47</sup> followed by DMAP-promoted addition of 76 afforded a 2:1 mixture of the desired  $C_{21}$  ester 81 and its  $C_{23}$  regioisomer 82. After chromatographic separation, 82 could be recycled by methanolysis to give back 76 and 28. Meanwhile, silyl protection of the  $C_{23}$  hydroxyl of 81, followed by silvlene removal and selective hydrolysis of the terminal methyl ester, then afforded the dimeric seco-acid 83. Note that without silyl protection of the C23 hydroxyl, competing cleavage of the  $C_{21}$  ester, and/or transesterification to the  $C_{23}$  position was observed during the final hydrolysis step. Cyclization of 83 was facile and high yielding (60-84%): subjection of 83 to the Yamaguchi macrolactonization conditions<sup>47</sup> at room temperature, and without the need for high dilution, afforded an 86: 14 mixture in favor of the desired 84 (acylation of the C21 hydroxyl) over the larger macrodiolide 85 (acylation of the  $C_{23}$  hydroxyl). As with **76**  $\rightarrow$  **77** + 78, the selectivity in ring size was sensitive to the macrolactonization conditions: use of the Keck protocol48 gave a 9:91 mixture of 84 and 85, permitting selective formation of the isoswinholide ring. Finally, total deprotection of 84 completed the first total synthesis of swinholide A (11). Similarly, isoswinholide A (18) was obtained upon deprotection of 85. Of particular note in these syntheses is the fact that the regioselectivity of macrolactonization was controlled without the need for differential hydroxyl protection. [Swinholide A (11): 0.4% overall yield from **34**; 25 steps longest linear sequence; 43 steps total;  $\sim 3$  steps per stereogenic center, allowing for  $C_2$  symmetry].

#### 2. Nicolaou Seament Syntheses18

The strategy proposed by Nicolaou and co-workers for achieving a synthesis of swinholide A is outlined in Scheme 4. Ring closure by Horner-Emmons reaction of the ketophosphonate-aldehyde 86 is planned,49 in contrast to the macrolactonization approach adopted by Paterson et al. The dimer 86 is expected to be obtained from the monomer 87 by a sequence of (i) esterification of the  $C_{21}$  hydroxyl of 87 to give ketophosphonate 88;50 (ii) Horner-Emmons coupling of 88 with another monomer unit corresponding to the  $C_{3'}$  aldehyde derived from 87; and, finally, (iii) esterification of the  $C_{21}$  hydroxyl. Formation of the  $C_3-C_{32}$  segment 87 by coupling of the  $C_3-C_{17}$  segment 89 and the  $C_{18}-C_{32}$  segment 90 is envisaged. At the time of writing, the preparation of 89 and 90 has been reported.

a.  $C_3-C_{17}$  Segment Synthesis. <sup>18a</sup> The  $C_3-C_{17}$  segment **89** was prepared from (S)-dimethyl malate (**91**) as outlined in Scheme 5. Thus, directed reduction of **91** to give the  $C_{12}$ ,  $C_{13}$  diol<sup>51</sup> was followed by sequential silylation with TBDPSCl and TBSOTf; DIBAL reduction at  $C_{15}$  then supplied the aldehyde **92**.  $C_{15}-C_{16}$  bond construction by syn-crotylboration

Scheme 4

of 92 using the Brown chiral crotylboron reagent 67,43 followed by O-methylation, furnished the desired homoallylic ether 93 with >95% ds (cf. similar diastereoselectivity was obtained by Paterson et al. for the transformation  $29 \rightarrow 68 \rightarrow 71$ : Scheme 3). Ozonolysis of 93 followed by a reductive workup gave the corresponding  $C_{17}$  alcohol which was protected as its PMB ether. Complete desilylation followed by regioselective reprotection at  $C_{12}$  then afforded alcohol **94**. After mesylation of the  $C_{13}$  hydroxyl of **94**, treatment with TBAF effected cyclization to give the epoxide **95** with inversion of configuration at  $C_{13}$ . The  $\alpha,\beta$ -unsaturated- $\delta$ -lactone **96** was prepared by using Ghosez's methodology, 52 involving reaction of epoxide 95 with the lithio derivative of methyl 3-(phenylsulfonyl)orthopropionate (97) and subsequent acid hydrolysis and DBU-induced elimination. Lactone 96 was reduced to the corresponding lactol 98, and ZnCl<sub>2</sub>-catalyzed C-glycosidation using silyl enol ether **39**<sup>25</sup> then supplied the aldehyde **99** with 80% ds. Chain extension at  $C_7$  was effected by means of a

90

# Scheme 5. Nicolaou Swinholide A $C_3-C_{17}$ Synthesis $^{18a\ a}$

 $^a$  (a) BH<sub>3</sub>·Me<sub>2</sub>S; cat. NaBH<sub>4</sub>; (b) TBDPSCl, Et<sub>3</sub>N, DMAP; (c) TBSOTf, 2,6-lutidine; (d) DIBAL; (e) **67** + **92**; H<sub>2</sub>O<sub>2</sub>, NaOH; (f) NaH, MeI; (g) O<sub>3</sub>; NaBH<sub>4</sub>; (h) Cl<sub>3</sub>CC(=NH)OPMB, CSA; (i) TBAF; (j) TBDPSCl, Et<sub>3</sub>N, DMAP; (k) MsCl, Et<sub>3</sub>N; (l) TBAF; (m) **97**, DMPU,  $^n$ BuLi; **95**; H<sub>2</sub>SO<sub>4</sub>; p-TsOH; Et<sub>3</sub>N, DBU; (n) DIBAL; (o) **39** + **98**, ZnCl<sub>2</sub>; (p) **99** + **100**, BF<sub>3</sub>·OEt<sub>2</sub>; (q) TBSOTf, 2,6-lutidine; (r) DDQ, H<sub>2</sub>O; (s) Swern oxidation; (t) HS(CH<sub>2</sub>)<sub>3</sub>SH, TiCl<sub>4</sub>; (u) DIBAL; (v) TBSOTf, 2,6-lutidine.

BF<sub>3</sub>·OEt<sub>2</sub>-promoted vinylogous Mukaiyama aldol reaction between 99 and the silyl ketene acetal 100,53 which provided 101 with 58% ds in favor of the required stereochemistry at  $C_7$  and with the correct (E)- $\alpha,\beta$ -unsaturated ester terminus. Note that a similar aldol reaction utilized in the Paterson synthesis (40  $\rightarrow$  42 in Scheme 2) proceeded with higher diastereoselectivity (81% ds). The degree of stereoinduction arising in these aldol reactions appears to be highly sensitive to subtle changes in substrate structure. With all of the stereogenic centers present in the  $C_3-C_{17}$  segment of swinholide A now installed, 101 was converted to a derivative suitable for coupling to the  $C_{18}$ - $C_{32}$  segment **90**. Thus, protection of the  $C_7$  hydroxyl of **101** was followed by deprotection at  $C_{17}$  and oxidation to give the  $C_{17}$  aldehyde; dithiane formation then afforded **102**. Reduction of the ester at C<sub>3</sub> followed by protection of the resulting hydroxyl then gave the desired  $C_3-C_{17}$  segment 89. Of the four newly created stereogenic centers in 89, two were set up in a single reagent-controlled reaction  $(92 \rightarrow 93)$ , and the remaining two arose from two substrate-controlled reactions (98  $\rightarrow$  99  $\rightarrow$  101). [C<sub>3</sub>- $C_{17}$  segment **89**: 3.3% overall yield from **91**; 22 steps; ~4 steps per stereogenic center.]

b. C<sub>18</sub>-C<sub>32</sub> Segment Synthesis. 18b The C<sub>18</sub>-C<sub>32</sub> segment 90 was prepared from L-rhamnose (103) as outlined in Scheme 6. Thus, peracetylation of 103 was followed by C-glycosidation<sup>54</sup> with allyltrimethylsilane to give exclusively the α-glycoside. Complete deacetylation and subsequent regioselective methylation of the C<sub>29</sub> hydroxyl, using <sup>n</sup>Bu<sub>2</sub>SnO and methyl iodide in the presence of cesium fluoride,55 then supplied 104. Barton deoxygenation  $^{37}$  at  $C_{28}$  and  $C_{30}$ of 104 was followed by reductive ozonolysis to give the C<sub>25</sub> primary alcohol, which was then transformed into iodide 105. Enders alkylation<sup>56</sup> using iodide 105 and the SAMP hydrazone 106, followed by ozonolytic removal of the chiral auxiliary, then furnished the ketone 107 with high diastereoselectivity at the newly formed stereogenic center at C24. Meanwhile syn crotylboration of aldehyde 108, using the Brown chiral crotylboron reagent 109,43 gave the homoallylic ether 110 with the correct configurations at C<sub>19</sub> and  $C_{20}$ . Benzylation and ozonolysis then provided the aldehyde 111, which, using the Evans protocol,<sup>57</sup> underwent a stereoselective aldol reaction with the chlorotitanium enolate derived from ketone 107 to give the syn-aldol adduct 112. A samarium-catalyzed, intramolecular Tischenko-Evans<sup>58</sup> reduction of  $\beta$ -hydroxy ketone 112 then furnished the corresponding monoprotected C<sub>21</sub>,C<sub>23</sub>-anti-diol 113. Silylation of the  $C_{23}$  hydroxyl of 113 then gave 114. Note that use of the Tischenko-Evans reduction enabled differential protection of the C21 and C23 hydroxyls, which should permit esterification of the C21 hydroxyl later in the synthesis of swinholide A, as required. This is in marked contrast to the route of Paterson et al., wherein such differential protection was not employed. With all of the stereogenic centers present in the  $C_{18}$ – $C_{32}$  segment of swinholide A now installed, 114 was converted to a derivative suitable for coupling to the  $C_3$ - $C_{17}$  segment 89. Thus hydrogenolysis of the benzyl ethers at  $C_{18}$  and  $C_{19}$  of 114 was followed by selective monotosylation of the resulting diol, and treatment with base then gave the  $C_{18}$ – $C_{32}$  epoxide segment 90. Thus, three of the nine stereogenic centers present in 90 originated from a carbohydrate. while the other six stereogenic centers were installed using a combination of reagent-controlled (108 + 109  $\rightarrow$  110), auxiliary-controlled (105 + 106  $\rightarrow$  107), and substrate-controlled (107 + 111  $\rightarrow$  112  $\rightarrow$  113) reactions.  $[C_{18}-C_{32} \text{ segment } 90: 3.2\% \text{ overall yield}$ from 103; 16 steps longest linear sequence; 21 steps total;  $\sim 2$  steps per stereogenic center.]

### 3. Nakata Segment Syntheses 19

Nakata *et al.* have recently completed the synthesis of a  $C_{11}-C_{32}$  segment (115) corresponding to the polyol portion of swinholide A via stereoselective, auxiliary-controlled aldol coupling of the  $C_{11}-C_{23}$  and  $C_{24}-C_{32}$  segments 116 and 117 (Scheme 7).

a.  $C_{11}$ – $C_{23}$  Segment Synthesis. <sup>19a</sup> The  $C_{11}$ – $C_{23}$  segment **116** was obtained via desymmetrization of the  $C_2$  symmetric ketone **118**, which was prepared according to the method of Nakata and Oishi for the stereocontrolled synthesis of 1,3-polyols (Scheme 8). <sup>59</sup> Thus, (S)-malic acid (**119**) was converted into alcohol **120**. <sup>60</sup> After protection of the  $C_{11}$  hydroxyl of **120**, cleavage of the acetonide gave the  $C_{13}$ ,  $C_{14}$  diol;

#### Scheme 6. Nicolaou Swinholide A C<sub>18</sub>-C<sub>32</sub> Synthesis 18b a

 $^a$  (a) Ac<sub>2</sub>O, Et<sub>3</sub>N, DMAP; (b) H<sub>2</sub>C=CHCH<sub>2</sub>TMS, BF<sub>3</sub>·OEt<sub>2</sub>, TMSOTf; (c) NaOMe, MeOH; (d)  $^n$ Bu<sub>2</sub>SnO; CsF, MeI; (e) NaH, imidazole; CS<sub>2</sub>; MeI; (f)  $^n$ Bu<sub>3</sub>SnH, AIBN; (g) O<sub>3</sub>; NaBH<sub>4</sub>; (h) I<sub>2</sub>, PPh<sub>3</sub>, imidazole; (i) **106**, LDA; **105**; (j) O<sub>3</sub>; (k) NaH, BnBr,  $^n$ Bu<sub>4</sub>NI, imidazole; (l) O<sub>3</sub>; Me<sub>2</sub>S; (m) **108** + **109**; H<sub>2</sub>O<sub>2</sub>, NaOH; (n) KH, BnBr; (o) O<sub>3</sub>; PPh<sub>3</sub>; (p) TiCl<sub>4</sub>, **107**, Et<sub>3</sub>N; **111**; (q) PhCHO, SmI<sub>2</sub>; (r) TBSOTf, 2,6-lutidine; (s) H<sub>2</sub>, 10% Pd−C; (t) TsCl, Et<sub>3</sub>N, DMAP; (u) K<sub>2</sub>CO<sub>3</sub>, MeOH.

#### Scheme 7

selective tosylation of the C<sub>14</sub> primary hydroxyl and treatment with base then furnished the epoxide 121. Regioselective opening of epoxide 121 by attack of lithiated 1,3-dithiane at  $C_{14}$ , followed by protection of the resulting  $C_{13}$  hydroxyl and subsequent dithiane hydrolysis, afforded the aldehyde 122. Double nitroaldol reaction between nitromethane and 2 equiv of aldehyde 122 then gave the mixture of diols 123. Protecting group exchange, followed by hydrolysis of the nitro group, 61 supplied a mixture of diastereomeric ketones 124. By analogy with the earlier work of Stork on erythronolide A,62 treatment of this mixture with potassium carbonate in methanol effected epimerization to give exclusively the C<sub>13</sub>,C<sub>15</sub>syn- $C_{17}$ , $C_{19}$ -syn  $C_2$  symmetrical ketone 118. This was predicted to be the thermodynamically most favorable epimer, since the acetonide rings both adopt chair conformations with the alkyl side chains at  $C_{13}$ ,  $C_{15}$ ,  $C_{17}$ , and  $C_{19}$  all equatorially disposed.

Introduction of a methyl group at  $C_{16}$  and desymmetrization was now required. Thus, Wittig methylenation at  $C_{16}$  of **118** supplied **125**, and replacement of the acetonide protecting groups by p-methoxybenzylidene acetals then afforded **126**. Reaction of **126** 

with excess BH<sub>3</sub>·Me<sub>2</sub>S led to hydroboration of the C<sub>16</sub> exomethylene group and concomitant differential reductive cleavage<sup>63</sup> of one of the p-methoxybenzylidene acetals, furnishing the fully differentiated diol 127 as a single diaster eomer. Note that attempts to perform the same reaction on 125 resulted in lower yields due to the diminished reduction potential of the acetonide moiety compared to the p-methoxybenzylidene acetal. The authors have proposed that, after hydroboration, the reaction proceeds via the transition state 128 in which the boron atom can coordinate with only one oxygen atom of the two acetals (that at C<sub>15</sub>), thus activating the corresponding acetal C-O bond to reductive cleavage. Regioselective deoxygenation of 127 was accomplished by selective tosylation of the primary alcohol and subsequent LiAlH<sub>4</sub> reduction. Methylation of the remaining C<sub>15</sub> hydroxyl then afforded 129. After deprotection of the benzyl groups in 129 with Raney nickel and acid hydrolysis of the p-methoxybenzylidene acetal, treatment with DDQ led to cyclization<sup>42a</sup> of the C<sub>13</sub> PMB ether onto the hydroxyl revealed at  $C_{11}$  to supply triol 130. Selective oxidation of the primary alcohol of 130 with Ag<sub>2</sub>CO<sub>3</sub> on celite gave the  $\beta$ -hydroxy  $\delta$ -lactone **131**, and stereoselective methylation<sup>64</sup> at the  $C_{20}$   $\alpha$  carbon then afforded 132. This completed the introduction of the stereogenic centers spanning C<sub>13</sub>-C<sub>20</sub> of swinholide A. A five-step sequence of protecting group exchange and adjustment of oxidation level at C21 transformed 132 into aldehyde 133, and a diastereoselective Evans syn aldol reaction of 133 with the (Z)-enol borinate derived from chiral oxazolidinone 13465 correctly installed the remaining stereogenic centers at  $C_{21}$  and  $C_{22}$ . Removal of the chiral auxiliary from the aldol adduct 135 and transformation into the  $C_{23}$ aldehyde, along with protection of the  $C_{21}$  hydroxyl, then furnished the  $C_{11}$ - $C_{23}$  segment 116 of swinholide A, in readiness for aldol coupling to the  $C_{24}$ - $C_{32}$  segment 117. [ $C_{11}$ - $C_{23}$  segment 116: 1.2%

## Scheme 8. Nakata Swinholide A C<sub>11</sub>-C<sub>23</sub> Synthesis 19a a

 $\begin{array}{l} ^{a} \text{ (a) BH}_{3}\text{'Me}_{2}\text{S; (b) Me}_{2}\text{CO, H}^{+}; \text{ (c) BnBr, NaH, } ^{n}\text{Bu}_{4}\text{NI; (d) H}_{2}\text{SO}_{4}; \text{ (e) TsCl, py; (f) } \text{K}_{2}\text{CO}_{3}, \text{ MeOH; (g) } ^{n}\text{BuLi, 1,3-dithiane; (h) DHP, } \\ p\text{-TsOH; (i) HgO, HgCl}_{2}, \text{H}_{2}\text{O}; \text{ (j) } \textbf{122} \text{ (2 equiv), MeNO}_{2} \text{ (1 equiv), Et}_{3}\text{N, 5.5 kbar; (k) AcOH, H}_{2}\text{O}; \text{ (l) } \text{(MeO)}_{2}\text{CMe}_{2}, \text{CSA; (m) } ^{i}\text{BuONa, } \\ \text{KMnO}_{4}, \text{MgSO}_{4}, \text{H}_{2}\text{O}; \text{ (n) K}_{2}\text{CO}_{3}, \text{MeOH; (o) Ph}_{3}\text{P}^{+}\text{MeI}^{-}, ^{n}\text{BuLi; (p) AcOH, H}_{2}\text{O}; \text{ (q) } \\ p\text{-MeO(C}_{6}\text{H}_{4}\text{)CH(OMe)}_{2}, \text{CSA; (r) BH}_{3}\text{'Me}_{2}\text{S (3 equiv); } \\ \text{H}_{2}\text{O}_{2}, \text{NaOH; (s) TsCl, py; (t) LAH; (u) KH, MeI; (v) H}_{2}, \text{Raney Ni; (w) AcOH, H}_{2}\text{O}; \text{ (x) DDQ; (y) Ag}_{2}\text{CO}_{3}\text{-celite; (z) LDA, MeI, HMPA; (a') LiBH}_{4}; \text{ (b') TBDPSCl, imidazole; (c') (MeO)}_{2}\text{CMe}_{2}, \text{PPTS; (d') TBAF; (e') Swern oxidation; (f') } \textbf{134}, ^{n}\text{Bu}_{2}\text{BOTf}, ^{i}\text{Pr}_{2}\text{NEt; } \textbf{133}; \text{H}_{2}\text{O}_{2}; \text{ (g') LiOH, H}_{2}\text{O}_{2}; \text{ (h') CH}_{2}\text{N}_{2}; \text{ (i') TESOTf, 2,6-lutidine; (j') DIBAL; (k') PDC.} \\ \end{array}$ 

overall yield from 121; 37 steps;  $\sim$ 5 steps per stereogenic center.]

b.  $C_{24}-C_{32}$  Segment Synthesis. 19b The route to the  $C_{24}-C_{32}$  segment 117 began with (S)-methyl 3-hydroxybutyrate (136), which was prepared from Dthreonine (137) according to the method of Larchevêque (Scheme 9).66 Thus, deamination of 137 in the presence of bromide ion gave 138, and esterification followed by reduction then supplied 136. A Claisen reaction between ester 136 and the lithium enolate of tert-butyl acetate afforded  $\beta$ -keto ester 139, and subsequent lactonization and simultaneous protection of the  $C_{29}$  carbonyl furnished **140**. Reduction of **140** delivered the corresponding lactol as a 4:1 mixture of  $\alpha$ - and  $\beta$ -anomers, which upon silylation gave a single isomer 141 having the  $C_{27}$  silyloxy group equatorial. This fixing of the anomeric group in the equatorial position was essential in order to achieve complete stereoselectivity in the subsequent introduction of the C<sub>29</sub> stereogenic center, which was achieved by means of ketone reduction.<sup>67</sup> Thus, after deprotection of the thioacetal to give ketone 142, reduction with LiAlH4 took place on the less-hindered  $\beta$ -face, resulting in exclusive formation of the corresponding  $C_{29}$ - $\alpha$  alcohol;<sup>67</sup> methylation then afforded 143. Stereoselective introduction of the  $C_{27}$  side chain was accomplished by BF<sub>3</sub>·OEt<sub>2</sub>-mediated allylsilane addition to 143, which gave exclusively the allyl glycoside 49, having the allyl group axially

disposed.<sup>54</sup> Note that a similarly stereoselective transformation was used in the Paterson synthesis (48  $\rightarrow$  49 in Scheme 2).<sup>17a</sup> After ozonolysis of 49, to afford aldehyde 144, extension of the side chain by Wittig olefination, followed by hydrogenation, then furnished the  $C_{32}-C_{24}$  segment 145. Finally, conversion to acyl chloride 146 and treatment with the lithium salt of 147<sup>65</sup> gave the imide 117.

c. C<sub>11</sub>-C<sub>32</sub> Segment Synthesis. 19b Coupling of the  $C_{11}-C_{23}$  and  $C_{24}-C_{32}$  segments was achieved by an Evans syn aldol reaction 65 between aldehyde 116 and the (Z)-enol borinate derived from imide 117. This supplied the  $\beta$ -hydroxy imide **148**, having the correct configurations at  $C_{23}$  and  $C_{24}$  (Scheme 9). After an exchange of protecting groups to give 149, removal of the chiral auxiliary 68 and adjustment of oxidation state afforded the alcohol **150**. In order to complete the synthesis of the  $C_{11}$ – $C_{32}$  segment, conversion of the  $C_{24}$  hydroxymethyl group of **150** into a methyl group was now required. Procedures for effecting this transformation via the coresponding mesulate. tosylate, xanthate, or iodide were found to be unsatisfactory. However, conversion to the 2-pyridyl sulfide 151,69 followed by reductive cleavage of the carbon-sulfur bond upon treatment with Ranev nickel,<sup>70</sup> furnished the C<sub>11</sub>-C<sub>32</sub> segment **115** of swinholide A. In this synthesis, three of the stereogenic centers in 115 (those at  $C_{13}$ ,  $C_{19}$ , and  $C_{31}$ ) originated from the chiral pool, the four stereogenic centers

# Scheme 9. Nakata Swinholide A $C_{11}$ – $C_{32}$ Synthesis 19b a

 $\begin{tabular}{ll} $^a$ (a) NaNO_2, HBr, KBr; (b) MeOH, $H^+$; (c) $H_2$, $Pd-C$; (d) MeCO_2$'Bu, LDA; $136$; (e) HS(CH_2)_2SH, $BF_3$'OEt_2$; (f) DIBAL; (g) TBDPSCl, imidazole; (h) NBS, $AgNO_3$, $Na_2CO_3$, $H_2O$; (i) LAH; (j) KH, $MeI$; (k) $H_2$C=CHCH_2TMS, $BF_3$'OEt_2$; (l) $O_3$; $Me_2S$; (m) $Ph_3$P=CHCO_2Me$; (n) $H_2$, $10\% Pd-C$; (o) LiOH, $H_2O$; (p) (COCl)_2$; (q) $147$, $^BuLi$; $146$; (r) $117$, $^Bu_2BOTf$, $^Pr_2NEt$; $116$; $H_2O_2$; (s) $HF_Py$, $py$; (t) $(MeO)_2CMe_2$, $PPTS$; (u) LiOH, $H_2O_2$, $H_2O$; (v) $CH_2N_2$; (w) $LAH$; (x) $2,2$'-dipyridyl disulfide, $^Bu_3P$; (y) Raney Ni. $$ \end{tabular}$ 

spanning  $C_{21}-C_{24}$  were introduced by two auxiliary-controlled reactions (133 + 134  $\rightarrow$  135 and 116 + 117  $\rightarrow$  149), and the remaining six stereogenic centers were set up by reactions relying on substrate control of asymmetric induction. [ $C_{11}-C_{32}$  segment 115: 0.5% overall yield from 121; 45 steps longest linear sequence; 62 steps total;  $\sim$ 5 steps per stereogenic center.]

### B. The Halichondrins

The halichondrins (152-160 in Figure 3) are a series of complex polyether macrolides, originally isolated from the marine sponge Halichondria okadai Kadota,71 which show potent in vitro and in vivo antitumor activity.71-73 Halichondrin B (152) has been selected by the NCI for development as an anticancer drug, and most synthetic efforts have focused on this member of the class. Due to the scarcity of the halichondrins obtained from sponge extracts, total synthesis is highly desirable to augment the natural supply. In 1992, Kishi and coworkers reported the first total synthesis of halichondrin B (and also of norhalichondrin B (155)).74e Significant segments have also been prepared by the groups of Salomon<sup>75</sup> and Horita and Yonemitsu. <sup>76a-d</sup> Burke et al. have also reported the synthesis of two halichondrin B segments.77

#### 1. Kishi Total Synthesis<sup>74</sup>

The landmark synthesis of halichondrin B (152) by Kishi and co-workers was based on the assembly of

Figure 3. Structures of the halichondrins.

three segments: a  $C_1$ – $C_{13}$  segment (161), a  $C_{14}$ – $C_{38}$  segment (162), and a  $C_{39}$ – $C_{54}$  segment (163) (Scheme 10). Kishi–Nozaki Ni(II)/Cr(II)-mediated coupling reactions<sup>78</sup> were used to construct the  $C_{13}$ – $C_{14}$  and  $C_{38}$ – $C_{39}$  bonds, and a macrolactonization reaction was used to close the macrocycle.

a.  $C_1-C_{13}$  Segment Synthesis.<sup>74f</sup> The most efficient and most recent synthesis of the  $C_1-C_{13}$  segment **161** is outlined in Scheme 11. The route began with

#### Scheme 10

Scheme 11. Kishi Halichondrin  $C_1$ – $C_{13}$  Synthesis<sup>74f a</sup>

<sup>a</sup> (a) Acetonide protection; (b) DIBAL; (c) 'BuOK, MeOCH<sub>2</sub>-PPh<sub>3</sub>+Cl⁻; (d) OsO<sub>4</sub>, ('PrNHCH<sub>2</sub>)<sub>2</sub>; (e) Ac<sub>2</sub>O, DMAP, py; (f) H<sub>2</sub>C=CHCH<sub>2</sub>TMS, TMSOTf; (g) catecholborane, RhCl(PPh<sub>3</sub>)<sub>3</sub>; (h) PCC/alumina; (i) Ph<sub>3</sub>P=CHCO<sub>2</sub>Me; Triton B methoxide; (j) p-TsOH; (k) NaIO<sub>4</sub>; (l) trans-¬BuCH=CHI, NiCl<sub>2</sub> (1.1%)−CrCl<sub>2</sub>; (m) FeCl<sub>3</sub>, SiO<sub>2</sub>; (n) TBSOTf, 2,6-lutidine; (o) O<sub>3</sub>; (p) CHI<sub>3</sub>, CrCl<sub>2</sub>.

L-mannonic  $\gamma$ -lactone (**164**), whose four stereocenters match those at  $C_8-C_{11}$  of halichondrin. Acetonide protection, reduction to the C<sub>7</sub> aldehyde and Wittig olefination converted **164** to **165**. Osmylation of enol ether **165** then introduced the correct hydroxyl stereochemistry at  $C_7$  (16:1 in favor of the desired), in accordance with the Kishi empirical rule. 79 After acetalization to give 166, C-allylation provided exclusively the expected<sup>54a</sup> axial allyl glycoside 167 with the desired configuration at C<sub>6</sub>. Following conversion to 168 and Wittig elefination at  $C_3$  to give 169, an in situ intramolecular hetero-Michael reaction then provided 170 and its C<sub>3</sub> epimer in 1:1 ratio. Upon treatment of the mixture with Triton B methoxide, however, complete equilibration to the thermodynamically more stable desired epimer 170 occurred. Thus the transformation  $168 \rightarrow 170$  was achieved in one pot. Selective hydrolysis of one of the two acetonide groups in 170 furnished the  $C_1-C_{12}$  segment 171. A six-step sequence involving Ni(II)/Cr-(II)-mediated vinyl addition to the C<sub>11</sub> aldehyde (vide infra)<sup>78</sup> then provided the C<sub>1</sub>-C<sub>13</sub> segment **161** used

# Scheme 12. Kishi Halichondrin $C_1-C_{13}$ Synthesis<sup>80</sup> a

<sup>a</sup> (a) BzCl, py; (b) p-TsOH, MeOH; (c) TBSOTf, Et<sub>3</sub>N; (d) MeONa; (e) Swern oxidation; (f) IC $\equiv$ CTMS, NiCl<sub>2</sub> (0.01%)-CrCl<sub>2</sub>; (g) AgNO<sub>3</sub>; (h) <sup>n</sup>Bu<sub>3</sub>SnH, AIBN; (i) I<sub>2</sub>; (j) TBSOTf, Et<sub>3</sub>N.

in the total synthesis. Thus, in the synthesis of **161**, the four stereogenic centers spanning  $C_8-C_{11}$  originated in the chiral pool, and the other three stereogenic centers were installed using reactions relying on substrate control of asymmetric induction.

An earlier, and longer, synthesis<sup>80</sup> of  $C_1-C_{13}$  segment 161 is summarized in Scheme 12. Alcohol 172 was synthesized from D-glucose diacetonide (173) in 20 steps.81 This involved transformation to L-talofuranoside to provide the B ring,82 allyl glycoside formation at C<sub>6</sub>, and formation of the A ring by an intramolecular hetero-Michael reaction (analogous to  $169 \rightarrow 170$  in Scheme 11). After a sequence of protecting group exchange and oxidation to provide the C<sub>11</sub> aldehyde 174, a Ni(II)/Cr(II)-mediated coupling<sup>78</sup> with an alkynyl iodide then gave the propargylic alcohol 175 with a selectivity of  $\sim$ 8:1 in favor of the desired, and anticipated,  $^{78c}$  configuration at  $C_{11}$ . Alcohol 175 was then converted into the vinyl iodide 161 in a further four steps. Note that the coupling methodology is well suited for labile aldehydes such as 174: in this case, no complications due to enolization, such as epimerization or dehydration, were observed. A similar stereoselective Ni(II)/Cr(II)mediated coupling reaction, this time involving a vinyl iodide, was used during the transformation 171  $\rightarrow$  **161** (Scheme 11). [C<sub>1</sub>-C<sub>13</sub> segment **161**: improved route-16 steps from 164; ~2 steps per stereogenic center; original route-30 steps from 173; ~4 steps per stereogenic center.]

b.  $C_{27}$ – $C_{38}$  Segment Synthesis. <sup>74b</sup> Synthesis of the C27-C38 segment 176 was accomplished using the Ireland-Claisen rearrangement, 83 Ni(II)/Cr(II)-mediated coupling<sup>78</sup> and intramolecular hetero-Michael reactions as key steps (Scheme 13). The synthesis began with D-galactose glycal (177), which was converted into its 4-O-benzyl-3,6-O-dipropionate derivative (178). Ireland-Claisen rearrangement of 178 under appropriate conditions (LiHMDS, TBSCl, HMPA/THF to generate the ketene silyl acetal; then reflux) gave the expected<sup>84</sup> product 179 with  $\sim$ 8:1 stereoselectivity. Iodolactonization of this mixture, followed by reductive removal of the iodine, then afforded the  $\gamma$ -lactone **180**; the minor diastereomer was removed by chromatography or recrystallization at this stage. In contrast, Ireland-Claisen rearrang-

#### Scheme 13. Kishi Halichondrin C<sub>27</sub>-C<sub>38</sub> Synthesis<sup>74b a</sup>

 $\begin{tabular}{ll} $^a$ (a) TBSCl, imidazole; (b) BnBr, NaH; (c) TBAF; (d) (EtCO)_2O, Et_3N; (e) LiHMDS, TBSCl, HMPA; $\Delta;$ (e') LDA; TBSCl; $\Delta;$ (f) NaOH; (g) I_2, KI, NaHCO_3; (h) $^nBu_3SnH$, AIBN; (i) DIBAL; (j) $p$-TsOH, MeOH; (k) Tf_2O, py; NaCN; (l) DIBAL; NaBH_4; (m) $H_2/Pd(OH)_2-C;$ (n) EtSH, BF_3-OEt_2; (o) TBSOTf, Et_3N; (p) I_2, NaHCO_3, H_2O; (q) $trans-MeO_2CCH=CHI, NiCl_2 (1.0%)-CrCl_2; (r) PPh_3, $p$-O_2NC_6H_4CO_2H; EtO_2CN=NCO_2Et; $K_2CO_3$, MeOH; (s) $Cl_3CC(=NH)OPMB, BF_3-OEt_2; (t) HF-Py; (u) $Me_2C(OMe)_2$, PPTS; (v) TBAF; (w) PPTS, MeOH; (x) TBSOTf, Et_3N; (y) LAH; (z) Dess-Martin periodinane. \end{tabular}$ 

ment of the isomeric ketene silvl acetal generated from 178 in the absence of HMPA gave the epimeric 181 as the major product with 5:1 stereoselectivity; this was converted to  $\gamma$ -lactone 182. Whereas the stereochemistry of 180 matches the C<sub>31</sub>-C<sub>36</sub> portion of the halichondrins, that of 182 matches the  $C_{46}$ C<sub>51</sub> portions of the norhalichondrins and homohalichondrins. Thus the configuration at C<sub>3</sub> (galactose numbering) of 178 (indicated \*) sets up the C<sub>32</sub> (or  $C_{47}$ ) stereocenter, and the stereochemistry of the ester enolate determines the  $C_{31}$  (or  $C_{46}$ ) configuration. Note that use of 3,4,6-tripropionate-D-galactose glycal could eliminate three steps needed for differential protection of the hydroxyls. Experimentally, however, the rates of the Ireland-Claisen rearrangements of the C<sub>3</sub> and C<sub>4</sub> propionates were found to be similar, and selective protection of the C<sub>4</sub> hydroxyl was required, as in 178.

 $\gamma$ -Lactone 180 was converted into the  $C_{30}$  aldehyde 183, and then a Ni(II)/Cr(II)-mediated coupling reaction<sup>78</sup> was used to construct the  $C_{29}-C_{30}$  bond. Unfortunately, a 2:1 mixture of the two possible diastereomers 184 and 185 was formed, favoring the desired 184. The minor undesired 185 could be converted to 184 by using the Mitsunobu inversion procedure.85 A sequence of protecting group interconversions transformed 184 into 186, and then a fluoride ion-mediated intramolecular hetero-Michael reaction was used to close the F ring with greater than 20:1 stereoselectivity at C25, in favor of the desired 187. Note that Michael reaction of the corresponding triol initially yielded the desired diastereomer as the major product, but it rapidly isomerized to the undesired diastereomer. Protecting group exchange and partial reduction at  $C_{27}$  then gave the  $C_{27}$ - $C_{38}$  segment 176. Thus, in this synthesis, two of the seven stereogenic centers in 176 originated in the chiral pool ( $C_{35}$  and  $C_{36}$ ), and the remaining five stereogenic centers were installed using reactions relying on substrate control of asymmetric induction. Compound 176 was also synthesized from methyl L-glucopyranoside, by a longer route, which served to confirm the stereochemistry. 74b

[ $C_{27}$ - $C_{38}$  segment 176: 25 steps from 177;  $\sim 3-4$  steps per stereogenic center.]

c. C<sub>39</sub>-C<sub>54</sub> Segment Synthesis. 74c Scheme 14 outlines the synthesis of the  $C_{39}-C_{54}$  segment 163. Conjugate addition of methyl cuprate to the  $\alpha,\beta$ unsaturated lactone 188, prepared from L-ascorbic acid (189),86 afforded the single C46 stereoisomer 190, which was then transformed into epoxide 191. Construction of the C<sub>49</sub>-C<sub>50</sub> bond by Yamaguchi coupling<sup>87</sup> of **191** with alkyne **192**, obtained from (R)malic acid (193), followed by Lindlar reduction of the coupled product, gave the cis-alkene 194. VO(acac)2catalyzed<sup>88</sup> epoxidation of **194** (employing aromatic solvents for optimum stereoselectivity) and subsequent acid treatment then gave the tetrahydrofuran **195** with 7-8:1 stereoselectivity. The stereochemistry of the epoxidation was assigned on the basis of literature precedents.89 Note that at this time there was still ambiguity concerning the C<sub>50</sub>, C<sub>51</sub>, and C<sub>53</sub> configurations of halichondrin B. By using both alkyne 192 and its antipode, by generating either cis or trans alkenes at C<sub>50</sub>-C<sub>51</sub>, and by employing either  $VO(acac)_2^{88}$  or m-CPBA $^{90}$  epoxidation, the Kishi route allowed the preparation of all the stereoisomers at the  $C_{50}$ ,  $C_{51}$ , and  $C_{53}$  centers. These were prepared and their  $^1H$  NMR spectra compared with the reported<sup>71</sup> spectrum for halichondrin B. The data for 195 and 196 matched well with the reported values. and so these two diastereomers were separately taken on to halichondrin B (and its diastereomer). In this way the stereochemistry of 195 was established as that of the natural product. This study elegantly illustrates the use of total synthesis to probe the stereochemical configuration of structurally complex natural products, as well as revealing the powerful advantage offered by using acyclic methods of stereocontrol, whereby the preparation of stereochemical analogs is made synthetically viable by simple changes of reagent.

Coupling of the  $C_{44}$  aldehyde **197** (derived from **195**) with the  $C_{43}$  alkyllithium derived from bromide **198** (in turn obtained from methyl (S)-3-hydroxy-2-methylpropionate (**199**)) provided the  $C_{39}-C_{54}$  seg-

# Scheme 14. Kishi Halichondrin $C_{39}-C_{54}$ Synthesis<sup>74c</sup> a

<sup>a</sup> (a) Me<sub>2</sub>CuLi, TMSCl; (b) LAH; (c) PivCl, py; (d) PMBBr, KH; (e) AcOH, H<sub>2</sub>O; (f) NaH; N-tosylimidazole; (g) BH<sub>3</sub>·SMe<sub>2</sub>, B(OMe)<sub>3</sub>; (h) Me<sub>2</sub>CO, p-TsOH; (i) Swern oxidation; (j) (MeO)<sub>2</sub>P(=O)CHN<sub>2</sub>, 'BuOK; (k) 192, 'BuLi; 191, BF<sub>3</sub>·OEt<sub>2</sub>; (l) H<sub>2</sub>, Lindlar catalyst, quinoline; (m) 'BuOOH, VO(acac)<sub>2</sub>; (n) TFA; (o) AcOH, H<sub>2</sub>O; (p)TBSOTf, Et<sub>3</sub>N; (q) LAH; (r) Dess-Martin periodinane; (s) 194, 'BuLi; 197; (t) AgNO<sub>3</sub>, HMDS; (u) 'Bu<sub>3</sub>SnH, AIBN; (v) I<sub>2</sub>; (w) Dess-Martin periodinane; (a') DHP, H+; (b') LAH; (c') Swern oxidation; (d') LiC≡CTMS; (e') Cl<sub>3</sub>CC(=NH)OPMB, BF<sub>3</sub>·OEt<sub>2</sub>; (f') CSA, MeOH; (g') MsCl, Et<sub>3</sub>N; (h') LiBr.

ment **200**. Oxidation at  $C_{44}$ , and transformation of the alkynylsilane functionality to the vinyl iodide then provided segment **163**. Thus, in this synthesis, four of the eight stereogenic centers in **163** originated in the chiral pool ( $C_{42}$ ,  $C_{47}$ ,  $C_{48}$ , and  $C_{53}$ ), and the other four stereogenic centers were installed using reactions relying on substrate control of asymmetric induction. [ $C_{39}$ – $C_{54}$  segment **163**: 19 steps from **188** longest linear sequence; 31 steps total;  $\sim$ 4 steps per stereogenic center.]

d.  $C_{39}$ – $C_{53}$  Norhalichondrin and  $C_{39}$ – $C_{55}$  Homohalichondrin Segment Syntheses.74d Synthesis of the C<sub>39</sub>-C<sub>53</sub> segment **201** of the norhalichondrins began with the previously prepared  $\gamma$ -lactone 182 which bore the correct stereochemical configuration at C<sub>46</sub>- $C_{51}$  (Scheme 15). Compound 182 was converted into acetal 202 using the transformations already performed on the epimeric  $\gamma$ -lactone 180 (Scheme 13), and then the protecting groups changed and the oxidation level at C<sub>45</sub> adjusted to give alcohol 203, which was subsequently homologated to the C44 aldehyde **204**. The same five-step sequence that was used to complete the C<sub>39</sub>-C<sub>54</sub> halichondrin segment (i.e. 197  $\rightarrow$  163 in Scheme 14) was employed to provide 205, and finally conversion to the  $C_{53}$  methyl ester gave the  $C_{39}-C_{53}$  norhalichondrin segment **201**.

Acetal 202 was also converted into the C<sub>39</sub>-C<sub>55</sub> homohalichondrin segment 206. Oxidation at C53 of 202, followed by Still Horner-Emmons homologation<sup>91</sup> and subsequent reduction, gave the C<sub>45</sub>-C<sub>55</sub> segment 207, which was then submitted to Sharpless asymmetric epoxidation. 92 In situ acid-catalyzed cyclization of the epoxide 208 then provided the completed M-ring compound 209. Note that the C<sub>53</sub> and C<sub>54</sub> configurations of the homohalichondrins were unknown at the time. Thus, preparation of both alkene 207 and its trans isomer and use of both enantiomers of diethyltartrate ligand in the Sharpless epoxidation allowed all possible stereoisomers to be made. 93 Comparison of <sup>1</sup>H NMR data with that reported for homohalichondrin A<sup>71b</sup> identified tetrahydrofuran 209 as having the natural configuration. Elaboration of 209 to the complete C<sub>39</sub>-C<sub>55</sub> homohalichondrin segment 206 was accomplished as for the norhalichondrin segment (cf.  $202 \rightarrow 205$ ).

#### Scheme 15. Kishi Norhalichondrin C<sub>39</sub>-C<sub>53</sub> and Homohalichondrin C<sub>39</sub>-C<sub>55</sub> Syntheses<sup>74d a</sup>

<sup>a</sup> (a) DIBAL; (b) p-TsOH, MeOH; (c) Tf<sub>2</sub>O, py; NaCN; (d) DIBAL; NaBH<sub>4</sub>; (e) H<sub>2</sub>, Pd(OH)<sub>2</sub>-C; (f) EtSH, BF<sub>3</sub>·OEt<sub>2</sub>; (g) TBSOTf, Et<sub>3</sub>N; (h) I<sub>2</sub>, NaHCO<sub>3</sub>; NaBH<sub>4</sub>; (i) MsCl, Et<sub>3</sub>N; NaCN; (j) DIBAL; (k) **198**, <sup>t</sup>BuLi; the C44 aldehyde for norhalichondrin or for homohalichondrin; (l) AgNO<sub>3</sub>, HMDS; (m) <sup>n</sup>Bu<sub>3</sub>SnH, AIBN; (n) I<sub>2</sub>; (o) Dess-Martin periodinane; (p) CSA; (q) Dess-Martin periodinane; (r) NaClO<sub>2</sub>, NaH<sub>2</sub>PO<sub>4</sub>, CH<sub>2</sub>N<sub>2</sub>; (s) Dess-Martin periodinane; (t) (CF<sub>3</sub>CH<sub>2</sub>O)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Me, KHMDS, 18-crown-6; the aldehyde from step(s); (u) DIBAL; (v) <sup>t</sup>BuOOH, (+)-DET, Ti(O<sup>t</sup>Pr)<sub>4</sub>; (w) p-TsOH, wet CHCl<sub>3</sub>.

#### Scheme 16. Kishi Halichondrin B and Norhalichondrin B Syntheses<sup>74e a</sup>

 $^a$  (a) 211, NaH; 210; (b) [(Ph\_3P)CuH]\_6, H\_2; (c) NaBH\_4; (d) the more polar alcohol, Ms\_2O, Et\_3N; (e) 176 + 215, NiCl\_2 (0.5%)—CrCl\_2; (f) KH; (g) 162, LAH; (h) Dess—Martin periodinane; (i) the aldehyde from previous step + 161, NiCl\_2 (0.1%)—CrCl\_2; (j) Dess—Martin periodinane; (k) DDQ; (l) LiOH, H\_2O; (m) 2,4,6-Cl\_3C\_6H\_2COCl, Et\_3N; DMAP,  $\Delta$ ; (n) TBAF; (o) PPTS; (p) p-O\_2NC\_6H\_4COCl, py; (q) TBSOTf, Et\_3N; (r) K\_2CO\_3, MeOH; (s) 220, Dess—Martin periodinane; (t) the aldehyde from the previous step + 163 (for halichondrin B) or 201 (for norhalichondrin B), NiCl\_2 (0.1%)—CrCl\_2; (u) Dess—Martin periodinane; (v) TBAF; (w) DDQ; (x) CSA; (y) LiOH, H\_2O.

Thus, in the synthesis of **201**, three of the seven stereogenic centers in the target molecule originated in the chiral pool ( $C_{42}$ ,  $C_{50}$ , and  $C_{51}$ ), and the other four stereogenic centers were installed using reactions relying on substrate control of asymmetric induction. In the synthesis of **201**, the two additional stereogenic centers in the target molecule, at  $C_{53}$  and  $C_{54}$ , were constructed using a chiral reagent. [ $C_{39}-C_{53}$  norhalichondrin segment **201**: 26 steps from **177** longest linear sequence; 34 steps total;  $\sim$ 5 steps per stereogenic center;  $C_{39}-C_{55}$  homohalichondrin segment **206**: 27 steps from **177** longest linear sequence; 35 steps total;  $\sim$ 4 steps per stereogenic center.]

e. Completion of the Total Syntheses of Halichondrin B and Norhalichondrin B. The Assembly of the segments and completion of the total synthesis of halichondrin B (152) and norhalichondrin B (155) is outlined in Scheme 16. The  $C_{14}-C_{21}$  and  $C_{22}-C_{26}$  segments 21094 and 21195 were prepared from the precursors 212 and 213, respectively. The  $C_{21}-C_{22}$  bond construction was accomplished by a Horner-

Emmons reaction between 210 and 211, under carefully controlled conditions, followed by conjugate reduction using the Stryker reagent ([(Ph<sub>3</sub>P)CuH]<sub>6</sub>/ H<sub>2</sub>),96 to afford 214. Note that no double-bond isomerization from the  $C_{19}$  exocyclic to the  $C_{19}-C_{20}$ endocyclic position was observed during these transformations. Unfortunately, hydride reduction of 214 gave a 1:1 ratio of C<sub>23</sub> alcohol epimers. These were interconvertible via Mitsunobu inversion,85 but because their stereochemistry could not be firmly established, both were transformed separately into the corresponding mesylates and used in the next coupling reaction. A Ni(II)/Cr(II)-mediated coupling<sup>78</sup> of 215 and the  $C_{27}$ - $C_{38}$  segment 176 yielded a 6:1 mixture of the two possible allylic alcohols, which were immediately subjected to base-catalyzed E-ring cyclization to give 162 (and the undesired minor diastereomer). The  $C_{23}$  and  $C_{27}$  configurations were now established by nOe experiments.

A Ni(II)/Cr(II) coupling<sup>78</sup> of the  $C_{14}$  aldehyde derived from **162** with the  $C_1$ – $C_{13}$  segment **161**, fol-

lowed by Dess-Martin oxidation, 27 gave the enone 216. After removal of the C<sub>30</sub> PMB ether<sup>42b,c</sup> and hydrolysis of the C<sub>1</sub> methyl ester, a Yamaguchi macrolactonization<sup>47</sup> provided the macrocycle 217. TBAF-mediated deprotection of 217 led to hetero-Michael cyclization of exclusively the C9 hydroxyl onto C<sub>12</sub> with 5-6:1 stereoselectivity in the desired sense to generate saturated ketone 218, bearing the C ring, which was then cyclized using PPTS to give the complete  $C_1-C_{38}$  portion **219** of halichondrin B. At this stage, the undesired diastereomer from the hetero-Michael cyclization could be separated and recycled under TBAF conditions. Selective protection of the C<sub>35</sub> hydroxyl via temporary protection of the C<sub>38</sub> hydroxyl then gave 220. A Ni(II)/Cr(II)-mediated coupling<sup>78</sup> of the C<sub>38</sub> aldehyde derived from **220** with the  $C_{39}-C_{54}$  halichondrin segment 163 and subsequent Dess-Martin oxidation<sup>27</sup> gave the enone **221**. A three-step sequence, without isolation of intermediates, finally converted 221 into halichondrin B (152). <sup>1</sup>H NMR analysis after the first step (TBAF) indicated the partial structure 222, suggesting initial deprotection of the C<sub>48</sub> TBS ether, hemiacetal formation with the C<sub>44</sub> ketone (→ K ring), and hetero-Michael addition of the hemiacetal hydroxyl onto C<sub>40</sub>  $(\rightarrow J \text{ ring})$ . Simultaneously, deprotection of the  $C_{35}$ TBS ether led to hemiacetal formation with the C<sub>38</sub> ketone ( $\rightarrow$  H ring). Deprotection of the C<sub>41</sub> PMB ether  $(DDQ)^{42b,c}$  and acid-catalyzed spiroacetalization with the C<sub>38</sub> hemiacetal then completed the I ring. Note that the differential protection of the  $C_{48}$  hydroxyl prevents formation of the alternative 5,5spiroacetal between the  $C_{44}$  ketone and  $C_{41}$  and  $C_{48}$ hydroxyls.

The synthesis of norhalichondrin B (155) was carried out in an analagous manner using the  $C_{39}-C_{53}$  norhalichondrin segment 201, except that an additional final step was required, viz. hydrolysis of the  $C_{53}$  methyl ester of 223.

The masterful Kishi total syntheses of 152 and 155 are noteworthy for their repeated use of Ni(II)/Cr-(II)-mediated coupling reactions<sup>78</sup> to assemble complex structures, and for the several examples of hetero-Michael intramolecular ring closures. Carbohydrate-based stereocontrol strategies<sup>8</sup> supplied many of the stereogenic centers in the target molecules. Improvements to the route are now being explored by Kishi and co-workers in an attempt to enhance the synthetic supply of halichondrin B. [Halichondrin B (152): 45 steps from 177 longest linear sequence; 120 steps total; ~4 steps per stereogenic center; norhalichondrin B (155): 46 steps from 177 longest linear sequence; 124 steps total; 4 steps per stereogenic center.]

#### 2. Horita/Yonemitsu Segment Syntheses<sup>76</sup>

Horita, Yonemitsu, and co-workers have synthesized the four halichondrin B segments depicted in Scheme 17:  $C_1-C_{15}$  segment **224**,  $C_{16}-C_{26}$  segment **225**,  $C_{27}-C_{36}$  segment **226**, and  $C_{37}-C_{54}$  segment **227**. The total synthesis has not yet been reported, but assembly of the segments in the order [(**225** + **226**) + **227**] + **224** or [(**226** + **227**) + **225**] + **224** and final macrolactonization has been proposed.

a.  $C_1$ – $C_{13}$  Segment Synthesis. The Horita/Yonemitsu synthesis of a  $C_1$ – $C_{13}$  halichondrin B

Scheme 17

segment (228) involved the construction of the A and B rings by intramolecular hetero-Michael reactions under thermodynamic and kinetic conditions, respectively  $(229 \rightarrow 228 \text{ and } 230 \rightarrow 231 \text{ in Scheme } 18)$ . Treatment of epoxide 232, obtained in nine steps from D-glucose diacetonide (173), with acetic acid led to a 5-exo<sup>97</sup> C-ring cyclization to afford tetrahydrofuran 233. Differential protection of the  $C_{13}$  and  $C_{11}$ hydroxyls of 233, dithiane hydrolysis98 to reveal the C<sub>8</sub> aldehyde, and a Horner-Emmons reaction then gave  $\alpha,\beta$ -unsaturated ester 234. After reduction at C<sub>6</sub>, a Sharpless asymmetric epoxidation<sup>92</sup> gave epoxide 235, whose ring opening into triol 236 via a carbamate was attained only by employing Roush's method.99 A 13-step sequence involving protecting group exchange, inversion at C<sub>11</sub> and extension by two carbon units at C<sub>6</sub> then gave the B ring precursor **230**. The  $C_7$ ,  $C_8$  diol acetonide in **230** fixes the  $\alpha$ ,  $\beta$ unsaturated side chain in a favorable conformation for cyclization to the B ring, and so TBAF-mediated  $C_{10}$  deprotection of **230** and subsequent kinetically controlled cyclication led to the desired  $C_6$ ,  $C_{10}$ -trans tetrahydropyran 231. Note that on treatment with alkali, 231 isomerized to the thermodynamically more stable tetrahydropyran 6-epi-231.100 A six-step sequence converted 231 to the A-ring precursor 229. Cyclization under thermodynamic conditions was expected to provide the desired  $C_3$ ,  $C_7$ -cis tetrahydropyran. Brief treatment of 229 with TBAF, followed by trityl protection of the C<sub>13</sub> primary hydroxyl, gave a 2:1 mixture of the C<sub>3</sub>,C<sub>7</sub>-cis tetrahydropyran 228 and its C<sub>3</sub> epimer. Prolonged exposure of this mixture to TBAF increased the ratio to 19:1, in favor of the desired 228. Elaboration of 228 to the  $C_1-C_{15}$ segment **224** requires a two-carbon extension at  $C_{13}$ , but has not yet been reported. Thus, in the synthesis of 228, two stereogenic centers originated in the chiral pool ( $C_9$  and  $C_{10}$ ), one stereogenic centers was constructed using a chiral reagent (C<sub>8</sub>), and the remaining five stereogenic centers were installed using reactions relying on substrate control of asymmetric induction.  $[C_1-C_{13}]$  segment **228**: 2.2% overall yield from 173; 44 steps;  $\sim 5-6$  steps per stereogenic center.

### Scheme 18. Horita/Yonemitsu Halichondrin B C<sub>1</sub>-C<sub>13</sub> Synthesis<sup>76a</sup> a

 $^{a}\left(a\right) BnCl, NaH; (b) 0.8 N H_{2}SO_{4}, MeOH; (c) BzCl, py; (d) 6 N HCl; (e) HS(CH_{2})_{3}SH, ZnCl_{2}; (f) PMPCH(OMe)_{2}, CSA; (g) MsCl, Et_{3}N; (h) K_{2}CO_{3}; (i) 'BuOK; (j) 80\% AcOH; (k) TrCl, DMAP, Et_{3}N; (l) PMBCl, NaH; (m) CSA; (n) TBDPSCl, imidazole; (o) HgO, BF_{3}OEt_{2}; (p) (^{i}PrO)_{2}P(=O)CH_{2}CO_{2}Et, 'BuOK; (q) DIBAL; (r) (-)-DET, Ti(^{i}OPr)_{4}, 'BuOOH, molecular sieves; (s) PhNCO, Et_{3}N; (t) BF_{3}OEt_{2}; (u) K_{2}CO_{3}; (v) PivCl, DMAP, Et_{3}N; (w) H_{2}C=C(OMe)Me, PPTS; (x) DDQ; (y) Swern oxidation; (z) NaBH_{4}; (a') TBAF; (b') H_{2}C=C(OMe)Me, p-TsOH; (c') Na, liquid NH_{3}; (d') PivCl, DMAP, py; (e') TESCl, imidazole, DMAP; (f') DIBAL; (g') Swern oxidation; (h') (^{i}PrO)_{2}P(=O)CH_{2}CO_{2}Et, ^{i}BuOK; (i') TBAF; (j') 'BuOK; (k') LAH; (l') TsCl, DMAP, Et_{3}N; (m') NaCN; (n') DIBAL; (o') Ph_{3}P=CHCO_{2}Me; (p') CSA; (q') TBAF; (r') TrCl, DMAP, Et_{3}N; (s') TBAF.$ 

#### Scheme 19. Horita/Yonemitsu Halichondrin B C<sub>16</sub>-C<sub>26</sub> Synthesis<sup>76b</sup> a

 $^a$  (a) TBDPSCl, Et<sub>3</sub>N; (b) Ca(BH<sub>4</sub>)<sub>2</sub>; (c) TsCl, Et<sub>3</sub>N; (d) NaCN; (e) DIBAL; H<sup>+</sup>; (f) Ph<sub>3</sub>P=CHCO<sub>2</sub>Me; (g) DIBAL; (h) (−)-DET, Ti(O<sup>i</sup>Pr)<sub>4</sub>, <sup>i</sup>BuOOH; (h) (+)-DET, Ti(O<sup>i</sup>Pr)<sub>4</sub>, <sup>i</sup>BuOOH; (i) Red-Al; (j) PMPCH(OMe)<sub>2</sub>, p-TsOH; (k) DIBAL; (l) Swern oxidation; (m) MeOH, H<sup>+</sup>; (n) BH<sub>3</sub>'Me<sub>2</sub>S; cat. NaBH<sub>4</sub>; (o) H<sub>2</sub>C=C(Me)OMe, PPTS; (p) (MeO)<sub>2</sub>P(=O)Me, <sup>n</sup>BuLi; (q) **247**, <sup>n</sup>BuLi; **245** or **246**; (r) LiI, LAH; (s) AcOH, MeOH; (t) I<sub>2</sub>, NaHCO<sub>3</sub>; (u) NaH; (v) Raney Ni, H<sub>2</sub>; (w) BzCl, py; (x) Swern oxidation; (y) Ph<sub>3</sub>P=CH<sub>2</sub>; (z) TBAF; (a') TsCl, Et<sub>3</sub>N, DMAP; (b') NaCN; (c') K<sub>2</sub>CO<sub>3</sub>, MeOH; (d') SEMCl, <sup>i</sup>Pr<sub>2</sub>NEt; (e') DIBAL; 1 N HCl; (f') Jones oxidation; (g') CH<sub>2</sub>N<sub>2</sub>.

b.  $C_{16}-C_{26}$  Segment Syntheses. The Scheme 19 outlines the syntheses of two  $C_{16}-C_{26}$  segments 225 and 237, epimeric at  $C_{23}$ , in which the D ring was constructed by iodoetherification (238  $\rightarrow$  239 and 240  $\rightarrow$  241, respectively). Horita and Yonemitsu envisage coupling of  $C_{16}-C_{26}$  and  $C_{27}-C_{36}$  segments by aldol construction of the  $C_{26}-C_{27}$  bond, followed by cyclization to form ring E. Cyclization with inversion at  $C_{23}$  would require 225 as the choice of  $C_{16}-C_{26}$  segment, cyclization with retention would require 237. The syntheses of 225 and 237 both began with the allylic alcohol 242, obtained from methyl (S)-3-hydroxy-2-methylpropionate (199) in seven steps. Sharpless epoxidation of 242, using the (-)-diethyl tartrate ligand, and subsequent Red-Al reduction 29 gave the

23R diol **243**. Likewise, use of the antipodal Sharpless catalyst afforded the 23S epimer **244**. Formation of the p-methoxybenzylidene acetal, regioselective reductive cleavage to the C<sub>21</sub> alcohol, and oxidation then provided the C<sub>21</sub> aldehydes **245** (from **243**) and **246** (from **244**). Next, construction of the C<sub>20</sub>-C<sub>21</sub> bond by Horner-Emmons reaction of aldehyde **245** and β-keto phosphonate **247**, derived from (R)-malic acid (**193**), gave the C<sub>16</sub>-C<sub>26</sub> segment **248**. Stereoselective reduction of **248** at C<sub>23</sub> under chelation-controlled conditions<sup>102</sup> and removal of the acetonide then provided the D-ring precursor **238**. Iodoetherification ( $I_2$ , NaHCO<sub>3</sub>) of **238** gave exclusively the desired C<sub>17</sub>,C<sub>20</sub>-trans tetrahydrofuran **239**. Note that, in preliminary studies, the TES-protected ana-

### Scheme 20. Horita/Yonemitsu Halichondrin B C<sub>27</sub>-C<sub>36</sub> Synthesis<sup>76c a</sup>

 $\begin{tabular}{ll} $a$ (a) $p$-MeO(C_6H_4)CH(OMe)_2, $p$-TsOH; (b) LAH; (c) TBSCl, imidazole; (d) DIBAL; (e) CSA, MeOH; (MeO)_2CMe_2, $p$-TsOH; (f) Swern oxidation; (g) (CF_3CH_2O)_2P(=O)CH_2CO_2Me, 18-crown-6, KHMDS; (h) 1 N HCl, MeOH; (i) TBDPSCl, imidazole; (j) $p$-TsOH; (k) DIBAL; (l) CSA, MeOH; (m) DDQ; (n) $m$-CPBA, $261; (o) MeMgCl, MeLi; (p) Ac_2O, Et_3N, DMAP; (q) TBAF; (r) TsCl, Et_3N; (s) K_2CO_3, MeOH; (t) TBSOTf, Et_3N; (u) NaCN; (v) DIBAL; (w) 1 N HCl; (x) Ph_3P=CHCO_2Me; (y) DIBAL; (z) (-)-DET, Ti(O^iPr)_4, ^BuOOH; (a') TsCl, Et_3N, DMAP; (b') NaI, NaHCO_3; (c') ^BuLi; (d') Ac_2O, Et_3N, DMAP; (e') OsO_4, NMO; (f') NaIO_4; (g') NaBH_4; (h') Ac_2O, Et_3N, DMAP; (i') H_2C=CHCH_2TMS, BF_3*OEt_2, TMSOTf; (j') TBSOTf, Et_3N; (k') K_2CO_3, MeOH; (l') Me_2C(OMe)_2, CSA; (m') OsO_4, NMO; (n') NaIO_4. \end{tabular}$ 

logue (249) of 248 was prepared. Unfortunately, reduction of 249 (using NaBH<sub>4</sub>-CeCl<sub>3</sub>) gave a 1.8:1 mixture of  $C_{19}$  epimers **250** and **251**. Whereas iodoetherification of 250 gave exclusively 239, iodoetherification of 251 (or the derived triol 19-epi-238) gave the unwanted  $C_{17}$ ,  $C_{20}$ -cis tetrahydrofuran isomer ( $\sim$ 2:1 mixture at  $C_{21}$ ) as the major product.<sup>103</sup> With 239 in hand, from 248, reduction of the iodide via an olefin and protection of the primary hydroxyl afforded 252. Swern oxidation<sup>38</sup> at C<sub>19</sub> and Wittig methylenation then gave 253, which was converted into the  $C_{16}$ – $C_{26}$  segment **225** in a further eight steps. In the same manner, Horner-Emmons coupling of 247 and the aldehyde 246 gave 240 which was transformed, via iodoetherification to 241, into the  $C_{23}$ -epimeric,  $C_{16}$ - $C_{26}$  segment 237. Thus, in the synthesis of 225, two stereogenic centers originated in the chiral pool ( $C_{17}$  and  $C_{25}$ ), one stereogenic center was constructed using a chiral reagent ( $C_{23}$ ), and the remaining stereogenic center was installed using substrate control of asymmetric induction  $(C_{20})$ .  $[C_{16}-C_{26} \text{ segment } \mathbf{225}: 16\% \text{ overall yield from } \mathbf{199};$ 29 steps longest linear sequence; 33 steps total; ~8 steps per stereogenic center.]

c.  $C_{27}$ - $C_{36}$  Segment Syntheses. 76c The route to the C<sub>27</sub>-C<sub>36</sub> segment 226, in which the F ring is constructed by a stereoselective C-glycosidation (254 → 255), is outlined in Scheme 20. Alcohol 256, prepared in five steps from dimethyl L-tartrate (257), 104 was converted to the F-ring precursor 258 by a four-step sequence including a Z-selective Horner-Emmons reaction using the procedure of Still.91 Sequential lactonization, reduction to the lactol, methylation, and, finally, oxidative removal of the C<sub>32</sub> PMB ether, 42b,c then gave allylic alcohol 259. An m-CPBA epoxidation of **259**, directed by the  $C_{32}$  hydroxyl, exclusively afforded the  $\beta$ -epoxide 260. Note that no reaction occurred unless the radical scavenger phenol 261 was also present. 106 Trans-diaxial opening of epoxide 260 with "Me2Mg" (obtained from the supernatant of a mixture of MeMgCl and salt-free

MeLi in ether and THF) gave 262 exclusively. Compound **262** was then converted into bis(acetate) 263. Unfortunately, C-glycosidation of 263 using allyltrimethylsilane, in the presence of boron trifluoride etherate, 54 gave a mixture of  $\alpha$  and  $\beta$  epimers (264). Replacement of the acetyl groups with larger TBS groups, however, as in 265, led to exclusive α-allylation (with in situ loss of the TBS groups) to give **266**, but only in low yield. The selective  $\alpha$ -glycosidation of **265** prompted a search for a substrate that would give higher yields. Thus 265 was transformed into allylic alcohol 267; Sharpless epoxidation<sup>92</sup> and iodination then gave **268**. After lithiumhalogen exchange on 268, in situ epoxide opening and acetylation of the product gave 269, which was converted into **254**. C<sub>29</sub>-Allylation of **254** was now completely α-selective, and high yielding (89% cf. 38%) for **265**), and the resulting diol was reprotected with TBS groups to give 255. Finally, protecting group exchange at  $C_{35}$  and  $C_{36}$  and oxidative cleavage of the double bond then gave the  $C_{27}$ - $C_{36}$  segment **226**. Thus, in this synthesis, two stereogenic centers originated in the chiral pool ( $C_{32}$  and  $C_{33}$ ), one stereogenic center was constructed using a chiral reagent (C<sub>35</sub>), and the remaining three stereogenic centers were installed using reactions relying on substrate control of asymmetric induction.  $[C_{27}-C_{36}]$ segment 226: 40 steps from 257; ~7 steps per stereogenic center.]

d.  $C_{37}$ – $C_{54}$  Segment Syntheses. <sup>76d</sup> Scheme 21 outlines the synthesis of the  $C_{37}$ – $C_{54}$  segment 227 involving construction of the three consecutive JKL rings. Alcohol 256, derived from dimethyl L-tartrate <sup>104</sup> and used as a starting material in the  $C_{27}$ – $C_{36}$  segment synthesis (Scheme 20), was converted into its homologue 270 and thence to the L-ring precursor 271. Sharpless epoxidation <sup>92</sup> of 271 was accompanied by an *in situ* 5-exo <sup>97</sup> cyclization of epoxide 272 to give the tetrahydrofuran 273 directly. Inversion at  $C_{51}$  and two-carbon extension to provide  $C_{53}$  and  $C_{54}$  was now required. Conversion to epoxide

#### Scheme 21. Horita/Yonemitsu Halichondrin B C<sub>37</sub>-C<sub>54</sub> Synthesis<sup>76d a</sup>

a (a) Swern oxidation; (b) Ph<sub>3</sub>PMe<sup>+</sup>Br<sup>-</sup>, 'BuOK; (c) (Sia)<sub>2</sub>BH; H<sub>2</sub>O<sub>2</sub>, NaOH; (d) Swern oxidation; (e) (EtO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Me, NaH; (f) 1 N HCl, MeOH; (g) TBSCl, imidazole; (h) DIBAL; (i) (−)-DET, Ti(O<sup>i</sup>Pr)<sub>4</sub>, 'BuOOH; (j) BzCl, py; (k) MsCl, Et<sub>3</sub>N, DMAP; (l) K<sub>2</sub>CO<sub>3</sub>, MeOH; (m) 'BuOK; (n) H<sub>2</sub>C=CHMgBr, CuI; (o) H<sub>2</sub>C=CH(OEt), PPTS; (p) OsO<sub>4</sub>, NMO; (q) PivCl, py; (r) Swern oxidation; (s) PPTS, MeOH; (t) Et<sub>2</sub>BOMe, NaBH<sub>4</sub>; (u) H<sub>2</sub>C=CH(OMe), PPTS; (v) TBAF; (w) Swern oxidation; (x) Ph<sub>3</sub>P, CBr<sub>4</sub>; (y) LDA; (z) K<sub>2</sub>CO<sub>3</sub>, MeOH; (a') TBDPSCl, inidazole; (b') <sup>n</sup>BuLi; ClCO<sub>2</sub>Me; (c') MeMgCl, CuI; (d') DDQ; (e') NaI; (f') H<sub>2</sub>, 10% Pd−C; (g') PhCH(OMe)<sub>2</sub>, CSA; (h') Swern oxidation; (i') (EtO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Me, NaH; (j') DIBAL; (k') (−)-DET, 'BuOOH, Ti(O<sup>i</sup>Pr)<sub>4</sub>; (l') Red-Al; (m') H<sub>2</sub>C=C(OMe)Me, PPTS; (n') Na, liquid NH<sub>3</sub>; (o') TBDPSCl, imidazole; (p') PMBCl, KHMDS; (q') TBAF; (r') TsCl, Et<sub>3</sub>N, DMAP; (s') NaI, NaHCO<sub>3</sub>; (t') 286, 'BuLi, CeCl<sub>3</sub>; 282; (u') p-TsOH; (v') Swern oxidation; (w') NaClO<sub>2</sub>, NaH<sub>2</sub>PO<sub>4</sub>; (x') TMSCHN<sub>2</sub>; (y') (MeO)<sub>2</sub>P(=O)Me, "BuLi.

**274** followed by a regioselective cuprous iodidecatalyzed addition of vinylmagnesium bromide achieved both these aims, and protection of the homoallylic alcohol product then gave **275**. Osmylation of alkene 275 afforded a diastereomeric mixture that was converted via oxidation and protecting group exchange to the single  $\beta$ -hydroxy ketone **276**. A Narasaka<sup>46a,b</sup> reduction of **276** then correctly set up the C<sub>53</sub> stereogenic center with 20:1 diastereoselectivity, to afford diol 277. Introduction of the C<sub>46</sub> methyl substituent was achieved via alkyne 278. Thus **277** was transformed into dibromoalkene **279**, followed by reaction with LDA to generate an alkyne, which was then methoxycarbonylated to provide 278. Reaction with dimethylcopper resulted in a cis carbocupration<sup>107</sup> which furnished alkene **280**. After transformation of **280** to the  $\alpha,\beta$ -unsaturated lactone 281, heterogeneous hydrogenation exclusively on the convex face delivered the C<sub>44</sub>-C<sub>54</sub> coupling segment **282** with the correct configuration at  $C_{46}$ , thus completing construction of the K ring. Meanwhile, the known triol 283, derived from D-tartaric acid, 108 was converted into allylic alcohol 284. Sharpless epoxidation<sup>92</sup> and Red-Al reduction<sup>29</sup> then gave the 1,3-diol 285 selectively. A sequence of acetonide protection at C<sub>38</sub> and C<sub>40</sub>, protecting group exchange at  $C_{41}$  and  $C_{43}$ , and conversion to the  $C_{43}$  iodide then gave the C<sub>38</sub>-C<sub>43</sub> coupling segment 286. Lithiumhalogen exchange in 286 and addition to lactone 282, in the presence of cerium trichloride, 109 then provided the C<sub>38</sub>-C<sub>54</sub> segment **287**. Acid-catalyzed acetonide

deprotection of 287 and in situ stereoselective spiroacetalization at  $C_{44}$  afforded **288**, which completed the construction of the JKL ring system. Finally, conversion of 288 to the  $\beta$ -keto phosphonate 227 provided the desired C<sub>37</sub>-C<sub>54</sub> segment. Thus, in this synthesis, three stereogenic centers originated in the chiral pool (C<sub>41</sub>, C<sub>47</sub>, and C<sub>48</sub>), three stereogenic centers were constructed using chiral reagents (C<sub>40</sub>,  $C_{50}$ , and  $C_{51}$ ), and the remaining four stereogenic centers were installed using reactions relying on substrate control of asymmetric induction. Coupling of all the halichondrin segments **224–227** to complete a total synthesis is currently in progress by Horita, Yonemitsu, and co-workers.  $[C_{37}-C_{54}]$  segment **227**: 2.7% overall yield from **257**; 43 steps longest linear sequence; 56 steps total; ~6 steps per stereogenic center.]

#### 3. Salomon Segment Syntheses<sup>75</sup>

The stereogenic centers in the halichondrins are located in several discrete regions of the molecule, and this can be exploited in the synthetic plan. Salomon and co-workers have synthesized each stereochemically isolated segment of halichondrin B starting from a variety of inexpensive commercially available carbohydrates: a  $C_1-C_{15}$  segment from D-ribose, a  $C_1-C_{21}$  segment **289** from D-ribose and D-glucose, a  $C_{27}-C_{35}$  segment **290** from D-glucose, and a  $C_{37}-C_{51}$  segment **291** from D-mannitol (Scheme 22). a.  $C_{27}-C_{35}$  Segment Syntheses. Scheme 23

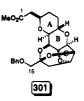
a.  $C_{27}$ – $C_{35}$  Segment Syntheses. Scheme 23 outlines the route to the  $C_{27}$ – $C_{35}$  segment 290, in

#### Scheme 22

Scheme 23. Salomon Halichondrin B  $C_{27}$ – $C_{35}$  Synthesis<sup>75a</sup> a

 $^a$  (a) MeOH; (b) PhCHO, ZnCl2; (c) N-tosylimidazole, NaOMe; (d) NaH; (e) MeMgCl; (f) TFAA, DMSO; Et3N; (g) Et3N; (h) LAH; (i) H2SO4, H2O; (j) MeCOMe, ZnCl2, H3PO4; (k) AcOH, H2O; (l) NaIO4; (m) H2C=C(OTBS)S'Bu, TiCl4; (n) 0.2 N NaOH; (o) TFA, H2O; (p) Ph3P=CHCO2'Bu; (q) Na; (r) Cl3CC(=NH)OPMB, TfOH.

which D-glucose was used to provide the F ring. This necessitated replacement of hydroxyl with methyl at  $C_{31}$ , homologation at  $C_{34}$  and  $C_{29}$ , and epimerization at C<sub>33</sub>. Initial selective protection of all but the *trans* hydroxyls at  $C_{30}$  and  $C_{31}$  of D-glucose (292)<sup>110</sup> was followed by selective to sylation of the  $C_{30}$  hydroxyl and formation of epoxide  ${\bf 293}.^{111}$  Regioselective opening of this epoxide at C<sub>31</sub> with MeMgCl (axial attack) introduced the methyl group with the incorrect configuration. Oxidation to the ketone at  $C_{30}$  and base-catalyzed epimerization of the C<sub>31</sub> methyl to the thermodynamically preferred equatorial position, however, afforded 294. Stereoselective reduction at  $C_{30}$  (axial attack) and protecting group hydrolysis then gave tetrol  $\bf 295.^{112}$  Inversion of the stereochemistry at C<sub>33</sub> and extension of the side chain was achieved by a C-C bond cleavage sequence exploiting the pyranose to furanose interconversion which occurred on acetonide protection of 295.113 Thus selective mono deacetalization at C<sub>33</sub>-C<sub>34</sub> of 296 and oxidative cleavage of the resulting diol gave the C<sub>33</sub> aldehyde 297, so destroying the incorrect stereogenic center. Entirely stereoselective (>99:1) generation of the correct C<sub>33</sub> configuration then ensued from chelation-controlled addition of a silyl ketene thioacetal, 114 which supplied carbon atoms 34 and 35, to



afford thioester 298. After base-catalyzed thioester

Figure 4. Halichondrin C<sub>1</sub>-C<sub>15</sub> subunit.

hydrolysis, acid-catalyzed acetonide cleavage led to interconversion of furanose back to pyranose. In situ lactonization then provided 299. Finally, homologation at C29 by Wittig olefination of 299 and intramolecular hetero-Michael reaction of the intermediate  $\alpha,\beta$ -unsaturated ester 300 was followed by PMB protection of the C<sub>30</sub> hydroxyl to supply the C<sub>27</sub>-C<sub>35</sub> segment 290. Note that the hetero-Michael cyclization to form ring F occurred with 97% ds, under the thermodynamic conditions employed, in favor of the desired C<sub>29</sub>,C<sub>33</sub>-trans tetrahydropyran which adopts a chair conformation with the  $C_{29}$  side chain equatorial. Thus, in the synthesis of 290, one stereogenic center originated in the chiral pool (C<sub>32</sub>), and the other four stereogenic centers were installed using reactions relying on substrate control of asymmetric induction.  $[C_{27}-C_{35} \text{ segment } 290: 7.9\% \text{ overall yield}$ from 292; 18 steps; ~4 steps per stereogenic center.] b.  $C_1$ - $C_{15}$  Segment Syntheses. 75b The synthesis of a  $C_1-C_{15}$  segment (301 in Figure 4) is depicted in Scheme 24. D-Ribose was used to provide the B ring carbons, and intramolecular hetero-Michael reactions were employed to construct both the A  $(302 \rightarrow 301)$ and C (303  $\rightarrow$  304) rings. Oxidation of the commercially available ribofuranoside **305** to the  $C_{10}$ aldehyde<sup>115</sup> was followed by Wittig olefination. An early variant of the Sharpless asymmetric dihydroxylation<sup>116</sup> then afforded diol **306** with 71% ds. The configuration at  $C_{11}$  needed to be inverted, and this was accomplished by conversion to the cyclic sulfate and regioselective nucleophilic substitution<sup>117</sup> at C<sub>11</sub> to provide **307**. A three-step sequence of debenzoylation, acetonide formation, and DIBAL reduction then afforded the  $C_{12}$  aldehyde 308 which underwent Wittig olefination to supply enone 309. TFA-mediated acetonide hydrolysis of 309 produced a tetrol and stronger acid (Dowex 50W) opened the furanoside to generate intermediate 303. In situ intramolecular hetero-Michael reaction, to close the C ring, and hemiacetal formation, to close the B ring, then took place; acetylation provided the product **304**. Note that the wrong configuration resulted at  $C_{12}$ , and all attempts to epimerize this center at this stage led to decomposition. Allylation of 304 with allyltrimethylsilane using a trityl perchlorate-catalyzed reaction<sup>118</sup> stereoselectively gave **310** (axial attack). Compound **310** was then converted into  $\alpha,\beta$ -unsaturated ester 311 in three steps. Epimerization at  $C_{12}$  was now feasible. Methoxide-catalyzed transesterification of 311 and hydroxyl deprotection led to partial epimerization to the desired  $C_{12}$  epimer 312 (2:1 mixture of 312 and 313). Treatment of this mixture with PPTS then effected spiroacetalization at  $C_{14}$  of 312 to generate the desired tetracycle 302. This acetal proved separable from the unreacted 313, which upon treatment with sodium methoxide provided more **312** by epimerization. Finally, methox-

## Scheme 24. Salomon Halichondrin B C<sub>1</sub>-C<sub>15</sub> Synthesis<sup>75b a</sup>

 $^a$  (a) DMSO, TFAA, Et<sub>3</sub>N; (b) Ph<sub>3</sub>P=CHCO<sub>2</sub>Et; (c) OsO<sub>4</sub>, NMO, dihydroquinidine p-chlorobenzoate; (d) SOCl<sub>2</sub>, Et<sub>3</sub>N; (e) RuCl<sub>3</sub>, NaIO<sub>4</sub>; (f)  $^n$ Bu<sub>4</sub>NOCOPh; (g) H<sub>2</sub>SO<sub>4</sub>, H<sub>2</sub>O; (h) Ba(OMe)<sub>2</sub>; (i) Me<sub>2</sub>C(OMe)<sub>2</sub>, PPTS; (j) DIBAL; (k) Ph<sub>3</sub>P=CHCOCH<sub>2</sub>OBn; (l) TFA, H<sub>2</sub>O; (m) Dowex 50W, H<sub>2</sub>O; (n) Ac<sub>2</sub>O, py, DMAP; (o) H<sub>2</sub>C=CHCH<sub>2</sub>SiMe<sub>3</sub>, Ph<sub>3</sub>CClO<sub>4</sub>; (p) (Sia)<sub>2</sub>BH; H<sub>2</sub>O<sub>2</sub>, NaOH; (q) py-SO<sub>3</sub>, DMSO; Et<sub>3</sub>N; (r) Ph<sub>3</sub>P=CHCO<sub>2</sub>Me; (s) NaOMe, MeOH; (t) PPTS; (u) Me<sub>3</sub>BnNOMe.

## Scheme 25. Salomon Halichondrin B C<sub>37</sub>-C<sub>51</sub> Synthesis<sup>75c a</sup>

 $^a$  (a)  $p\text{-MeO}(C_6H_4)\text{CHO}$ , molecular sieves; (b) TBSCl, imidazole; (c) NaH, imidazole; CS2; MeI; (d)  $^n\text{Bu}_3\text{SnH}$ , AIBN; (e) TBAF; (f) BnBr, NaH; (g) DIBAL; (h) DDQ, anhydrous; (i) Swern oxidation; (j) (MeO)2P(=O)CH2COTBS, NaH; (k) Me2Cu(CN)Li2, TMSCl; (l) NaOH, 30% H2O2; HCl; (m) PhOP(=O)Cl2; PhSH; (n) Ph3P=CH2; (o) **320** + **323**,  $\Delta$ ; (p) Me2Cu(CN)Li2, TMSCl; (q) (NH4)2Ce(NO3)6; (r) TBSOTf, Et3N; (s) Raney Ni, H2; (t) TBSCl, imidazole; (u) DDQ; (v) Swern oxidation; (w) **323** + **328**,  $\Delta$ ; (x) Me2Cu(CN)Li2, TMSCl; (y) DDQ; (z) TBAF; (a') 1% HCl.

ide-induced intramolecular hetero-Michael reaction of **302** (thermodynamic conditions) closed the A ring and completed the synthesis of the  $C_1$ – $C_{15}$  segment **301**. The correct configuration of the  $C_3$  stereogenic center was generated in the cyclization reaction, with the carbomethoxymethyl substituent adopting an equatorial orientation in the chair conformation of tetrahydropyran A. Thus, in the synthesis of **301**, three stereogenic centers originated in the chiral pool  $(C_7$ – $C_9$ ), one stereogenic center was installed using a chiral reagent  $(C_{10})$ , and the other five stereogenic centers were installed using reactions relying on substrate control of asymmetric induction.  $[C_1$ – $C_{15}$ 

segment 301: 0.7% overall yield from 305; 21 steps;  $\sim$ 2 steps per stereogenic center.]

c.  $C_{37}$ – $C_{51}$  Segment Synthesis. The  $C_{37}$ – $C_{51}$  segment **291**, containing the IJKL tetracyclic array, was synthesized as outlined in Scheme 25. Owing to the double anomeric effect and the preference for diequatorial disposition of the  $C_{42}$  and  $C_{46}$  methyl substituents on the J and K rings, Salomon and coworkers reasoned that the required IJKL array might be obtained by diastereoselective spiroacetalization, under thermodynamic control, at  $C_{44}$  of ketodiol intermediates such as **314**. Diastereoselective conjugate addition of two methyl nucleophiles to a

"a (a) BnBr, NaH; (b) AcOH; (c) Ac<sub>2</sub>O, py; (d) Ba(OH)<sub>2</sub>, MeOH; (e) TsCl, py; (f) HCl, MeOH, (g) PhOC(=S)Cl, DMAP; (h) "Bu<sub>3</sub>SnH, AIBN; (i) TFA, H<sub>2</sub>O; (j) NaBH<sub>4</sub>; (k) NaI; (l) TBSCl, imidazole; (m) Ph<sub>3</sub>P=CHCOCH<sub>2</sub>Li; (n) 331 + 336; (o) (NH<sub>4</sub>)<sub>2</sub>Ce(NO<sub>3</sub>)<sub>6</sub>, H<sub>2</sub>O; (p) AcOH, H<sub>2</sub>O; (q) Triton B; (r) Ac<sub>2</sub>O, py, DMAP; (s) H<sub>2</sub>C=CHCH<sub>2</sub>TMS, HClO<sub>4</sub>; (t) (Sia)<sub>2</sub>BH; H<sub>2</sub>O<sub>2</sub>, NaOH; (u) Swern oxidation; (v) Ph<sub>3</sub>P=CHCO<sub>2</sub>Me; (w) NaOMe, MeOH; (x) NaOMe, MeCN; PPTS; (y) TBSCl, imidazole; (z) Triton B methoxide.

dienone was envisaged to construct  $314.^{120}$  Due to the  $C_2$  symmetry of the IJKL portion of halichondrin B, the rings I and L are identical, as are J and K. The route pursued cleverly exploits this symmetry feature.

Using known chemistry, acid-catalyzed cyclization of D-mannitol (315) gave tetrol 316.121 A sequence of (i) selective formation of the monoacetal at C<sub>40</sub> and  $C_{42}$  of **316**; (ii) selective silvlation of the  $C_{37}$  primary hydroxyl; (iii) Barton deoxygenation<sup>37</sup> at C<sub>39</sub>; and finally, (iv) protecting group exchange at C<sub>37</sub>, then gave 317 which bears all the stereogenic centers of both the I and L rings. Reductive cleavage of 317 with DIBAL<sup>122</sup> generated a 6:4 ratio of the differentially protected regioisomers 318 and 319. Either could be quantitatively recycled to 317 by oxidation with DDQ under anhydrous conditions. 42a Swern oxidation<sup>38</sup> of 318 provided the C<sub>42</sub> aldehyde 320. A Horner-Emmons reaction then provided the  $\alpha,\beta$ unsaturated acyl silane 321, which underwent diastereoselective 1,4-addition with Me<sub>2</sub>Cu(CN)Li<sub>2</sub>, <sup>123</sup> in the presence of TMSCl, 124 to provide 322. A sequence of (i) oxidative desilylation, (ii) conversion of the resulting acid to the thioester, (iii) acylation of methylenetriphenylphosphorane, and (iv) condensation of the resulting ylid 323 with aldehyde 320 then gave enone 324. A second diastereoselective 1,4methyl addition gave the key  $C_2$ -symmetric ketone **325**. Note that use of the  $\alpha,\beta$ -unsaturated acyl silane **321** was required, since the corresponding  $\alpha,\beta$ unsaturated ester was completely unreactive with methyl cuprates, even in the presence of TMSCl;<sup>124</sup> and although the corresponding  $\alpha,\beta$ -unsaturated methyl ketone underwent conjugate addition, aldol addition of the product with aldehyde 320 did not generate enone 324. Note also that this stepwise approach for construction of the IJKL-ring carbon skeleton 325 (i.e.  $320 \rightarrow 321 \rightarrow 322 \rightarrow 324 \rightarrow 325$ ) proved to be necessary since dienones such as 326 proved to be unreactive toward Me<sub>2</sub>CuLi, and the use of TMSCl to promote the conjugate addition<sup>124</sup> led to undesired side reactions such as Nazarov cyclizations. 125 Upon treatment of 325 with ceric ammonium nitrate, oxidative removal of the PMB ethers<sup>42b,c</sup> gave dihydroxy ketone 314, which was

followed by spiroacetalization to afford the  $C_2$ -symmetric IJKL segment 327.

Since the symmetry of the IJKL segment must be broken upon incorporation into halichondrin B, a nonsymmetric analogue of **327** was required. This was obtained by employing both the regioisomers 318 and 319. Thus 319 was converted to aldehyde 328, which involved selective hydrogenolysis with Raney nickel of a benzyl in the presence of a p-methoxybenzyl ether. 126 Wittig coupling of 328 with phosphorus ylide 323, already obtained from 320, generated an enone and then diastereoselective conjugate methyl addition provided the unsymmetrical ketone 329. Deprotection of 329 with DDQ42b,c and then TBAF gave triol 330, which afforded the IJKL C<sub>37</sub>-C<sub>51</sub> segment **291** upon treatment with dilute hydrochloric acid. Thus, in this synthesis, six stereogenic centers originated in the chiral pool, and the other three stereogenic centers  $(C_{42}, C_{44}, and C_{46})$  were installed using reactions relying on substrate control of asymmetric induction.  $[C_{37}-C_{51}]$  segment **291**: 19 steps longest linear sequence from 315; 24 steps total; ~3 steps per stereogenic center.]

d.  $C_1$ – $C_{21}$  Segment Synthesis. The synthesis of the  $C_1$ – $C_{21}$  segment **289** is outlined in Scheme 26. A D-ring segment **331** was obtained from D-glucose, the B ring was derived from D-ribose, and intramolecular hetero-Michael reactions were employed to construct the A and C rings.

Benzyl protection  $^{127}$  of the  $C_{19}$  hydroxyl of D-glucose diacetonide (173) followed by selective cleavage of the  $C_{16}-C_{17}$  acetonide and tosylation of the resulting diol gave 332. After acid-catalyzed transacetalization and intramolecular O-alkylation to afford 333, Barton deoxygenation  $^{37}$  at  $C_{16}$  then provided the D-ring segment 334. Adjustment of oxidation level at  $C_{21}$  followed by protection at  $C_{21}$  and iodine introduction at  $C_{16}$  gave 335, which was elaborated into the  $C_{13}-C_{21}$  segment 331. Wittig olefination of 331 with the  $C_{12}$  aldehyde 336 (obtained in an analogous manner to the preparation of 308 from D-ribose outlined in Scheme 24) then gave  $C_6-C_{21}$  segment 337. Selective removal of the  $C_6$  PMB group was accomplished using ceric amonium nitrate to give 338, followed by mild acid hydrolysis of the acetonides and silyl group.

#### Scheme 27

Treatment with Triton B, followed by acetylation of the crude product, gave the BC-ring segment **339**. It appears that after acetonide hydrolysis, the furanoside (as in **338**) to pyranoside (as in the B ring of **339**) interconversion is essentially complete, and the intramolecular hetero-Michael reaction of the revealed  $C_9$  hydroxyl onto  $C_{12}$  is facile under the basic conditions of Triton B. Note that a similar transformation of the methyl furanoside  $309 \rightarrow 304$  (Scheme 24) was much lower yielding, due to side reactions occurring under the more strongly acidic conditions needed to hydrolyse the methyl furanoside of **309** compared to the *p*-methoxybenzyl furanoside of **337**.

Conversion of 339 to 340 was carried out as for the analogous conversion of 304 - 311 (Scheme 24), involving axial allylation at C<sub>6</sub> of 339; reaction of 340 with sodium methoxide in methanol then afforded a tetrol (cf. 312, Scheme 24). Completion of the  $C_1$ - $C_{21}$  segment **289** required a modified procedure in which base-catalyzed intramolecular hetero-Michael cyclization to form the A ring was effected prior to spiroacetalization at C<sub>14</sub> (cf. these reactions were performed in the opposite order in the transformation  $312 \rightarrow 302 \rightarrow 301$ , Scheme 24). After silylation of the remaining hydroxyl, a C3 epimer was converted to the natural isomer by treatment with Triton B methoxide to deliver the desired  $C_1-C_{21}$  segment 289. Thus, in this synthesis, six stereogenic centers originated in the chiral pool, two stereogenic centers were constructed using a chiral reagent ( $C_{10}$  and  $C_{11}$ ), and the remaining four stereogenic centers were installed using reactions relying on substrate control of asymmetric induction. At the time of writing, coupling of the segments 289, 290, and 291 to complete a synthesis of halichondrin B had not been reported by Salomon and co-workers.  $[C_1-C_{21}]$  segment **289**: 4.8% overall yield from 174; 26 steps longest linear sequence; >36 steps total; ~3 steps per stereogenic center.]

#### 4. Burke Segment Syntheses<sup>77</sup>

Burke *et al.* have reported syntheses of two halichondrin segments: a shorter route to the  $C_1-C_{15}$  segment **301** of Salomon, and an ingenious synthesis of a  $C_{22}-C_{34}$  segment **341** (Scheme 27).

a.  $C_{22}-C_{34}$  Segment Synthesis. Ta Burke et al. noticed that the target  $C_{22}-C_{34}$  segment 341 was a single epimerization away from being a meso compound. Hence, the strategy they adopted involved the asymmetric desymmetrization the meso biscallylic alcohol) 342 (Scheme 28). A four-step sequence of (i) bisco-alkylation) of meso-2-cyclopenten-1,4-diol (343)<sup>129</sup> with tert-butyl bromoacetate under phase-transfer conditions, 130 (ii) ozonolytic ring cleavage, (iii) Wittig homologation, 131 and, finally, (iv) borohydride reduction, provided the biscallylic alco-

# Scheme 28. Burke Halichondrin B $C_{22}-C_{34}$ Synthesis<sup>77a a</sup>

 $^a$  (a) BrCH<sub>2</sub>CO<sub>2</sub>'Bu, NaOH,  $^n$ Bu<sub>4</sub>NHSO<sub>4</sub>; (b) O<sub>3</sub>, Ph<sub>3</sub>P; (c) Ph<sub>3</sub>P=C(Me)CHO, K<sub>2</sub>CO<sub>3</sub>; (d) NaBH<sub>4</sub>; (e) (+)-DET, Ti(O'Pr)<sub>4</sub>,  $^t$ BuOOH; (f) MsCl, Et<sub>3</sub>N; (g) NaI; (h) TFA; (i) LHMDS; TMSCl-Et<sub>3</sub>N;  $\Delta$ ; H<sup>+</sup>; CH<sub>2</sub>N<sub>2</sub>.

hol) **342**. Desymmetrization of **342** was achieved by using the Sharpless asymmetric epoxidation<sup>92</sup> to provide the bis(epoxy alcohol) 344 with high diastereo- and enantiomeric purity. 128,132,133 Dimesylation of 344 followed by displacement with NaI, wherein excess iodide effected reductive opening of the epoxide, then afforded 345 in high yield; TFA-mediated lactonization subsequently gave the bis(dioxonone) **346**. Finally, kinetic enolization of **346** using LH-MDS and trapping with TMSCl provided the bis-(silylketene acetal) 347, which on heating underwent two stepwise Ireland-Claisen [3,3] sigmatropic rearrangements, 83,134 thus forming **348**. On work up and esterification, 348 provided the  $C_{22}-C_{34}$  segment 341. Thus, in the synthesis of **341**, the two newly created stereogenic centers at  $C_{23}$  and  $C_{33}$  were installed using a combination of reagent control of asymmetric induction (Sharpless asymmetric epoxidation) and substrate-controlled transfer of chirality (Claisen rearrangement). At the time of writing, the elaboration of segment 341 into a completely functionalized C<sub>22</sub>-C<sub>34</sub> halichiondrin intermediate had not yet been reported by Burke et al. [C<sub>22</sub>-C<sub>34</sub> segment 341: 15% overall yield from 343; 9 steps; ~2 steps per stereogenic center].

b.  $C_1$ – $C_{15}$  Segment Synthesis. The Burke et al. have reported a shorter route to the Salomon  $C_1$ – $C_{15}$  segment **301** (Scheme 29). The synthesis began with the the commercially available carbohydrate **349** (Deglycero-Degluco-hepto- $\gamma$ -lactone) which requires inversion at  $C_{11}$ , but which has the correct configuration for  $C_8$ – $C_{10}$  of halichondrin B. In this latter respect, it is similar to the Kishi synthesis of a  $C_1$ – $C_{13}$  segment **162** from **165**, depicted in Scheme 11. Regioselective bis(acetalization) of **349** with 3-pentanone gave **350**. Note that bis(acetonide) formation

# Scheme 29. Burke Halichondrin B $C_1$ – $C_{14}$ Synthesis<sup>77b a</sup>

 $^a$  (a) EtCOEt, H<sub>2</sub>SO<sub>4</sub>; (b) PDC, AcOH; (c) Zn(BH<sub>4</sub>)<sub>2</sub>; (d) TBSOTf, 2,6-lutidine; (e) **352**,  $^n$ BuLi, **351**, 1.8 N HCl; (f) MsCl, Et<sub>3</sub>N, DMAP; (g) EtMgBr,  $\Delta$ ; (h) DIBAL; (i) O<sub>3</sub>, PPh<sub>3</sub>, Ph<sub>3</sub>P=CHCO<sub>2</sub>Me; (j) BnMe<sub>3</sub>NOMe; (k) 80% AcOH; (l) NaIO<sub>4</sub>; (m) Ph<sub>3</sub>P=CHCOCH<sub>2</sub>OBn; (n) 52% aqueous HF; (a')  $^i$ PrMgCl,  $^n$ Bu<sub>3</sub>SnH, Galvinoxyl; (b') (R)-BINAL; (c') CH<sub>3</sub>CH(OEt)Cl, Me<sub>2</sub>NPh.

is known to proceed with a different regioselectivity. 135b,c Oxidation of 350 to the C11 ketone and chelation-controlled reduction 136 with Zn(BH<sub>4</sub>)<sub>2</sub> afforded the epimerized C<sub>11</sub> alcohol which was then protected as its TBS ether to provide 351. Meanwhile, stannane 352 was prepared via asymmetric reduction of the acyl stannane derived from 4-pentenal. 137 Transmetalation of 352, to afford the corresponding α-alkoxyorganolithium reagent, 138 followed by reaction with ketone 351 and subsequent mesylation then supplied the indicated diastereomer of  $C_3-C_{12}$  segment **353**. Treatment of this with ethylmagnesium bromide initiated a pinacol rearrangement<sup>139</sup> to form the pyranone, which was subsequently stereoselectively reduced at C7 to afford the B-ring product **354**. Ozonolysis of the double bond of **354**, followed by a reductive work up and in situ Wittig olefination, then provided the A-ring precursor 355. The now standard intramolecular hetero-Michael reaction, mediated by methoxide ion, closed the A ring to supply 356 with the correct (thermodynamic) configuration at C<sub>3</sub>. Selective hydrolysis<sup>135b-d</sup> of the  $C_{12}$ ,  $C_{13}$  acetonide of **356** and periodate cleavage of the ensuing vicinal diol was then followed by Wittig homologation<sup>65b</sup> to afford enone **357**. Finally, HF-mediated acetonide and TBS ether cleavage was followed by in situ intramolecular hetero-Michael addition of the  $C_9$  hydroxyl onto  $C_{12}$ , to close the Cring, and spiroacetalization of the C<sub>8</sub> and C<sub>11</sub> hydroxyls at  $C_{14}$  then furnished the  $C_1-C_{15}$  segment **301**. Thus, in this synthesis, four of the nine stereogenic centers in the target molecule originated from

the chiral pool  $(C_8-C_{11})$ , one was created using a chiral reagent  $(\rightarrow 352)$ , and the remaining four stereogenic centers were introduced using substrate-controlled reactions.  $[C_1-C_{15} \text{ segment } 301: 1.0\%$  overall yield from 349; 14 steps longest linear sequence; 17 steps total;  $\sim 2$  steps per stereogenic center.

In surveying the various synthetic approaches to the halichondrins, it is clear that carbohydrate-based strategies<sup>8</sup> have proved overwhelmingly popular, despite the large number of protecting group manipulations that are frequently required when adopting this approach, and the ensuing length of some of the resulting syntheses. Besides the construction of many stereogenic centers, the other major synthetic challenge associated with the halichondrins has been the formation of spiroacetal, 140 tetrahydropyran, 141 and tetrahydrofuran<sup>141</sup> ring systems. While the spiroacetal rings have generally been synthesized using acid-catalyzed acetalization reactions of hydroxy ketones, intramolecular hetero-Michael reactions have been repeatedly used to construct the tetrahydropyran and tetrahydrofuran rings.

## C. Aplasmomycin

Aplasmomycin (358 in Scheme 30), isolated from a strain of  $Streptomyces\ griseus$  found in shallow sea mud, is a boron-containing ionophoric antibiotic that exhibits activity against Gram-positive bacteria  $in\ vitro$  and  $Plasmodia\ berghei\ in\ vivo.^{142a}$  It has a completely symmetrical  $C_2$  structure  $^{142b}$  and belongs to the family of borate-bridged macrodiolides of which boromycin (359), produced by the terrestrial actinomycete  $Streptomyces\ antibioticus$ , was the first known member.  $^{143,144}$  The first total synthesis of aplasmomycin was reported by Corey  $et\ al.$  in 1982.  $^{145}$  White  $et\ al.$  have also completed a total synthesis,  $^{146a}$  and Nakata, Oishi and co-workers  $^{147}$  and Matsuda  $et\ al.$   $^{148}$  have each achieved a formal total synthesis of aplas-

#### Scheme 30

momycin by preparation of a key intermediate (360) used in Corev's synthesis.

# 1. Corey Total Synthesis 145

The synthesis of aplasmomycin (358) by Corey et al. was based on the construction of a  $C_3-C_{17}$  segment (360 in Scheme 30) from  $C_3-C_{10}$  and  $C_{11}-C_{17}$  segments (361 and 362). Chain extension of 360 with dimethyl oxalate then provided the entire  $C_1-C_{17}$  sequence of aplasmomycin. Direct dimerization to form the macrodiolide, or, alternatively, sequential coupling and macrolactonization, was followed by introduction of the borate to furnish the natural product.

a.  $C_3$ – $C_{17}$  Segment Synthesis. <sup>145a</sup> The  $C_{11}$ – $C_{17}$  segment **362** was prepared from D-mannose (Scheme 31). Thus, reaction of D-mannose diacetonide (**363**)<sup>149</sup> with methyllithium afforded exclusively the diol **364** resulting from chelation control by the  $C_{15}$  oxygen.

Selective tosylation of the less-hindered C<sub>16</sub> hydroxyl of 364 and in situ S<sub>N</sub>2 displacement by the C<sub>13</sub> hydroxyl, with inversion of configuration at  $C_{16}$ , provided the tetrahydrofuran 365. The side chain acetonide of 365 was selectively hydrolyzed, and oxidative cleavage of the resulting diol then supplied the  $C_{12}$  aldehyde **366**, which was converted into the alkyne 367 via the dichloroolefin 368.150 A five-step sequence involving acetonide cleavage, selective silylation of the C<sub>15</sub> hydroxyl, and deoxygenation at C<sub>14</sub> (via triflate ester formation, displacement by iodide. 151 and reduction with tributyltin hydride 152) then provided alkyne 369. Radical-mediated reaction of 369 with tributylstannane gave the desired trans vinylstannane C<sub>11</sub>-C<sub>17</sub> segment 362, together with a smaller amount of the undesired cis isomer 370 (trans/cis = 5:1) which could be thermally equilibrated to provide more of **362** (trans/cis = 85:15).

Scheme 31. Corey Aplasmomycin C<sub>3</sub>-C<sub>17</sub> Synthesis 145a a

"a (a) MeLi; (b) TsCl, py; (c) HCl, H<sub>2</sub>O; (d) NaIO<sub>4</sub>, NaHCO<sub>3</sub>; (e) CBrCl<sub>3</sub>, (Me<sub>2</sub>N)<sub>3</sub>P; (f) "BuLi; (g) HCl, MeOH; (h) TIPSCl, DMAP; (i) Tf<sub>2</sub>O, py; (j) "Bu<sub>4</sub>NI; (k) NaBH<sub>4</sub>, "Bu<sub>3</sub>SnCl,  $h\nu$ ; (l) "Bu<sub>3</sub>SnH, AIBN; (m)  $\Delta$ ; (n) H<sub>2</sub>C=CHMgBr, CuI; (o) NaOMe, MeOH; (p) OsO<sub>4</sub>, NMO; (q) LAH; (r) Me<sub>2</sub>CO, p-TsOH; (s) PCC, 3 Å molecular sieves; (t) m-CPBA; (u) Me<sub>3</sub>Al, HS(CH<sub>2</sub>)<sub>3</sub>SH; (v) O<sub>3</sub>; DMS; BF<sub>3</sub>·OEt<sub>2</sub>, HS(CH<sub>2</sub>)<sub>3</sub>SH; (w) (MeO)<sub>2</sub>CMe<sub>2</sub>, p-TsOH; (x) DMSO, Ac<sub>2</sub>O, AcOH; (x') MOMCl, Et<sub>3</sub>N, DMAP; (y) AcOH, H<sub>2</sub>O; (z) PhCOCN, Et<sub>3</sub>N; MsCl, Et<sub>3</sub>N; (a') "Bu<sub>4</sub>NOH, MeOH; (b') 362, "BuLi; CuCN; 361 or 380; (c') TBSOTf, 2,6-lutidine; (d') AgNO<sub>3</sub>, 2,6-lutidine, H<sub>2</sub>O; (e') TBSOTf, 2,6-lutidine.

The  $C_3-C_{10}$  segment **361** was prepared from (R)pulegone (371). Conjugate addition of a vinylmagnesiocuprate to **371** gave a 1:1 mixture of trans- and cis-cyclohexanones, which was equilibrated using sodium methoxide to provide an 85:15 mixture in favor of the trans isomer 372 having the required configuration at  $C_7$ . Osmylation of **372** occurred stereoselectively on the more accessible si face of the olefin to set up the C<sub>9</sub> stereogenic center with unnatural configuration, and in situ hemiacetal formation then afforded 373. Reduction of 373 to generate a triol and selective acetonide protection of the C<sub>9</sub> and C<sub>10</sub> hydroxyls was followed by reoxidation to give ketone 374, which underwent Baeyer-Villiger oxidation to supply lactone 375. Reaction of 375 with  $trimethylaluminum/propane-1,3-dithiol^{153}$  then gave the ketenethioacetal triol 376. Ozonolysis of 376 and subsequent thioacetalization at C<sub>3</sub>, followed by selective acetonide protection of the C<sub>7</sub> and C<sub>9</sub> hydroxyls, then gave 377 which bore the complete  $C_3-C_{10}$  chain for incorporation into aplasmomycin. The C7 hydroxyl of 377 was protected in two alternative ways: as the (methylthio)methyl (MTM) ether (378), 154 and as the methoxymethyl (MOM) ether (379). Compound 378 was transformed into the epoxide  $C_3-C_{10}$ segment 361 by a three-step sequence of (i) selective hydrolysis of the acetonide of 378 to give a 1,2-diol, (ii) selective benzoylation of the primary hydroxyl followed by mesylation of the secondary hydroxyl, and (iii) benzoate cleavage and in situ epoxide closure with inversion to provide the natural configuration at  $C_9$ . In a similar manner, 379 was converted into the alternative  $C_3-C_{10}$  segment 380.

Coupling of the  $C_3-C_{10}$  and  $C_{11}-C_{17}$  segments **361** and **362** was accomplished via transmetalation of stannane **362** to generate the corresponding organocuprate<sup>155</sup> and addition of epoxide **361**. Exchange of protecting groups in the resulting **381** then provided the  $C_3-C_{17}$  segment **360**. Likewise, coupling of **380** with the organocuprate derived from **362** gave the alternative  $C_3-C_{17}$  segment **382**.  $[C_3-C_{17}$  segment **360**: 6.2% overall yield from **371**; 18 steps longest linear sequence; 30 steps total; 5 steps per stereogenic center;  $C_3-C_{17}$  segment **382**: 15 steps longest linear sequence; 27 steps total;  $\sim$ 4-5 steps per stereogenic center.]

b. Completion of the Total Synthesis of Aplasmomycin.  $^{145b}$  Two complementary routes were developed by Corey et al. for the completion of the synthesis of aplasmomycin. In one route, coupling of two  $C_1-C_{17}$  segments and subsequent macrolactonization generated the macrodiolide; in the other approach, the coupling and cyclization were accomplished in a single step (Scheme 32).

In the first route, lithiation of the dithiane moiety of the  $C_3-C_{17}$  segment **360** and reaction with di-

Scheme 32. Corey Aplasmomycin Synthesis 145b a

 $^{\alpha}$  (a)  $^{n}$ BuLi, TMEDA; HMPA, (CO<sub>2</sub>Me)<sub>2</sub>; (b) LiI, 2,6-lutidine; (c) TBAF; (d) **384** + **385**, bis(2-oxo-3-oxazolidinyl)phosphinic chloride (BOPCl), Et<sub>3</sub>N; (e) LiI, 2,6-lutidine; (f) TBAF; (g) BOPCl, Et<sub>3</sub>N; (h) NaBH<sub>4</sub>; (i) HF; (i') ( $^{i}$ PrS)<sub>2</sub>BBr; (j) HgCl<sub>2</sub>, CaCO<sub>3</sub>, H<sub>2</sub>O; (k) B(OMe)<sub>3</sub>.

methyl oxalate provided the complete  $C_1-C_{17}$  segment 383 with both its carboxyl (C1) and hydroxyl (C<sub>15</sub>) termini protected. Ester cleavage at C<sub>1</sub> then afforded acid 384, whereas selective desilylation at  $C_{15}$  gave alcohol **385**. Coupling of **384** and **385** was then accomplished by esterificaton using the Palomo-Coll protocol (BOP chloride/triethylamine)<sup>156</sup> to afford **386**. Cleavage of both the  $C_1$  ester and  $C_{15}$  silyl ether of 386 gave a seco-acid, which was macrolactonized using the Palomo-Coll procedure 156 to supply the macrodiolide 387 in good yield (64% from the two C<sub>1</sub>- $C_{17}$  segments). Reduction of the  $\alpha$ -keto groups of **387** and subsequent desilylation, followed by dithiane cleavage and in situ hemiacetal formation at C3 and  $C_{3'}$  then gave deboraplasmomycin **388**, as a mixture of diastereomers differing in configuration at C2 and  $C_2$ . Treatment of this mixture with trimethyl borate afforded diastereomerically pure aplasmomycin (358). Note that borate bridging is accompanied by equilibration at C2 and C2 via enolization, and the natural configuration at these centers must be the thermodynamically most favorable stereochemistry.

For the second approach, the alternatively protected C<sub>3</sub>-C<sub>17</sub> segment 382 was employed. Twocarbon homologation at C<sub>3</sub> to provide 389 was performed in exactly the same way as for  $360 \rightarrow 383$ . Cleavage of both the C<sub>1</sub> ester and C<sub>15</sub> silyl ether of **389** then supplied the  $\omega$ -hydroxy acid  $C_1-C_{17}$  segment 390. Subjection of 390 to the Palomo-Coll esterification protocol<sup>156</sup> then gave directly the desired macrodiolide 391. Although the yield of 391 was only moderate (25%), the various byproducts underwent saponification with base to regenerate **390**, making this one-step coupling-cyclization procedure highly effective due to its extreme economy of steps. Cleavage of the MOM ethers of 391 by use of diisopropylthioboron bromide<sup>157</sup> was followed by ketone reduction at  $C_2$  and  $C_2$  to give **392**, again as a mixture of diastereomers at these centers. Note that the cleavage of the C7 MOM ethers could not be achieved by conventional acid-catalyzed hydrolysis since participation of the C<sub>9</sub> hydroxyl resulted in formation of a six-membered cyclic methylene acetal. In the case of diisopropylthioboron bromide, the  $C_9$ hydroxyl forms a diisopropylthioborate ester which facilitates MOM ether cleavage by coordination to the C7 oxygen. 158 After dithiane hydrolysis and in situ hemiacetal formation as for the first route, borate complexation then afforded the natural product in diastereomerically pure form. Thus, in this synthesis, three stereogenic centers in the target molecule originated from the chiral pool  $(C_4, C_{13}, and C_{15})$ , and the remaining five stereogenic centers were introduced using substrate-controlled reactions. [Aplasmomycin (358): first route-2.2% overall yield from **371**; 28 steps longest linear sequence; 41 steps total;  $\sim$ 5 steps per stereogenic center, allowing for  $C_2$ symmetry; second route-23 steps longest linear sequence; 35 steps total; ~4 steps per stereogenic center, allowing for  $C_2$  symmetry.]

#### 2. White Total Synthesis 146a

In the synthesis of aplasmomycin by White et al., the macrodiolide **393** was constructed by ring contraction of the key intermediate **394** in a novel

#### Scheme 33

application of the Chan reaction<sup>159</sup> (Scheme 33). Compound **394** was obtained from the  $C_3-C_{17}$  segment **395**, which was in turn constructed from  $C_3-C_{10}$  and  $C_{11}-C_{17}$  segments (**396** and **397**).

The most recent, and most efficient, route146b to the  $C_3-C_{10}$  segment **396** began with (R)-pulegone (**371** in Scheme 34). As in the Corey synthesis, 145a conjugate addition of a vinylmagnesiocuprate to 371 followed by base-catalyzed equilibration gave ketone **372** with 85% ds at C<sub>7</sub>. Oxidative cleavage<sup>160</sup> of the vinyl group of 372 and esterification of the resulting carboxylic acid then gave 398. More expediently, hydrocyanation of (R)-pulegone gave **399** with > 97% ds at  $C_7$ , 161 and methanolysis then supplied **398**. In common with the Corey synthesis, a Baeyer-Villiger reaction was used to introduce an oxygen atom at C7. Accordingly, lactone 400 was obtained from ketone **398** with high regioselectivity. Two sequences were then developed to transform 400 into its lower homologue, the  $C_3-C_9$  segment 401. Thus, methanolysis of 400 and silvlation of the resulting C<sub>7</sub> hydroxyl gave diester **402**. Selective reduction of the less-hindered ester then afforded primary alcohol 403, and elimination of the derived o-nitrophenyl selenoxide<sup>162</sup> supplied olefin 404. Oxidative cleavage<sup>160</sup> of the double bond of 404 to give the C<sub>3</sub> carboxylic acid was followed by HF-mediated desilvlation at  $C_7$  and in situ lactonization to provide **401**. Alternatively, a variant of the Barbier-Wieland degradation<sup>163</sup> was used to convert **400** to **401**. Thus, reaction of lactone 400 with phenylmagnesium bromide gave diol 405, together with some of the ketone resulting from only monoaddition of Grignard reagent, and acid-catalyzed dehydration of 405 then supplied alkene 406. Temporary protection of the  $C_7$ hydroxyl of 406 was followed by oxidative cleavage 160 of the double bond to give the C<sub>3</sub> carboxylic acid. Deprotection at  $C_7$  and lactonization then afforded **401**. Note that both routes provided **401** in diastereoand enantiomerically pure form. Reaction of 401 with (2R,3R)-2,3-butanediol furnished the corresponding ortholactone and condensation with the lithio anion of methyl phenyl sulfone then gave the  $C_3-C_{10}$  segment **396**.

## Scheme 34. White Aplasmomycin Synthesis 146 a

 $\begin{array}{l} ^a\text{ (a) } H_2\text{C=}\text{CHMgBr, CuBr; (b) KOH, EtOH; (c) RuCl_3, NaIO_4; (d) CH_2N_2; (e) NaCN, NH_4\text{Cl; (f) MeOH, } H_2\text{SO_4; (g) } \text{CF}_3\text{CO}_3\text{H; (h) } K_2\text{CO}_3, \\ \text{MeOH; (i) TBSOTf, 2,6-lutidine; (j) LAH; (k) } o\text{-}O_2\text{NC}_6\text{H}_4\text{SeCN, } ^n\text{Bu}_3\text{P; (l) } H_2\text{O}_2; (m) \text{ RuCl}_3, \text{ NaIO}_4; (n) \text{ HF; (o) PhMgBr; (p) PPTS; (q) } Ac_2\text{O}, \\ \text{py, DMAP; (r) RuCl}_3, \text{NaIO}_4; (s) K_2\text{CO}_3, \text{MeOH; (t) 1 N HCl; (u) } (2R,3R)\text{-butanediol, } p\text{-TsOH; (v) } \text{MeSO}_2\text{Ph, } ^n\text{BuLi; (a') } \text{Ti}(O^i\text{Pr})_4, (-)\text{-DIPT, } ^i\text{BuOOH; (b') THPOCH}_2\text{C=}\text{CH, } ^n\text{BuLi; (c') } (\text{MeO})_2\text{CMe}_2, p\text{-TsOH, } \text{MeOH; (d') LAH, } \text{AlCl}_3; (e') \text{NCS, DMS; (f') } \textbf{396, } ^n\text{BuLi, } \text{KI; } \textbf{397}; \\ \text{(g') Al-Hg; (h') LAH; (i') } \text{Ac}_2\text{O, DMAP; (j') } p\text{-TsOH, } \text{H}_2\text{O}; (k') \text{NaOH, } \text{H}_2\text{O}; (l') \\ 5\% \text{ HCl; (m') PhSeCl; (n') } \text{H}_2\text{O}_2; (o') \text{ TBSOTf, 2,6-lutidine; } \\ \text{(p') NaOH, } \text{H}_2\text{O}; (q') \text{ TBSOTf, 2,6-lutidine; } (r') \text{ K}_2\text{CO}_3, \text{ H}_2\text{O}; (s') \text{ TBAF; (t') BrCH}_2\text{CO}_2(\text{CH}_2)_2\text{TMS, } \text{K}_2\text{CO}_3; (u') \text{ BrCH}_2\text{COCl, py, DMAP; } (v') \\ \textbf{395} + \textbf{418}, \text{ K}_2\text{CO}_3; (w') \text{ TBAF; } (x') \text{ 2-chloropyridinium methiodide, } \text{Et}_3\text{N}; (y') \text{ LDA; TMSOTf; } (z') \text{ HF; (a'') B(OMe)}_3. \\ \end{array}$ 

Note that the preparation of 396 from (R)-pulegone (11 steps, 25% overall yield) is more stereochemically

efficient than the original route, 144e which began with the aldol reaction 164 between the lithio dianion (407)

of tiglic acid and aldehyde **408** (prepared in three steps from isobutyraldehyde and formaldehyde) and which generated **409** in racemic form. Transformation of **409** to **401** (obtained as a 4:1 mixture of racemic trans and cis isomers) was then followed by resolution with (2R,3R)-2,3-butanediol, chromatographic separation, and sulfone introduction to provide enantiomerically pure **396** (12 steps, <11% overall yield). 144e

The  $C_{11}-C_{17}$  segment **397** was prepared<sup>144f</sup> by kinetic resolution of racemic 3-buten-2-ol via Sharpless asymmetric epoxidation.<sup>28a</sup> The resulting epoxide **410** (91% ee) underwent regioselective ring opening with an alkynyllithium to provide diol **411**, and an exchange of protecting groups then gave the propargylic alcohol **412**. After reduction of **412** with LiAlH<sub>4</sub>/AlCl<sub>3</sub> to provide the *trans* allylic alcohol,<sup>165</sup> transformation into the allylic chloride<sup>166</sup> furnished the  $C_{11}-C_{17}$  segment **397**.

Construction of the  $C_{10}-C_{11}$  bond was achieved<sup>144f</sup> via alkylation of the lithium enolate of keto sulfone 396 with chloride 397. Reductive removal of the sulfone moiety then afforded ketone 413, which underwent highly stereoselective reduction (96% ds)146a due to chelation control by the pyran oxygen at C7. Acetylation of the resulting alcohol then provided 414, and acid-catalyzed hydrolysis effected acetonide cleavage at C<sub>15</sub> and C<sub>16</sub> and simultaneous opening of the ortholactone at  $C_3$  to provide ester **415**. Saponification of 415 gave a carboxylic acid at C<sub>3</sub> and acid-catalyzed lactonization then afforded 416. Regioselective intramolecular oxyselenation 167 of alkene 416, involving 5-exo-trig<sup>97</sup> attack of the C<sub>16</sub> hydroxyl at the  $C_{13}$  terminus of the double bond, followed by oxidative elimination of the resulting selenide provided, after silylation of the C<sub>15</sub> hydroxyl, the tetrahydrofuran 417 together with its C<sub>13</sub> epimer in a 1:1 ratio. Saponification of 417, silvlation of the resulting dihydroxy acid, and subsequent selective cleavage of the C<sub>15</sub> silyl ether then furnished the  $\omega$ -hydroxy acid C<sub>3</sub>-C<sub>17</sub> segment **395**. Reaction of the potassium salt of **395** with 2-(trimethylsilyl)ethyl α-bromoacetate was followed by esterification with α-bromoacetyl chloride to give 418, which was then coupled with the potassium salt of carboxylic acid 395 to give the cyclization precursor 419. After removal of the (trimethylsilyl)ethyl ester protecting group at C<sub>1</sub> of **419**, macrolactonization according to the Mukaiyama protocol<sup>168</sup> provided **394** in excellent yield. Treatment of 394 with LDA followed by TMSOTf then initiated the key "double-Chan" reaction, 159 providing the macrodiolide 393 in good yield. HFmediated desilylation of 393 and in situ hemiacetalization then supplied deboraplasmomycin 388. Finally, reaction of 388 with trimethyl borate, 144c, 145 furnished the natural product. Thus, in this synthesis, one stereogenic center in the target molecule originated from the chiral pool  $(C_4)$ , two stereogenic centers were installed using asymmetric induction from a chiral catalytic reagent ( $C_{15}$  and  $C_{16}$ ), and the remaining five stereogenic centers were introduced using substrate-controlled reactions. [Chan reaction precursor 394: 2.6% overall yield from 371; aplasmomycin (358): 34 steps longest linear sequence; 39

Scheme 35

steps total;  $\sim 5$  steps per stereogenic center, allowing for  $C_2$  symmetry.]

## 3. Nakata and Oishi Formal Total Synthesis 147

Nakata, Oishi, and co-workers have achieved a formal total synthesis of aplasmomycin by preparation of the  $C_3-C_{17}$  segment **360**, which was an intermediate in Corey's earlier total synthesis. Compound **360** was constructed by a Julia olefination reaction between the  $C_3-C_{11}$  and  $C_{12}-C_{17}$  segments **420** and **421** (Scheme 35). Stereoselective ketone reductions were used several times to set up key stereogenic centers.

The  $C_3-C_{11}$  segment **420** was prepared from commercially available (S)-pantolactone (422), which supplied the C<sub>9</sub> stereogenic center (Scheme 36). Thus a three-step sequence of reduction, acetonide protection, and oxidation provided aldehyde 423 from 422.169 Aldol reaction of 423 with the lithium enolate of *tert*-butyl acetate then gave  $\beta$ -hydroxy ester **424** as a mixture of  $C_7$  epimers. Acetonide deprotection and in situ lactonization followed by selective silylation of the C<sub>10</sub> primary hydroxyl then led to lactone **425**. After temporary protection of the  $C_7$  hydroxyl of 425, sequential aldol addition of the lithium enolate of isopropyl propionate to C5, methoxylation of the resulting hemiacetal, and simultaneous deprotection at  $C_7$  followed by oxidation at  $C_7$  then gave ketone **426**. Stereoselective reduction at  $C_7$  of **426** by L-selectride then furnished exclusively the axial alcohol 427.170b Acid-catalyzed hydrolysis of 427 was followed by NaBH<sub>4</sub> reduction at C<sub>5</sub> to provide 428, as a mixture of diastereomers at C4 and C5. Acidmediated lactonization of the  $C_7$  hydroxyl of **428** and subsequent acetylation of the remaining hydroxyls was followed by DBU-induced elimination across C<sub>4</sub>- $C_5$  to afford the  $\alpha,\beta$ -unsaturated lactone **429**. After DIBAL reduction of 429, acetalization at C<sub>3</sub> and desilylation then gave diol 430. Heterogeneous hydrogenation of 430 proceeded stereoselectively from the less-hindered  $\alpha$  face to generate 431 with 93% ds at  $C_4$ . 171 After silvlation of the primary hydroxyl at C<sub>10</sub>, chromatographic removal of the minor C<sub>4</sub>-α epimer afforded 432; desilylation then provided diastereomerically pure 431. On treatment with NaH and TsCl, or simply excess KH, 431 was converted into epoxide 433, which underwent regioselective BF<sub>3</sub>·OEt<sub>2</sub>-mediated addition<sup>87</sup> of the lithio anion of methyl phenyl sulfone to afford 434. Transacetalization with 1,3-propanedithiol provided the C<sub>3</sub> thioacetal and silvlation of the resulting diol then supplied the  $C_3-C_{11}$  segment 420.

The  $C_{12}-C_{17}$  segment **421** was derived from (S)-malic acid (**119**), which supplied the  $C_{13}$  stereogenic center. Thus **119** was converted into aldehyde **435**, 60 and aldol addition of the lithium enolate of isopropyl propionate then gave  $\beta$ -hydroxy ester **436**, as a

## Scheme 36. Nakata/Oishi Aplasmomycin C<sub>3</sub>-C<sub>17</sub> Synthesis<sup>147 a</sup>

"a (a) LAH; (b) Me<sub>2</sub>CO, p-TsOH; (c) PCC; (d) MeCO<sub>2</sub>'Bu, LDA; 423; (e) HCl, MeOH; (f) TBDPSCl, imidazole; (g) H<sub>2</sub>C=CHOEt, PPTS; (h) EtCO<sub>2</sub>'Pr, LDA; (i) CSA, MeOH; (j) PCC; (k) L-selectride; (l) HCl; (m) NaBH<sub>4</sub>; (n) CSA; (o) Ac<sub>2</sub>O, py, DMAP; (p) DBU; (q) DIBAL; (r) PPTS, 'BuOH; (s) TBAF; (t) H<sub>2</sub>, 5% Rh-Al<sub>2</sub>O<sub>3</sub>; (u) TBDPSCl, imidazole; (v) TBAF; (w) KH or NaH, TsCl; (x) MeSO<sub>2</sub>Ph, "BuLi, HMPA, BF<sub>3</sub>·OEt<sub>2</sub>; (y) HS(CH<sub>2</sub>)<sub>3</sub>SH, BF<sub>3</sub>·OEt<sub>2</sub>; (z) TBSOTf, 2,6-lutidine; (a') EtCO<sub>2</sub>'Pr, LDA; 435; (b') HCl, MeOH; (c') TBDPSCl, imidazole; (d') CSA; (e') DIBAL; (f') CSA, CH(OMe)<sub>3</sub>; (g') PCC; (h') NaOMe; (i') NaBH<sub>4</sub>, CeCl<sub>3</sub>; (j') O<sub>3</sub>; DMS; (k') Zn(BH<sub>4</sub>)<sub>2</sub>; (l') K<sub>2</sub>CO<sub>3</sub>, MeOH; (m') Me<sub>2</sub>CO, p-TsOH; (n') MsCl, py; (o') TBAF; (p') NaOMe; (q') AcOH, H<sub>2</sub>O; (r') BzCl, py; (s') TIPSCl, imidazole; (t') K<sub>2</sub>CO<sub>3</sub>, MeOH; (u') PCC; (v') 420, "BuLi, HMPA; 421; (w') BzCl, py, DMAP; (x') 6% Na-Hg.

mixture of  $C_{15}$  and  $C_{16}$  epimers. Protecting group exchange and lactonization then furnished 437 which, after reduction to the lactol and acetalization with methanol, provided 438. Oxidation at  $C_{15}$  of 438 and subsequent base-induced elimination then gave the enone 439 which underwent a highly stereoselective Luche reduction<sup>23</sup> to correctly set the  $C_{15}$  stereogenic center. Ozonolysis of 440 gave  $\alpha$ -hydroxy ketone 441, which underwent a moderately selective chelation-controlled reduction<sup>170a</sup> with  $Z_{10}(BH_4)_2$ . After deformylation at  $C_{13}$  and selective acetonide protection

of the  $C_{15}$  and  $C_{16}$  hydroxyls, the desired diastereomer 442 was obtained, together with its  $C_{16}$  epimer, in a ratio of 79:21. Mesylation of the  $C_{13}$  hydroxyl of 442 was followed by cleavage of the  $C_{12}$  silyl ether. Treatment with base then generated the epoxide 443 with inversion of configuration at  $C_{13}$ . Exposure of 443 to acid then led to cleavage of the acetonide, and cyclization of the resulting  $C_{16}$  hydroxyl onto the epoxide, with inversion at  $C_{13}$  again, to give tetrahydrofuran 444. Protecting group exchange and oxidation then afforded the  $C_{12}$ – $C_{17}$  segment 421.

A Julia olefination reaction between the lithio anion of sulfone **420** and aldehyde **421** gave the  $C_3$ – $C_{17}$  segment **360**, the Corey intermediate, and so completed a formal total synthesis of aplasmomycin. Thus, in this synthesis of **360**, two of the five stereogenic centers in the target molecule originated from the chiral pool ( $C_9$  and  $C_{13}$ ); the remaining three stereogenic centers were introduced using substrate-controlled reactions (**429**  $\rightarrow$  **430**  $\rightarrow$  **431** for  $C_4$ , **425**  $\rightarrow$  **426**  $\rightarrow$  **427** for  $C_7$ , **439**  $\rightarrow$  **440** for  $C_{15}$ , and **441**  $\rightarrow$  **442** for  $C_{16}$ ). [ $C_3$ – $C_{17}$  segment **360**: 2.8% overall yield from **422**; 29 steps longest linear sequence; 51 steps total;  $\sim$ 8–9 steps per stereogenic center.]

#### 4. Matsuda Formal Total Synthesis 148

Matsuda *et al.* have also achieved a formal total synthesis of aplasmomycin by preparation of the Corey intermediate **360**. In common with the earlier work of Nakata, Oishi, and co-workers, <sup>147</sup> **360** was constructed from the  $C_3-C_{11}$  and  $C_{12}-C_{17}$  segments

420 and 421 (Scheme 35), and many of the key stereogenic centers were again set up by means of stereoselective ketone reductions.

The  $C_{12}$ – $C_{17}$  segment **421** was derived from (S)malic acid (119), which supplied the C<sub>15</sub> stereogenic center (Scheme 37). Note that Nakata, Oishi, and co-workers also used (S)-malic acid to prepare 421, but in that case the acid supplied the C<sub>13</sub> stereogenic center. By using Still's procedure, 172 Matsuda et al. converted 119 into 2-hydroxybutanolide (445). After protection of the hydroxyl, reaction with methyllithium and subsequent protection of the resulting C<sub>13</sub> hydroxyl furnished ketone 446. Chelationcontrolled reduction<sup>170a</sup> of 446 using Zn(BH<sub>4</sub>)<sub>2</sub> then set up the C<sub>16</sub> stereogenic center with high diastereoselectivity (94% ds). After silvlation of the C<sub>16</sub> hydroxyl of 447, deprotection at  $C_{13}$  was followed by oxidation to the C<sub>13</sub> aldehyde. Reaction with [(benzyloxy)methyl]lithium, 173 followed by oxidation, then gave ketone 448. Stereoselective reduction<sup>174</sup> at C<sub>13</sub>

Scheme 37. Matsuda Aplasmomycin C<sub>3</sub>-C<sub>17</sub> Synthesis 148 a

 $\begin{array}{l} ^a\text{ (a) (MeO)}_2\text{CMe}_2, p\text{-TsOH; (b) BH}_3; \text{ (c) } \\ \text{H}^+; \text{ (d) MEMCl, }^i\text{Pr}_2\text{NEt; (e) MeLi; (f) BOMCl, }^i\text{Pr}_2\text{NEt; (g) } \\ \text{Zn}(\text{BH}_4)_2; \text{ (h) TBSCl, imidazole; (i) Li, liquid NH}_3; \text{ (j) CrO}_3^i\text{-2py; (k) BnOCH}_2\text{Li; (l) CrO}_3^i\text{-2py; (m) LiAlH}(O^i\text{-Bu})_3; \text{ (n) MeLi; TsCl; (o) TBAF; (p) HCl, MeOH; (q) TIPSCl, DMAP; (r) Na, liquid NH}_3; \text{ (s) Swern oxidation; (t) } \\ \text{($2R,3R$)-1,4-dimethoxy-2,3-butanediol, $p\text{-TsOH; (u) PCC, NaOAc; (v) } \\ \text{H}_2\text{C=CHCH}_2\text{MgBr; (w) Jones oxidation; (t) LAH, LiBr; (y) BnCl, NaO^i\text{Am; (z) OSO4, NaIO4; (a') NaBH}_4; (b') BnCl, NaO^i\text{Am; (c') HCl, Me2CO; (d') } \\ \text{H}_2\text{C=CHCH}_2\text{MgBr; (e') Jones oxidation; (f') LAH; (g') (EtCO)}_2\text{O, py, DMAP; (h') OSO4, NaIO4; (i') NaBH}_4; (j') TsCl, Et_3\text{N, DMAP; (k') KI; (l') LDA; (m') KOMe, MeOH; $p\text{-TsOH; (n') DIBAL; (o') CSA, MeOH; (p') Na, liquid NH}_3; (q') TsCl, Et_3\text{N; (r') LiSPh; (s') TBSOTf, 2,6-lutidine; (t') $m\text{-CPBA; (u') recrystallize; (v') HS(CH}_2)_3\text{SH, BF}_3\cdot\text{OEt}_2; (w') TBSOTf, 2,6-lutidine; (x') 420, $^n\text{BuLi, HMPA; 421; (y') BzCl, py, DMAP; (z') 6\% Na-Hg.} \\ \end{array}$ 

of 448 using LiAl(O'Bu)<sub>3</sub>H then gave 449 with 91% ds. Note that the reduction of ketone 446 apparently involves 1,2-asymmetric induction from an  $\alpha$ -alkoxy group, viz. the  $C_{15}$  MEM ether, whereas reduction of 448 relies on 1,3-asymmetric induction from a  $\beta$ -alkoxy group, namely the same  $C_{15}$  MEM ether. The  $C_{15}$  hydroxyl stereochemistry originating from (S)-malic acid is thus used to direct the introduction of the stereogenic centers at both  $C_{13}$  and  $C_{16}$ . After tosylation of the  $C_{13}$  hydroxyl of 449, silyl ether cleavage at  $C_{16}$  led to in situ cyclization, with inversion of configuration at  $C_{13}$ , to give the tetrahydrofuran 450. Protecting group exchange and oxidation at  $C_{12}$  then gave aldehyde 421.

The C<sub>3</sub>-C<sub>11</sub> segment **420** was derived from 3-hydroxy-2,2-dimethylpropanal (451). Acetalization of **451** with (2R,3R)-1,4-dimethoxy-2,3-butanediol<sup>175</sup> was followed by oxidation to the aldehyde at C<sub>9</sub>, addition of allylmagnesium bromide, and reoxidation at C9 to give the ketone 452 bearing a  $C_2$ -symmetric chiral auxiliary. 174 LiAlH<sub>4</sub> reduction of 452 in the presence of LiBr under carefully controlled conditions furnished 453 and introduced the C9 stereogenic center with 93% ds. Note that the stereocontrol imparted on the reduction by chelation of the chiral auxiliary is an example of 1,5-asymmetric induction. After protection of the C<sub>9</sub> hydroxyl of the diastereomeric mixture 453, oxidative cleavage of the alkene was followed by reduction to the  $C_{11}$  primary alcohol and subsequent protection of the hydroxyl. Acid-catalyzed removal of the chiral auxiliary then supplied the C<sub>7</sub> aldehyde; reaction with allylmagnesium bromide, followed by oxidation, then afforded the ketone **454**. This was accordingly obtained in 86% ee, since **453** was of 86% de. Chelation-controlled reduction<sup>174</sup> of  $454 \rightarrow 455$  then set up the C<sub>7</sub> stereogenic center with 94% ds. After protection of the  $C_7$  hydroxyl of 455 as its propionate, oxidative cleavage of the double bond was followed by a three-step conversion to C<sub>5</sub> iodide 456. Treatment of 456 with excess LDA effected ring closure, and kinetically controlled protonation of the resulting lithium enolate furnished a 1:1 mixture of C<sub>4</sub> epimers. Upon exposure to methanolic potassium methoxide, this mixture was equilibrated to provide 457 with 94% ds in favor of the desired C<sub>4</sub> stereochemistry. DIBAL reduction of 457 to give the lactol, followed by acid-catalyzed methoxylation, gave a 66:34 mixture of acetal epimers 458 and 459. Only 458 was taken on to sulfone 420, but 459 could be equilibrated under acidic conditions to provide more of **458**. Cleavage of the benzyl ethers of **458**; selective monotosylation of the resulting diol at C11; and subsequent thiophenolate displacement, silylation at C<sub>13</sub>, and subsequent oxidation then afforded the sulfone 459. This was obtained in 86% ee, but recrystallization allowed the isolation of 459 in enantiomerically pure form. Transacetalization with 1,3-propanedithiol and protection of the resulting  $C_7$  hydroxyl then furnished the  $C_3$ - $C_{11}$  segment **420**.

Finally, Julia olefination of aldehyde **421** and sulfone **420**, according to the procedure of Nakata, Oishi, and co-workers, <sup>147b</sup> supplied the Corey intermediate **360** and completed a formal total synthesis of aplasmomycin. Thus, in this synthesis of **360**, one

$$S:R = 52:48$$

OH

NMe2

OH

Aplyronine A (460)

Aplyronine B (461)

Aplyronine C (462)

Figure 5. Structures of the aplyronines.

of the five stereogenic centers in the target molecule originated from the chiral pool  $(C_{15})$ , one stereogenic center was installed using asymmetric induction from a chiral auxiliary (452  $\rightarrow$  453 for  $C_9$ ), and the remaining four stereogenic centers were introduced using substrate-controlled reactions (456  $\rightarrow$  457 for  $C_4$ , 454  $\rightarrow$  455 for  $C_7$ , 448  $\rightarrow$  449  $\rightarrow$  450 for  $C_{13}$ , and 446  $\rightarrow$  447 for  $C_{16}$ ).  $[C_3-C_{17}$  segment 360: 1.4% overall yield from 452; 33 steps longest linear sequence; 52 steps total;  $\sim$ 9 steps per stereogenic center.]

## D. The Aplyronines

Aplyronine A (460 in Figure 5) and its congeners aplyronines B (461) and C (462) are potent antitumor macrolides, isolated from the Japanese sea hare Aplysia kurodai, which were reported by Yamada and co-workers in 1993.176a-d In addition to the 24membered macrocycle, the structures are interesting due to the presence of a terminal N-methyl-Nvinylformamide unit, as found in several other antitumor marine macrolides, 177 and, in particular, because of the presence of two scalemic 178 amino acid residues. Note that the N,N,O-trimethylserine moiety on  $C_7$  exists as a 2-1.1:1 mixture of S and Rconfigurations, whereas the N,N-dimethylalanine moiety on  $C_{29}$  exists as a 6-3:1 mixture of S and Rconfigurations. 176a Since these isomer ratios vary slightly according to the animal sample employed, it is possible that partial epimerization of the amino acid residues occurs during isolation from the sponge extracts. Yamada and co-workers established the absolute stereochemistry of aplyronine A by enantioselective synthesis of degradation products, 176c,d and in 1994 completed the total synthesis of alpyronine A itself. 176e,f

## 1. Yamada Total Synthesis 176e,f

Yamada and co-workers designed a highly convergent route to aplyronine A (Scheme 38). Thus a  $C_5$ –  $C_{20}$  segment (463) was constructed by the sequential connection of three segments (464–466), and a  $C_{21}$ –  $C_{34}$  segment (467) was assembled from two segments (468 and 469). Julia coupling 179 of 463 and 467 followed by addition of the  $C_1$ – $C_4$  portion by a Horner–Emmons reaction 180 gave the seco-acid 470, which was then macrolactonized. The extensive use of sulfone additions in the synthesis is noteworthy, as is the use of acyclic methods of stereocontrol. Evans aldol reactions 55 and Sharpless asymmetric

#### Scheme 38

epoxidations<sup>92</sup> were used to construct the three sets of four contiguous stereogenic centers at  $C_7-C_{10}$  (as in **464**),  $C_{23}-C_{26}$  (as in **468**), and  $C_{29}-C_{32}$  (as in **469**). a.  $C_{21}-C_{34}$  Segment Synthesis. <sup>176e</sup> Yamada and coworkers observed that the relative stereochemistry of the two sets of four contiguous stereogenic centers spanning  $C_{23}-C_{26}$  and  $C_{29}-C_{32}$  of aplyronine A is the same, *i.e.* syn-anti-anti. However, the absolute stereochemistry of one set is opposite to that of the other. Accordingly, the introduction of the stereogenic centers for the  $C_{21}-C_{27}$  segment **468** and the  $C_{28}-C_{34}$  segment **469** was accomplished using identical methodology, but starting from antipodal starting materials (Scheme 39).

The synthesis of 469 began with the Evans aldol reaction<sup>65</sup> between imide **471** and aldehyde **108** to give 472 with extremely high diastereoselectivity. Transamidation to the Weinreb (N-methyl-N-methoxy) amide,181a protection of the C29 hydroxyl, and reduction<sup>181b</sup> at C<sub>31</sub> then provided the aldehyde **473**, which was homologated at C<sub>31</sub> by means of a Horner-Emmons reaction and subsequently reduced at C<sub>33</sub> to give the allylic alcohol 474. Sharpless asymmetric epoxidation<sup>92</sup> of 474 afforded 475, which underwent stereo- and regionselective (10:1) opening<sup>182</sup> at  $C_{32}$ upon treatment with Me<sub>2</sub>CuLi to give diol 476, which bore the complete syn-anti-syn stereorelationship. Transformation of the C<sub>33</sub> primary hydroxyl of 476 to a cyano group was followed by reduction to give the C<sub>34</sub> aldehyde and spontaneous formation of a hemiacetal with the C<sub>31</sub> hydroxyl. Acetalization with acidic methanol and simultaneous deprotection at C<sub>29</sub> then afforded a separable 4:5 mixture of epimeric acetals 477a and 477b. Equilibration of the minor acetal 477a provided more of 477b. Protecting group exchange and subsequent oxidation at C28 then furnished the  $C_{28}$ – $C_{34}$  segment **469**.

The synthesis of 468 employed the same synthetic strategy as that used for 469. Thus, an Evans aldol reaction<sup>65</sup> between imide 134 and aldehyde 478 afforded 479, which was led onto aldehyde 480 in the standard manner. Compound 480 was converted into diol 481 using the same sequence of reactions used to obtain diol 476 from 473, except that the antipodal Sharpless epoxidation catalyst was now used. Note that the regionselectivity of epoxide opening was only 3:1 in this case. Transformation of the  $C_{27}$  primary hydroxyl of 481 to a sulfide group,

Scheme 39. Yamada Aplyronine A C<sub>21</sub>-C<sub>34</sub> Synthesis<sup>176e a</sup>

"a (a) 471, "Bu<sub>2</sub>BOTf, Et<sub>3</sub>N; 108; (b) Me<sub>2</sub>AlN(Me)OMe; (c) TBSCl, imidazole; (d) DIBAL; (e) ('PrO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Et, 'BuOK; (f) DIBAL; (g) Ti(O'Pr)<sub>4</sub>, (+)-DET, 'BuOOH; (h) Me<sub>2</sub>CuLi; (i) p-TsCl, py; (j) NaCN; (k) DIBAL; (l) CSA, MeOH; (m) Na, liquid NH<sub>3</sub>; (n) TBDPSCl, imidazole; (o) BnBr, NaH; (p) TBAF; (q) Swern oxidation; (r) 468, "BuLi; 469; (s) 6% Na-Hg; (t) Ca, liquid NH<sub>3</sub>; (u) H<sub>2</sub>, 5% Rh-Al<sub>2</sub>O<sub>3</sub>; (v) PhSSPh, "Bu<sub>3</sub>P; (w) m-CPBA; (x) m,p-(MeO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>CH<sub>2</sub>OCH<sub>2</sub>Cl, 'Pr<sub>2</sub>NEt; (a') 134, "Bu<sub>2</sub>BOTf, Et<sub>3</sub>N; 478; (b') Me<sub>2</sub>AlN(Me)OMe; (c') TESCl, imidazole; (d') DIBAL; (e') ('PrO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>Et, 'BuOK; (f') DIBAL; (g') Ti(O'Pr)<sub>4</sub>, (-)-DET, 'BuOOH; (h') Me<sub>2</sub>CuLi; (i') PhSSPh, "Bu<sub>3</sub>P; (j') TESCl, imidazole; (k') m-CPBA.

#### Scheme 40. Yamada Alplyronine A C<sub>5</sub>-C<sub>20</sub> Synthesis<sup>176f a</sup>

 $^a (a) \ \textbf{471}, ^n Bu_2 BOTf, Et_3 N; \ \textbf{483}; (b) \ Me_2 AlN(Me)OMe; (c) \ TBSOTf, 2,6-lutidine; (d) \ DIBAL; (e) (^i PrO)_2 P(-O)CH_2 CO_2 Et, ^i BuOK; (f) \ DIBAL; (g) \ Ti(O^i Pr)_4, (+)-DET, ^i BuOOH; (h) \ Red-Al; (i) \ PivCl, py; (j) \ H_2, 10\% \ Pd-C; (k) \ PhSSPh, ^n Bu_3 P; (l) \ TESCl, imidazole; (m) $m$-CPBA; (n) \ \textbf{464}, \ LDA; \ \textbf{465}, \ HMPA; (o) 5\% \ Na-Hg; (p) \ H_2, 10\% \ Pd-C; (q) \ Dess-Martin periodinane; (r) \ Me_2 CuLi; (s) \ Dess-Martin periodinane; (t) \ \textbf{466}, ^n BuLi; \ \textbf{490}; (u) 6\% \ Na-Hg; (v) \ AcOH, \ H_2O; (w) \ DMSO, \ Ac_2O, \ AcOH; (x) \ HCO_2 H; (y) \ Dess-Martin periodinane; (a') \ BnBr, \ NaH; (b') \ HCl, \ H_2O; (c') \ TBSCl, \ Et_3N, \ DMAP; (d') \ MeI, \ NaH; (e') \ TBAF; (f') \ p-TsCl, \ py; (g') \ NaI; (h') \ TrCl, \ py; (i') \ MeI, \ LDA; (j') \ LAH; (k') \ TBDPSCl, \ imidazole; (l') \ MeI, \ NaH; (m') \ TBAF; (n') \ p-TsCl, \ py; (o') \ PhSO_2 Me, ^n BuLi. \$ 

protection of the  $C_{23}$  secondary hydroxyl, and subsequent oxidation to the  $C_{27}$  sulfone then furnished the  $C_{21}-C_{27}$  segment 468. Note that Yamada and coworkers did not opt for differential protection of the  $C_{23}$  and  $C_{25}$  hydroxyls, and thus a regioselective lactonization was required later in the synthesis of aplyronine A.

A Julia olefination reaction<sup>179</sup> between the carbanion of sulfone **468** and aldehyde **469**, followed by reductive cleavage of the  $C_{21}$  and  $C_{29}$  benzyl ethers and hydrogenation of the  $C_{27}$ – $C_{28}$  double bond, provided the  $C_{21}$ – $C_{34}$  segment **482**. Protection of the  $C_{29}$  hydroxyl and transformation to the sulfone at  $C_{21}$  then gave **467** [16% overall yield from **471**; 24 steps longest linear sequence; 35 steps total; ~4 steps per stereogenic center].

b.  $C_5$ - $C_{20}$  Segment Synthesis. <sup>176f</sup> The construction of the set of four contiguous stereogenic centers spanning C7-C10 of 464 (Scheme 40) was accomplished using methodology similar to that used to set up the stereochemistry of 468 and 469. Aldehyde **483** was prepared from commercially available (R)methyl 3-hydroxy-2-methylpropionate, 44b and its Evans aldol reaction<sup>65</sup> with imide 471 then provided 484. Transformation of 484 to epoxide 485 was accomplished via aldehyde 486, as for  $472 \rightarrow 473 \rightarrow 475$ (Scheme 39), and regioselective reduction<sup>29</sup> at C<sub>6</sub> of **485** afforded diol **487**. Protection of the  $C_5$  and  $C_7$ hydroxyls and conversion into the sulfone at  $C_{11}$  then gave 464. Alkylation<sup>183</sup> of the carbanion of sulfone 464 with iodide 465 (prepared in seven steps from the glycerol derivative 488) and subsequent reductive removal of the sulfonyl group furnished the  $C_5-C_{14}$ segment 489, which was then transformed into the  $C_{14}$  ketone **490** in a further four steps. The Julia olefination reaction<sup>179</sup> between ketone 490 and sulfone **466** (obtained in eight steps from lactone **491**) provided the desired trans olefin 492 (44% yield) along with the undesired cis isomer (20%) and the C<sub>14</sub> tertiary alcohol (23%). Protecting group exchange at  $C_7$  of 492 and deprotection and oxidation at  $C_{20}$ then gave the  $C_5$ – $C_{20}$  segment 463 [9.0% overall yield

from 471; 25 steps longest linear sequence; 40 steps total;  $\sim$ 6 steps per stereogenic center].

c. Completion of the Total Synthesis of Aplyronine  $A.^{176f}$  The union of the  $C_5-C_{20}$  segment 463 with the  $C_{21}-C_{34}$  segment 467 and the completion of the total synthesis of aplyronine A (460) by Yamada and coworkers is summarized in Scheme 41. Once again, a Julia olefination reaction<sup>179</sup> was employed, this time to join 463 and 467 to give alkene 493 with high stereoselectivity ( $C_{20}-C_{21}$  trans/cis = 10:1). Deprotection and Dess-Martin oxidation<sup>27</sup> at C<sub>5</sub> of 493, followed by installation of the C1-C4 section by a Horner-Emmons reaction, <sup>180</sup> then gave **494**. Silyl ether cleavage at C<sub>23</sub> and C<sub>25</sub> of **494** and hydrolysis at the C<sub>1</sub> terminus provided seco-acid 470; Yamaguchi macrolactonization<sup>47</sup> using the modified conditions of Yonemitsu<sup>184</sup> then afforded the desired 24membered macrolide 495 (42% yield) and a 26membered lactone (28%). The latter could be isomerized to **495** (2.5:1 equilibrium ratio in favor of **495**, 60-65% isolated yield of **495**) in the presence of Ti(O<sup>i</sup>Pr)<sub>4</sub>. After silylation of the C<sub>25</sub> hydroxyl of **495**, hydrolysis and reduction at  $C_{34}$  gave diol **496**, which was converted into aldehyde 497 in a further four steps. Reaction of 497 with N-methylformamide in the presence of PPTS and DDQ introduced the terminal *trans-N*-methyl-*N*-vinylformamide moiety to give 498. After oxidative cleavage of the [(m,p)]dimethoxybenzyl)oxy]methyl ether at C29 of 498 (note that a [(p-methoxybenzyl)oxy]methyl ether at  $C_{29}$ could not be cleaved without decomposition of the conjugated lactone), the resulting hydroxyl was acylated<sup>48</sup> with N,N-dimethylalanine (S/R = 3:2) to give a diastereomeric mixture of dimethylalanine esters (S/R = 4:1). Similarly, deprotection at C<sub>7</sub> and acylation<sup>48</sup> with N,N,O-trimethylserine (S/R = 5:2) afforded a diastereomeric mixture of trimethylserine esters (S/R = 4:3). Note that the use of optically pure amino acids also afforded diastereomeric mixtures, implying that partial epimerization occurs during introduction of the amino acid residues. It is possible that some kinetic resolution also takes place when

Scheme 41. Yamada Aplyronine A Synthesis 176f a

 $^a$  (a) 467,  $^n$ BuLi; 463; (b) Ac<sub>2</sub>O, DMAP, py; (c) 5% Na-Hg; (d) DIBAL; (e) Dess-Martin periodinane; (f) (EtO)<sub>2</sub>P(=O)CH<sub>2</sub>CH=CHCO<sub>2</sub>Et, LDA; (g) HF·py, py; (h) LiOH; (i) Cl<sub>3</sub>C<sub>6</sub>H<sub>2</sub>COCl, Et<sub>3</sub>N, DMAP; (j) Ti(O<sup>i</sup>Pr)<sub>4</sub>; (k) TBSCl, imidazole; (l) HCl, H<sub>2</sub>O; (m) NaBH(OMe)<sub>3</sub>; (n) TrCl, py; (o) Ac<sub>2</sub>O, DMAP, py; (p) HCO<sub>2</sub>H; (q) Dess-Martin periodinane; (r) MeNHCHO, PPTS, hydroquinone; (s) DDQ; (t) N,N-dimethylalanine (S/R = 3:2), DCC, DMAP, CSA; (u) AgNO<sub>3</sub>, 2,6-lutidine, H<sub>2</sub>O; (v) N,N-O-trimethylserine (S/R = 5:2), DCC, DMAP, CSA; (w) HF·py, py.

using the scalemic amino acid samples. Finally, silyl ether cleavage at  $C_9$  and  $C_{25}$  furnished aplyronine A (460). Thus, in this synthesis, five of the stereogenic centers in the target molecule originated from the chiral pool ( $\rightarrow$  C<sub>10</sub>, C<sub>13</sub>, C<sub>19</sub>, and the two amino acid stereogenic centers), six stereogenic centers were constructed in three auxiliary-controlled Evans aldol reactions ( $\rightarrow$  C<sub>8</sub>, C<sub>9</sub>, C<sub>23</sub>, C<sub>24</sub>, C<sub>29</sub>, and C<sub>30</sub>), three stereogenic centers were installed using chiral reagents via three Sharpless epoxidation reactions (→  $C_7$ ,  $C_{25}$ , and  $C_{31}$ ), and the remaining three stereogenic centers ( $C_{17}$ ,  $C_{26}$ , and  $C_{32}$ ) were set up by reactions relying on substrate control of asymmetric induction. [Aplyronine A (**460**): 0.5% overall yield from **471**; 47 steps longest linear sequence; 98 steps total; ~6-7 steps per stereogenic center.]

#### E. The Scytophycins

The scytophycins (23–27 in Figure 2) are a class of macrolides which exhibit potent cytotoxicity against human tumor cell lines, as well as displaying broad spectrum antifungal activity. They are isolated from the terrestial blue-green alga Scytonema pseudohofmanni and are therefore not marine natural products, and thus, strictly speaking, lie outside the scope of this review. However, in view of the close structural homology the between the  $C_1-C_{26}$  portions of swinholide A (11) and scytophycin C (23), this latter congener does merit some discussion here. Indeed, this structural homology implies a genetic link between the organisms producing the scytophycins and the swinholides (vide supra).

At the time of writing, the total synthesis of scytophycin C has not been reported. Paterson *et al.* have proposed a synthesis involving regioselective macrolactonization of the seco-acid **499**, followed by addition of the vinyl formamide moiety at  $C_{32}$  (Scheme 42). <sup>187,188</sup> Two routes to **499** have been outlined.

#### Scheme 42

Thus, construction of the  $C_{18}-C_{19}$  bond by stereoselective aldol union of the silyl enol ether **72** and aldehyde **500**, followed by stereoselective ketone reduction at  $C_{17}$  is analogous to the transformation **30** + **72**  $\rightarrow$  **73**  $\rightarrow$  **74** used in the synthesis of swinholide A (Scheme 3). The Alternatively, the  $C_{16}-C_{17}$  bond might be constructed by stereoselective aldol union of the ethyl ketone **44** and aldehyde **501**, followed by stereoselective ketone reduction at  $C_{15}$ . The Alternative transformation of the ethyl ketone **44** and aldehyde **501**, followed by stereoselective ketone reduction at  $C_{15}$ .

## Scheme 43. Paterson Scytophycin C C<sub>17</sub>-C<sub>32</sub> Synthesis<sup>187 a</sup>

"a (a) Cl<sub>3</sub>CC(=NH)OBn, TfOH; (b) LAH; (c) Swern oxidation; (d)  $\mathbf{502} + \mathbf{503}$ ;  $\mathbf{H}_2\mathbf{O}_2$ , NaOH; (e) MeI, NaH; (f)  $\mathbf{9}$ -BBN;  $\mathbf{H}_2\mathbf{O}_2$ , NaOH; (g) TIPSCl, imidazole; (h)  $\mathbf{H}_2$ , Pd-C; (i) Swern oxidation; (j) (MeO)<sub>2</sub>P(=O)Me, "BuLi; (k) PDC; (l) Cl<sub>3</sub>CC(=NH)OBn, TfOH; (m) Me(MeO)NH·AlMe<sub>3</sub>; (n) EtMgBr; (o) "Hex<sub>2</sub>BCl, Et<sub>3</sub>N;  $\mathbf{H}_2\mathbf{C}$ =C(Me)CHO;  $\mathbf{H}_2\mathbf{O}_2$ ; (p) Me<sub>4</sub>NBH(OAc)<sub>3</sub>; (q) "Bu<sub>2</sub>Si(OTf)<sub>2</sub>, 2,6-lutidine; (r) 9-BBN;  $\mathbf{H}_2\mathbf{O}_2$ , NaOH; (s) Swern oxidation; (t)  $\mathbf{507} + \mathbf{512}$ , Ba(OH)<sub>2</sub>; (u)  $\mathbf{H}_2$ , Pd-C; (v) Dess-Martin periodinane; (w)  $\mathbf{H}_2\mathbf{C}$ =CHCH<sub>2</sub>TMS, BF<sub>3</sub>·OEt<sub>2</sub>; (x) MeOTf, 2,6-di-tert-butylpyridine; (y) O<sub>3</sub>, NaHCO<sub>3</sub>; Me<sub>2</sub>S.

The syntheses of **44** and **72** have already been described (Schemes 2 and 3, respectively). The preparation of aldehydes **500** and **501** is outlined in Scheme 43.

Anti crotylboration of aldehyde 502, obtained in three steps from (S)-methyl 3-hydroxy-2-methylpropionate (199),44b using the Brown chiral crotylboron reagent 503,43b afforded homoallylic alcohol 504 with 95% ds, thus setting up the three contiguous stereogenic centers spanning C<sub>28</sub>-C<sub>30</sub> of scytophycin C (Scheme 43). After methylation of the  $C_{29}$  hydroxyl, hydroboration of the alkene and subsequent protection of the resulting alcohol gave 505. Debenzylation and oxidation to the C27 aldehyde was followed by addition of lithiated methyl dimethylphosphonate to provide  $\beta$ -hydroxy phosphonates **506**, as a 2:1 mixture of  $C_{27}$  epimers. Oxidation of this mixture then supplied the  $\beta$ -keto phosphonate **507**. Meanwhile, substrate-controlled aldol addition of methacrolein to the (E)-dicyclohexyl enol borinate **52** derived<sup>33</sup> from ethyl ketone 53, itself obtained from (S)-methyl 3-hydroxy-2-methylpropionate (199),<sup>34</sup> gave  $\beta$ -hydroxy ketone 508, having the desired configuration at C<sub>22</sub> and C<sub>23</sub>, with 98% ds.<sup>32</sup> Stereoselective reduction of 508 using the Saksena-Evans reagent<sup>35</sup> gave the C<sub>21</sub>,C<sub>23</sub>-anti diol 509 with 94% ds, which was protected as its di-tert-butylsilylene derivative 510. Note that, as was the case in the synthesis of swinholide A, 17a Paterson et al. did not opt for differential protection of the  $C_{21}$  and  $C_{23}$  hydroxyls, and thus a regioselective lactonization will be required later in the synthesis of scytophycin C.

Introduction of the  $C_{24}$  stereogenic center was effected by hydroboration of **510** to give **511** with 93% ds. This completed the synthesis of the stereopentad spanning  $C_{20}-C_{24}$ . Oxidation of alcohol **511** then afforded the aldehyde 512 and C25-C26 bond construction was accomplished by a Horner-Emmons coupling<sup>180</sup> of **507** and **512** to give exclusively the (E)-enone **513**. Note the rare use of barium hydroxide as the base in this reaction, 189,190 which proceeded cleanly without  $\beta$ -elimination in the aldehyde **512** or epimerization of either 507 or 512 occurring. 191 Use of the existing Masamune-Roush (LiCl, 'Pr2NEt or DBU)192a or Rathke (LiBr or MgBr<sub>2</sub>, Et<sub>3</sub>N)<sup>192b</sup> protocols was unsuccessful in this case. Catalytic heterogeneous hydrogenation of 513 led to debenzylation and 1,4reduction of the enone to give the alcohol 514. Dess-Martin<sup>27</sup> oxidation then supplied the  $C_{19}-C_{32}$  aldehyde segment **500**. The  $C_{19}$  stereogenic center was set up by BF<sub>3</sub>·OEt<sub>2</sub>-promoted addition of allyltrimethylsilane to 500, which afforded the desired Felkin-Anh diastereomer 515 with  $\geq 97\%$  ds. Note that use of TiCl4 in this reaction led to much lower diastereoselectivity, providing a 2:1 mixture of C<sub>19</sub> epimers. Finally, O-methylation<sup>193</sup> of 515 and ozonolysis then furnished the  $C_{17}$ – $C_{32}$  aldehyde segment 501. In this synthesis, five of the seven newly created stereogenic centers were installed by a series of four substrate-controlled reactions (53  $\rightarrow$  508, 508  $\rightarrow$  509, 510  $\rightarrow$  511, and 500  $\rightarrow$  515); the remaining two were set up in a single reagent-controlled reaction (502 + 503  $\rightarrow$  504). [C<sub>17</sub>-C<sub>32</sub> segment 501: 9.3% overall yield from 199; 17 steps longest linear sequence; 25 steps total;  $\sim$ 3 steps per stereogenic center.]

# F. The Ulapualides and Halichondramides

The ulapualides, e.g. ulapualide A (516 in Figure 6), are a class of tris(oxazole)-containing macrolides which were first isolated from egg masses of the marine nudibranch Hexabranchus sanguineas. 194 Similar structurally related macrolides, which have variously been called kabiramides, 195 mycalolides, 196 and halichondramides, 195b, 197 e.g. 517, have been isolated from other nudibranches and also from marine sponges. 198 This family of marine metabolites exhibits a wide variety of biological activities, including antifungal, antileukemic, and ichthyotoxic properties. Such a biological profile may in part be associated with the capacity of the metabolites to sequester and transport metal ions in vivo using the several oxygen and nitrogen ligand binding sites in their structures.199

The relative stereochemistries of the ulapualides and halichondramides have not yet been unequivocally established. However, Pattenden and co-workers have remarked<sup>200c,d</sup> that both classes of molecules bear side chains terminating in formyl enamine residues which are very similar to side chains found in the scytophycins, e.g. 23,16,201 and the aplyronines, e.g. 460. 176a-d Since the absolute and relative stereochemistries of 23 and 460 have been secured by partial synthesis and/or X-ray crystallography, Pattenden and co-workers have suggested identical configurations for the side chains of ulapualide A  $(516)^{200b,d}$  and halichondramide  $(517)^{200c}$  at the coincident stereogenic centers, as indicated in Figure 6. In addition, Pattenden and co-workers have also proposed a complete stereochemical assignment for ulapualide A (516) based on computer modeling studies.200d They speculated that if ulapualide A is indeed an ionophore, then the natural stereoisomer will be the one best able to form a metal chelate, *i.e.* that stereoisomer which shows the lowest strain energy for a ulapualide-metal complex. On the basis of molecular mechanics calculations on a ulapualide A-Co(III) complex, the stereochemistry indicated in

Figure 6 was predicted. Note that the stereochemical prediction for the side chain so obtained matched exactly that expected by comparison with the side chain of scytophycin C.

At the time of writing, no total syntheses of either ulapualide A or halichondramide have been reported. However, both Yoo<sup>202</sup> and Pattenden and coworkers<sup>200a</sup> have prepared tris(oxazole) segments.<sup>203</sup> In addition, Pattenden and co-workers have come extremely close to synthesizing the ulapualide A macrocycle,<sup>200b</sup> and the side chain of halichondramide has also been prepared by the same research group.<sup>200c</sup>

The first synthetic work performed on the ulapualides concerned the preparation of the unprecedented tris(oxazole) segment. The probable biosynthetic pathway to the tris(oxazole) moiety involves cyclization and oxidation of a substituted tris(serine) tripeptide intermediate, and the first synthesis of a contiguous tris(oxazole) (518 in Scheme 44), by Pattenden and co-workers, 200a employed three molecules of serine in three sequential oxazoline cyclization-oxidation sequences (the first cyclization being  $519 \rightarrow 520$ ), according to the method of Meyers and co-workers.<sup>204</sup> Later, Yoo prepared the segment **521**, using three sequential [3+2]-cycloaddition reactions of nitriles with dimethyl diazomalonate (the first cycloaddition being  $522 + 523 \rightarrow 524$ , <sup>202</sup> following precedent from the work of Helquist and co-workers.<sup>205</sup>

## 1. Yoo Tris(oxazole) Segment Synthesis<sup>202</sup>

The Yoo tris(oxazole) segment synthesis began with the Rh<sub>2</sub>(OAc)<sub>4</sub>-mediated cycloaddition reaction between dimethyl diazomalonate (**523**) and cyanohydrin **522**, obtained from pivaldehyde (**525**), which afforded oxazole **526** (Scheme 45).<sup>205</sup> Reductive removal of the 5-methoxy group and simultaneous reduction of the 4-methoxycarbonyl moiety then supplied oxazole **527**, and a three-step sequence converted **527** into the nitrile **524**. Repetition of the cycloaddition-reduction-nitrile formation sequence provided the bis(oxazole) **528**. After a third cycloaddition and reduction to give alcohol **529**, adjustment of hydroxyl-protecting groups then furnished the tris-

**Figure 6.** Proposed structures for ulapualide A and halichondramide in comparison with known structures of aplyronine A and scytophycin C.

Scheme 45. Yoo Halichondramide/Ulapualide A Tris(oxazole) Synthesis  $^{202}$   $^a$ 

 $^\alpha$  (a) KCN; (b) TBSCl, imidazole; (c)  $523,\ Rh_2(OAc)_4;$  (d) LAH; (e) Swern oxidation; (f) NH2OH+HCl, K2CO3; (g) Tf2O, Et3N; (h) Ac2O, Et3N; (i) TBAF.

(oxazole) segment **521** [3.3% overall yield from **525**; 16 steps].

# 2. Pattenden Tris(oxazole) Segment Synthesis<sup>200a</sup>

The synthesis of a tris(oxazole) segment by Pattenden and co-workers began with the condensation

## Scheme 46. Pattenden Halichondramide/ Ulapualide A Tris(oxazole) Synthesis<sup>200a</sup> a

 $^a$  (a)  $\bf 530+531, Et_3N;$  (b) NiO<sub>2</sub>,  $\Delta;$  (c) KOH, H<sub>2</sub>O; (d) SOCl<sub>2</sub>; (e)  $\bf 530, Et_3N;$  (f) SOCl<sub>2</sub>; (g) AgOTf; (h) NiO<sub>2</sub>,  $\Delta.$ 

## Scheme 47

between L-serine ethyl ester hydrochloride (530) and ethyl acetimidate hydrochloride (531) in the presence of base, which afforded oxazoline 532 (Scheme 46). Oxidation of 532 using nickel peroxide, according to the procedure of Meyers, 204 then gave the oxazole 520. Saponification of 520 and subsequent formation of the acid chloride 533 was followed by amide formation using a second molecule of 530 to supply 534. After reaction of 534 with thionyl chloride to provide 535, AgOTf-induced cyclization 206 furnished 536 and oxidation then gave the bis(oxazole) 537. Repetition of the sequence of reactions used to transform 520 into 537 then afforded the tris(oxazole) segment 518 [~6% overall yield from 531; 14 steps].

# 3. Pattenden C<sub>1</sub>-C<sub>24</sub> Segment Synthesis<sup>200b</sup>

Besides the synthesis of the tris(oxazole) segment **518** spanning  $C_9-C_{19}$  of ulapualide A (*vide supra*), <sup>200a</sup> Pattenden and co-workers have also prepared the  $C_1-C_8$  and  $C_{20}-C_{24}$  segments **538** and **539** (Scheme 47). Sequential coupling of the segments **518**, **539**, and **538** has been achieved, <sup>200b</sup> to provide a  $C_1-C_{24}$ 

## Scheme 48. Pattenden Ulapualide A C<sub>1</sub>-C<sub>24</sub> Synthesis<sup>200b a</sup>

"a (a) (+)-DET, Ti(O'Pr)<sub>4</sub>, 'BuOOH; (b) Me<sub>2</sub>CuLi; (c) MOMCl, 'Pr<sub>2</sub>NEt; (d) NaH, MeI; (e) TBAF; (f) Swern oxidation; (g) NBS, AIBN,  $h\nu$ ; (h) PPh<sub>3</sub>; (i) **545**, 'BuOK; **539**; (j) DIBAL; (k) Dess-Martin periodinane; (l) **548**, LDA; H<sub>2</sub>C=CHCH<sub>2</sub>Br; (m) Bakers' yeast; (n) TBDPSCl, imidazole; (o) BH<sub>3</sub>; H<sub>2</sub>O<sub>2</sub>; (p) Jones oxidation; (q) (COCl)<sub>2</sub>; (r) Ph<sub>3</sub>P=CH<sub>2</sub>; (s) **538**, base; **547**.

segment of ulapualide A. Cyclization of this segment has not yet been reported; neither has the synthesis of the  $C_{25}-C_{37}$  side-chain segment **540**. However, the side chain of halichondramide has been prepared by the same research group (*vide infra*).<sup>200c</sup>

Synthesis of the  $C_{20}-C_{24}$  segment **539** began with Sharpless asymmetric epoxidation<sup>92</sup> of the allylic alcohol **541**, which afforded epoxide **542** (Scheme 48). Directed ring opening with lithium dimethylcuprate<sup>182</sup> then supplied the diol **543** with the required configurations at both  $C_{22}$  and  $C_{23}$ . Protecting group manipulations and adjustment of oxidation state then gave the aldehyde **539**. Meanwhile, bromination of the tris(oxazole) segment **518** at the carbon  $\alpha$  to the terminal oxazole furnished the bromide **544**, which was transformed into the phosphonium salt **545**. After an *E*-selective Wittig coupling of **539** and **545**, adjustment of oxidation state in the resulting **546** provided the  $C_9$  aldehyde **547**.

Synthesis of the  $C_1$ – $C_8$  segment **538** began with monoalkylation of the dianion of methyl acetoactetate (**548**) to afford the ketone **549**. Incubation with Bakers' yeast then supplied the corresponding  $C_3$  alcohol with the configuration required for ulapualide A, and protection of the hydroxyl provided **550**. After regioselective hydroboration of **550**, oxidation of the resulting primary alcohol gave the carboxylic acid **551**. This was converted to the corresponding acid chloride, and then to the  $\beta$ -keto phosphonium salt

**538**. Wittig coupling of **547** and the ylide derived from **538** furnished the  $C_1-C_{24}$  enone segment **552**. Deprotection at both  $C_1$  and  $C_{24}$  of **552** to afford the seco-acid, followed by macrolactonization to supply the truncated macrolide **553**, has not yet been reported. Stereoselective introduction of the  $C_9$  methyl group is envisaged via Michael addition of lithium dimethylcuprate to enone **553**, which is expected to proceed in the required stereochemical sense as a consequence of macrocyclic stereocontrol. <sup>207</sup> Pattenden and co-workers have not specified whether the  $C_{25}-C_{37}$  side chain is to be introduced before or after macrolactonization.

Thus, in the synthesis of **552**, one of the three stereogenic centers  $(C_{22})$  was installed using a reagent-controlled reaction  $(\mathbf{541} \rightarrow \mathbf{542})$ ; a second stereogenic center  $(C_3)$  was set up using an enzyme-mediated reaction  $(\mathbf{549} \rightarrow \mathbf{550})$ ; and the third stereogenic center  $(C_{23})$  was introduced via a reaction relying on substrate-controlled asymmetric induction  $(\mathbf{542} \rightarrow \mathbf{543})$ . Introduction of a fourth stereogenic center, at  $C_9$  of **553**, is envisaged using macrocyclic stereocontrol.  $[C_1-C_{24} \text{ segment } \mathbf{552}$ : 20 steps longest linear sequence; 33 steps total.]

## 4. Pattenden Halichondramide Side Chain Synthesis<sup>200c</sup>

Pattenden and co-workers have synthesized the  $C_{18}-C_{37}$  side chain **554** of halichondramide (Scheme 49).<sup>200c</sup> Compound **554** was obtained via elaboration

of the  $C_{20}-C_{35}$  segment **555**, which was prepared by Horner-Emmons coupling of the  $C_{20}-C_{28}$  and  $C_{29}-C_{35}$  segments **556** and **557**.

The key step in the synthesis of the  $C_{20}-C_{28}$  segment **556** involved an Evans aldol reaction<sup>65</sup> between the chiral oxazolidinone **558** and the  $C_{24}-C_{28}$  aldehyde segment **559** (Scheme 50). Synthesis of **559** began with Sharpless asymmetric epoxidation<sup>92</sup> of the allylic alcohol **560**, which afforded

epoxide 561.208 Directed ring-opening208 with methylmagnesium bromide in the presence of catalytic CuI then supplied the diol 562 with the required configurations at both  $C_{26}$  and  $C_{27}$ . Protecting group manipulations and adjustment of oxidation state gave the aldehyde **559**, and addition to the (Z)-enol dibutylborinate derived from imide 558 delivered the aldol adduct 563 having the required configurations at both C23 and C24. After protection of the C24 hydroxyl of **563**, reductive removal<sup>209</sup> of the chiral auxiliary gave 564. Conversion to the corresponding C<sub>23</sub>-methyl compound was effected via sequential mesylation followed by hydride displacement; deprotection at C28 followed by oxidation then furnished the aldehyde 556 [7.4% overall yield from 560; 13 steps;  $\sim 3$  steps per stereogenic center].

Meanwhile, the  $C_{29}-C_{35}$  segment **557** was prepared from allylic alcohol **565**. Thus, **565**<sup>29b</sup> was converted into diol **566**, having the required configurations at  $C_{31}$  and  $C_{32}$ , via a two-step sequence already used in the transformation of **560** into **562**. Protecting group manipulations and adjustment of oxidation state then gave the ester **567**, which, following a further protecting group interconversion, was used to acylate the anion derived from diethyl methylphosphonate, thus providing **557** [12% overall yield from **565**; 10 steps; 5 steps per stereogenic center].

Horner-Emmons coupling of aldehyde **556** and the  $\beta$ -keto phosphonate **557** proceeded smoothly to afford the (E)-alkene; hydrogenation then effected simultaneous reduction of the double bond and hydrogenolysis of the  $C_{20}$  ether to supply alcohol **555**. After

Scheme 50. Pattenden Halichondramide C<sub>18</sub>-C<sub>37</sub> Synthesis<sup>200c a</sup>

 $\begin{tabular}{ll} $^a$ (a) (-)-DET, $Ti(O^iPr)_4$, $^iBuOOH; (b) MeMgBr, $CuI; (c) TBDPSCl$, $imidazole; (d) NaH, MeI; (e) $H_2$, $Pd(OH)_2-C; (f) Swern oxidation; (g) $558$, $^iBu_2BOTf, $Et_3N$; $559; $H_2O_2$; (h) MOMCl$, $^iPr_2NEt; (i) LiBH_3(OMe)$; (j) $MsCl$, $^iPr_2NEt; (k) LiBH_3(OMe)$, $\Delta$; (l) $TBAF$; (m) Swern oxidation; (n) $(-)-DET$, $Ti(O^iPr)_4$, $^iBuOOH$; (o) $MeMgBr$, $CuI$; (p) $TBDPSCl$, $imidazole$; (q) NaH, $MeI$; (r) $TBAF$; (s) $PDC$; (t) $AcCl$, $MeOH$; (u) $H_2$, $Pd(OH)_2-C$; (v) $TBSOTf$, $2,6$-lutidine$; (w) $(EtO)_2P(=O)Me$, $^iBuLi$; (x) $557$, $KHMDS$; $556$; (y) $H_2$, $Pd(OH)_2-C$; (z) $Swern oxidation$; (a') $Ph_3P=CHCO_2Me$; (b') $TBAF$; (c') oxidize$; (d') $HCONHMe$, $H^+$.} \end{tabular}$ 

Figure 7. Structures of the bryostatins.

oxidation to the C<sub>20</sub> aldehyde, Wittig reaction with methyl (triphenylphosphoranylidene)acetate led to the (E)- $\alpha,\beta$ -unsaturated ester. Deprotection at  $C_{35}$ and oxidation then afforded aldehyde **568**. Finally, reaction with N-methylformamide under mild acid catalysis (conditions not specified) furnished the halichondramide  $C_{18}$ – $C_{35}$  side-chain segment **554**. In this synthesis, two of the six stereogenic centers were introduced in two reagent-controlled reactions (560  $\rightarrow$  **561** and the analogous transformation of **565**), two were installed using substrate-controlled asymmetric induction (561  $\rightarrow$  562 and the analogous preparation of **566**), and two were constructed in a single auxiliarycontrolled reaction (558 + 559  $\rightarrow$  563). [C<sub>18</sub>-C<sub>37</sub> segment **554**: 20 steps longest linear sequence from **560**; 30 steps total; 5 steps per stereogenic center.]

# G. The Bryostatins

The bryostatins constitute a class of 17 macrolides, isolated from the marine bryozoans Bugula neritina Linnaeus and Amanthia convoluta, which exhibit exceptionally high levels of antineoplastic activity against lymphocytic leukemia and ovarian carcinoma,210 and which have recently reached phase 2 clinical trials.<sup>211</sup> Except for the C<sub>20</sub>-deoxy analogues, such as bryostatin 11 (569 in Figure 7), 210h they differ only in the nature of the ester functions at  $C_7$  and  $C_{20}$ . Bryostatins 1 (570)<sup>210a</sup> and 7 (571),<sup>210e</sup> in particular, have attracted synthetic interest, and the first total synthesis of bryostatin 7 was completed by Masamune and co-workers in 1990.<sup>212e</sup> No other total syntheses have been completed, but several groups<sup>213-218</sup> have reported the synthesis of segments, including significant contributions from the groups of Vandewalle, 213 Roy, 214 Hale, 215 Evans, 216 and Nishiyama and Yamamura.<sup>217</sup>

## 1. Masamune Total Synthesis<sup>212</sup>

Initial efforts by Masamune and co-workers to synthesize bryostatin 7 culminated in the preparation of the seco-acid derivative **572**, which was obtained via coupling of  $C_1$ – $C_{10}$ ,  $C_{11}$ – $C_{16}$ , and  $C_{17}$ – $C_{27}$  segments **573**, **574**, and **575**, respectively (Scheme 51).  $^{212a-c}$  Unfortunately, cleavage of the  $C_3$  MOM ether of **572**, which had been introduced at an early stage, was problematic. Thus the original synthetic route was revised such that creation of the  $C_3$  stereogenic center was postponed until the end of the seco-acid synthesis.  $^{219}$  Accordingly, sequential connection of the  $C_3$ – $C_{10}$  segment **576** (instead of the  $C_1$ – $C_{10}$  segment **573**), segments **574** and **575**, and finally the  $C_1$ – $C_2$  segment **577**, followed by depro-

TBDPSO CHO

16

TBDPSO

TDDPSO

tection, afforded seco-acid **578**. Macrolactonization then completed the synthesis of bryostatin 7.<sup>212e</sup>

a.  $C_1$ – $C_{10}$  and  $C_3$ – $C_{10}$  Segment Syntheses. <sup>212b–d,220</sup> The original route designed by Masamune and coworkers for preparation of the  $C_1-C_{10}$  segment 573 exploited methodology developed by Masamune and Sharpless for achieving stereoselective synthesis of 1,3-diols via directed reduction of chiral epoxides.<sup>29a</sup> Thus, Horner-Emmons reaction of aldehyde 579, obtained from diol 580, followed by reduction gave the allylic alcohol 581 (Scheme 52). Sharpless asymmetric epoxidation<sup>92</sup> then afforded epoxide **582** with 92% ee. A three-step sequence of oxidation to the C<sub>5</sub> aldehyde, formylolefination, and reduction supplied the allylic alcohol 583, and Sharpless asymmetric epoxidation<sup>92</sup> then afforded the bis(epoxide) **584** with >99% ds. Note that this second epoxidation using a chiral substrate and a chiral catalyst leads to product **584** of enhanced enantiomeric purity as a result of diastereomer formation. Directed double

 $\begin{tabular}{ll} $^a$ (a) PhCHO, $H^+$; (b) $BH_3$'Me_2S; (c) PCC; (d) (EtO)_2P(=O)CH_2CO_2Et, NaH; (e) DIBAL; (f) (-)-DET, $Ti(O'Pr)_4$, $'BuOOH; (g) Swern oxidation; (h) $Ph_3P=CHCHO; (i) NaBH_4$; (j) (+)-DET, $Ti(O'Pr)_4$, $'BuOOH; (k) Red-Al; (l) $'BuCOCl$, $py; (l') $TBDPSCl$, $imidazole; (m) $(MeO)_2CMe_2$, $PPTS; (n) DIBAL; (o) Swern oxidation; (p) $(EtO)_2P(=O)CH_2CO_2Et$, $NaH; (q) DIBAL; (r) (-)-DET, $Ti(O'Pr)_4$, $'BuOOH; (s) Red-Al; (t) $TBDPSCl$, $imidazole; (u) $MOMBr$, $'Pr_2NEt; (v) Raney $Ni; (v') Na$, $liquid $NH_3$; (w) Swern oxidation; (x) $MeLi; (y) Swern oxidation; (a') $TBDPSCl$, $imidazole; (b') $O_3$; $Me_2S$; (c') $595 + $(R_1R_1)-596$, $'Pr_2NEt; $592$; $H_2O_2$; (d') $(MeO)_2CH_2$, $P_2O_5$; (e') $LiCuMe_2$; (f') $598 + $(S_1S_2)-596$, $'Pr_2NEt; $579$; $H_2O_2$; (g') $Me_4NBH(OAc)_3$; (h') $(MeO)_2CMe_2$, $PPTS$.} \end{tabular}$ 

reduction<sup>29</sup> of **584** using Red-Al gave the  $C_5$ , $C_7$ -anti diol **585**. A series of protecting group manipulations then furnished **586**, and a second sequence of Swern oxidation,<sup>38</sup> Horner–Emmons olefination, reduction, Sharpless epoxidation,<sup>92</sup> and Red-Al reduction<sup>29</sup> then afforded diol **587** (*i.e.* **586**  $\rightarrow$  **588**  $\rightarrow$  **589**  $\rightarrow$  **587**, *cf.* **582**  $\rightarrow$  **583**  $\rightarrow$  **584**  $\rightarrow$  **585**). After differential protection of the  $C_1$  and  $C_3$  hydroxyls of **587** to give **590**,<sup>221</sup> a four-step sequence of (i) deprotection at  $C_9$ , (ii) oxidation to the  $C_9$  aldehyde, (iii) methyllithium addition, and (iv) reoxidation at  $C_9$  provided the  $C_1$ – $C_{10}$  segment **573**.

Due to the number of functional group transformations and protecting group manipulations, the above linear route to **573** involved 25 steps [3.9% overall yield from **580**], which was considered too many for a target containing only three stereogenic centers. However, by applying Masamune's asymmetric aldol methodology using chiral boron reagents, <sup>222</sup> a shorter

and more efficient convergent synthesis of 573 was developed. Thus, aldehyde **592**, which was obtained from *cis*-3-hexen-1-ol (**593**), underwent an enantioselective aldol reaction with the chiral enol borinate 594 derived from thioacetate 595 and the chiral boron triflate reagent (R,R)-596,<sup>222</sup> to provide the aldol adduct 597 with 89% ee (Scheme 52). Compound 594 thus served as a chiral acetate equivalent. After protection of the C<sub>3</sub> hydroxyl of **597**, treatment with lithium dimethylcuprate<sup>223</sup> supplied the methyl ketone **598**. Regioselective kinetic enolization of **598** in the presence of the chiral boron triflate reagent (S,S)-596 afforded the enol borinate 599, and aldol addition to aldehyde 579 then delivered the aldol adduct 600 with 80% ds in favor of the required configuration at  $C_7$ . Note that ald reaction using the antipodal chiral reagent (R,R)-596 proceeded with equal and opposite diasteroselectivity, furnishing the epimeric adduct 7-epi-600 with 80% ds. Note

# Scheme 53. Masamune Bryostatin 7 C<sub>3</sub>-C<sub>16</sub> Synthesis<sup>212c-e a</sup>

 $^a$  (a) DHP, PPTS; (b)  $^n$ BuLi, HCHO; (c) Red-Al; I2; (d) TBDPSCl, imidazole; (e) H2C=CHCH2MgBr, CuI; (f) PPTS, EtOH; (g) (py)2CrO3; (h) 576 + (R,R)-596,  $^i$ Pr2NEt; 574; H2O2; (i) (MeO)3CH, MeOH, PPTS, (j) Hg(OAc)2; KCl; (k) Ac2O, py, DMAP; (l) NaBH4, O2; (m) Swern oxidation; (n) Al2O3, H2O.

also that use of the achiral meso reagent (R,S)-596 resulted in an approximately 1:1 ratio of 600 and 7-epi-600. Hence, there is negligible substrate control of asymmetric induction in this reaction, and control comes almost entirely from the chiral reagents employed. In addition, by using a chiral boron reagent, the aldol reaction of ketone 598 leads to product 600 of enhanced enantiomeric purity as a result of diastereomer formation. Reduction of  $\beta$ -hydroxy ketone 600 using the Saksena-Evans reagent<sup>35</sup> gave the C<sub>5</sub>,C<sub>7</sub>-anti diol 601 which was then protected as its acetonide 590. Compound 590 was converted into the  $C_1-C_{10}$  segment **573** as before. In this revised synthetic route, 573 was obtained in approximately half the previous number of steps [12] steps longest linear route; 15 steps total; 5 steps per stereogenic center and in 31% overall yield from 592. In the initial route to 573, all three stereogenic centers were installed using chiral reagents via three Sharpless epoxidation reactions. In this latter approach, two of the stereogenic centers were constructed using reagent control (592 + 595  $\rightarrow$  597 and  $579 + 598 \rightarrow 600$ ) and the third was set up using substrate control ( $600 \rightarrow 601$ ).

The  $C_1-C_{10}$  segment **573** was used in the synthesis of protected seco-acid **572** in the initial approach to bryostatin 7 (*vide supra*). <sup>212a-c,219</sup> However, in the revised synthetic route, the  $C_3-C_{10}$  segment **576** was required. This was obtained via a modification of the initial route to **573**. <sup>220</sup> Thus, selective silylation <sup>221</sup> of the primary hydroxyl of **585** and subsequent acetonide protection of the secondary hydroxyls provided **591**, which was transformed into the methyl ketone **576** in an analogous manner to the conversion

of **590** into **573**. Accordingly, the  $C_3-C_{10}$  segment **576** was prepared in 17 steps from **580** [*i.e.* 8-9 steps per stereogenic center].

b.  $C_{11}$ – $C_{16}$  Segment Synthesis and Coupling to  $C_3$ – $C_{10}$  Segment.  $^{212c-e}$  Preparation of the  $C_{11}$ – $C_{16}$  segment **574** required stereoselective construction of the  $C_{13}$  exocyclic double bond. This was achieved by application of Corey's methodology for the synthesis of trisubstituted olefins.  $^{224a}$  Thus, after protection of the hydroxyl of **602**, alkynyllithium formation, and addition to formaldehyde provided propargylic alcohol **603** (Scheme 53). Reduction with Red-Al<sup>224b</sup> followed by iodination then gave selectively the (Z) iodoolefin **604**, and hydroxyl protection followed by copper-catalyzed allyl Grignard addition afforded **605** with the required configuration of the  $C_{13}$  exocyclic double bond. Deprotection at  $C_{11}$  followed by oxidation then gave the  $C_{11}$ – $C_{16}$  segment **574**.

Stereoselective construction of the  $C_{10}-C_{11}$  bond was accomplished by means of an aldol reaction between aldehyde **574** and the enol borinate **606** derived from enolization of ketone **576** using the chiral boron reagent (R,R)-**596**.  $^{212d,222}$  Accordingly, the  $\beta$ -hydroxy ketone **607** was obtained with 89% ds in favor of the required configuration at  $C_{11}$ . Note that this is a *matched* double-diastereodifferentiating reaction:  $^{9b}$  use of the achiral *meso* reagent (R,S)-**596** still afforded **607** as the major adduct, but with diminished diastereoselectivity (75% ds), indicating the degree of substrate control of asymmetric induction. In the *mismatched* case using the antipodal chiral reagent (S,S)-**596**, reagent control dominated substrate control and the epimeric product 11-epi-

## Scheme 54. Masamune Bryostatin 7 C<sub>17</sub>-C<sub>27</sub> Synthesis<sup>212b a</sup>

"a (a) NaNO<sub>2</sub>, H<sub>2</sub>SO<sub>4</sub>, H<sub>2</sub>O; (b) MeOH, AcCl; (c) (MeO)<sub>2</sub>CMe<sub>2</sub>, p-TsOH; (d) DIBAL; (e) Ph<sub>3</sub>P=CH<sub>2</sub>; (f) Sia<sub>2</sub>BH; H<sub>2</sub>O<sub>2</sub>; (g) PCC; (h) allenyl-ZnBr; (i) NaH, PMBCl; (j) 'BuLi; ClCO<sub>2</sub>Me; (k) "Bu<sub>3</sub>SnCu-LiBr-Me<sub>2</sub>S, MeOH; (l) DIBAL; (m) TBDPSCl, imidazole; (n) I<sub>2</sub>; (o) (+)-DET, Ti(O'Pr)<sub>4</sub>, 'BuOOH; (p) PhNCO, Et<sub>3</sub>N; (q) BF<sub>3</sub>·OEt<sub>2</sub>, H<sub>2</sub>SO<sub>4</sub>, H<sub>2</sub>O; (r) K<sub>2</sub>CO<sub>3</sub>, MeOH; (s) (MeO)<sub>2</sub>CMe<sub>2</sub>, PPTS or (MeO)<sub>3</sub>CH, p-TsOH; (t) NaH, m,p-dimethoxybenzyl chloride; (u) Raney Ni; (v) MsCl, Et<sub>3</sub>N; PhSNa; (w) HCl, MeOH; (x) NaIO<sub>4</sub>; (y) **613**, "BuLi; **612**; (z) TESOTf, 2,6-lutidine; (a') DDQ (1 equiv); (b') DMSO, Ac<sub>2</sub>O; Et<sub>3</sub>N; (c') MoO<sub>5</sub>·HMPA·H<sub>2</sub>O; (d') DDQ; SiO<sub>2</sub>; (e') TMSOTf, TMSOMe or BF<sub>3</sub>·OEt<sub>2</sub>, MeOH.

607 was produced as the major diaster eomer with  $67\%~ds.^{212d}$ 

Acid-catalyzed cleavage of the  $C_5$ ,  $C_7$  acetonide in **607** triggered simultaneous acetalization at C<sub>9</sub> to provide 608 with the correctly assembled A ring. Formation of the B ring was achieved by oxymercuration of the terminal double bond of **608**; acetylation of the  $C_7$  hydroxyl then furnished **609** as a mixture of C<sub>15</sub> epimers. After oxidative demercuration<sup>225</sup> to give alcohol 610, Swern oxidation38 afforded aldehyde **611** as a 1:1 mixture of  $C_{15}$  epimers. Upon exposure to alumina, however, equilibration was effected to afford a 9:1 ratio in favor of the desired equatorially disposed aldehyde 611. Thus, in the synthesis of the  $C_3-C_{16}$  segment **611**, one of the three newly created stereogenic centers was constructed using reagent control of asymmetric induction (574 + 606  $\rightarrow$  607 for  $C_{11}$ ); the other two were installed by substratecontrolled reactions (607  $\rightarrow$  608 for C<sub>9</sub>, and thermodynamic equilibration of aldehyde **611** for  $C_{15}$ ). [ $C_3$ –  $C_{16}$  segment **611**: <5% overall yield from **580**; 24 steps longest linear sequence; 31 steps total;  $\sim$ 6 steps per stereogenic center.]

c.  $C_{17}$ – $C_{27}$  Segment Synthesis. <sup>212b</sup> The  $C_{17}$ – $C_{27}$  segment **575** was constructed via the coupling of  $C_{17}$ – $C_{20}$  and  $C_{21}$ – $C_{27}$  segments **612** and **613** (Scheme 54). Compound **613** was prepared from L-threonine (**614**). Thus, deamination of **614** gave **615**, with retention of configuration at  $C_{25}$ , <sup>225</sup> and esterification followed by acetonide formation then supplied **616**. <sup>226</sup> After reduction to the corresponding  $C_{24}$  aldehyde, homologation to provide **617** was effected by a three-step sequence of Wittig olefination, hydroboration, and oxidation. Chelation-controlled addition of allenylzinc bromide to **617** then furnished alkyne **618** with

89% ds in favor of the required configuration at  $C_{23}$ . After protection of the  $C_{23}$  hydroxyl, alkynyllithium formation, and addition to methyl chloroformate afforded the acetylenic ester **619**. Stereoselective introduction of the exocyclic double bond at  $C_{21}$  was accomplished by using Piers' method.<sup>228</sup> Thus conjugate addition of a (tributylstannyl)cuprate to **619** gave **620**. After reduction of the ester and protection of the resulting alcohol, replacement of the tributylstannyl group with iodine then afforded **613**.

Meanwhile, aldehyde **612** was prepared from allylic alcohol **581**, which was used in the synthesis of the  $C_3-C_{10}$  segment **576**, via a sequence of reactions developed by Masamune and Sharpless for saccharide synthesis. Thus, Sharpless epoxidation of **581** afforded **621** with 92% ee; **621** was then converted into the phenylurethane **622** and exposure to  $BF_3$ • $OEt_2$  led to formation of the carbonate **623** with inversion of configuration at  $C_{20}$ . After a sequence of protecting group manipulations to give **624**, deprotection at  $C_{17}$  was followed by mesylation and thiophenolate displacement. Acetonide removal followed by oxidative glycol cleavage then furnished the  $C_{17}-C_{20}$  segment **612**.

The stereogenic center at  $C_{19}$  of **612** was used to direct the introduction of stereochemistry at  $C_{20}$ , and, having served its purpose, was then destroyed through oxidation. Thus, chelation-controlled coupling of aldehyde **612** with the lithio anion derived from iodide **613** afforded the  $C_{17}-C_{27}$  segment **625** with 86% ds in favor of the required configuration at  $C_{20}$ . After protection of the  $C_{20}$  hydroxyl, selective cleavage of the  $C_{19}$  m.p-dimethoxybenzyl ether in the presence of the  $C_{23}$  p-methoxybenzyl ether<sup>42b,230</sup> was followed by oxidation at  $C_{19}$  to supply ketone **626**.

# Scheme 55. Masamune Bryostatin 7 Synthesis<sup>212e</sup> a

<sup>a</sup> (a) 575, PhLi; 611; BzCl, DMAP; (b) Na-Hg, Na<sub>2</sub>HPO<sub>4</sub>; (c) TBAF; (d) TBSCl, imidazole; (e) Ac<sub>2</sub>O, py, DMAP; (f) TBAF; (g) MnO<sub>2</sub>; MeOH, NaCN, AcOH; (h) Swern oxidation; (i) 595 + (S,S)-596, <sup>i</sup>Pr<sub>2</sub>NEt; 629; H<sub>2</sub>O<sub>2</sub>; (j) CSA, MeOH; (k) TESOTf, 2,6-lutidine; (l) Hg(O<sub>2</sub>CCF<sub>3</sub>)<sub>2</sub>, Na<sub>2</sub>HPO<sub>4</sub>; (m) HF·py; (n) DCC, PPTS, py; (o) K<sub>2</sub>CO<sub>3</sub>, MeOH; 5% HCl, H<sub>2</sub>O; (p) TBSCl, Et<sub>3</sub>N, DMAP; (q) Ac<sub>2</sub>O, py; (r) HF.

Deprotection at  $C_{23}$  and subsequent intramolecular hemiacetal formation then furnished 627. Finally, methoxylation of 627 under forcing conditions<sup>231</sup> gave the  $C_{17}-C_{27}$  segment **575** having all stereogenic centers in the C ring correctly installed. Thus, in the synthesis of **575**, two stereogenic centers originated in the chiral pool, and the remaining three were installed by substrate-controlled reactions. One of the substrate-controlled reactions  $(612 + 613 \rightarrow 625)$ relied on asymmetric induction from a stereogenic center, created using reagent control, which was later destroyed. Note that in 575, Masamune and coworkers opted to forego the possibility of differential protection of the C<sub>26</sub> and C<sub>25</sub> hydroxyls. Hence a regioselective macrolactonization would be required later in the synthesis.  $[C_{17}-C_{27} \text{ segment } 575: 1.3\%]$ overall yield from 614; 22 steps longest linear sequence; 36 steps total; ~7 steps per stereogenic center.]

d. Completion of the Total Synthesis of Bryostatin 7.  $^{212e}$  The  $C_3$ – $C_{16}$  and  $C_{17}$ – $C_{27}$  segments **611** and **575** were coupled by means of a Julia–Lythgoe reaction  $^{179}$  using phenyllithium as base (Scheme 55). Note that the choice of base was critical: because of the steric congestion around  $C_{17}$ , weaker bases such as lithium amide bases (LDA, LiNEt<sub>2</sub>) were ineffective; stronger carbon bases ( $^n$ BuLi,  $^t$ BuLi) led to concomitant formation of arylic anions. A reductive elimination reaction, performed in the presence of  $Na_2HPO_4$  in order to retain the  $C_7$  acetate group,  $^{232}$  then furnished **628** with the required trans double bond at  $C_{16}$ – $C_{17}$ . After a series of protecting group manipulations and adjustment of oxidation state to afford **629**, introduction of the  $C_1$ – $C_2$  unit was effected by means of a

reagent-controlled222 double diastereodifferentiating aldol reaction. Thus addition of the chiral enol borinate **577** to aldehyde **629** provided  $\beta$ -hydroxy thioester **630** with 75% ds in favor of the required configuration at C<sub>3</sub>. Note that **577** and **629** constitute a mismatched pair,9b and hence reagent control from 577 is required to overturn the intrinsic diastereofacial preference of **629**. With the carbon skeleton of bryostatin 7 now intact, selective cleavage of the acetonide of 630 gave the seco-acid derivative 631. All attempts to achieve direct macrolactonization of 631 using a thiophilic metal cation<sup>233</sup> failed, and hence 631 was converted into the seco-acid 578. Macrolactonization of 578 was effected by employing a modification of the procedure of Boden and Keck<sup>48</sup> (DCC, with pyridine and PPTS in place of DMAP). Note that cyclization occurred selectively at  $C_{25}$ , without the need for protection of the C<sub>26</sub> hydroxyl. The  $C_9$  methyl acetal and  $C_7$  acetate were hydrolyzed under the reaction conditions, affording macrolide **632**. The  $C_{19}$  methyl acetal could not be hydrolyzed (note that forcing conditions<sup>231</sup> were required for its creation: 627 - 575 in Scheme 54) unless the electron-withdrawing acetate group at C20 was first removed. Selective silvlation of the C26 hydroxyl of 633 followed by reacetylation and desilylation then furnished bryostatin 7 (571) [5  $\times$  10<sup>-3</sup>% overall yield from **614**; 42 steps longest linear sequence; 80 steps total;  $\sim$ 7 steps per stereogenic center].

## 2. Vandewalle Segment Syntheses<sup>213</sup>

Vandewalle and co-workers have synthesized the three bryostatin 11 segments depicted in Scheme 56:  $C_1$ – $C_9$  segment **634**,  $C_{11}$ – $C_{16}$  segment **635**, and

Note: PMB = p-methoxybenzyl; DMB = m,p-dimethoxybenzyl

 $C_{17}-C_{27}$  segment **636**. Preparation of a  $C_{17}-C_{27}$  segment **637**, suitable for bryostatin 7, was also attempted. The total synthesis has not yet been reported, but assembly of segments **634** and **635** using a  $\beta$ -keto phosphonate to introduce  $C_{10}$ , subsequent construction of the  $C_{16}-C_{17}$  bond by a Julia–Lythgoe olefination reaction, <sup>179</sup> and final macrolactonization has been proposed. <sup>213c</sup> Vandewalle and coworkers adopted the "chiron" approach, whereby **634–636** were all prepared from starting materials available from the chiral pool, as indicated in Scheme 56.

a.  $C_1$ – $C_9$  Segment Synthesis. <sup>213a,c</sup> Vandewalle and co-workers investigated two different routes to the  $C_1$ – $C_9$  segment **634**, both starting from epoxide **638** (Scheme 57). The first, abortive, approach centered on the successive coupling of epoxides **638** and **639** 

to dithiane, which provided C<sub>5</sub>. After dithiane cleavage, the stereogenic center at C5 was to be installed by a 1,3-anti selective reduction of a  $\beta$ -hydroxy ketone. Epoxide 639 was available from (S)-malic acid (119) via alcohol 120<sup>60</sup> (cf. 119  $\rightarrow$  120  $\rightarrow$  121 in Scheme 8). Epoxide 638, meanwhile, was derived from (R)-pantolactone (640) following the route of Lavallée et al.<sup>234</sup> Thus, reduction of **640** followed by acetalization provided exclusively the C<sub>6</sub>,C<sub>7</sub>-pentylidene acetal 641; protection of the C<sub>9</sub> hydroxyl, acetal hydrolysis, and subsequent tosylimidazole-mediated epoxide formation then supplied 638. Reaction of 638 with 2-lithiodithiane followed by protection of the resulting C7 hydroxyl furnished 642. Unfortunately, coupling of the lithio anion of 642 with epoxide 639 afforded only very low yields of 643, even under optimized conditions. Accordingly, the first synthetic strategy was abandoned.

In the second approach, only the  $C_7$  stereogenic center originated in the chiral pool. Construction of the C<sub>5</sub> stereogenic center was accomplished by a chelation-controlled aldol reaction between 644 and **645**; a 1,3-anti selective reduction of  $\beta$ -hydroxy ketone 646 then set up the C3 stereogenic center. Production of aldehyde 644 by cleavage of dithiane **642** proved troublesome. Instead, **644** was obtained by vinylcuprate addition to epoxide **638** followed by hydroxyl protection and subsequent two-step oxidative alkene cleavage. The chelation-controlled aldol reaction between aldehyde 644 and ketone 645, obtained from 1,3-butanediol (647), afforded exclusively the  $\beta$ -hydroxy ketone **646** with the required configuration at C<sub>5</sub>. However, stereoselective reduction to provide 648 was accomplished only by using LiAl(O'Bu)<sub>3</sub>H in the presence of lithium iodide.<sup>235</sup> Note that the Saksena-Evans reagent<sup>35</sup> proved completely unselective in this case. Finally, ac-

Scheme 57. Vandewalle Bryostatin 11 C<sub>1</sub>-C<sub>9</sub> Synthesis<sup>213a,c a</sup>

 $^a$  (a) LAH; (b) Et<sub>2</sub>CO, p-TsOH; (c)  $^t\text{BuOK}$ , PMBCl; (d) HCl; (e) NaH; N-tosylimidazole; (f) 1,3-dithiane,  $^n\text{BuLi}$ ; (g)  $^t\text{BuOK}$ , BnBr; (h) BH<sub>3</sub>·Me<sub>2</sub>S; (i) Me<sub>2</sub>CO, H<sup>+</sup>; (j) TBDPSCl, imidazole; (k) H<sup>+</sup>; (l) TsCl, py; (m) K<sub>2</sub>CO<sub>3</sub>, MeOH; (n) **642**,  $^n\text{BuLi}$ , TMEDA; **639**, DMPU; (o) (H<sub>2</sub>C=CH)<sub>2</sub>Cu(CN)Li; (p)  $^t\text{BuOK}$ , BnBr; (q) OsO<sub>4</sub>, NMO, H<sub>2</sub>O; (r) Pb(OAc)<sub>4</sub>, py; (s) TBDPSCl, imidazole; (t) CrO<sub>3</sub>·py<sub>2</sub>; (u) **645**, LDA; **644**; (v) LiAl(OʻBu)<sub>3</sub>H, LiI; (w) (MeO)<sub>2</sub>CMe<sub>2</sub>, Amberlyst-15.

# Scheme 58. Vandewalle Bryostatin 11 $C_{11}-C_{16}$ Synthesis $^{213a}$ $^{a}$

<sup>a</sup> (a) Me<sub>2</sub>CO, ZnCl<sub>2</sub>, Na<sub>2</sub>SO<sub>4</sub>; (b) NaBH<sub>4</sub>; (c) NaIO<sub>4</sub>; (d) NaBH<sub>4</sub>; (e) TsCl, Et<sub>3</sub>N; (f) NaI; (g) 2-(trimethylsilyl)-1,3-dithiane, <sup>n</sup>BuLi; (h) TBAF; (i) **652**, <sup>n</sup>BuLi, HMPA; BrCH<sub>2</sub>CH(OMe)<sub>2</sub>; (j) HCl; (k) **655**, <sup>n</sup>BuLi; **635**.

etonide protection gave the  $C_1$ - $C_9$  segment **634** [30% overall yield from **640**; 12 steps longest linear sequence; 14 steps total;  $\sim 5$  steps per stereogenic center].

b.  $C_{11}$ – $C_{16}$  Segment Synthesis. <sup>213a</sup> The  $C_{11}$ – $C_{16}$  segment **635** was obtained from L-erythrulose (**649**), which provided the  $C_{15}$  stereogenic center (Scheme 58). Thus, conversion of **649** into the  $C_{15}$ ,  $C_{16}$  acetonide **650** was followed by reduction at  $C_{14}$  and cleavage of the resulting glycol to afford the corresponding  $C_{14}$  aldehyde. <sup>236</sup> Reduction and subsequent tosylation then furnished **651**, which was trans-

formed into dithiane **652** via **653**.<sup>237</sup> Note that direct formation of **652** from **651** was low yielding. Alkylation of the  $C_{13}$  lithio anion of **652** supplied **654**, and acid-catalyzed transacetalization provided the  $C_{11}$ – $C_{16}$  segment **635** as an anomeric mixture. Note that **635** has a protected carbonyl group at  $C_{13}$ . Vandewalle and co-workers envisage formation of the  $C_{13}$  exocyclic  $\alpha,\beta$ -unsaturated ester via a stereoselective Horner–Emmons reaction<sup>238,239</sup> later in the synthesis. [ $C_{11}$ – $C_{16}$  segment **635**: 32% overall yield from **649**; 10 steps.]

Vandewalle and co-workers have proposed union of the  $C_1-C_9$  and  $C_{11}-C_{16}$  segments **634** and **635** using a  $\beta$ -keto phosphonate to introduce  $C_{10}$ . Model studies have been performed using  $\beta$ -keto phosphonate **635**. Thus, deprotonation of **655** and addition to **635** afforded **656** as the sole product. The stereoselective construction of the B ring in **656** occurs via the *in situ* intramolecular hetero-Michael reaction of intermediate **657**, in which formation of the new  $C_{11}$  stereogenic center is controlled by the existing  $C_{15}$  stereogenic center.

c.  $C_{17}$ – $C_{27}$  Segment Synthesis. <sup>213b.c</sup> The bryostatin 11  $C_{17}$ – $C_{27}$  segment **636** was constructed from  $C_{17}$ – $C_{20}$  and  $C_{21}$ – $C_{27}$  segments **658** and **659** (Scheme 59). Compound **659** was prepared from (R)-isobutyl lactate (**660**), which supplied the  $C_{26}$  stereogenic center. Two chelation-controlled allylation reactions were then used to install the  $C_{25}$  and  $C_{23}$  stereogenic centers. Compound **658**, meanwhile, was obtained from (R)-pantolactone (**640**). Although the  $C_{19}$  stereogenic center in **640** was destroyed in the synthesis of **636**, it has the potential for controlling the installation of a stereogenic center at  $C_{20}$  in a synthesis of the bryostatin 7  $C_{17}$ – $C_{27}$  segment **637**.

Thus, after protection of the hydroxyl of **660**, adjustment of oxidation state led to aldehyde **661**.  $\alpha$ -Chelation-controlled allylstannane addition<sup>240</sup> to **661** then afforded exclusively **662** having the required configuration at  $C_{25}$ . Oxidative cleavage of the

Scheme 59. Vandewalle Bryostatin 11  $C_{17}$ – $C_{27}$  Synthesis<sup>213b,c</sup> a

"a (a) MOMCl,  ${}^{i}Pr_{2}NEt$ ; (b) LAH; (c) Swern oxidation; (d)  $H_{2}C$ =CHCH $_{2}Sn^{n}Bu_{3}$ , MgBr $_{2}OEt_{2}$ ; (e)  ${}^{i}BuOK$ , PMBCl; (f) OsO $_{4}$ , NaIO $_{4}$ ; (g)  $H_{2}C$ =CHCH $_{2}Sn^{n}Bu_{3}$ , MgBr $_{2}OEt_{2}$ ; (h)  ${}^{i}BuOK$ ,  $m_{s}p$ -dimethoxybenzyl chloride; (i) OsO $_{4}$ , NMO; (j) NaIO $_{4}$ ,  ${}^{n}Bu_{4}NBr$ ,  $H_{2}O$ ; (k) (MeO) $_{2}P$ (=O)CHN $_{2}$ ,  ${}^{i}BuOK$ ; (l)  ${}^{n}BuLi$ ; ClCO $_{2}Me$ ; (m)  ${}^{n}Bu_{3}SnCuMe_{2}SLiBr$ , MeOH; (n) DIBAL; (o) TBDPSCl, imidazole, DMAP; (p) I $_{2}$ ; (q) LAH; (r) Et $_{2}CO$ , p-TsOH; (s) TsCl, py, DMAP; (t)  ${}^{i}BuOK$ , PhSH; (u)  $H_{2}SO_{4}$ , MeOH; (v) NaH, TsCl; (w)  ${}^{i}BuPh(MeO)SiBr$ , Et $_{3}N$ ; (x) TBSOTf, 2,6-lutidine; (y) HF-py; (z) TPAP, NMO; (a') **659**,  ${}^{i}BuLi$ ; 2-thienyl-CuCNLi; **658**, BF $_{3}OEt_{2}$ ; (b') DMSO, SO $_{3}$ -py; Et $_{3}N$ ; (c') **659**,  ${}^{i}BuLi$ ; **671**.

double bond of 662 provided aldehyde 663, and a  $\beta$ -chelation-controlled allylstannane addition provided 664 as the sole product. Note the much lower level of diastereoselectivity (71% ds) observed for the corresponding reaction of the C25, C26 acetonideprotected analogue of 663, wherein each acetal oxygen could be involved in chelation. After protection of the  $C_{23}$  hydroxyl of **664**, oxidative double-bond cleavage supplied the aldehyde **665**. Conversion to the alkyne 666 was best effected by employing the Seyferth reagent (dimethyl diazomethylphosphonate).<sup>241</sup> Lithiation of **666** and addition to methyl chloroformate then gave 667. As in the Masamune synthesis (vide supra), 212b stereoselective introduction of the exocyclic double bond at  $C_{21}$  was accomplished by using Piers' method. Thus, conjugate organostannylcuprate addition to 667 gave the (E)vinylstannane 668. After reduction of the ester and protection of the resulting alcohol, replacement of the tributylstannyl group with iodine then afforded **659**  $(cf. 619 \rightarrow 620 \rightarrow 613 \text{ in Scheme } 54).$ 

Meanwhile, tosylation of alcohol 641, derived from (R)-pantolactone (640), followed by thiophenolate displacement and acetal cleavage supplied diol 669. Conversion to epoxide **658** was then effected via the primary tosylate. Reaction of 658 with the mixed higher order cuprate derived from iodide **659** led to the  $C_{17}$ – $C_{27}$  segment **670**, albeit in low yield, together with substantial amounts of dehalogenated 659. Although oxidation at C<sub>19</sub> was envisaged after C<sub>16</sub>-C<sub>17</sub> coupling, Vandewalle and co-workers sought to confirm that oxidation could be performed without concomitant migration of the C21 exocyclic double bond. In fact, oxidation of 670 led smoothly to 636 [3.2% overall yield from **660**; 18 steps longest linear sequence: 24 steps total; 8 steps per stereogenic center].

The potential for controlled installation of a stereogenic center at C20 via the coupling of 659 with aldehyde 671 was also investigated. Unfortunately, reaction of 671, prepared from diol 669, with the lithio anion derived from iodide 659 furnished the  $C_{17}$ - $C_{27}$  segment **672** as a 2:1 ratio of  $C_{20}$  epimers. The degree of Cram control in this reaction was lower than expected, and better stereocontrol is required for efficient synthesis of the bryostatin 7 segment 637. Note that Masamune and co-workers were able to achieve much higher diastereoselectivity (86% ds) in the  $C_{20}-C_{21}$  bond construction by employing the antipode of aldehyde 671 and by using a dimethoxybenzyl ether rather than a silyl ether as the protecting group at  $C_{19}$ , such that chelation control could be used effectively in the coupling reaction (612 + **613**  $\rightarrow$  **625** in Scheme 54).  $^{212b}$ 

## 3. Roy Segment Syntheses<sup>214</sup>

Roy et al. have synthesized the three bryostatin 11 segments depicted in Scheme 60:  $C_1-C_9$  segment 673,  $C_{17}-C_{20}$  segment 674, and  $C_{21}-C_{27}$  segment 675. These segments were all prepared according to the "chiron" approach,<sup>8</sup> *i.e.* using starting materials available from the chiral pool, as indicated in the scheme

a.  $C_1-C_9$  Segment Synthesis. 214b In the synthesis of the  $C_1-C_9$  segment 673, the  $C_5$  stereogenic center

Scheme 60

was installed by a stereoselective Mukaiyama aldol reaction between silyl enol ether 676 and the fivecarbon chiral building block 677, which was obtained in high enantiomeric excess via enzymatic hydrolysis of a prochiral precursor (Scheme 61).242 Stereoselective reduction of the resulting  $\beta$ -hydroxy ketone then introduced the remaining C<sub>7</sub> stereogenic center. Thus, borohydride reduction<sup>243</sup> of dimethyl 3-ketoglutarate (678) followed by protection of the resulting hydroxyl gave the prochiral enzyme substrate 679. Selective hydrolysis of the pro-S ester group of 679 upon incubation with  $\alpha$ -chymotrypsin then supplied the mono acid 680,242 and borohydride reduction of the mixed anhydride of 680 followed by PCC oxidation furnished aldehyde 677. Meanwhile, reaction of diketene (681) with tert-butyl mercaptan gave β-keto thioester **682**.<sup>244</sup> Bis(methylation) at C<sub>8</sub> followed by silvl enol ether formation then afforded 676.

 $\beta$ -Chelation-controlled Mukaiyama aldol reaction<sup>245</sup> between 676 and 677 was expected<sup>246</sup> to give selectively the  $\beta$ -hydroxy ketone **683** with the required configuration at  $C_5$  (si-face attack). However, the use of both TiCl4 and SnCl4 as chelating Lewis acids led to formation of the unwanted isomer 684 as the major product (683/684 = 40:60). This behavior may be a consequence of alternative modes of chelation—for instance, additional coordination by the  $C_1$  carbomethoxy group, as proposed by Roy, or, alternatively, coordination by both oxygen atoms of the  $C_3$ MOM ether—resulting in cagelike structures which favor re-face attack. In contrast, by employing the nonchelating Lewis acid BF<sub>3</sub>·OEt<sub>2</sub> in the Mukaiyama aldol reaction, the desired isomer 683 could be obtained as the major product (683/684 = 66:34). Note that stereochemical control in this reaction arises solely from electrostatic repulsion.<sup>247</sup> Reduction of the inseparable mixture of 683 and 684 using the Saksena-Evans reagent<sup>35</sup> occurred with >98% ds, to afford the corresponding C5, C7-anti diols, and acetonide protection then gave 673 and 685 which could be separated. Besides synthesizing the  $C_1-C_9$ segment 673, Roy et al. also prepared a C<sub>1</sub>-C<sub>9</sub> segment (686) suitable for biological activity studies. Thus, selective cleavage of the acetonide of 673 and lactonization of the resulting diol furnished 687; acetylation of the C7 hydroxyl and cleavage of the C3

## Scheme 61. Roy Bryostatin 11 C<sub>1</sub>-C<sub>9</sub> Synthesis<sup>214b a</sup>

 $^{a}$  (a)  $^{t}$ BuSH, NaH; (b)  $^{t}$ BuOK, MeI; (c) TMSOTf, Et<sub>3</sub>N; (d) NaBH<sub>4</sub>; (e) (MeO)<sub>2</sub>CH<sub>2</sub>, P<sub>2</sub>O<sub>5</sub>; (f) α-chymotrypsin; (g) EtOCOCl, Et<sub>3</sub>N; (h) NaBH<sub>4</sub>; (i) PCC; (j) **676** + **677**, BF<sub>3</sub>·OEt; (k) Me<sub>4</sub>NBH(OAc)<sub>3</sub>; (l) (MeO)<sub>2</sub>CMe<sub>2</sub>, p-TsOH; (m) PPTS, MeOH; (n) Hg(CF<sub>3</sub>CO<sub>2</sub>)<sub>2</sub>; (o) Ac<sub>2</sub>O, py; (p) TiCl<sub>4</sub>.

MOM ether then gave **686**.  $[C_1-C_9]$  segment **673**: 20% overall yield from **678**; 9 steps longest linear sequence; 12 steps total; 4 steps per stereogenic center.]

b.  $C_{17}$ - $C_{20}$  and  $C_{21}$ - $C_{27}$  Segment Syntheses. 214a The  $C_{17}$ – $C_{20}$  segment **674** was prepared from (*R*)pantolactone (640) by a two-step sequence of hydroxyl protection followed by reduction (Scheme 62). The  $C_{19}$  stereogenic center in **674** was expected to permit diastereoselective addition of the C<sub>21</sub>-C<sub>27</sub> dithiane **675**, thus allowing controlled installation of the  $C_{20}$ stereogenic center in a synthesis of a bryostatin 7 C<sub>17</sub>-C<sub>27</sub> segment. Indeed, model studies<sup>214c</sup> revealed that 2-lithiodithiane added to 674 with high diastereoselectivity (96% ds) to afford the Cram adduct 688 having the required configuration at  $C_{20}$ . Note that use of the unprotected analogue of 674, namely 689, led to completely the opposite sense of stereochemical induction during the dithiane addition, yielding 690 as a consequence of chelation control.

The  $C_{21}$ – $C_{27}$  segment **675** was obtained from D-galactono-1,4-lactone (691), which supplied the stereogenic centers at  $C_{23}$ ,  $C_{25}$ , and  $C_{26}$ . Deoxygenation at  $C_{24}$  and  $C_{27}$  was therefore required. Thus, one-pot bromination and acetylation of **691** supplied **692**.<sup>248</sup> Heterogeneous hydrogenation in the presence of triethylamine then afforded 693 via reaction of the intermediate enol acetate 694 on its less-hindered β-face.<sup>248</sup> After borohydride reduction to give **695**, sequential bis(acetonide) formation and kinetic monoacetonide cleavage<sup>249</sup> furnished the diol **696**. Conversion to the epoxide 697 was then effected via the primary tosylate. Regioselective opening of epoxide **697** with 2-lithiodithiane followed by protection of the resulting  $C_{23}$  hydroxyl then provided the  $C_{21}$ -C<sub>26</sub> segment **675**. On the basis of the model studies (vide supra),  $^{214c}$  coupling of 675 and the  $C_{17}-C_{20}$ segment 674 is expected to proceed with high diastereoselectivity to afford a  $C_{17}$ – $C_{27}$  segment of bryostatins 1 and 7.  $[C_{21}-C_{26} \text{ segment } \mathbf{675}: 11\% \text{ overall}$ yield from 691; 9 steps; 3 steps per stereogenic center.]

# Scheme 62. Roy Bryostatin 11 $C_{17}$ – $C_{20}$ and $C_{21}$ – $C_{27}$ Syntheses<sup>214a a</sup>

 $^a$  (a) TBSCl, DMAP, Et<sub>3</sub>N; (b) DIBAL; (b') BH<sub>3</sub>·THF; (c) 1,3-dithiane,  $^n$ BuLi; (d) HBr, AcOH; Ac<sub>2</sub>O; (e) H<sub>2</sub>, 5% Pd-C, Et<sub>3</sub>N; (f) LiBH<sub>4</sub>; (g) (MeO)<sub>2</sub>CMe<sub>2</sub>, p-TsOH; (h) p-TsOH, MeOH or I<sub>2</sub>, MeOH; (i) TsCl, py; (j) K<sub>2</sub>CO<sub>3</sub>, MeOH; (k) 1,3-dithiane,  $^n$ BuLi; (l) BzCl, py.

# 4. Nishiyama/Yamamura Segment Syntheses<sup>217</sup>

Nishiyama, Yamamura, and co-workers have synthesized a  $C_1$ - $C_{16}$  segment **698** of the bryostatins via

coupling of the  $C_5-C_9$  and  $C_{10}-C_{16}$  segments **699** and **700**, followed by addition of a chiral enolate representing  $C_1-C_4$  synthon **701** (Scheme 63). Stereocontrolled introduction of the  $\alpha,\beta$ -unsaturated ester at the  $C_{13}$  position was then attempted using a simple Horner-Emmons reaction.

a.  $C_5-C_9$  and  $C_{10}-C_{16}$  Segment Syntheses.<sup>217</sup> The route to the  $C_5-C_9$  segment **699** began with epoxy

alcohol **582**, which was an intermediate in the Masamune synthesis. Directed reduction with Red-Al then afforded the 1,3-diol **702** (Scheme 64). After protection of the  $C_5$  and  $C_7$  hydroxyls, deprotection at  $C_9$  followed by oxidation supplied aldehyde **703**, and thioacetal formation then gave **699**.

Meanwhile, synthesis of the  $C_{10}-\bar{C}_{16}$  segment 700 began with stereoselective conjugate addition of a vinyl group to enone 704, the hetero-Diels-Alder adduct of the Danishefsky diene (705) and the glyceraldehyde derivative 706.<sup>250</sup> After protection of the  $C_{13}$  carbonyl group of the resulting 707, ozonolysis supplied the aldehyde 708. Epimerization at the  $C_{15}$   $\alpha$ -carbon was then effected, to afford the thermodynamically more favorable  $C_{11}$ ,  $C_{15}$ -cis isomer 709. Reduction at  $C_{16}$  and protection of the resulting hydroxyl furnished 710; selective removal of the acetonide and subsequent glycol cleavage followed by reduction then gave alcohol 711. Introduction of iodine at  $C_{10}$  was accomplished under standard conditions to afford 700 in readiness for coupling.

b.  $C_1$ - $C_{16}$  Segment Synthesis.<sup>217</sup> Lithiation of the sterically encumbered dithiane **699** was best effected using tert-butyllithium and HMPA; addition to iodide

Scheme 64. Nishiyama/Yamamura Bryostatin 1 C<sub>1</sub>-C<sub>16</sub> Synthesis<sup>217 a</sup>

 $\begin{tabular}{ll} $^a$ (a) $\it{\bf 705}$ + $\it{\bf 706}$, $ZnCl_2$; $CF_3CO_2H$; (b) $H_2C=CHMgBr$, $CuI$, $TMSCl$, $DMPU$; (c) $(MeO)_2CMe_2$, $MeOH$, $PTS$; (d) $O_3$; $Me_2S$; (e) $K_2CO_3$, $MeOH$; (f) $NaBH_4$; (g) $BnBr$, $NaH$; (h) $Amberlite $IR-120 B$ ($H^+$)$, $MeOH$; (i) $NaIO_4$; (j) $NaBH_4$; (k) $I_2$, $Ph_3P$, $imidazole$; (k') $TsCl$, $py$; $NaI$; (l) $Red-Al$; (m) $TBSCl$, $imidazole$; (n) $H_2$, $Pd-C$; (o) $Swern oxidation$; (p) $HS(CH_2)_3SH$, $MgBr_2OEt_2$; (q) $\it{\bf 699}$, $^tBuLi$, $HMPA$; $\it{\bf 700}$; (r) $PPTS$, $MeOH$; (s) $DMSO$, $SO_3$; $py$; $Et_3N$; (t) $MeCOCH_2CO_2$'Bu$, $Ac_2O$, $H_2SO_4$; (u) $\it{\bf 714}$, $LDA$; $\it{\bf 713}$, $LiI$; (v) $EtOH$, $\Delta$; (w) $Me_4NBH(OAc)_3$; (x) $H_3O^+$; (y) $(MeO)_2CMe_2$, $PPTS$, $Me_2CO$; (z) $(MeO)_2P(=O)CH_2CO_2Me$, $NaH$; (a') $HgCl_2$, $HgO$, $H_2O$; (b') $PPTS$, $MeOH$. } \end{tabular}$ 

**700** then afforded a high yield of the  $C_5-C_{16}$  segment 712. After selective cleavage of the C<sub>5</sub> silyl ether of 712, oxidation supplied aldehyde 713. Meanwhile, enolization of the chiral enone 714,251 derived from (-)-menthone (715), afforded the lithium enolate 716. In the presence of lithium iodide, 102 716 underwent a highly diastereoselective aldol addition to aldehyde 713 to furnish the  $C_1-C_{16}$  segment 717 with the required configuration at  $C_5$  (96% ds). Note that in the absence of the additive, the sense of diastereoselectivity of the coupling reaction was reversed, such that the C<sub>5</sub> epimer of 717 was now the major product. Upon treatment of 717 with ethanol in refluxing toluene, removal of the chiral auxiliary via ketene **718** furnished the  $\beta$ -keto ester **719**. After reduction to the corresponding C3, C5-anti diol using the Saksena-Evans reagent, 35 deprotection at C13 followed by acetonide protection at C<sub>3</sub> and C<sub>5</sub> supplied the ketone **720**. Horner-Emmons reaction of **720** then gave a mixture of the corresponding  $C_{13}$  exocyclic  $\alpha,\beta$ -unsaturated esters. After removal of the  $C_9$  thioacetal group, formation of the A ring via acid-catalyzed acetalization afforded a 62:38 ratio of **698** and **721**. In this synthesis, one stereogenic center  $(C_7)$  in the  $C_1-C_{16}$  segment **698** was introduced using reagent control  $(\rightarrow 582)$ . The remaining five stereogenic centers were installed using substrate-controlled reactions, including the use of a chiral auxiliary to construct the  $C_5$  stereogenic center (**713** + **716**  $\rightarrow$  **717**).  $[C_1-C_{16}$  segment **698**: 2.3% overall yield from **706**; 22 steps longest linear sequence; 28 steps total;  $\sim$ 4 steps per stereogenic center.]

# 5. Hale C<sub>17</sub>-C<sub>27</sub> Segment Synthesis<sup>215</sup>

Hale *et al.* have recently synthesized a  $C_{17}-C_{27}$  segment (**722** in Scheme 65) of bryostatin 1 via Claisen coupling of the  $C_{19}-C_{27}$  segment **723** and an ester enolate representing synthon **724**. Compound **723** was obtained by a Wittig olefination of the  $C_{21}-C_{27}$  segment **725**, followed by Sharpless asymmetric dihydroxylation<sup>252</sup> to introduce the  $C_{20}$  and  $C_{21}$  stereogenic centers. A combination of Sharpless asymmetric epoxidation (AE)<sup>92</sup> and a Sharpless asymmetric dihydroxylation (AD) was used to prepare segment **725**.

The synthesis of **725** began with regionselective Sharpless asymmetric dihydroxylation of the disubstituted double bond of (E)-1,4-hexadiene  $(\mathbf{726})$ , which introduced the  $C_{25}$  and  $C_{26}$  stereogenic centers with 94% ee (Scheme 66). Protection of the resulting diol then afforded **727**. After oxidative cleavage of the double bond of **727** to give the  $C_{23}$  aldehyde, Wittig olefination and subsequent reduction supplied the *trans* allylic alcohol **728**. Sharpless asymmetric epoxidation then afforded epoxy alcohol **729** with >96% ee. Note that this second asymmetric reaction using a chiral substrate and a chiral catalyst led to product **729** of enhanced enantiomeric purity as a result of diastereomer formation. Directed reduc-

## Scheme 66. Hale Bryostatin 1 C<sub>17</sub>-C<sub>27</sub> Synthesis<sup>215 a</sup>

 $^{\alpha}$  (a) AD-mix- $\beta$ ; (b) TBSCl, imidazole; (c) OsO<sub>4</sub>, NaIO<sub>4</sub>; (d) Ph<sub>3</sub>P=CHCO<sub>2</sub>Et; (e) DIBAL; (f) (-)-DET, Ti(O<sup>i</sup>Pr)<sub>4</sub>, 'BuOOH; (g) Red-Al; (h) p-MeO(C<sub>6</sub>H<sub>4</sub>)CH(OMe)<sub>2</sub>, PPTS; (i) DIBAL; (j) Swern oxidation; (k) Ph<sub>3</sub>P=CHCO<sub>2</sub>Et; (l) AD-mix- $\beta$ , MeSO<sub>2</sub>NH<sub>2</sub>; (m) (MeO)<sub>2</sub>CMe<sub>2</sub>, PPTS; (n) MeO<sub>2</sub>CCHMe<sub>2</sub>, LDA; **723**; (o) HF·py; (p) PivCl, py; (q) DDQ, H<sub>2</sub>O; (r) Amberlyst-15 (H<sup>+</sup>), MeOH; (s) AcCl, MeOH; (t) RuCl<sub>3</sub>, NaIO<sub>4</sub>; (u) Ph<sub>3</sub>P=CHCO<sub>2</sub>Me.

tion<sup>29</sup> of **729** with Red-Al provided diol **730** with the required configuration at  $C_{23}$ , and protecting group manipulation and adjustment of oxidation state then furnished aldehyde 725. Wittig homologation of 725, to afford the  $\alpha,\beta$ -unsaturated ester **731**, was followed by a second Sharpless AD reaction; protection of the resulting diol then supplied 723. After a Claisen reaction between ester 723 and the lithium enolate of methyl isobutyrate, to provide the  $\beta$ -keto ester **732**, exchange of protecting groups gave the C-ring precursor 733. Treatment of 733 with Amberlyst resin effected hemiacetal formation at C<sub>19</sub> to form the C ring, along with acetonide cleavage and simultaneous cyclization onto  $C_{17}$  to give the  $\gamma$ -butyrolactone; Fischer glycosidation<sup>254</sup> of the resulting **734** then furnished the acetal 735 having the required configuration at C<sub>19</sub>. Finally, oxidation<sup>160</sup> of **735** afforded the corresponding C21 ketone, and a nonstereoselective Wittig reaction then supplied the desired  $C_{17}$ - $C_{27}$  segment **722** along with its double-bond isomer 736 in a 1:1 ratio. Thus, in this synthesis of 722, four of the five stereogenic centers were installed using asymmetric induction from chiral catalytic reagents; the fifth stereogenic center, at  $C_{19}$ , was constructed using substrate control. [ $C_{17}$ – $C_{27}$  segment **722**: 1.1% overall yield from **726**; 21 steps;  $\sim$ 4 steps per stereogenic center.]

# 6. Evans Segment Syntheses<sup>216b</sup>

Evans et al. observed that the acetate-derived oxygenation pattern of the bryostatin backbone was especially apparent in seco-acid **737**, obtained by replacement of the unsaturated esters at  $C_{13}$  and  $C_{21}$  of the bryostatin skeleton (Scheme 67). The identification of recurring structural motifs was further enhanced by substitution of the  $C_{13}$  carbonyl with hydroxyl as in **738**. Thus inspection of **738** revealed that both  $C_1-C_6$  and  $C_{11}-C_{16}$  segments could be obtained from the same triol ester **739**, while the  $C_{21}-C_{27}$  segment might be derived from the one-carbon homologue **740** containing an additional stereogenic center. A unified synthetic approach to all stereoisomers of **739** and **740** was developed.

a.  $C_1$ – $C_6$  and  $C_{11}$ – $C_{16}$  Segment Syntheses. <sup>216b</sup> The syntheses of **739** and **740** were based upon the enantioselective asymmetric epoxidation/kinetic resolution of cinnamyl alcohols<sup>255</sup> and the use of metasubstituted anisyl rings as masked  $\beta$ -keto ester synthons<sup>256</sup> (Scheme 68). Thus, Sharpless epoxidation<sup>92,255</sup> of allylic alcohol **741**, derived from reduction of trans-cinnamate **742**, afforded epoxide **743** with 94% ee. After protection of the hydroxyl, Birch reduction gave dihydroanisole **744**. Ozonolysis then furnished the  $\beta$ -keto ester **745**. Finally, directed reduction of **745** using the Saksena–Evans reagent<sup>35</sup> afforded the anti 1,3-diol **739** with 93% ds, suitable for  $C_1$ – $C_6$  and  $C_{11}$ – $C_{16}$  segments of the bryostatins.

b.  $C_{21}-C_{27}$  Segment Synthesis. <sup>216b</sup> The racemic allylic alcohol **746** was prepared by aldol condensation of m-anisaldehyde (**747**) with acetone, followed by ketone reduction. <sup>23</sup> After Sharpless kinetic resolution of **746**, which afforded the anti epoxy alcohol **748** with >90% ee, Mitsunobu inversion <sup>85</sup> and saponification of the resulting benzoate ester supplied the syn epoxy alcohol **749**. Protection of the hydroxyl then gave **750**. Birch reduction of **750** was accompanied by silyl group migration. However, mi-

Scheme 68. Evans Bryostatin 1 C<sub>1</sub>-C<sub>6</sub>, C<sub>11</sub>-C<sub>16</sub>, and C<sub>21</sub>-C<sub>27</sub> Syntheses<sup>216b</sup> a

 $^{a}\left(a\right) DIBAL; (b) (+)-DIPT, Ti(O^{i}Pr)_{4}, ^{i}BuOOH; (c) TIPSCl, imidazole, DMAP; (d) Li, liquid NH_{3}, ^{i}BuOH; (e) O_{3}; Me_{2}S; (f) Me_{4}NHB(OAc)_{3}; (g) Me_{2}CO, NaOH; (h) NaBH_{4}, CeCl_{3}; (i) (+)-DIPT, Ti(O^{i}Pr)_{4}, ^{i}BuOOH; (j) DEAD, PPh_{3}, PhCO_{2}H; (k) K_{2}CO_{3}, MeOH; (l) TIPSOTf, Et_{3}N; (m) Pd-BaSO_{4}, H_{2}; (n) DIBAL; (o) Li, liquid NH_{3}, ^{i}PrOH; (p) O_{3}; Me_{2}S; (q) Me_{4}NHB(OAc)_{3}.$ 

gration was avoided by performing consecutive reductions under more controlled conditions. Thus, reductive cleavage of the epoxide via hydrogenolysis afforded **751**, and subsequent Birch reduction of the derived dialkylaluminum (**752**) cleanly supplied dihydroanisole **753**. Ozonolysis then furnished the  $\beta$ -keto ester **754**. Finally, directed reduction of **754** using the Saksena–Evans reagent<sup>35</sup> afforded the *anti* 1,3-diol **740** with 90% ds, suitable for a  $C_{21}-C_{27}$  segment of the bryostatins.

# 7. Evans C<sub>1</sub>-C<sub>16</sub> Segment Synthesis<sup>216a</sup>

The stereoselective construction of the exocyclic  $\alpha,\beta$ -unsaturated esters present at  $C_{13}$  and  $C_{21}$  of the bryostatins represents one of the key synthetic challenges posed by this class of natural products. The Masamune total synthesis<sup>212</sup> and the Vandewalle C<sub>17</sub>-C<sub>27</sub> fragment synthesis<sup>213b,c</sup> both successfully applied existing methodology<sup>224,228</sup> for the stereoselective construction of trisubstituted double bonds (vide supra). In contrast, Nishiyama, Yamamura, and co-workers achieved modest stereocontrol for the introduction of the  $C_{13}$   $\alpha,\beta$ -unsaturated ester by using a simple Horner-Emmons reaction.217 Evans and Carreira have investigated an entirely novel strategy for controlling the exocyclic olefin geometry at  $C_{13}$ , whereby a tethered phosphonate reagent anchored to a hydroxyl at  $C_{16}$  of the bryostatin fragment **755** underwent a highly selective Horner-Emmons macroolefination reaction<sup>49</sup> to afford macrolide 756 (Scheme 69). An analogous strategy could be used to control the olefin geometry at  $C_{21}$ , *i.e.* **757**  $\rightarrow$  **758**. <sup>257</sup>

Parameters such as tether length and cyclization conditions were defined using the model substrate 759 (Scheme 70). Preparation of racemic 759 began with  $\beta$ -hydroxy ketone **760**, derived from aldol reaction of ethyl acetoacetate (761) with 3-methyl-2butenal. Stereoselective reduction of 760 using the Saksena-Evans reagent<sup>35</sup> afforded the C<sub>13</sub>,C<sub>15</sub>-anti diol **762**; protection of the  $C_{13}$  and  $C_{15}$  hydroxyls and reduction at  $C_{11}$  then supplied aldehyde **763**. After Horner-Emmons olefination using a model  $\beta$ -keto phosphonate, and subsequent deprotection to provide 764, closure of the B ring via an intramolecular hetero-Michael reaction gave tetrahydropyran 765 with the required stereochemistry at  $C_{11}$ . Reduction and hydroxyl protection at  $C_9$ , and oxidation at  $C_{13}$ , then afforded **766** which was used in subsequent reactions as a 1:1 mixture of C<sub>9</sub> epimers. Ozonolysis of **766** supplied the corresponding ketoaldehyde;

chemoselective reduction of the aldehyde group<sup>258</sup> then furnished alcohol 767. Molecular modeling revealed that a six-carbon tether was optimal, since it would lead to formation of a 14-membered macrocycle in which the Z configuration of the double bond was calculated to be thermodynamically more stable than the E geometry. Thus, attachment of the corresponding  $\beta$ -keto phosphonate tether **768**, derived from hexane-1,6-diol (769) via 770, supplied the olefination precursor 759. Intramolecular Horner-Emmons reaction<sup>49</sup> of **759** was then effected using lithium chloride and triethylamine, affording the macrocycle 771 as a single olefin stereoisomer. Having served its function, the tether was then removed via methanolysis, to afford 772 with configuration of the  $C_{13}$   $\alpha,\beta$ -unsaturated ester as required for the bryostatins.

A similar cyclization was performed on an advanced intermediate (755) in the synthesis of a  $C_{1-}$   $C_{16}$  segment of bryostatin 1. Thus, Horner–Emmons reaction of 755<sup>259</sup> afforded the 14-membered macrodiolide 756 as a single olefin stereoisomer. Selective methanolysis of the saturated lactone supplied a primary alcohol at  $C_{16}$ ; oxidation then furnished the  $C_1-C_{16}$  aldehyde segment 773, in readiness for an envisaged coupling with a  $C_{17}-C_{27}$  sulfone segment via a *trans*-selective Julia–Lythgoe olefination. At the time of writing, no further work has been reported.

## 8. Thomas C<sub>10</sub>-C<sub>16</sub> Segment Synthesis<sup>218</sup>

Munt and Thomas have developed a novel route to  $C_{10}$ - $C_{16}$  segment 774 corresponding to the B ring of the bryostatins in which the geometry of the exocyclic double bond at C13 is established via cyclization of a vinyl radical (Scheme 71). Thus, Yamaguchi coupling<sup>87</sup> of epoxide 775 with methyl lithiopropynoate afforded alcohol 776, which was converted into alkoxymalonate 777 upon treatment with dimethyl diazomalonate and Rh<sub>2</sub>(OAc)<sub>4</sub>. After alkylation with Me<sub>2</sub>N=CH<sub>2</sub>+I<sup>-</sup> to give 778, N-methylation and decarboxylative elimination supplied enol ether 779 as described by Ganem. 260 Conjugate addition of an organostannylcuprate, according to the procedure of Piers, 228 gave 780; subsequent iodination then furnished the (E)-vinyl iodide 781. Radicalmediated cyclization of 781 was effected upon treatment with tributyltin hydride and AIBN to afford an 80:20 mixture of exocyclic double bond isomers 782 and 783. Note that the vinyl radical 784 equilibrates

## Scheme 70. Evans Bryostatin 1 C<sub>1</sub>-C<sub>16</sub> Synthesis<sup>216a a</sup>

<sup>a</sup> (a) **761**, LDA; 3-methyl-2-butenal; (b) Me<sub>4</sub>NBH(OAc)<sub>3</sub>; (c) PMBOMe, DDQ; (d) LAH; (e) Swern oxidation; (f) (MeO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sup>t</sup>Bu, LiCl, <sup>i</sup>Pr<sub>2</sub>NEt; (g) AcOH, H<sub>2</sub>O; (h) <sup>i</sup>BuOK; (i) Ac<sub>2</sub>O, py, DMAP; (j) NaBH<sub>4</sub>; (k) TIPSOTf, Et<sub>3</sub>N; (l) K<sub>2</sub>CO<sub>3</sub>, MeOH; (m) PDC, pyridinium trifluoroacetate; (n) O<sub>3</sub>; Me<sub>2</sub>S; (o) LiAlH(OCEt<sub>3</sub>)<sub>3</sub>; (p) NaH, TBSCl; (q) (MeO)<sub>2</sub>P(=O)CH<sub>2</sub>CO<sub>2</sub>H, DCC, DMAP; (r) Jones oxidation; (s) **767** + **768**, DCC, DMAP; (t) LiCl, Et<sub>3</sub>N; (u) K<sub>2</sub>CO<sub>3</sub>, MeOH; (u') Li<sub>2</sub>CO<sub>3</sub>, MeOH; (v) oxidation.

## Scheme 71. Thomas Bryostatin 1 C<sub>10</sub>-C<sub>16</sub> Synthesis<sup>218 a</sup>

 $^{a}$  (a) LiC≡CCO<sub>2</sub>Me, BF<sub>3</sub>·OEt<sub>2</sub>; (b) (MeO<sub>2</sub>C)<sub>2</sub>CN<sub>2</sub>, Rh<sub>2</sub>(OAc)<sub>4</sub>; (c) Me<sub>2</sub>N=CH<sub>2</sub><sup>+</sup>I<sup>-</sup>, Et<sub>3</sub>N; (d) MeI; (e)  $^{n}$ Bu<sub>3</sub>SnCu·LiBrMe<sub>2</sub>S; (f) I<sub>2</sub>; (g)  $^{n}$ Bu<sub>3</sub>SnH, AIBN,  $\Delta$ ; (h) NaBH<sub>4</sub>.

prior to cyclization. The preferential formation of **782**, in which the terminal methoxycarbonyl group is *trans* to the newly formed carbon—carbon single bond, may be accounted for by cyclization of the (Z)-vinyl radical occurring faster than cyclization of the E isomer. Note also that selective formation of the  $C_{11}$ ,  $C_{15}$ -cis-disubstituted products **782** and **783** implies stereoselective H-atom abstraction from the axial direction. Finally, chemoselective reduction of the saturated ester provided the  $C_{10}$ – $C_{16}$  bryostatin segment **774** [11% overall yield from **775**; 8 steps; 4 steps per stereogenic center].

### H. The Macrolactins

The macrolactins are a group of polyene macrolides isolated from a taxonomically undefined deep sea

bacterium.<sup>261</sup> Macrolactin A (**785** in Scheme 72), the parent aglycon, exhibits a number of interesting biological properties. These include selective antibacterial activity, inhibition of murine melanoma cancer cells, inhibition of mammalian  $Herpes\ simplex$  viruses, and protection of T-lymphoblast cells against human HIV replication.<sup>261</sup> Rychnovsky  $et\ al.$  have determined the absolute stereochemistry of macrolactin B (**786**) by a combination of spectral analysis, oxidative degradation, and chemical correlation studies.<sup>262</sup> Macrolactin A, the aglycon of macrolactin B, is assumed to have the same configuration. At the time of writing, no total synthesis of macrolactin A has been reported.<sup>263</sup> However, Grée and co-workers have prepared two differently protected  $C_{15}$ – $C_{24}$ 

segments, **787** and **788**, <sup>264</sup> and Donaldson *et al.* have prepared the model  $C_1-C_{11}$  and  $C_{16}-C_{24}$  segments, **789** and **790**. <sup>265</sup> Both groups independently adopted similar synthetic strategies exploiting the properties of diene-tricarbonyl complexes, whereby the Fe(CO)<sub>3</sub> group is used both as a temporary protecting group for 1,3-diene functionality, and as a means of directing stereocontrol in C-C bond formation at adjacent carbon atoms. <sup>266</sup>

## 1. Gree Segment Synthesis<sup>264</sup>

Grée and co-workers synthesized the C<sub>15</sub>-C<sub>24</sub> segments 787 and 788 starting from the optically pure diene-tricarbonyl complex 791, obtained via resolution,  $^{267}$  and (3R)-butanediol (647), which supplied the C<sub>23</sub> stereogenic center (Scheme 73). Thus, selective monobromination of 647 gave 792; hydroxyl protection and halogen exchange then supplied the iodide **793**. Reaction of **791** with the organolithium derived from 793 provided the diastereomeric alcohols 794 and **795** in a ratio of 80:20 in favor of the  $\psi$ -exo isomer 794.  $^{268}$  Deoxygenation at  $C_{20}$  was required in order to obtain the macrolactin A intermediates 787 and 788. This was accomplished by means of ionic hydrogenation<sup>269</sup> via the Fe(CO)<sub>3</sub>-stabilized pentadienyl cation;<sup>266e</sup> triethylsilane and trifluoroacetic acid proved to be the reagents of choice. Under these conditions, deoxygenation of the  $\psi$ -exo diastereomer

**794** proceeded smoothly to afford **796**, in which the  $C_{23}$  silyl ether had been cleaved. The  $\psi$ -endo diastereomer **795** could also be used in the synthesis. However, in order for deoxygenation of **795** to proceed cleanly, exchange of the  $C_{23}$  TBS ether protecting group for the more robust TBDPS group was first required; subsequent deoxygenation then afforded **797**. Finally, reprotection of the  $C_{23}$  hydroxyl of **796**, either as the PMB ether (**798**) or the silyl ether (**799**), followed by adjustment of oxidation state<sup>270</sup> at  $C_{15}$  supplied the  $C_{15}$ – $C_{23}$  segments **787** and **788**. These were obtained in overall yields of 12% and 15%, respectively, over the eight steps from **647**. Note that in the synthesis, the  $C_{23}$  stereogenic center originated in the chiral pool.

## 2. Donaldson Segment Syntheses<sup>265</sup>

The synthesis of the model  $C_{16}-C_{24}$  segment **790** by Donaldson et al. resembled the synthesis of the C<sub>15</sub>-C<sub>24</sub> segments **787** and **788** by Grée and coworkers insofar as the key reactions involved construction of the C20-C21 bond by organometallic addition to an aldehyde, and subsequent deoxygenation at  $C_{20}$  by ionic reduction. However, the Donaldson synthesis differed in that the C23 stereogenic center was installed by employing asymmetric induction from the remote Fe(CO)<sub>3</sub> group, via the temporary installation of a stereogenic center at C<sub>20</sub>, instead of relying upon the chiral pool. Thus, reaction of the racemic diene-tricarbonyl complex  $(\pm)$ -800 with the achiral Grignard reagent **801** provided the diastereomeric racemic alcohols 802 and 803 in almost equal amounts (Scheme 74). The lack of diastereoselectivity for Grignard addition was of no consequence, since the C20 stereogenic center was epimerized in the following step. Thus, acid-catalyzed hydrolysis of either pure 802, pure 803, or a mixture of both, afforded a mixture of diastereomeric lactols 804 and 805 in a ratio of 75:25. Subjection of the undesired  $\psi$ -endo isomers (804) to the hydrolysis conditions afforded more of the mixture of 804 and 805. The equilibration of 804 and 805 may be rationalized by ionization of the C<sub>20</sub> lactol C-O bond under the acidic conditions to generate the  $Fe(CO)_3$ -stabilized pentadienyl cation. Rotation about the  $C_{19}$ - $C_{20}$  and attack of oxygen on the face opposite to the Fe(CO)<sub>3</sub> group then effects epimerization.<sup>271</sup>

Scheme 73. Grée Macrolactin A C<sub>15</sub>-C<sub>24</sub> Synthesis<sup>264</sup> a

a (a) Ph<sub>3</sub>P, Br<sub>2</sub>; (b) TBSCl, imidazole; (c) NaI, CuI (cat.); (d) **793**, <sup>t</sup>BuLi; **791**; (e) PPTS, EtOH; (f) TBDPSCl, imidazole; (g) Et<sub>3</sub>SiH, CF<sub>3</sub>CO<sub>2</sub>H; (h) Cl<sub>3</sub>CC(=NH)OPMB, TfOH; (h') TBSCl, imidazole; (i) DIBAL; (j) <sup>n</sup>PrMgBr; azodicarbonyldipiperidine.

# Scheme 74. Donaldson Macrolactin A $C_{16}$ – $C_{24}$ Synthesis<sup>265</sup> a

$$(CO)_{3}Fe \xrightarrow{CHO} CHO CIMg \xrightarrow{21} 230$$

$$(\pm) -800 \qquad 801$$

$$(CO)_{3}Fe \xrightarrow{20} OH \qquad (CO)_{3}Fe \xrightarrow{20} OH \qquad (CO)_{3}$$

(a) H<sub>2</sub>SO<sub>4</sub>; (b) MeTi(O<sup>i</sup>Pr)<sub>3</sub>; (c) p-TsOH; (d) NaBH<sub>3</sub>CN, BF<sub>3</sub>·OEt<sub>2</sub>.

Installation of the  $C_{23}$  stereogenic center was accomplished via treatment of the  $\psi$ -exo isomers **805** with MeTi(O<sup>i</sup>Pr)<sub>3</sub>,<sup>272</sup> which furnished the  $C_{20}$ , $C_{23}$ -syn diol **806** as a single diastereomer. Note that the same reaction of the  $\psi$ -endo isomers **804** supplied the corresponding  $C_{20}$ , $C_{23}$ -syn diol **807** as a single diastereomer. The  $\psi$ -exo configuration at  $C_{20}$  of **806** 

positions the hydroxyl appropriately for ionization with anchimeric assistance from the metal center. Thus, treatment of **806** with acid led to loss of the  $C_{20}$  hydroxyl, and participation of the  $C_{23}$  hydroxyl then afforded the tetrahydrofuran **808**. Finally, ionic reduction<sup>269</sup> of **808** via the pentadienyl cation provided the  $C_{16}-C_{24}$  segment **790** in an overall yield of 13% over five steps. Thus, in this synthesis, the Fe- $(CO)_3$  group was used to control the installation of the temporary stereogenic center at  $C_{20}$ , which in turn was used to direct the formation of the  $C_{23}$  stereogenic center; the stereogenic center at  $C_{20}$  was then removed.

In the synthesis of the model  $C_1-C_{11}$  segment **789** by Donaldson et al., the dienylic C7 stereogenic center was installed by employing asymmetric induction from the neighboring Fe(CO)<sub>3</sub> group (Scheme 75). Thus, TiCl4-induced hetero-Diels-Alder reaction of the racemic diene-tricarbonyl complex  $(\pm)$ -800 with the Danishefsky diene 705 provided the diastereomeric dihydropyrones **809** and **810** in a a ratio of 79: 21.273 Note that use of BF<sub>3</sub>·OEt<sub>2</sub> as the Lewis acid in this reaction led to a turnover in stereoselectivity. affording 809 and 810 in a ratio of 25:75. After stereoselective DIBAL reduction of the  $\psi$ -endo isomer (809), acid-catalyzed ring opening of the resulting 811 then supplied the enal 812. Finally, Horner-Emmons reaction of 812 using Still's conditions<sup>91</sup> gave the required (Z,E)- $\alpha,\beta,\gamma,\delta$ -unsaturated ester **789**, together with the corresponding E,E isomer 813 in a ratio of 75:25. Thus, the model  $C_1-C_{11}$  segment 789 was obtained in an overall yield of 7.0% over four steps. Note that the diene-tricarbonyl complex 800 was chosen for the model studies due to its ready availability. Preparation of the C1-C11 fragment 814, having the Z configuration at  $C_{10}$  required for synthesis of macrolactin A, would require hetero-Diels-Alder cycloaddition of the (E,Z)-dienal complex 815. Such a reaction has previously been demonstrated by Donaldson  $et\ al.^{274}$ 

Note also that both **789** and **790** were obtained in racemic form, since their syntheses began with the racemate of the diene-tricarbonyl complex  $(\pm)$ -800.

Scheme 75. Donaldson Macrolactin A  $C_1$ – $C_{11}$  Synthesis<sup>265 a</sup>

Note: Absolute configuration of amphidinolide A is, at present, undetermined.

An enanticontrolled synthesis would be possible by using the resolved<sup>275</sup> form of **800**.

# I. The Amphidinolides

The amphidinolides are a class of cytotoxic polyene macrolides isolated from dinoflagellates of the genus Amphidinium, which are symbionts of Okinawan marine flatworm Amphiscolops sp.276 All of these compounds exhibit significant in vitro activity against murine leukemia cells, and some congeners also display activity toward rabbit skeletal muscle actomyosin ATPase. The gross chemical structures of the various members of the amphidinolide family have been deduced by Kobayashi and co-workers. 276 The same group has proposed relative stereochemical assignments for a few congeners, e.g. amphidinolide A (816 in Scheme 76), on the basis of spectroscopic studies.<sup>277</sup> However, absolute stereochemistry has been determined in only two cases, via synthesis of oxidative degradation fragments.<sup>278</sup> At the present time, no total syntheses of any of the amphidinolides have been reported. However, a  $C_{10}-C_{19}$  segment of amphidinolide A has been prepared by O'Connor and Williard, <sup>279</sup> and Boden and Pattenden have reported the formation of the macrocyclic skeleton of amphidinolide A using a novel palladium-catalyzed macrocyclization reaction.280

# 1. Williard Segment Synthesis<sup>279</sup>

At the time that the synthesis of a  $C_{10}$ – $C_{19}$  segment of amphidinolide A was reported by O'Connor and Williard, 279 neither the relative nor the absolute stereochemistry of the natural product was known. Accordingly, the researchers sought to develop a synthetic strategy that could be used to prepare all of the possible diastereomers (*vide infra*). The  $C_{10}$ – $C_{19}$  segment 817 was obtained via Julia coupling of the  $C_{10}$ – $C_{13}$  and  $C_{14}$ – $C_{19}$  segments 818 and 819, which were each available in either enantiomeric configuration from chiral pool starting materials (Scheme 76).

Construction of the  $C_{14}-C_{19}$  segment **819** began with (R)-methyl 3-hydroxy-2-methylpropionate (ent-199), which supplied the  $C_{18}$  stereogenic center (Scheme 77).<sup>281</sup> Note that the (S) enantiomer (199) is also commercially available. Thus, protection of the hydroxyl of ent-199 and subsequent reduction of the ester gave the alcohol **820**; iodination then

# Scheme 77. Williard Amphidinolide A $C_{10}-C_{19}$ Synthesis<sup>279</sup> a

 $^a$  (a) MOMCl,  $^i Pr_2 NEt;$  (b) LAH; (c) Ph<sub>3</sub>P, I<sub>2</sub>, imidazole; (d)  $^i BuLi;$  9-BBN-OMe; (e) PhSO<sub>2</sub>Me,  $^n BuLi;$  824; (f) 822 + 823, 3 mol % (DPPF)PdCl<sub>2</sub>; (g) MeOH, H<sup>+</sup>, molecular sieves; (h) (MeO)<sub>2</sub>CMe<sub>2</sub>, H<sup>+</sup>, molecular sieves; (i) KOH, MeOH; (j) BH<sub>3</sub>THF; (k) TBSCl, imidazole; (l) 819 (1 equiv),  $^n BuLi;$  818 (0.5 equiv); LDA; 818 (0.5 equiv); (m) NaBH<sub>4</sub>; (n) MsCl, Et<sub>3</sub>N; (o) Na−Hg, Na<sub>2</sub>HPO<sub>4</sub>.

provided **821**. After lithium—halogen exchange, reaction with 9-BBN-OMe afforded the borane **822**. Note that **822** is not available in sufficiently high enantiomeric purity via an alternative route involving asymmetric hydroboration of a methallyl alcohol derivative.<sup>282</sup> Suzuki coupling<sup>283</sup> of **822** with vinyl bromide **823**, derived from 2,3-dibromopropene (**824**),<sup>284</sup> then furnished **819**.

Meanwhile, the C<sub>10</sub>-C<sub>13</sub> segment 818 was obtained from L-tartaric acid (825) according to the procedure of Musich and Rapoport. 285 Thus, bis(esterification) of 825 and subsequent acetonide protection was followed by selective saponification of one of the ester groups to afford 826. Chemoselective reduction of the carboxylic acid group of 826 and protection of the resulting alcohol then gave 818. Julia coupling of ester 818 and sulfone 819 under carefully controlled conditions<sup>286</sup> furnished the  $\beta$ -keto sulfone 827 as a mixture at C14 epimers. Reduction to the corresponding hydroxy sulfone followed by mesylation and reductive elimination then afforded the C<sub>10</sub>-C<sub>19</sub> segment 817 with the required E double bond. Note that the stereochemistry of 817 is in agreement with the relative stereochemical assignment for amphidinolide A subsequently proposed by Kobayashi et al. 277a Note also that, in the synthesis of 817, all three stereogenic centers originated in the chiral pool. The use of D-tartaric acid and (S)-methyl 2-hydroxy-3-methylpropionate, therefore, would allow synthesis of the enantiomer of 817. The absolute configuration

#### Scheme 78a

 $^a$  (a)  $\rm H_2O_2, Na_2CO_3; HCl;$  (b)  $\rm (MeO)_2CMe_2, {\it p-TsOH};$  (c) NaOMe, MeOH; (d) TBSCl, imidazole.

of amphidinolide A is, at present, undetermined. [ $C_{10}-C_{19}$  segment 817: 31% overall yield from *ent*-199; 9 steps longest linear sequence; 15 steps total; 5 steps per stereogenic center.]

The "chiron strategy" adopted by O'Connor and Williard was sufficiently flexible to be able to supply all the diastereomeric possibilities for the  $C_{10}-C_{19}$  segment. Thus, if it had proved necessary, upon determination of the relative stereochemistry of amphidinolide A, to use the  $C_{10}-C_{13}$  segment 828 (Scheme 78), then the depicted enantiomer of this intermediate was available by synthesis from Disoascorbic acid (829) via lactone 830.<sup>287</sup> Either enantiomer of 829 was also available by resolution of racemic DL-erythronic lactone 830.<sup>288</sup>

## 2. Pattenden Macrocycle Synthesis<sup>280</sup>

Boden and Pattenden have prepared the macrocycle 831, a model for the ring skeleton of amphidinolide A, from the vinyl stannane-allylic chloride 832 using a novel intramolecular Stille coupling procedure which results in the formation of 1,4-dienes (Scheme 79). 280 Thus, ozonolysis of cyclononene (833) in acidic methanol, followed by reductive workup, provided the aldehyde-acetal 834. After Horner-Emmons olefination, subsequent reduction followed by acetal hydrolysis supplied the aldehyde 835. A second Horner-Emmons olefination and subsequent saponification then afforded the acid 836. Meanwhile, alcohol 837 was prepared by stannyl cupration of allene (838) followed by trapping with ethylene oxide according to the procedure of Fleming and Pulido.<sup>289</sup> After transformation to iodide 839, coupling with the carboxylate anion derived from 836 furnished **840**. Introduction of chloride at  $C_{15}$ , via the mesylate derived from 840, gave the cyclization precursor **832**. Finally, treatment of **832** with Pd(0) in the presence of triphenylarsine<sup>290</sup> led to smooth cyclization to afford the 20-membered macrocycle 831 with the required E configuration at  $C_{13}$  (38% cyclization yield), together with a small amount of the corresponding 13Z isomer, and a trace of the 18membered macrocycle resulting from allylic isomerization of the starting material. Note that although the model 831 contains all the endocyclic double bonds found in amphidinolide, it lacks several other structural features present in the natural product: namely the  $C_{19}$  side chain, the exomethylene groups at  $C_7$  and  $C_{10}$ , and the four hydroxyls at  $C_8$ ,  $C_9$ ,  $C_{11}$ ,

# Scheme 79. Pattenden Amphidinolide A Macrocyclization Study<sup>280 a</sup>

 $^a$  (a) O<sub>3</sub>, MeOH, p-TsOH; NaHCO<sub>3</sub>, Ph<sub>3</sub>P; (b) (EtO)<sub>2</sub>-P(=O)CH<sub>2</sub>CO<sub>2</sub>Et, LDA; (c) DIBAL: (d) p-TsOH, H<sub>2</sub>O; (e) (E)-(MeO)<sub>2</sub>P(=O)CH<sub>2</sub>CH=CHCO<sub>2</sub>Me, LDA; **835**; (f) NaOH; (g) (^nBu<sub>3</sub>Sn)<sub>2</sub>CuLi; ethylene oxide; (h) I<sub>2</sub>, PPh<sub>3</sub>, imidazole; (i) **836** + **839**, DBU; (j) MsCl, Et<sub>3</sub>N, LiCl; (k) Pd<sub>2</sub>dba<sub>3</sub>, Ph<sub>3</sub>As,  $\Delta$ .

and  $C_{12}$ . All of these features might be expected to assist the cyclization to form the real macrocycle, by inducing a favorable conformation in the cyclization precursor. Accordingly, cyclization in the real system could proceed in a yield higher than that observed for the model **831**. The Pattenden cyclization methodology should also be applicable to the synthesis of other polyene macrolides possessing 1,4-diene functionality.

## J. Tedanolide

Tedanolide (**841** in Scheme 80) is a potent cytotoxic macrolide, isolated from the Caribbean sponge  $Tedania\ ignis$ , which inhibits KB human carcinoma and PS lymphocytic leukemia  $in\ vitro.^{291}$  At the time of writing, no total synthesis of tedanolide has been reported. However, Yonemitsu has described the macrolactonization of the seco-acid derivative **842**, to form the advanced intermediate **843**. The protecting group arrangement of **842** was selected as a result of molecular modeling studies ( $vide\ infra$ ), and **842** was synthesized from (R)- and (S)-2-hydroxy-3-methylpropionic acids via four segments:  $C_1-C_7$ ,  $C_8-C_{11}$ ,  $C_{13}-C_{17}$ , and  $C_{18}-C_{21}$ . At the time of writing, no further details have been published.

# 1. Yonemitsu Macrolide Synthesis 76e

To avoid the possibility of decomposition of the flexible acyclic precursors occurring as a result of retro-aldol cleavage reactions, Yonemitsu opted to replace some of the aldol relationships in tedanolide with protected 1,3-diols. Selective deprotection and

# Scheme 80. Yonemitsu Tedanolide Macrolide Synthesis $^{76e}$ $^a$

<sup>a</sup> (a) 2,4,6-Cl<sub>3</sub>C<sub>6</sub>H<sub>2</sub>COCl, Et<sub>3</sub>N; DMAP.

oxidation would then be required, after macrolactonization, in order to obtain the natural product.

In determining the arrangement of protecting groups, Yonemitsu identified two criteria which must be met: firstly, that the conformation of the secoacid should resemble as closely as possible the conformation of the corresponding macrolide, in order for macrolactonization to be efficient; and, secondly, that the conformation of the macrolactonization product must be similar to the conformation of tedanolide itself, in order for the necessary selective oxidation to be favorable. Computer-aided conformational analysis indicated that these requirements were likely to be satisfied by the seco-acid 842 and the corresponding macrolide 843, in which the C<sub>5</sub> and  $C_{15}$  carbonyls were reduced to hydroxy groups and protected as m,p-dimethoxybenzylidene acetals. Thus, the calculated lowest energy conformation of 844 (a model for 842) was very similar to that computed for **845** (a model for **843**). In turn, the calculated lowest energy conformation of 845 closely resembled the conformation of the lactone portion of tedanolide (840) as revealed by X-ray analysis.<sup>291</sup>

In practice, seco-acid **842** cyclized smoothly, without the need for high dilution, using the Yonemitsu modification<sup>184</sup> of the Yamaguchi procedure,<sup>47</sup> to afford an 89% yield of the corresponding macrolide **843**. The deprotection and selective oxidation of **843** to provide tedanolide has not yet been reported.

### K. The Latrunculins

Latrunculins A and B (846 and 847 in Figure 8), originally isolated from the Red Sea sponge  $Latrunculia\ magnifica\ (Keller)\ in\ 1980,^{292a-c,293}\ were\ the\ first$ 

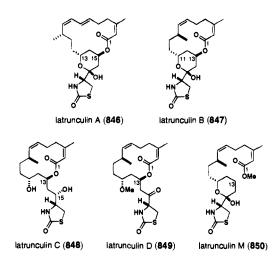


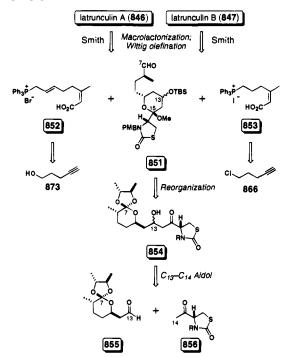
Figure 8. Structures of the latrunculins.

macrolides to be discovered from a truly marine source and the first natural products to embody the 2-thiazolidinone moiety.<sup>294</sup> Other congeners have since been found, such as latrunculins C (848), 292d,295 D (849), 292d and M (850). 292g Biological interest in the latrunculins stems from their powerful inhibition of the polymerization of the cytoskeletal protein actin, 2925, 296a and on their ability to reversibly disrupt microfilament organization. 292b,h,296b The total synthesis of latrunculin B was reported by Smith and co-workers in 1986; $^{297a,c}$  and in 1990, Smith et al. $^{297b,c}$ and White and Kawasaki<sup>298</sup> independently completed total syntheses of latrunculin A. Kashman and coworkers have reported chemical modifications of the natural products<sup>292c-g</sup> and the de novo synthesis<sup>292d,e</sup> of model latrunculin tetrahydropyran ring systems.

# 1. Smith Total Syntheses<sup>297</sup>

The syntheses of latrunculin A (846) and latrunculin B (847) by Smith and co-workers employed the same advanced intermediate 851 (Scheme 81). Wit-

#### Scheme 81



## Scheme 82. Smith Latrunculin B Synthesis<sup>297a,c</sup> a

 $^a$  (a) m-CPBA, NaHCO3; (b) LDA; MeI; (c) (+)-(R, R)-2,3-butanediol, CSA; (d) O3; PPh3; (e) COCl2; (f) PMBBr, NaH; (g) 1 N KOH; H3O+; (h) NaH, (COCl)2; MeMgBr; (i) 856,  $^n\mathrm{Bu}_2\mathrm{BOTf}$ ,  $^i\mathrm{Pr}_2\mathrm{NE}$ ; 855; (j) p-TsOH, MeOH; (k) 2 N HCl; (l) CSA, MeOH; (m) TBSCl, imidazole; (n) DIBAL; (o) 852 or 853, KHMDS; 851; (p) HF-py; (q) PPh3, DEAD; (r) 0.25 M (NH4)2Ce(NO3)6, MeCN; (s) AcOH, H2O; (t) (NH4)2Ce(NO3)6, MeCN, H2O; (u) NaBH4; (v) MeOH, BF3·OEt2; (w) Et3SiH, BF3·OEt2; (x) AcOH; (y) CH2N2; (a')  $^n\mathrm{BuLi}$ , ClCO2Me; (b') Me2CuLi; (c') 1 N LiOH; (d') NaI; (e') PPh3; (f') Swern oxidation; (g') (EtO)2P(=O)CH2CO2Et, NaH; (h') DIBAL; (i') DHP, PPTS; (j')  $^n\mathrm{BuLi}$ , ClCO2Me; (k') Me2CuLi; (l') Amberlyst-15; (m') LiOH; (n') NBS, Me2S; (o') PPh3.

tig reaction of **851** with either **852** or **853**, followed by macrolactonization, led to latrunculins A and B, respectively. Compound **851** was obtained by a novel acid-catalyzed reorganization of ortho ester **854**, which was in turn derived from aldol union of aldehyde **855** and methyl ketone **856**. Since aldol reactions of methyl ketones are often poorly stereoselective, <sup>299</sup> the synthesis was designed to accommodate both configurations at  $C_{13}$  in **854** (latrunculin B numbering). Thus, the  $\alpha$  epimer of **854** required macrocyclization with inversion of configuration at  $C_{15}$  via the Mitsunobu reaction; <sup>85</sup> the  $\beta$  epimer would

be lactonized with retention via carboxyl activation. 300

a. Smith Latrunculin B Total Synthesis.  $^{297a,c}$  Aldehyde **855** was prepared in four steps from racemic **857** (Scheme 82). Thus, Baeyer–Villiger oxidation followed by  $\alpha$ -methylation gave **858** as a 1:1 mixture of diastereomers; ortho ester formation with (+)-(R,R)-2,3-butanediol then effected both resolution and equilibration, increasing the trans/cis ratio to 6:1.  $^{301}$  HPLC separation of the diastereomers provided isomer (-)-**859**, which was ozonolyzed to provide the  $C_7$ - $C_{13}$  aldehyde **855** in enantiopure form. Ketone

856 was derived from L-cysteine ethyl ester (860). Thus, formation of the thiazolidinone, PMB protection of the amide nitrogen, and ester hydrolysis gave acid 861. Reaction of the derived acid chloride with MeMgBr then afforded methyl ketone 856.

The  $C_{13}$ – $C_{14}$  bond was constructed by aldol coupling of the dibutylboron enolate<sup>302</sup> of ketone 856 with aldehyde **855**, giving  $\beta$ -hydroxy ketone **854** as an inseparable 4:1 ( $\alpha/\beta$ ) mixture of  $C_{13}$  epimers. Note that use of the lithium or zinc enolates of 856 gave similar diastereoselectivity but lower yields. Rearrangement of ortho ester 854 catalyzed by p-TsOH in methanol furnished the methyl acetals 862 as a 4:1 mixture of  $C_{13}$  epimers. However, use of aqueous HCl effected  $C_{13}$  equilibration as well, supplying hemiacetal 863 as a 12:1 mixture of  $\alpha$  and  $\beta$ epimers.303,304 Smith has postulated that equilibration occurs via equatorial addition of the C<sub>13</sub> hydroxyl to oxonium ion 864. The C<sub>15</sub> hemiacetal stereochemistry is governed by the anomeric effect. Methanolysis of 863 $\alpha$  followed by TBS protection of the  $C_{13}$ hydroxyl gave 865 which was reduced with DIBAL to afford the key C<sub>9</sub> aldehyde **851**.

The phosphonium salt 853, required for the  $C_1$ -C<sub>6</sub> segment of latrunculin B, was prepared in five steps from 5-chloro-1-pentyne (866). Thus, carbalkoxylation of 866 gave 867, which underwent stereoselective<sup>107a</sup> carbocupration to afford 868. Hydrolysis at C<sub>1</sub> and phosphorus introduction at C<sub>6</sub> then supplied 853. A completely Z-selective Wittig coupling of aldehyde 851 with the dianion derived from **853**, followed by cleavage of the  $C_{13}$  silyl ether, furnished the seco-acid 869. Macrocyclization of 869 using the Mitsunobu protocol,85 with inversion of configuration at  $C_{13}$ , then afforded the lactone 870. Deprotection of the thiazolidinone nitrogen using aqueous ceric ammonium nitrate (CAN)305 gave 871 and subsequent acetal hydrolysis provided latrunculin B (847). The natural product could also be directly obtained from 870 by treatment with a more concentrated solution of CAN. Note that the choice of the *p*-methoxybenzyl group for the thiazolidinone nitrogen was critical: both the N-benzyl and N-(m,pdimethoxybenzyl) derivatives of latrunculin B were synthesized, but the protecting groups could not be removed.

Kashman and co-workers have converted natural latrunculin B (847) into latrunculin C (848) and its  $C_{15}$  epimer by reduction with NaBH<sub>4</sub>.<sup>292d</sup> They have also prepared latrunculin M (850) from 847 via acetalization at  $C_{15}$  with methanol and reductive opening of the macrolide with  $Et_3SiH/BF_3\cdot OEt_2$  to give 872, followed by deacetalization at  $C_{15}$  and, finally, methyl ester formation at  $C_{1}$ .<sup>292g</sup> The Smith total synthesis of latrunculin B thus achieves formal syntheses of two additional members of the latrunculin family. [Latrunculin B (847): 2.0% overall yield from ( $\pm$ )-857; 14 steps longest linear sequence; 23 steps total;  $\sim$ 4–5 steps per stereogenic center.]

b. Smith Latrunculin A Total Synthesis.  $^{297b,c}$  The phosphonium salt **852** (Scheme 82), required for the  $C_1-C_8$  segment of latrunculin A, was prepared in 10 steps from 5-hydroxy-1-pentyne (873). Thus, Swern oxidation and Horner-Emmons olefination to give 874 was followed by 1,2-reduction and hydroxyl

Scheme 83. Smith Latrunculin A Synthesis<sup>297b,c</sup> a

 $^a$  (a) LAH; (b)  $^tBuLi;$  O2; (c) PCC, Al2O3; (d) TMSCH2CH2OCOCl,  $^iPr_2NEt,$  DMAP; (e) **852**, KHMDS; **878**: (f) HF py; (g) PPh3, DEAD; (h) TBAF; (i) 3 N HCl.

protection to afford 875. Transformation of the terminal alkyne to the  $\alpha,\beta$ -unsaturated ester **876** was accomplished as for the earlier conversion of 866 → **868**. Finally, hydrolysis at  $C_1$  and phosphorus introduction at C<sub>8</sub> then supplied 852. In an analogous manner to the synthesis of latrunculin B, Wittig coupling of aldehyde 851 with the dianion derived from **852**, followed by cleavage of the  $C_{15}$  silyl ether and macrocyclization, afforded the lactone 877. Unfortunately, deprotection of the thiazolidinone nitrogen in 877 proved to be impossible due to interference from the sensitive diene moiety (not present in the latrunculin B analogue 870). A change of protecting group earlier in the synthesis was required. Accordingly, 865 was converted in four steps, including removal of the PMB group using the Williams procedure (tert-butyllithium, O<sub>2</sub>),<sup>306</sup> into the aldehyde 878 in which the thiazolidinone nitrogen atom was protected as its  $[\beta$ -(trimethylsilyl)ethoxy]carbamate (Scheme 83). Wittig olefination of 852 and 878 followed by cleavage of the C<sub>15</sub> silvl ether gave secoacid 879, and macrocyclization under Mitsunobu conditions<sup>85</sup> then provided macrolide 880. The TEOC nitrogen-protecting group was removed using fluoride ion, and then final hydrolysis of the C<sub>17</sub> methyl acetal of 881 gave latrunculin A (846).

Thus, in this synthesis, the stereogenic center at  $C_{18}$  of latrunculin A was obtained from the chiral pool, the  $C_{10}$  and  $C_{13}$  stereogenic centers were constructed by using a combination of substrate control and resolution (( $\pm$ )-856  $\rightarrow$  ( $\pm$ )-858  $\rightarrow$  (-)-859), and the stereogenic centers at  $C_{15}$  and  $C_{17}$  were set up using substrate-controlled reactions (854  $\rightarrow$  863, and subsequent inversion at  $C_{15}$  during 879  $\rightarrow$  880). [Latrunculin A (846): 0.6% overall yield from ( $\pm$ )-857; 17 steps longest linear sequence; 31 steps total;  $\sim$ 6 steps per stereogenic center.]

# 2. White Latrunculin A Total Synthesis<sup>298</sup>

The synthesis of latrunculin A (846) by White and Kawasaki was designed to illustrate new methodology<sup>307</sup> for the synthesis of (E,Z)-1,3-dienes (*vide infra*). Accordingly, the target was divided into the

three principal segments 882–884 depicted in Scheme 84. A Wittig olefination was used for the  $C_8-C_9$  bond construction, and an aldol reaction was employed for the  $C_{15}-C_{16}$  bond construction. After acetalization at  $C_{17}$ , macrocyclization was accomplished by means of the Mitsunobu reaction. Note that the contemporaneous Smith synthesis of latrunculin  $A^{297b,c}$  employed the same principal disconnections, but used a different order of segment assembly.

The  $C_9$ – $C_{15}$  aldehyde segment representing synthon **883** was assembled from two four-carbon segments, **885** and **886** (Scheme 85). Epoxide **885** was prepared from (S)-(-)-malic acid (**119**). An initial six-step sequence of protecting group exchanges and adjustment of oxidation level gave **887**, and acetonide deprotection and selective primary tosylation of the derived diol then furnished **888**. Treatment of **888** with base formed the epoxide and simultaneously cleaved the  $C_{15}$  acetate; finally, TBS protection at  $C_{15}$  provided the desired **885**. Meanwhile, sulfone **886** 

was obtained in five routine steps from methyl (R)-3-hydroxy-2-methylpropionate (ent-199). Coupling of the lithio anion of 886 with 885 followed by reductive removal of the sulfone group provided the  $C_9$ - $C_{15}$  segment 889. After protection of the  $C_{13}$  alcohol, deprotection at  $C_9$  and Swern oxidation<sup>38</sup> then gave the key aldehyde 890.

The White methodology for construction of (E,Z)-1,3-dienes involves addition of an enolate dianion to a dienylphosphonium salt followed by an in situ Z-selective Wittig reaction<sup>309</sup> of the derived E-ylide with an aldehyde.<sup>307</sup> Thus, the dilithio anion **891** of β-keto ester 892 was alkylated with diene 893.310 which was in turn obtained via deprotonation of the phosphonium bromide 894,311 to give the  $C_1-C_8$  (E)ylide 882. Reaction of 882 with aldehyde 890 then afforded the  $C_1-C_{15}$  segment (E,Z)-895, along with a trace of the (E,E)-diene. Stereoselective formation of the (E)-enolate of  $\beta$ -keto ester 895<sup>313</sup> and trapping with diethylphosphorochloridate gave the (E)-enol phosphate 896, which reacted with a methylmagnesiocuprate<sup>314</sup> with retention of configuration at C<sub>3</sub> to give the ester 897. Selective silyl ether cleavage at C<sub>15</sub>, followed by Swern oxidation, <sup>38</sup> then furnished aldehyde 898.

Methyl ketone **884** was obtained in three steps from L-cysteine methyl ester (**899**), involving initial thiazolidinone formation by reaction with CO and O<sub>2</sub> in the presence of selenium. Aldol reaction of the mixed lithio-cerio dianion of ketone **884** with alde-

Scheme 85. White Latrunculin A Synthesis<sup>298</sup> a

 $^a$  (a) MeOH, AcCl; (b) DHP, H+; (c) LAH; (d) MeOH, H+; (e) Me<sub>2</sub>CO, H+; (f) Ac<sub>2</sub>O, py; (g) AcOH, H<sub>2</sub>O; (h) p-TsCl, py; (i) K<sub>2</sub>CO<sub>3</sub>, MeOH; (j) TBSCl, imidazole; (k) Cl<sub>3</sub>CC(=NH)OBn, TfOH; (l) LAH; (m) p-TsCl, py; (n) NaI; (o) PhSO<sub>2</sub>Na; (p) 886,  $^n$ BuLi; 885; (q) Na(Hg); (r) SEMCl,  $^i$ Pr<sub>2</sub>NEt; (s) H<sub>2</sub>, 10% Pd-C; (t) Swern oxidation; (u) LDA; (v) (EtO)<sub>2</sub>POCl,  $^i$ Pr<sub>2</sub>NEt, DMAP; HMPA; (w) MeCu, MeMgCl; (x) MeOH, H+; (y) Swern oxidation; (z) 884, LDA; CeCl<sub>3</sub>; 898; (a') HF; (b') MeOH, H+; (c') TBAF; (d') Ph<sub>3</sub>P, DEAD; (e') AcOH, H<sub>2</sub>O; (f') CO, O<sub>2</sub>, Se; (g') AcOH, HCl; (h') MeLi, MeMgCl.

hyde 898, without protection of the nitrogen atom in **884** (vide infra), gave  $C_1-C_{19}$  segment **900** as an inseparable 1:1 mixture of C<sub>15</sub> epimers. Selective cleavage of the C<sub>13</sub> SEM ether by acidic hydrolysis then led to a spontaneous ring closure to form the hemiacetal at  $C_{17}$ , and subsequent treatment with acidic methanol furnished the methyl acetals 901 and 15-epi-901. After separation, the  $15\alpha$  alcohol (901) was taken on to latrunculin A (846). Thus, cleavage of the (trimethylsilyl)ethyl ester from 901 gave secoacid 902. Cyclization under Mitsunobu conditions<sup>85</sup> occurred with inversion at C<sub>15</sub>, and subsequent hydrolysis of the C<sub>17</sub> methyl acetal then gave the natural product. In an analogous manner the  $15\beta$ alcohol (15-epi-901) was transformed into 15-epilatrunculin A. Note that, in theory, 15-epi-901 could also supply the natural product, i.e. latrunculin A, by means of carboxyl-activated macrolactonization<sup>300</sup> which would be expected to occur with retention of configuration at  $C_{15}$ . This avenue was apparently not explored by White and Kawasaki.

Thus, in this synthesis, three of the five stereogenic centers in latrunculin A were obtained from the chiral pool ( $C_{10}$ ,  $C_{13}$ , and  $C_{18}$ ), the  $C_{17}$  stereogenic center was constructed using a substrate-controlled reaction ( $900 \rightarrow 901$ ), and the  $C_{15}$  stereogenic center was not controlled (901 and its  $C_{15}$  epimer were separated). [Latrunculin A (846): 0.3% overall yield from ( $\pm$ )-857; 26 steps longest linear sequence; 35 steps total; 7 steps per stereogenic center.]

Finally, note that, as in the Smith synthesis of latrunculin A, the choice of group used to protect the thiazolidinone nitrogen atom proved to be critical. White and Kawasaki found it advantageous to use no protecting group at all. Use of the MOM protecting group, for instance, on thiazolidinone 884 led, after aldol coupling and hemiacetal formation at  $C_{17}$ , to participation of the formaldehyde unit thus forming the  $N_0$ -acetal 903. This acetal proved to be extremely resistant to hydrolysis and 903 could not be converted to latrunculin A. Other protecting groups on the nitrogen atom of 884 proved to be difficult to remove after lactonization. The identification of suitable protecting groups thus continues to be a nontrivial issue when contemplating the synthesis of complex polyfunctional natural products.

## 3. Kashman Model Studies on Latrunculin Tetrahydropyran Ring<sup>292d,e</sup>

Starting from L-cysteine, Kashman et al. have synthesized some model 2-thiazolidinone-tetrahydropyran ring systems of the latrunculins (Scheme 86). Reaction of L-cysteine ethyl ester (860) with phosgene afforded the thiazolidinone, 316 and then protection of the nitrogen atom and transformation of the  $C_{15}$  ester to the acyl chloride gave **904**. Palladium-catalyzed coupling<sup>317</sup> of **904** with an alkynyl stannane supplied the tetrahydropyran ring precursor 905. After cleavage of the C<sub>11</sub> TBS ether, to afford 906, partial hydrogenation over Lindlar's catalyst and in situ hemiacetal formation gave 907 (as a 1:1 mixture of epimers at C<sub>15</sub>, together with 5% of the open  $cis \delta$ -hydroxy- $\alpha,\beta$ -unsaturated ketone **908**). Hetero-Michael addition to  $C_{13}$  of **908** (in the case  $R \neq H$ ) was suggested as a possible route to latrunculin B. Partial hydrogenation of 906 over Pd/ BaSO<sub>4</sub> furnished the *trans*  $\delta$ -hydroxy- $\alpha,\beta$ -unsaturated ketone 909, which could also be obtained by basic equilibration of **907**.

Michael addition of methanol, in the presence of K<sub>2</sub>CO<sub>3</sub>, to either **907** or **909** followed by acetalization of the resulting hemiacetal by addition of BF<sub>3</sub>·OEt<sub>2</sub> to the methanolic solution led to two out of the four possible C<sub>13</sub>,C<sub>15</sub> dimethoxy derivatives. Compounds 910 and 911 were obtained in a ratio of 3:1. Acid treatment of either 910 or 911 gave the corresponding hemiacetal: thus 910 provided 912 and 913 as a 3:2 equilibrium mixture. Methanol addition to 909 without acetalization also provided 912 and 913 as an equilibrium mixture. Note that none of the tetrahydropyrans 910-913, which lack an alkyl substituent at  $C_{11}$ , possess the correct configurations for  $C_{13}$  and  $C_{15}$  of the latrunculins as depicted in **914**. Note also that NMR studies on an analogue 915 of **914** ( $R = (CH_2)_2CH(CH_3)CH_2OH$ ), obtained by reduc-

Scheme 86. Kashman Latrunculin B THP Ring Model Studies<sup>292d,e</sup> a

a (a) COCl<sub>2</sub>; (b) BnBr, NaH; (c) H<sup>+</sup>; SOCl<sub>2</sub>; (d) TBSOCH<sub>2</sub>CH<sub>2</sub>C≡CSn<sup>n</sup>Bu<sub>3</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>; (e) H<sup>+</sup>; (f) H<sub>2</sub>, Lindlar catalyst; (g) H<sub>2</sub>, Pd−BaSO<sub>4</sub>, py; (h) MeOH, py; (i) MeOH, K<sub>2</sub>CO<sub>3</sub>; BF<sub>3</sub>·OEt<sub>2</sub>; (j) H<sup>+</sup>, SiO<sub>2</sub>; (k) MeOH, K<sub>2</sub>CO<sub>3</sub>; (l) MeOH, BF<sub>3</sub>·OEt<sub>2</sub>; (m) O<sub>3</sub>; NaBH<sub>4</sub>; (n) H<sup>+</sup>.

tive ozonolysis of latrunculin B, showed that it exists in the single cyclic hemiacetal conformation illustrated. Thus the studies revealed that the presence of an alkyl substituent at  $C_{11}$  (i.e.  $R \neq H$  for 908  $\rightarrow$  914) would be essential in order for the correct latrunculin configuration at  $C_{13}$  to be generated by hetero-Michael addition.

## L. The Octalactins

Octalactins A and B (916 and 917 in Scheme 87) are two eight-membered lactones which have been isolated from a marine-derived actinomycete found living on the surface of the Sea of Cortez gorgonian octocoral Pacifigorgia sp. 318 Whereas octalactin A exhibits strong cytotoxicity against certain melanoma and human colon tumor cell lines, octalactin B is completely inactive in the same assays. The octalactins are not, in the strictest sense, macrolides, but are included in this review owing to the fact that octalactin A represents one of the comparatively few examples of medium-ring marine lactones which displays significant biological activity.319 The first total syntheses of both octalactin A and B were accomplished by Buszek et al. in 1994;320 this work serving to establish the absolute configuration of the natural products. Total syntheses were also reported by Williams and Clardy later in the same year. 321

# 1. Buszek Total Syntheses 320

Buszek et al. synthesized octalactins A and B from a common intermediate (918 in Scheme 87), which was obtained by Ni(II)/Cr(II)-mediated coupling<sup>78</sup> of the  $C_1$ - $C_9$  and  $C_{10}$ - $C_{15}$  segments 919 and 920. The key step in the synthetic approach to 919 was intended to be lactonization of the unsaturated secoacid 921, which was expected to be facile owing to

### Scheme 87

the conformational rigidity imparted by the double bond, followed by hydrogenation of the *cis* olefin.<sup>322</sup> However, in the actual synthesis of **919**, an unprecedented lactonization of the corresponding saturated seco-acid **922** was found to be preferable.

Construction of the  $C_1-C_9$  segment **928** was based on coupling of the  $C_1-C_6$  and  $C_7-C_9$  segments **923** and 924 (Scheme 88). Compounds 923 and 924 were obtained from, respectively, the S and R enantiomers of methyl 3-hydroxy-2-methylpropionate, i.e. 199 and ent-199. Thus, 199 was converted to alkene 925 in five steps, with unspecified diastereoselectivity for the formation of the stereogenic center at  $C_3$ . Hydroboration of 925 followed by protection of the resulting primary alcohol and subsequent cleavage of the C<sub>5</sub> silyl ether furnished alcohol **926**. After oxidation of 926 to the C<sub>5</sub> aldehyde, condensation with Seyferths' reagent ((MeO)<sub>2</sub>P(=O)CHN<sub>2</sub>)<sup>241a</sup> gave alkyne 927, and iodination then supplied 923. Ni-(II)/Cr(II)-mediated coupling<sup>78</sup> of iodide **923** with aldehyde 924, obtained in three steps from ent-199, occurred with negligible stereoselectivity, affording the  $C_1-C_9$  segment **928** as an inseparable mixture of C<sub>7</sub> epimers. Hydrogenation using Lindlar's catalyst, followed by acetylation of the  $C_7$  hydroxyl and deprotection at C<sub>1</sub> then delivered alkene **929** with the correct configuration at  $C_7$ , together with its  $C_7$ epimer 930. After separation of 929 and 930, twostage oxidation at C<sub>1</sub> of **929** followed by deacetylation then supplied the unsaturated seco-acid 921. Similarly, 931 was obtained from 930. The undesired epimer 931 could be recycled via oxidation, to give **932**, followed by reduction, which gave a 2:1 ratio of 921 and 931. Cyclization of 921 to provide the unsaturated lactone 933 was accomplished by means of the Corey-Nicolaou "double-activation" protocol (formation of the 2-pyridyl thioester of 921 followed by AgBF<sub>4</sub>-mediated cyclization). 322,323 Unfortunately, all attempts to obtain the saturated lactone 934 by reduction of the double bond of 933 proved unsuccessful.

As an alternative, Buszek et al. investigated lactonization of the saturated seco-acid 922, which was obtained by hydrogenation of **921** prior to cyclization. By using the Corey-Nicolaou procedure, cyclization of 922 was in fact straightforward: the corresponding lactone **934** was obtained in a yield (73%) comparable to that achieved for the unsaturated analogue 933. Note that this represents the first example of a highyielding synthesis of an eight-membered lactone from a saturated seco-acid precursor. The use of an olefin moiety to provide extra conformational rigidity in the seco-acid appears not to be necessary in this particular case: the combined influences of the stereochemical arrangement of 922 and the sterically demanding protecting groups apparently lead to a preferred conformation in the presumed transition state<sup>324</sup> that sufficiently facilitates ring closure. Note that other diastereomeric acyclic precursors exhibited varying propensity for cyclization: thus, whereas **934** and its  $C_7$  epimer were both formed in 96 h, the  $C_3$ ,  $C_7$  bis-(epimer) was obtained after only 50 h and the C<sub>3</sub> epimer required 2 weeks.

Desilylation of 934 followed by oxidation gave the  $C_1-C_9$  aldehyde segment 919. Meanwhile, the  $C_{10}-$ 

## Scheme 88. Buszek Octalactin A and B Syntheses 320 a

a (a) DHP, p-TsOH; (b) DIBAL; H<sub>2</sub>C=CHMgBr; separate C<sub>3</sub> epimers; (c) PMBCl, KH; (d) PPTS, EtOH; (e) TBDPSCl, imidazole; (f) 9-BBN; H<sub>2</sub>O<sub>2</sub>, NaOH; (g) MMTrCl, Et<sub>3</sub>N; (h) TBAF; (i) Dess-Martin periodinane; (j) (MeO)<sub>2</sub>P(=O)CHN<sub>2</sub>, 'BuOK; (k) I<sub>2</sub>, morpholine; (l) TBDPSCl, imidazole; (m) DIBAL; (n) Dess-Martin periodinane; (o) **923** + **924**, NiCl<sub>2</sub> (1.0%)−CrCl<sub>2</sub>; (p) H<sub>2</sub>, Lindlar catalyst; (q) Ac<sub>2</sub>O, DMAP, py; (r) PPTS, MeOH; (s) Dess-Martin periodinane; (t) NaClO<sub>2</sub>, 2-methyl-2-butene, 'BuOH; (u) K<sub>2</sub>CO<sub>3</sub>, MeOH; (v) Dess-Martin periodinane; (w) L-selectride, CeCl<sub>3</sub>; (x) 2,2'-pyridine disulfide, PPh<sub>3</sub>; AgBF<sub>4</sub>; (y) H<sub>2</sub>, 10% Pd−C; (z) 2,2'-dipyridyl disulfide, PPh<sub>3</sub>; AgBF<sub>4</sub>; (a') TBAF, AcOH; (b') Dess-Martin periodinane; (c') NaNO<sub>2</sub>, HCl; (d') LAH; (e') KOH; (f') TMS−C≡C−Li, BF<sub>3</sub>·OEt<sub>2</sub>; (g') TBSCl, imidazole; (h') 1 N NaOH; (i') <sup>n</sup>BuLi; MeI; (j') Cp<sub>2</sub>ZrClH; I<sub>2</sub>; (k') **919** + **920**, NiCl<sub>2</sub> (0.1%)−CrCl<sub>2</sub>; (l') Dess-Martin periodinane; (m') HF; (n') DDQ, H<sub>2</sub>O; (o') VO(acac)<sub>2</sub>, 'BuOOH; (p') m-CPBA; (q') Mo(CO)<sub>6</sub>, 'BuOOH.

C<sub>15</sub> iodide segment **920** was prepared from L-valine (**935**). Thus, **935** was converted via **936** into epoxide **937**, according to the procedure of Koppenhoeffer and Schurig. Yamaguchi coupling of **937** with lithium (trimethylsilyl)acetylide followed by hydroxyl protection then gave **938**. Hydrolytic removal of the

trimethylsilyl group followed by methylation supplied alkyne **939**, which was regioselectively hydrozirconated and iodinated<sup>326</sup> to generate **920**. Ni(II)/Cr(II)-mediated coupling<sup>78</sup> of **919** and **920** then afforded a 1.5:1 ratio of  $C_9$  epimers **918** and **940**. Oxidation of both **918** and **940** followed by deprotection provided

octalactin B (917). Synthesis of the bioactive congener octalatin A required stereoselective epoxidation of the  $C_{10}$ - $C_{11}$  alkene. Thus, vanadium-mediated epoxidation88 of the major coupling adduct 918 furnished 941 as a single isomer with the required epoxide configuration. Oxidation of 941 then gave 942 and deprotection afforded octalactin A (916). The minor coupling adduct 940 could also be transformed to octalactin A. Whereas epoxidation of 940 with m-CPBA led to the undesired epoxide stereochemistry (943), reaction with molybdenum-hydroperoxide88a delivered a 1:1 mixture of epoxides 943 and 944. Oxidation of 944 followed by protecting group removal then supplied the bioactive natural product. In the synthesis of octalactin A by Buszek et al., the stereogenic centers at C4 and C8 originate from the chiral pool (from 199 and ent-199, respectively), and the remaining five stereogenic centers are introduced by substrate-controlled reactions. Unfortunately, the stereochemical efficiency of the route is at present limited by the low stereoselectivity encountered in two of these reactions: namely the two instances of Ni(II)/Cr(II)-mediated coupling reactions: **923** + **924**  $\rightarrow$  928 and 919 + 920  $\rightarrow$  918. [Octalactin B (917): 4.8% overall yield from 925 with no recycling; 26 steps longest linear sequence; 37 steps total; ~7 steps per stereogenic center; octalactin A (916): 3.2% overall yield from 925 with no recycling; 27 steps longest linear sequence; 38 steps total; ~5 steps per stereogenic center.

# 2. Clardy Total Syntheses321

The absolute configurations of the octalactins were unknown when Williams and Clardy embarked upon their synthetic studies. Upon completion of the total syntheses, it was apparent that they had arbitrarily prepared the unnatural antipodes, ent-916 and ent-**917**. The latter stages of the synthesis were similar to the route of Buszek in that the  $C_1-C_{15}$  segment 945 was prepared via coupling of C<sub>1</sub>-C<sub>9</sub> segment 946 and  $C_{10}$ – $C_{15}$  segment **947** (Scheme 89, cf. **918**  $\rightarrow$  **919** + **920** in Scheme 87). However, Williams and Clardy designed a route to 946 that was entirely different from the Buszek strategy, whereby formation of the eight-membered lactone was accomplished by a Baeyer-Villiger oxidation of ketone 948. Compound **949**, the precursor to ketone **948**, was itself obtained via double Baeyer-Villiger oxidation of the key diketone intermediate 950.

The synthesis of **950** began with (R)-citronellic acid (**951**), which supplied the  $C_3-C_7$  portion, bearing a stereogenic center at  $C_4$ , along with  $C_9$  (Scheme 90). Thus, esterification of **951** and subsequent ozonolysis gave the  $C_7$  aldehyde; methylenation according to the procedure of Osima (Zn,  $CH_2I_2$ ,  $AlMe_3$ )<sup>327</sup> then supplied the alkene **952**. After saponification of the ester and formation of the corresponding acid chloride,  $SnCl_4$ -induced cyclization afforded a mixture of  $\beta$ -chlorocycloheptanones (**953**); treatment with DBU then furnished cycloheptenone **954**. Kinetic deprotonation of **954** led to the cross-conjugated silyl dienol ether **955**, and a stereoselective Mukaiyama double-Michael reaction with methyl vinyl ketone <sup>328</sup> gave the diketone **950**, via intermediate **956**.

Double Baeyer-Villiger oxidation of **950** using peracetic acid afforded an 85:15 mixture of the

Scheme 89

regioisomeric lactones 949 and 957. Note that the use of more reactive peracids led to lower selectivity for formation of the required 949. Sequential saponification and acid-catalyzed lactonization of the inseparable mixture of 949 and 957 supplied the corresponding acyl-migrated hydroxylactones; protection of the hydroxyl then gave 958 and 959. After separation, alkylation on the convex face of 958<sup>329</sup> furnished **960** with >95% ds in favor of the required configuration at C<sub>8</sub>. Reduction of the lactone to afford the  $C_1, C_9$  diol was followed by selective protection of the  $C_9$  hydroxyl; oxidation at  $C_1$  then supplied the cycloheptanone 948. Baever-Villiger oxidation of 948 proved troublesome. However, after cleavage of the silyl ether at C<sub>3</sub>, the oxidation was straightforward, and under carefully controlled conditions provided the eight-membered lactone 961 with the correct configuration at all of its stereogenic centers. Reprotection of the C<sub>3</sub> hydroxyl under mildly acidic conditions,<sup>330</sup> ammonolytic deprotection at C<sub>9</sub>, and subsequent oxidation then afforded the C<sub>1</sub>-C<sub>9</sub> segment **946**.

Meanwhile, the  $C_{10}-C_{15}$  segment **947** was prepared from (S)-2-hydroxy-3-methylbutanoic acid (**962**). Thus, reduction of **962** followed by regioselective mesylation of the resulting diol and subsequent base-induced cyclization gave the epoxide *ent-***937**, which was elaborated to the alkynylsilane **963** via Yamaguchi coupling<sup>87</sup> with lithium (trimethylsilyl)acetylide followed by hydroxyl protection, as in the Buszek

# Scheme 90. Clardy ent-Octalactin A and B Syntheses<sup>321 a</sup>

 $^a$  (a) p-TsOH, MeOH; (b) O<sub>3</sub>; Me<sub>2</sub>S; (c) Zn, CH<sub>2</sub>I<sub>2</sub>, AlMe<sub>3</sub>; (d) LiOH, H<sub>2</sub>O; (e) (COCl)<sub>2</sub>; SnCl<sub>4</sub>; (f) DBU; (g) LDA; TMSCl; (h) MeCOCH=CH<sub>2</sub>, SnCl<sub>4</sub>; (i) CH<sub>3</sub>CO<sub>3</sub>H, AcOH, NaOAc; (j) KOH, MeOH; HCl; (k) TBDPSCl, imidazole; (l) LDA; MeI, HMPA; (m) LiBH<sub>4</sub>; (n) (ClH<sub>2</sub>CCO)<sub>2</sub>O, Et<sub>3</sub>N; (o) Swern oxidation; (p) HF; (q) CF<sub>3</sub>CO<sub>3</sub>H; (r) MeCOCH=C(Me)OTBS, p-TsOH; (s) liquid NH<sub>3</sub>; (t) Dess-Martin periodinane; (u) **947**, 'BuLi; **946**; (v) Dess-Martin periodinane; (w) HF; (x) VO(acac)<sub>2</sub>, 'BuOOH; (a') LAH; (b') MsCl, Et<sub>3</sub>N; (c') K<sub>2</sub>CO<sub>3</sub>, MeOH; (d') TMS-C≡C-Li, BF<sub>3</sub>·OEt<sub>2</sub>; (e') TBDPSCl, imidazole; (f') (c-C<sub>6</sub>H<sub>11</sub>)<sub>2</sub>BH; MeLi; MeI; (g') Br<sub>2</sub>; (h') MeONa, MeOH.

synthesis<sup>320</sup> (cf. 937  $\rightarrow$  938 in Scheme 88). Compound 963 was transformed into the vinylsilane 964 according to the method of Nozaki, <sup>331</sup> involving (i) regioselective cis hydroboration of 963, (ii) transmetalation of the resulting vinylborane to generate the corresponding vinyllithium, and (iii) alkylation with methyl iodide with retention of the double-bond configuration. Trans bromination of 964 followed by base-induced trans desilicobromination, according to the procedure of Miller, <sup>332</sup> then afforded vinyl bromide 947 with inversion of the double-bond configuration. Note that the Buszek synthesis also involved transformation of an alkynylsilane into a vinyl halide, but utilized different methodology (cf. 938  $\rightarrow$  939  $\rightarrow$  920 in Scheme 88).

Coupling of aldehyde **946** with the vinyllithium derived from **947** furnished a mixture of  $C_9$  epimers (**945**); oxidation of the mixture to provide **965**, followed by removal of the protecting groups, then supplied the unnatural antipode of octalactin B (*ent*-**917**). Finally, vanadium-mediated epoxidation<sup>88</sup> of *ent*-**917** proceeded with moderate stereoselectivity,

affording a 2:1 mixture of unnatural enantiomer of octalactin A (ent-916) and its  $C_{10}$ ,  $C_{11}$  bis(epimer) 966. In this synthesis, the  $C_4$  and  $C_{13}$  stereogenic centers originated in the chiral pool (cf. in the Buszek synthesis, the  $C_4$  and  $C_8$  stereogenic centers were obtained from the chiral pool). All the other stereogenic centers were installed using substrate-controlled asymmetric induction. [ent-Octalactin B (917): 0.2% overall yield from 951; 23 steps longest linear sequence; 31 steps total;  $\sim$ 6 steps per stereogenic center; ent-octalactin A (916): 0.1% overall yield from 951; 24 steps longest linear sequence; 32 steps total;  $\sim$ 3 steps per stereogenic center.]

## III. Concluding Remarks

The foregoing work demonstrates that rapid progress has been made in the field of organic chemistry concerned with the total synthesis of bioactive marine macrolides. Notably, most of these efforts have been concentrated over the last 5 years. The range of exquisite chemical structures fashioned by marine organisms, which seems to be limitless,

needs to be matched by the ingenuity and resourcefulness of synthetic chemists, and the challenge associated with macrolides like the halichondrins and swinholides is firmly at the cutting edge of contemporary synthetic organic chemistry. Such complex targets have provided an important impetus for the development of new methods and strategies.

Returning to the issues highlighted in the introduction, it can be seen that workable solutions have been developed, which have culminated in the completion of a significant number of total syntheses. The growing ascendancy of acyclic methods of stereocontrol has led to increasingly concise synthetic routes (1−2 steps per stereogenic center is now becoming a realistic goal). However, de novo chemical synthesis has not yet reached a level of efficiency to completely eclipse all other methods for obtaining supplies of marine macrolides. The only existing total syntheses of halichondrin B and bryostatin 7, for instance, require 120 and 80 steps, respectively. Although such syntheses represent impressive contributions to organic chemistry, even more practical synthetic routes must be developed if sufficient synthetic material is to be made available for clinical evaluation in cases where the natural supply is inadequate. This remains the key challenge for the future.

## Acknowledgments

We thank the EPSRC for financial support, Mrs. Cheryl Cook for assistance in obtaining copies of articles not held by the Cambridge University libraries, and Dr. Cameron Cowden, Dr. Klaus Fessner, and Miss Christine Watson for their careful proofreading of this manuscript.

#### References

- (1) For a recent collection of reviews surveying the immense variety of marine natural products, see: Chem Rev. 1993, 93 (5), 1671-1944
- For a complementary review concerning synthetic studies of biologically active marine cyclopeptides, see: Wipf, P. Chem. Rev. 1995, 95, 2115-2134 (accompanying article in this issue).
  (3) For reviews concerning the total synthesis of macrolides pro-
- duced by terrestrial organisms, see: (a) Paterson, I.; Mansuri, M. M. Tetrahedron 1985, 41, 3569. (b) Masamune, S.; McCarthy, P. A. In Macrolide Antibiotics: Chemistry, Biology and Practice; Omura, S., Ed.; Academic Press: Orlando, FL, 1984; Chapter
- (4) (a) Pettit, G. R.; Cichacz, Z. A.; Gao, F.; Herald, C. L.; Boyd, M. R.; Schmidt, J. M.; Hooper, J. N. A. J. Org. Chem. 1993, 58, 1302. (b) Pettit, G. R.; Cichacz, Z. A.; Gao, F.; Herald, C. L.; Boyd, M. R. J. Chem. Soc., Chem. Commun. 1993, 1166. (c) Pettit, G. R.; Herald, C. L.; Cichacz, Z. A.; Gao, F.; Schmidt, J. M.; Boyd, M. R.; Christie, N. D.; Boettner, F. E. J. Chem. Soc., Chem. Commun. 1993, 1805. (d) Pettit, G. R.; Herald, C. L.; Cichacz, Z. A.; Gao, F.; Boyd, M. R.; Christie, N. D.; Schmidt, J. M. Nat. Prod. Lett. 1993, 3, 239. (e) Pettit, G. R.; Cichacz, Z. A.; Herald, C. L.; Gao, F.; Boyd, M. R.; Schmidt, J. M.; Hamel, E.; Bai, R. J. Chem. Soc., Chem. Commun. **1994**, 1605.
- (5) (a) Kobayashi, M.; Aoki, S.; Sakai, H.; Kawazoe, K.; Kihara, N. Sasaki, T.; Kitagawa, I. Tetrahedron Lett. 1993, 34, 2795. (b) Kobayashi, M.; Aoki, S.; Kitagawa, I. Tetrahedron Lett. 1994, *35*, 1243.
- (6) Fusetani, N.; Shinoda, K.; Matsunaga, S. J. Am. Chem. Soc. 1**993**, *115*, 3977.
- (7) For a review of bioactive metabolites produced by symbiotic marine organisms, see: Kobayashi, J.; Ishibashi, M. Chem. Rev. 1993, 93, 1753.
- For a review of carbohydrate-based stereocontrol, see: Hanessian, S. Total Synthesis of Natural Products: the Chiron Approach; Pergamon Press: Oxford, 1983.
- For reviews of acyclic stereocontrol strategy, see: (a) Bartlett, P. A. Tetrahedron 1980, 36, 3. (b) Masamune, S.; Choy, W.; Petersen, J. S.; Sita, L. R. Angew. Chem., Int. Ed. Engl. 1985, 24, 1. (c) Hoffmann, R. W. Angew. Chem., Int. Ed. Engl. 1987, *2*6, 489.
- (10) For a recent review of macrolactonization methodology, see: Bartra, M.; Urpí, F.; Vilarassa, J. In Recent Progress in the

- Chemical Synthesis of Antibiotics and Related Microbial Prod-
- ucts; Kukacs, G., Ed.; Springer-Verlag: Berlin, 1993; Vol. 2. (11) Swinholide A: Carmely, S.; Kashman, Y. Tetrahedron Lett. 1985,
- (12) Swinholides B-G: (a) Kobayashi, M.; Tanaka, J.; Katori, T.; Kitagawa, I. Chem. Pharm. Bull. 1990, 38, 2960. (b) Tsukamoto, S.; Ishibashi, M.; Sasaki, T.; Kobayashi, J. J. Chem. Soc., Perkin Trans. 1 1991, 3185. (c) Todd, J. S.; Alvi, K. A.; Crews, P. Tetrahedron Lett. 1992, 33, 441.
- (13) (a) Kobayashi, M.; Tanaka, J.; Katori, T.; Matsuura, M.; Kitagawa, I. Tetrahedron Lett. 1989, 30, 2963. (b) Kitagawa, I.; Kobayashi, M.; Katori, T.; Yamashita, M.; Tanaka, J.; Doi, M.; Ishida, T. J. Am. Chem. Soc. 1990, 112, 3710. (c) Kobayashi, M.; Tanaka, J.; Katori, T.; Matsuura, M.; Yamashita, M.; Kitagawa, I. Chem. Pharm. Bull. 1990, 38, 2409. (d) Doi, M.; Ishida, T.; Kobayashi, M.; Kitagawa, I. J. Org. Chem. 1991, 56, 3629
- (14) This corresponds to the size of the macrocycle obtained by counting around the carbon skeleton of the dihydropyran. If the dihydropyran oxygens are counted instead, the ring size is reduced by two and four atoms for the monomer and dimer, respectively
- (15) (a) Sakai, R.; Higa, T.; Kashman, Y. Chem. Lett. 1986, 1499. (b) Kato, Y.; Fusetani, N.; Matsunaga, S.; Hashimoto, K.; Sakai, R.; Higa, T.; Kashman, Y. Tetrahedron Lett. 1987, 28, 6225. (c) Tanaka, J.; Higa, T.; Kobayashi, M.; Kitagawa, I. *Chem. Pharm. Bull.* **1990**, *38*, 2967. (d) Kobayashi, J.; Tsukamoto, S.; Tanabe, A.; Sasaki, T.; Ishibashi, M. J. Chem. Soc., Perkin Trans. 1 1991,
- 23/9.
  (16) Scytophycins A-E: (a) Ishibashi, M.; Moore, R. E.; Patterson, G. M. L.; Xu, C.; Clardy, J. J. Org. Chem. 1986, 51, 5300. (b) Moore, R. E.; Patterson, G. M. L.; Mynderse, J. S.; Barchi, J., Jr.; Norton, T. R.; Furusawa, E.; Furusawa, S. Pure Appl. Chem. 1986, 58, 263. (c) Moore, R. E.; Banarjee, S.; Bornemann, V.; Caplan, F. R.; Chen. J. L.; Corley, D. G.; Larsen, L. K.; Moore, B. S.; Patterson, G. M. L.; Paul, V. J.; Stewart, J. B.; Williams, D. E. Pure Appl. Chem. 1989, 61, 521.
  (17) (a) Peterson, L. Cumping, L. C. Totscholdre, Lett. 1992, 22, 2847.
- (17) (a) Paterson, I.; Cumming, J. G. Tetrahedron Lett. 1992, 33, 2847.
  (b) Paterson, I.; Smith, J. D. J. Org. Chem. 1992, 57, 3261. (c) Paterson, I.; Smith, J. D. Tetrahedron Lett. 1993, 34, 5351. (d) Paterson, I.; Cumming, J. G.; Smith, J. D.; Ward, R. A. Tetrahedron Lett. 1994, 35, 441. (e) Paterson, I.; Smith, J. D.; Ward, R. A.; Cumming, J. G. J. Am. Chem. Soc. 1994, 116, 2615. (f) Paterson, I.; Cumming, J. G.; Smith, J. D.; Ward, R. A.; Yeung, K.-S. Tetrahedron Lett. 1994, 35, 3405. (g) Paterson, I.; Yeung, K.-S.; Ward, R. A.; Cumming, J. G.; Smith, J. D. J. Am. Chem. Soc. 1994, 116, 9391
- (18) (a) Patron, A. P.; Richter, P. K.; Tomaszewski, M. J.; Miller, R. A.; Nicolaou, K. C. J. Chem. Soc., Chem. Commun. 1994, 1147. (b) Richter, P. K.; Tomaszewski, M. J.; Miller, R. A.; Patron, A. P.; Nicolaou, K. C. J. Chem. Soc., Chem. Commun. 1994, 1151.
  (19) (a) Nakata, T.; Komatsu, T.; Nagasawa, K.; Yamada, H.; Takahashi, T. Tetrahedron Lett. 1994, 35, 8225. (b) Nakata, T.;
- Komatsu, T.; Nagasawa, K. Chem. Pharm. Bull. 1994, 42, 2403.
- (20) The numbering systems adopted throughout this review for natural product segments corresponds to those used for the relevant natural products themselves.
- (21) Paterson, I.; Osborne, S. A. Tetrahedron Lett. 1990, 31, 2213.
- (22) Price, C. C.; Pappalardo, J. A. J. Am. Chem. Soc. 1950, 72, 2613. (23) (a) Luche, J.-L. J. Am. Chem. Soc. 1978, 100, 2226. (b) Gemal,
   A. L.; Luche, J.-L. J. Am. Chem. Soc. 1981, 103, 5454. (c) Danishefsky, S. J.; DeNinno, S.; Lartey, P. J. Am. Chem. Soc. 1987, 109, 2082.
- (24) (a) Ferrier, R. J. J. Chem. Soc. 1964, 5443. (b) Ferrier, R. J.;
- Prasad, N. J. Chem. Soc. (C) 1969, 570. (25) Jung, M. E.; Blum, R. B. Tetrahedron Lett. 1977, 3791.
- (26) Krohn, K.; Tolkiehn, K.; Lehne, V.; Schmalle, H. W.; Grützmacher, H.-F. Liebigs Ann. Chem. 1985, 1311.
- (27) (a) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4155. (b) Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 113, 7277.
  (28) (a) Martin, V. S.; Woodard, S. S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K. B. J. Am. Chem. Soc. 1981, 103, 6237. (b) Gao, V. Y.; Hanson, R. M.; Klunder, J. M.; Ko, S. Y.; Masamune, H.; Sharpless, K. B. J. Am. Chem. Soc. 1987, 109, 5765. (c) Roush,
- W. R.; Brown, R. J. *J. Org. Chem.* **1983**, *48*, 5093. (29) (a) Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Viti, S. M. J. Org. Chem. 1982, 47, 1378. (b) Finan, J. M.; Kishi, Y. Tetrahedron Lett. 1982, 23, 2719.
- (30) Hosomi, A.; Sakata, Y.; Sakurai, H. Tetrahedron Lett. 1984, 25,
- (31) Trippett, S.; Walker, D. M. J. Chem. Soc. 1961, 1266.
   (32) Paterson, I.; Goodman, J. M.; Isaka, M. Tetrahedron Lett. 1989, 30, 7121.
- (33) (a) Brown, H. C.; Dhar, R. K.; Bakshi, R. K.; Pandiarajan, P. K.; Singaram, B. J. Am. Chem. Soc. 1989, 111, 3441. (b) Brown, H. C.; Dhar, R. K.; Ganesan, K.; Singaram, B. J. Org. Chem. 1992, 57, 499,
- (34) (a) Paterson, I.; Lister, M. A. Tetrahedron Lett. 1988, 29, 585. (b) Paterson, I.; Norcross, R. D.; Ward, R. A.; Romea, P.; Lister, M. A. J. Am. Chem. Soc. 1994, 116, 11287.

- (35) (a) Saksena, A. K.; Mangiaracina, P. Tetrahedron Lett. 1983, 24, 273. (b) Evans, D. A.; Chapman, K. T.; Carreira, E. M. J. Am. Chem. Soc. 1988, 110, 3560.
  (36) (a) Still, W. C.; Barrish, J. C. J. Am. Chem. Soc. 1983, 105, 2487. (b) Paterson, I.; Perkins, M. V. Tetrahedron Lett. 1992, 33, 801.
  (37) (a) Barton, D. H. R.; McCombie, S. W. J. Chem. Soc., Perkin Trans. I 1975, 1574. (b) Rasmussen, J. R.; Slinger, C. J.; Kordish, R. J.; Newman-Evans, D. D. J. Org. Chem. 1981, 46, 4843. (c) Review: Hartwig, W. Tetrahedron 1983, 39, 2609.
  (38) (a) Mancuso, A. J.; Huang, S.-L.; Swern, D. J. Org. Chem. 1978, 43, 2480. (b) Review: Tidwell, T. T. Org. React. 1990, 39, 297.
  (39) Narayanan, B. A.; Bunnelle, W. H. Tetrahedron Lett. 1987, 28, 6261.
- 6261.
- (40) Paterson, I.; Goodman, J. M.; Lister, M. A.; Schumann, R. C.; McClure, C. K.; Norcross, R. D. Tetrahedron 1990, 46, 4663.
  (41) Evans, D. A.; Hoveyda, A. H. J. Org. Chem. 1990, 55, 5190.
- (42) (a) Oikawa, Y.; Yoshioka, T.; Yonemitsu, O. Tetrahedron Lett.
   1982, 23, 889. (b) Oikawa, Y.; Yoshioka, T.; Yonemitsu, O. Tetrahedron Lett.
   1982, 23, 885. (c) Horita, K.; Yoshioka, T.; Tanaka, T.; Oikawa, Y. Yonemitsu, O. Tetrahedron 1986, 42, 2002
- (43) (a) Brown, H. C.; Bhat, K. S. J. Am. Chem. Soc. 1986, 108, 5919. (b) Brown, H. C.; Bhat, K. S.; Randad, R. S. J. Org. Chem. 1989, 54, 1570.
- (44) (a) Roush, W. R.; Ando, K.; Powers, D. B.; Palkowitz, A. D.; Halterman, R. L. J. Am. Chem. Soc. 1990, 112, 6339. (b) Roush, W. R.; Palkowitz, A. D.; Ando, K. J. Am. Chem. Soc. 1990, 112,
- (45) (a) Tsuji, J.; Nagashima, H.; Nemeto, H. Org. Synth. 1984, 62, 9. (b) Review: Tsuji, J. Synthesis, 1984, 369.
  (46) (a) Narasaka, K.; Pai, F.-C. Tetrahedron 1984, 40, 2233. (b) Chen, K.-M.; Hardtmann, G. E.; Prasad, K.; Repič, O.; Shapiro, M. J. Tetrahedron Lett. 1987, 28, 155. (c) Reference 32.
  (47) Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Pull Chen, Soc. In. 1979, 52, 1989.

- (47) Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Bull. Chem. Soc. Jpn. 1979, 52, 1989.
  (48) Boden, E. P.; Keck, G. E. J. Org. Chem. 1985, 50, 2394.
  (49) (a) Stork, G.; Nakamura, E. J. Org. Chem. 1979, 44, 4010. (b) Nicolaou, K. C.; Seitz, S. P.; Pavia, M. R.; Petasis, N. A. J. Org. Chem. 1979, 44, 4011. (c) Nicolaou, K. C.; Chakraborty, T. K.; Daines, R. A.; Simpkins, N. S. J. Chem. Soc., Chem. Commun. 1986, 413. (d) Nicolaou, K. C.; Daines, R. A.; Chakraborty, T. K. J. Am. Chem. Soc. 1987, 109, 2208. K. J. Am. Chem. Soc. 1987, 109, 2208.
- (50) (a) Neises, B.; Steglich, W. Angew. Chem., Int. Ed. Engl. 1978, 522. (b) Ziegler, F. E.; Berger, G. D. Synth. Commun. 1979,
   539. (c) Nicolaou, K. C.; Seitz, S. P.; Pavia, M. R. J. Am. Chem. Soc. 1982, 104, 2030. (d) Reference 48.
- (51) (a) Saito, S.; Hasegawa, T.; Inaba, M.; Nishida, R.; Fujii, T.; Nomizu, S.; Moriwake, T. Chem. Lett. 1984, 1389. (b) Saito, S.; Ishikawa, T.; Kuroda, A.; Koga, K.; Moriwake, T. Tetrahedron 1**992**, 48, 4067
- (52) Carretero, J. C.; Ghosez, L. Tetrahedron Lett. 1988, 29, 2059 and references cited therein.
  (a) Savard, J.; Brassard, P. Tetrahedron 1984, 40, 3455. (b)
- Witkowski, R.; Bandermann, F. Macromol. Chem. 1989, 190,
- (54) (a) Lewis, M. D.; Cha, J. K.; Kishi, Y. J. Am. Chem. Soc. 1982, 104, 4976. (b) Kozikowski, A. P.; Sorgi, K. L.; Wang, B. C.; Xu, Z. Tetrahedron Lett. 1983, 24, 1563. (c) Giannis, A.; Sandhoff, E. Tetrahedron Lett. 1985, 26, 1479. (d) Nicolaou, K. C.; Hwang, C.-K.; Duggan, M. E. J. Am. Chem. Soc. 1989, 111, 6682.

  (a) Nagashima, N.; Ohno, M. Chem. Lett. 1987, 141. (b) Review: David, S.; Hanessian, S. Tetrahedron 1985, 41, 643.

  (a) Enders, D.; Eichenauer, H. Chem. Ber. 1979, 112, 2933. (b)
- Enders, D.; Eichenauer, H. Angew. Chem., Int. Ed. Engl. 1979, 18, 397.
- (a) Evans, D. A.; Clark, J. S.; Metternich, R.; Novack, V. J.; Sheppard, G. S. J. Am. Chem. Soc. 1990, 112, 866. (b) Evans, D. A.; Rieger, D. L.; Bilodeau, M. T.; Urpí, F. J. Am. Chem. Soc. 1991, 113, 1047.
- (58) Evans, D. A.; Hoveyda, A. H. J. Am. Chem. Soc. 1990, 112, 6447.
  (59) Nakata, T.; Suenaga, T.; Oishi, T. Tetrahedron Lett. 1989, 30,
- (60) (a) Hanessian, S.; Ugolini, A.; Therien, M. J. Org. Chem. 1983, 48, 4427. (b) Hanessian, S.; Ugolini, A.; Dubé, D.; Glamyan, A. Can. J. Chem. 1984, 62, 2146.
- Kornblum, N.; Erickson, A. S.; Kelly, W. J.; Henggeler, B. J. Org. Chem. 1982, 47, 4534.
- (62) Stork, G.; Paterson, I.; Lee, F. K. C. J. Am. Chem. Soc. 1982, 104, 4686.
- Horita, K.; Inoue, T.; Tanaka, K.; Yonemitsu, O. Tetrahedron
- Lett. 1992, 33, 5537.

  Fukui, M.; Okamoto, S.; Sano, T.; Nakata, T.; Oishi, T. Chem. Pharm. Bull. 1990, 38, 2890.

  Evans, D. A.; Bartroli, J.; Shih, T. L. J. Am. Chem. Soc. 1981, 102, 2197.
- 103, 2127.
- (66) Larchevêque, M.; Mambu, L.; Petit, Y. Synth. Commun. 1991,
- (a) Nakata, T.; Nagao, S.; Takao, S.; Tanaka, T.; Oishi, T. Tetrahedron Lett. 1985, 26, 73. (b) Nakata, T.; Nagao, S.; Oishi, T. Tetrahedron Lett. 1985, 26, 75.

- (68) Evans, D. A.; Britton, T. C.; Ellman, J. A. Tetrahedron Lett. 1987, 28.6141
- (69) Nakagawa, I.; Aki, K.; Hata, T. J. Chem. Soc., Perkin Trans. 1 1983, 1315.
- (70) Hanessian, S.; Cooke, N. G.; DeHoff, B.; Sakito, Y. J. Am. Chem. Soc. 1990, 112, 5276.
- (71) (a) Uemura, D.; Takahashi, K.; Yamamoto, T.; Katayama, C.; Tanaka, J.; Okumura, Y.; Hirata, Y. J. Am. Chem. Soc. 1985, 107, 4796. (b) Hirata, Y.; Uemura, D. Pure Appl. Chem. 1986, 58, 701.
- (72) (a) Halichondrin B (152) and homohalichondrin B (158) were subsequently also isolated from the Axinella sponge: Pettit, G. R.; Herald, C. L.; Boyd, M. R.; Leet, J. E.; Dufresne, C.; Doubek, D. L.; Schmidt, J. M.; Cerny, R. L.; Hooper, J. N. A.; Rützler, K. C. J. Med. Chem. 1991, 34, 3339. (b) Halichondrin B, homohalichondrin B, and 10a hydroxyhalichondrin B (called halistatin 1) have been isolated from the *Phakellia carteri* sponge: Pettit, G. R.; Tan, R.; Gao, F.; Williams, M. D.; Doubek, D. L.; Boyd, M. R.; Schmidt, J. M.; Chapuis, J.-C.; Hamel, E.; Bai, R.; Hooper, J. N. A.; Tackett, L. P. J. Org. Chem. 1993, 58, 2538.
- (73) Isohomohalichondrin B (160) has recently been isolated from the sponge Lissodendoryx sp. See: Litaudon, M.; Hart, J. B.; Blunt, J. W.; Lake, R. J.; Munro, M. H. G. Tetrahedron Lett. 1994, 35, 9435.
- 9435.
  (74) (a) Aicher, T. D.; Kishi, Y. Tetrahedron Lett. 1987, 28, 3463. (b) Aicher, T. D.; Buszek, K. R.; Fang, F. G.; Forsyth, C. J.; Jung, S. H.; Kishi, Y.; Scola, P. M. Tetrahedron Lett. 1992, 33, 1549. (c) Buszek, K. R.; Fang, F. G.; Forsyth, C. J.; Jung, S. H.; Kishi, Y.; Scola, P. M.; Yoon, S. K. Tetrahedron Lett. 1992, 33, 1553. (d) Fang, F. G.; Kishi, Y.; Matelich, M. C.; Scola, P. M. Tetrahedron Lett. 1992, 33, 1557. (e) Aicher, T. D.; Buszek, K. R.; Fang, F. G.; Forsyth, C. J.; Jung, S. H.; Kishi, Y.; Matelich, M. C.; Scola, P. M.; Spero, D. M.; Yoon, S. K. J. Am. Chem. Soc. 1992, 114, 3162. (f) Duan, J. J.-W.; Kishi, Y. Tetrahedron Lett. 1993, 34, 7541. 1993, 34, 7541.
- (75) (a) Kim, S.; Salomon, R. G. Tetrahedron Lett. 1989, 30, 6279.
  (b) Cooper, A. J.; Salomon, R. G. Tetrahedron Lett. 1990, 31, 3813. (c) DiFranco, E.; Ravikumar, V. T.; Salomon, R. G. Tetrahedron Lett. 1993, 34, 3247. (d) Cooper, A. J.; Pan, W.; Salomon, R. G. Tetrahedron Lett. 1993, 34, 8193.
- (76) (a) Horita, K.; Hachiya, S.; Nagasawa, M.; Hikota, M.; Yone-Hachiya, S.; Yonemitsu, O. Synlett 1994, 46. (e) Yonemitsu, O. J. Synth. Org. Chem. Jpn. 1994, 52, 946 (special issue in English).
- (77) (a) Burke, S. D.; Buchanan, J. L.; Rovin, J. D. Tetrahedron Lett. 1991, 32, 3961. (b) Burke, S. D.; Jung, K. W.; Phillips, J. R.; Perri, R. E. Tetrahedron Lett. 1994, 35, 703.
- (78) (a) Jin, H.; Uenishi, J.; Christ, W. J.; Kishi, Y. J. Am. Chem. Soc. 1986, 108, 5644. (b) Takai, K.; Tagashira, M.; Kuroda, T.; Oshima, K.; Utimoto, K.; Nozaki, H. J. Am. Chem. Soc. 1986, 108, 6048. (c) Kishi, Y.; Christ, W. J.; Taniguchi, M. In Natural Products and Biological Activities; Imura, H., Goto, T., Murachi, T., Nakajima, T., Eds.; University of Tokyo Press: Tokyo, 1986;
- pp 87-98.
  (79) (a) Cha, J. K.; Christ, W. J.; Kishi, Y. Tetrahedron Lett. 1983, 24, 3943. (a) Cha, J. K.; Christ, W. J.; Kishi, Y. Tetrahedron 1984, 40, 2247.
- (80) See footnote 15 in ref 74e.
- (81) The synthesis of alcohol 172 was originally studied by using the unnatural antipode obtained by starting with 1,6-anhydro-p-galactose-3,4-acetonide: see ref 74a.
- (82) The ring nomenclature system adopted by Horita, Yonemitsu, and co-workers using the letters A-L (ref 76) is that used in this review. Note that this differs from the letter designations adopted by Salomon and co-workers (ref 75).
- (83) (a) Ireland, R. E.; Mueller, R. H.; Willard, A. K. J. Am. Chem. Soc. 1976, 98, 2868. (b) Review: Pereira, S.; Srebnik, M. Aldrichimica Acta 1993, 26, 17.
- (84) For examples of similar Ireland-Claisen rearrangements, see: (a) Ireland, R. E.; Wuts, P. G. M.; Ernst, B. J. Am. Chem. Soc. 1981, 103, 3205. (b) Kozikowski, A. P.; Lee, J. J. Org. Chem. **1990**, *55*, 863.
- (85) For reviews, see: (a) Mitsunobu, O. Synthesis 1981, 1. (b) Castro, B. R. Org. React. 1983, 29, 1. (c) Hughes, D. L. Org. React. 1992, 42, 335.
- Vekemans, J. A. J. M.; Franken, G. A. M.; Dapperens, C. W. M.; Godefroi, E. F.; Chittenden, G. J. F. *J. Org. Chem.* **1988**, *53*, 627.
- (87) (a) Yamaguchi, M.; Hirao, I. Tetrahedron Lett. 1983, 24, 391.
  (b) Yamaguchi, M.; Hirao, I. J. Chem. Soc., Chem. Commun. **1984**, 202.
- (88) (a) Sharpless, K. B.; Michaelson, R. C. J. Am. Chem. Soc. 1973, 95, 6136. (b) Tanaka, S.; Yamamoto, H.; Nozaki, H.; Sharpless, K. B.; Michaelson, R. C.; Cutting, J. D J. Am. Chem. Soc. 1974,
- (89) For example, see: (a) Sharpless, K. B.; Verhoeven, T. R.

- Aldrichimica Acta 1979, 12, 63. (b) Kishi, Y. Aldrichimica Acta
- (90) m-CPBA epoxidation of these alkenes gave the inverse stereochemistry of VO(acac)<sub>2</sub>-catalyzed epoxidation.
  (91) Still, W. C.; Gennari, C. Tetrahedron Lett. 1983, 24, 4405.
- (92) (a) Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. 1980, 102, 5974. (b) Review: Pfenninger, A. Synthesis 1986, 89
- (93) Although the C<sub>50</sub> and C<sub>51</sub> configurations of the homohalichon-drins were also unknown, the stereochemistry at these centers was assumed to be the same as in norhalichondrin A, the structure of which was unambiguously established by X-ray crystallographic analysis (ref 71). Thus Kishi and co-workers focused only on the synthesis of all the possible stereoisomers with respect to the C53 and C54 positions of the homohalichon-
- (94) For preparation of 210 see footnote 6 in ref 74e.
- (95) For preparation of 211 see footnote 23 in ref 74e.
  (96) (a) Mahoney, W. S.; Brestensky, D. M.; Stryker, J. M. J. Am. Chem. Soc. 1988, 110, 291. (b) Mahoney, W. S.; Stryker, J. M. J. Am. Chem. Soc. 1989, 111, 8818.
  (97) (a) Baldwin, J. F. J. Chem. Soc. Chem. Commun. 1972, 734.
- J. Am. Chem. Soc. 1989, 111, 8818.
  (97) (a) Baldwin, J. E. J. Chem. Soc., Chem. Commun. 1976, 734.
  (b) Nicolaou, K. C.; Prasad, C. V. C.; Somers, P. K.; Hwang, C.-K. J. Am. Chem. Soc. 1989, 111, 5330.
  (98) Vedejs, E.; Fuchs, P. L. J. Org. Chem. 1971, 36, 366.
  (99) Roush, W. R.; Brown, R. J. J. Org. Chem. 1982, 47, 1371.
  (100) CONFLEX-MM2 calculations (ref 101) indicated that 231 was

- 2.48 kcal mol<sup>-1</sup> less stable than 6-epi-231: see refs 76a and 76e. (101) (a) Gōto, H.; Ōsawa, E. *Tetrahedron Lett.* 1992, 33, 1343. (b) Gōto, H.; Ōsawa, E. *J. Chem. Soc., Perkin Trans.* 2 1993, 187. (c) Gōto, H.; Ōsawa, E.; Yamato, M. *Tetrahedron* 1993, 49, 387.
- (102) Mori, Y.; Kuhara, M.; Takeuchi, A.; Suzuki, M. Tetrahedron Lett. **1988**, 29, 5419.
- Iodoetherification of  $\alpha, \gamma$ -anti dihydroxyolefins usually gives 2,5-Cis-tetrahydrofurans as the major products. See: (a) Tamaru, Y.; Hojo, M.; Kawamura, S.; Sawada, S.; Yoshida, Z. J. Org. Chem. 1987, 52, 4062. (b) Kim, Y. G.; Cha, J. K. Tetrahedron Lett. 1988, 29, 2011. Compare: (c) Marek, I.; Lefrançois, J.-M.; Normant, J.-F. Tetrahedron Lett. 1992, 33, 1747. (d) Rychnovsky,
- Normant, J.-F. Tetrahearon Lett. 1892, 55, 1747. (a) Rychnovsky, S. D.; Bartlett, P. A. J. Am. Chem. Soc. 1981, 103, 3963. (104) Horita, K.; Sakurai, Y.; Hachiya, S.; Nagasawa, M.; Yonemitsu, O. Chem. Pharm. Bull. 1994, 42, 683. (105) (a) Chamberlain, P.; Roberts, M. L.; Whitham, G. H. J. Chem. Soc. (B) 1970, 1374. (b) Henbest, H. B. Pro. Chem. Soc. 1963,
- (106) (a) Kishi, Y.; Aratani, M.; Tanino, H.; Fukuyama, T.; Goto, T.; Inoue, S.; Suguira, S.; Kakoi, H. J. Chem. Soc., Chem. Commun.
- 1972, 64. (b) Linde, R. G., II; Egbertson, M.; Coleman, R. S.; Jones, A. B.; Danishefsky, S. J. J. Org. Chem. 1990, 55, 2771. (107) (a) Corey, E. J.; Katzenellenbogen, J. A. J. Am. Chem. Soc. 1969, 91, 1851. (b) Siddall, J. B.; Biskup, M.; Fried, J. H. J. Am. Chem. Soc. 1969, 91, 1853. (c) Nicolaou, K. C.; Duggan, M. E.; Hwang,
- Soc. 1969, 91, 1853. (C) Nicolaou, R. C.; Duggan, M. E.; Hwang, C.-K. J. Am. Chem. Soc. 1989, 111, 6666.
  (108) Nicolaou, K. C.; Papahatjis, D. P.; Claremon, D. A.; Magolda, R. L.; Dolle, R. E. J. Org. Chem. 1989, 50, 1440.
  (109) Imamoto, T.; Kusumoto, T.; Tawarayama, Y.; Sugiura, Y.; Mita, T.; Hatanaka, Y.; Yokoyama, M. J. Org. Chem. 1984, 49, 3904.
  (110) Richtmyer, N. K. Methods Carbohydr. Chem. 1962, 1, 107.
  (111) Richtmyer, N. E. Methods Carbohydr. Chem. 1962, 1, 107.
- (111) Hicks, D. R.; Fraser-Reid, B. Synthesis 1974, 203. (112) Pougny, J.-R.; Sinay, P. J. Chem. Res. (S) 1982, 1; J. Chem. Res. (M) **1982**, 0186.

- (113) Murray, D. H.; Prokop, J. J. Pharm. Sci. 1965, 10, 1468.
  (114) Gennari, C.; Cozzi, P. G. Tetrahedron 1988, 44, 5965.
  (115) (a) Arrick, R. E.; Baker, D. C.; Horton, D. Carbohydr. Res. 1973, 26, 441. (b) Jones, G. H.; Moffatt, J. G. Methods Carbohydr.
- Chem. 1972, 6, 315.
  (116) (a) Jacobsen, E. N.; Markó, I.; Mungall, W. S.; Schröder, G.; Sharpless, K. B. J. Am. Chem. Soc. 1988, 110, 1968. (b) Wai, J. S. M.; Markó, I.; Svendsen, J. S.; Finn, M. G.; Jacobsen, E. N.;
- Sharpless, K. B. J. Am. Chem. Soc. **1989**, 111, 1123. (117) (a) Gao, Y.; Sharpless, K. B. J. Am. Chem. Soc. **1988**, 110, 7538. (b) Kim, B. M.; Sharpless, K. B. Tetrahedron Lett. **1989**, 30, 655.
- (118) Mukaiyama, T.; Kobayashi, S.; Shoda, S. Chem. Lett. 1984, 1529. (119) Suavé, G.; Schwartz, D. A.; Ruest, L.; Deslongchamps, P. Can.
- J. Chem. 1984, 62, 2929. (120) For literature precedents for high stereoselection in 1,4-addition of organocuprates to  $\gamma$ -alkoxy- $\alpha$ , $\beta$ -unsaturated ketones, see: Miller, D. B.; Raychaudhuri, S. R.; Avasthi, K.; Lal, K.; Levison, B.; Salomon, R. G. J. Org. Chem. 1990, 55, 3164 and references cited therein.
- (121) Koerner, T. A. W., Jr.; Voll, R. J.; Younathan, E. S. Carbohydrate Res. 1977, 59, 403.
  (122) Mikami, T.; Asano, H.; Mitsunobu, O. Chem. Lett. 1987, 2033.
  (123) Lipshold, B. H.; Wilhelm, R. S.; Kozlowski, J. Tetrahedron Lett.
- 1**982**, 23, 3755.
- (124) Matsuzawa, S.; Horiguchi, Y.; Nakamura, E.; Kuwajima, I. Tetrahedron 1989, 45, 349.
  TMSOTf promotes Nazarov cyclizations of 1,4-dien-3-ones. See:
- (a) Hirano, S.; Hiyama, T.; Nozaki, H. Tetrahedron Lett. 1974, 1429. (b) Kjeldsen, G.; Knudsen, J. S.; Ravn-Peterson, L. S.; Torssell, K. B. G. Tetrahedron 1983, 39, 2237. (c) Andrews, J.

- F. P.; Regan, A. C. Tetrahedron Lett. 1991, 32, 7731.

  (126) Oikawa, Y.; Tanaka, T.; Horita, K.; Yonemitsu, O. Tetrahedron Lett. 1984, 25, 5397.

  (127) Czernecki, S.; Georgoulis, C.; Provelenghiou, C. Tetrahedron Lett.
- 1976, 3535.
- (128) For a detailed discussion of this strategy, see: (a) Schreiber, S. L. Chem. Scr. 1987, 27, 563. (b) Hoye, T. R.; Suhadolnik, J. C. Tetrahedron 1986, 42, 2855.
- (129) Kaneko, C.; Sugimoto, A.; Tanaka, S. Synthesis 1974, 876.
  (130) (a) Pietraszkiewicz, M.; Jurczak, J. Tetrahedron 1984, 40, 2967.
  (b) Rabinovitz, M.; Cohen, Y.; Halpern, M. Angew. Chem., Int. Ed. Engl. 1986, 25, 960.
  (c) M. Tribus, C. Willer, D. W. J. Chem. Soc. 1961, 1966, (b)
- (131) (a) Trippett, S.; Walker, D. M. J. Chem. Soc. 1961, 1266. (b) Schlessinger, R. H.; Poss, M. A.; Richardson, S.; Lin, P. Tetra-hedron Lett. 1985, 26, 2391.
- (132) For a detailed discussion of the exceptionally high enantiomeric excesses available in related reactions involving enantiotopic group and diastereotopic face selectivity, see: (a) Schreiber, S. L.; Schreiber, T. S.; Smith, D. B. J. Am. Chem. Soc. 1987, 109, 1525. Note that the reactions are so selective because the inherent selectivities of the two asymmetric reactions are multiplicative, and because a stereochemical error in the first asymmetric reaction is negated by the second asymmetric reaction, whereby a diastereomer is formed. See also: (b) Kogure, T; Eliel, E. L. J. Org. Chem. 1984, 49, 576. (c) Midland, M. M.; Gabriel, J. J. Org. Chem. 1985, 50, 1143. (133) For a discussion of the negative and positive aspects of molecular
- (133) For a discussion of the negative and positive aspects of molecular desymmetrization in synthesis, see: (a) Bertz, S. H. Tetrahedron Lett. 1983, 24, 5577. (b) Bertz, S. H. J. Chem. Soc., Chem. Commun. 1984, 218. (c) Whitesell, J. K.; Allen, D. E. J. Org. Chem. 1985, 50, 3025.
  (134) (a) Burke, S. D.; Armistead, D. M.; Schoenen, F. J.; Fevig, J. M. Tetrahedron 1986, 42, 2787. (b) Burke, S. D.; Lee, K. C.; Santafianos, D. Tetrahedron Lett. 1991, 32, 3957.
  (135) (a) Masamuna S. Ma P. Olympto H.; Ellinghon I. W.; Ito V.
- (135) (a) Masamune, S.; Ma, P.; Okumoto, H.; Ellingboe, J. W.; Ito, Y.
  J. Org. Chem. 1984, 49, 2834. (b) Shing, T. K. M.; Tsui, H.-C.;
  Zhou, Z.-H.; Mak, T. C. W. J. Chem. Soc., Perkin Trans. 1 1992, 887. (c) Shing, T. K. M.; Tsui, H.-C.; Zhou, Z.-H. J. Chem. Soc., Chem. Commun. 1992, 810. (d) A 1,3-dioxolane ring cis-fused to another ring is more stable toward hydrolysis than that formed from a side chain: Haines, A. H. Adv. Carbohydr. Chem. Biochem. 1981, 39, 13.
  (136) Iida, H.; Yamazaki, N.; Kibayashi, C. J. Org. Chem. 1986, 51,
- 3769 and references cited therein.
- (137) Chan, P. C.-M.; Chong, J. M. J. Org. Chem. 1988, 53, 5584. (138) Still, W. C.; Sreekumar, C. J. Am. Chem. Soc. 1980, 102, 1201.
- (139) Suzuki, K.; Katayama, E.; Tsuchihashi, G. Tetrahedron Lett. 1983, 24, 4997.
- (140) For reviews concerning stereoselective synthesis of spiroacetals, see: (a) Kluge, A. F. *Heterocycles* 1986, 24, 1699. (b) Boivin, T. L. B. Tetrahedron 1987, 43, 3309. (c) Perron, F.; Albizati, K. F. Chem. Rev. 1989, 89, 1617.
- For reviews concerning stereocontrolled formation of tetrahy-
- drofuran and tetrahydropyran ring systems, see: (a) ref 140b. (b) Cardillo, G.; Orena, M. Tetrahedron 1990, 46, 3321. (142) (a) Okami, Y.; Okazaki, T.; Kitahara, T.; Umezawa, H. J. Antibiot. 1976, 29, 1019. (b) Nakamura, H.; Iitaka, Y.; Kitahara, T.; Takao, O.; Okami, Y. J. Antibiot. 1977, 30, 714.
- (143) (a) Hütter. R.; Keller-Schierlein, W.; Knüsel, F.; Prelog, V.; Rodgers, G. C., Jr.; Suter, P.; Vogel, G.; Voser, W.; Zähner, H. Helv. Chim. Acta 1967, 50, 1533. (b) Dunitz, J. D.; Hawley, D. M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, W.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D. N. J.; Berlin, Y.; Marušić, R.; Prelog, M.; M.; Marušić, R.; Prelog, M.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; M.; Mikloš, D.; White, D.; M.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; White, D.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; Mikloš, D.; White, D.; Mikloš, D.; White, D . Helv. Chim. Acta 1971, 54, 1709.
- (144) Since boromycin is a macrolide produced by a terrestrial organism, discussion of the efforts made to accomplish its total synthesis is outside the scope of this review. For synthetic work on boromycin, see: (a) Chen, T. S. S.; Chang, C.; Floss, H. G. J. Org Chem. 1981, 46, 2661. (b) Hanessian, S.; Tyler, P. C.; Demailly, G.; Chapleur, Y. J. Am Chem. Soc. 1981, 103, 6243. (c) Avery, M. A.; White, J. D.; Arison, B. H. Tetrahedron Lett. 1981, 22, 3123. (d) Hanessian, S.; Delorme, D.; Tyler, P. C.; Demailly, G.; Chapleur, Y. Can. J. Chem. 1983, 61, 634. (e) White, J. D.; Avery, M. A.; Choudry, S. C.; Dhingra, O. P.; Kang, M.; Whittle, A. J. J. Am. Chem. Soc. 1983, 105, 6517. (f) White, J. D.; Choudhry, S. C.; Kang, M. Tetrahedron Lett. 1984, 25, 3671.
- (145) (a) Corey, E. J.; Pan, B.-C.; Hua, D. H.; Deardorff, D. R. J. Am.
- (146) (a) Corey, E. J.; Fan, B.-C.; Hua, D. H.; Deardorn, D. R. J. Am. Chem. Soc. 1982, 104, 6816. (b) Corey, E. J.; Hua, D. H.; Pan, B.-C.; Seitz, S. P. J. Am. Chem. Soc. 1982, 104, 6818.
  (146) (a) White, J. D.; Vedananda, T. R.; Kang, M.; Choudrhy, S. C. J. Am. Chem. Soc. 1986, 108, 8105. (b) White, J. D.; Kuo, S.; Vedananda, T. R. Tetrahedron Lett. 1987, 28, 3061.
  (147) (a) Nakata, T.; Saito, K.; Oishi, T. Tetrahedron Lett. 1986, 27, 231, (b) Nobest, T. Saita, K.; Oishi, T. Tetrahedron Lett. 1986, 27, 2011.
- 6341. (b) Nakata, T.; Saito, K.; Oishi, T. Tetrahedron Lett. 1986, 27, 6345.
- (148) (a) Matsuda, F.; Tomiyoshi, N.; Yanagiya, M.; Matsumoto, T. Chem. Lett. 1987, 2097. (b) Matsuda, F.; Tomiyoshi, N.; Yanagiya, M.; Matsumoto, T. Tetrahedron 1990, 46, 3469. (c) Matsuda, F. J. Synth. Org. Chem. Jpn. 1993, 51, 744 (in Japanese).
  (149) Lee, J. B.; Nolan, T. J. Tetrahedron, 1967, 23, 2789.

- (150) Corey, E. J.; Fuchs, P. L. Tetrahedron Lett. 1972, 3769.
- (151) Binkley, R. W.; Ambrose, M. G.; Hehemann, D. G. J. Org. Chem. **1980**, 45, 4387.
- (152) Corey, E. J.; Suggs, J. W. J. Org. Chem. 1975, 40, 2554.
  (153) Corey, E. J.; Beames, D. J. J. Am. Chem. Soc. 1973, 95, 5829.
  (154) Pojer, P. M.; Angyal, S. J. Aust. J. Chem. 1978, 31, 1031.
- (155) Lipshutz, B. H.; Kozlowski, J.; Wilhelm, R. S. J. Am. Chem. Soc. **1982**, 104, 2305.
- (156) Diago-Meseguer, J.; Palomo-Coll, A. L.; Fernández-Lizarbe, J. R.; Zugaza-Bilbao, A. Synthesis 1980, 547
- (157) Goubeau, J.; Wittmeier, H. W. Z. Anorg. Allg. Chem. 1952, 270,
- (158) Corey, E. J.; Hua, D. H.; Seitz, S. P. Tetrahedron Lett. 1984, 25,
- (159) The Chan reaction is the rearrangement of  $\alpha$ -(acyloxy)acetates to form  $\alpha,\beta$ -bis(silyloxy)- $\alpha,\beta$ -unsaturated esters. See: Lee, S. D.; Chan, T. H.; Kwon, K. S.  $Tetrahedron\ Lett.\ 1984,\ 25,\ 3399.$
- Carlsen, P. H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936.
- (161) Hann, A. C. O.; Lapworth, A. Pro. Chem. Soc. London 1904, 20,
- (162) Grieco, P. A.; Gilman, S.; Nishizawa, M. J. Org. Chem. 1976, 41, 1485.
- (163) Shoppee, C. W. Rep. Progr. Chem. 1947, 44, 170.
- (164) Johnson, P. R.; White, J. D. J. Org. Chem. 1984, 49, 4424.
- (165) Raphael, R. A. Acetylenic Compounds in Organic Synthesis; Butterworth: London, 1955; p 29.
- (166) Corey, E. J.; Kim, C. U.; Takeda, M. Tetrahedron Lett. 1972, 4339
- (167) For related examples of tetrahydrofuran formation by intramolecular oxyselenation reactions, used during synthetic studies toward boromycin, see refs 144d and f.
- (168) Mukaiyama, T.; Usui, M.; Saigo, K. Chem. Lett. 1976, 49.
- (169) (a) Matsuo, T.; Mori, K.; Matsui, M. Tetrahedron Lett. 1976, 1979. (b) Dolle, R. E.; Nicolaou, K. C. J. Am. Chem. Soc. 1985, 107, 1691.
- (170) (a) Nakata, T.; Tanaka, T.; Oishi, T. Tetrahedron Lett. 1983, 24, 2653. (b) Nakata, T.; Takao, S.; Fukui, M.; Tanaka, T.; Oishi, T. *Tetrahedron Lett.* 1**983**, 24, 3873.
- (171) Schlessinger, R. H.; Poss, M. A. J. Am. Chem. Soc. 1982, 104,
- (172) Collum, D. B.; McDonald, J. H., III; Still, W. C. J. Am. Chem. Soc. 1980, 102, 2118.
  (173) Still, W. C. J. Am. Chem. Soc. 1978, 100, 1481.
  (174) Matsumoto, T.; Matsuda, F.; Hasegawa, K.; Yanagiya, M. Tetrahedron 1984, 40, 2337.

- (175) Haines, A. H.; Jenkins, C. S. P. J. Chem. Soc., Perkin Trans. 1 1**972**, 273.
- (176) (a) Yamada, K.; Ojika, M.; Ishigaki, T.; Yoshida, Y.; Ekimoto, H.; Arakawa, M. J. Am. Chem. Soc. **1993**, 115, **1**1020. (b) Ojika, M.; Kigoshi, H.; Ishigaki, T.; Yamada, K. Tetrahedron Lett. 1993, 34, 8501. (c) Ojika, M.; Kigoshi, H.; Ishigaki, T.; Nisiwaki, M.; Tsukada, I.; Mizuta, K.; Yamada, K. Tetrahedron Lett. 1993, 34, 1sukada, I.; Mizuta, K.; Iamada, K. Ietranedron Lett. 1995, 34, 8505. (d) Ojika, M.; Kigoshi, H.; Ishigaki, T.; Tsukada, I.; Tsuboi, T.; Ogawa, T.; Yamada, K. J. Am. Chem. Soc. 1994, 116, 7441. (e) Kigoshi, H.; Ojika, M.; Suenaga, K.; Mutou, T.; Hirano, J.; Sakakura, A.; Ogawa, T.; Nisiwaki, M.; Yamada, K. Tetrahedron Lett. 1994, 35, 1247. (f) Kigoshi, H.; Ojika, M.; Ishigaki, T.; Suenaga, K.; Mutou, T.; Sakakura, A.; Ogawa, T.; Yamada, K. J. Am. Chem. Soc. 1994, 116, 7443.
- J. Am. Chem. Soc. 1994, 116, 7443.

  (177) For instance, terminal N-methyl-N-vinylformamide units are also found in the scytophycins (ref 16), and the family of tris-(oxazole) macrolides (refs 194-197, vide infra).
- (178) The term scalemic is used to describe an unequal mixture of enantiomers. See: Heathcock, C. H.; Finkelstein, B. L.; Jarvi, E. T.; Radel, P. A.; Hadley, C. R. J. Org. Chem. 1988, 53, 1922. (179) (a) Julia, M.; Paris, J.-M. Tetraheon Lett. 1973, 4833. (b) Julia, M. Phara L. Chennelle Chennel
- (179) (a) Julia, M.; Paris, J.-M. Tetrahedron Lett. 1973, 4833. (b) Julia, M. Pure Appl. Chem. 1985, 57, 763.
  (180) (a) Sato, K.; Mizuno, S.; Hirayama, M. J. Org. Chem. 1967, 32, 177. (b) Review: Maryanoff, B. E.; Reitz, A. B. Chem. Rev. 1989, 89, 863. (c) Review: Kelley, S. E. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon Press: Oxford, 1991; Vol. 1, p 729.
  (181) (a) Levin, J. I.; Turos, E.; Weinreb, S. M. Synth. Commun. 1982, 12, 989. (b) Nahm, S.; Weinreb, S. M. Tetrahedron Lett. 1981, 22, 3815.
  (182) Johnson M. R.; Nakata, T.; Kieli, V. Turo, S.
- (182) Johnson, M. R.; Nakata, T.; Kishi, Y. Tetrahedron Lett. 1979,
- (183) Kondo, K.; Tunemoto, D. Tetrahedron Lett. 1975, 1007. (184) Hikota, M.; Sakurai, Y.; Horita, K.; Yonemitsu, O. Tetrahedron Lett. 1990, 31, 6367.
- Note that the 1.5:1 selectivity for 24- vs 26-membered lactonization observed for aplyronine A is comparable with the 4:1 selectivity obtained for 22- vs 24-membered Yamaguchi lactonization of the monomeric seco acid of swinholide A. See: ref 17g.
- (186) It has been suggested that the swinholides may in fact be metabolites of symbiotic blue-green algae associated with Theonella sp. See refs 7 and 13c.
- Paterson, I.; Yeung, K.-S. Tetrahedron Lett. 1993, 34, 5347.
- (188) For other synthetic work on scytophycin C, see ref 64.

- (189) Alvarez-Ibarra, C.; Arias, S.; Bañón, G.; Fernández, M. J.; Rodríguez, M.; Sinisterra, V. J. Chem. Soc., Chem. Commun. **1987**, 1509.
- (190) For the use of other weak inorganic bases in the Horner-Emmons reaction, see: Blackwell, C. M.; Davidson, A. H.; Launchbury, S. B.; Lewis, C. N.; Morrice, E. M.; Reeve, M. M.; Roffey, J. A. R.; Tipping, A. S.; Todd, R. S. J. Org. Chem. 1992, 57, 1935.
- (191) For further examples of barium hydroxide-promoted Horner-Emmons reactions in natural products synthesis, see: Paterson,
- I.; Yeung, K.-S.; Smaill, J. B. Synlett 1994, 774.

  (192) (a) Blanchette, M. A.; Choy, W.; Davis, J. T.; Essenfeld, A. P.;

  Masamune, S.; Roush, W. R.; Sakai, T. Tetrahedron Lett. 1984, 25, 2183. (b) Rathke, M. W.; Nowak, M. J. Org. Chem. 1985, 50, 2624
- (193) Evans, D. A.; Sheppard, G. S. J. Org. Chem. 1990, 55, 5192.
  (194) Roesener, J. A.; Scheuer, P. J. J. Am. Chem. Soc. 1986, 108, 846. (195) (a) Matsunaga, S.; Fusetani, N.; Hashimoto, K.; Koseki, K.;
- Noma, M. J. Am. Chem. Soc. 1986, 108, 847. (b) Matsunaga, S.; Fusetani, N.; Hashimoto, K.; Koseki, K.; Noma, M.; Noguchi, H.; Sankawa, U. J. Org. Chem. 1989, 54, 1360.
- (a) Fusetani, N.; Yasumuro, K.; Matsunaga, S.; Hashimoto, K.; Tetrahedron Lett. 1989, 30, 2809. (b) Fusetani, N.; Sugawara,
- T.; Matsunaga, S.; Hirota, H. *J. Org. Chem.* 1**991**, *56*, 4971. (197) (a) Kernan, M. R.; Faulkner, D. J. *Tetrahedron Lett.* **1987**, *28*, 2809. (b) Kernan, M. R.; Molinski, T. F.; Faulkner, D. J. J. Org. Chem. 1988, 53, 5014.
- (198) Note that nudibranches are known to feed on sponges, and sponges are symbiotically associated with blue-green algae. Thus, the ulapualides, kabiramides, mycalolides, and halichondramides are all strongly suggested to be metabolites of bluegreen algae. The swinholides and bistheonellides are likewise speculated to originate from blue-green algae, from which the structurally related scytophycins are obtained. See ref 7.
- (199) Review: Michael, J. P.; Pattenden, G. Angew. Chem., Int. Ed.
- Engl. 1993, 32, 1.
  (200) (a) Knight, D. W.; Pattenden, G.; Rippon, D. E. Synlett 1990, 36. (b) Pattenden, G. J. Hetereocycl. Chem. 1992, 29, 607. (c) Kiefel, M. J.; Maddock, J.; Pattenden, G. Tetrahedron Lett. 1992, 33, 3227. (d) Maddock, J.; Pattenden, G.; Wight, P. G. J. Comput. Aided Mol. Des. **1993**, 7, 573.
- (201) Note that a plausible biosynthetic pathway to ulapualide A, first proposed by Moore and co-workers, involves formation of the tris-(oxazole) moiety via three Beckmann rearrangments of a precursor having a 22-membered lactone ring, and that this is the same
- ring size as is found in scytophycin C. See ref 16a. (202) Yoo, S.-K. Tetrahedron Lett. 1992, 33, 2159.
- (203) For a recent synthesis of functionalized oxazoles which should
- (206) For a recent synthesis of infictionalized byazones which should also be applicable to the preparation of tris(oxazoles), see: Wipf, P.; Miller, C. P. J. Org. Chem. 1993, 58, 3604.
  (204) Evans, D. L.; Minster, D. K.; Jordis, U.; Hecht, S. M.; Mazzu, A. L., Jr.; Meyers, A. I. J. Org. Chem. 1979, 44, 497.
  (205) (a) Connell, R.; Scavo, F.; Helquist, P.; Åkermark, B. Tetrahedron Lett. 1986, 27, 5559. (b) Connell, R. D.; Tebbe, M.; Helquist, P.; Åkermark, B. Tetrahedron Lett. 1991, 32, 17.
  (206) Hamada V.; Shibata M.; Shipiri T. Tatrahedron Lett. 1985, 26.
- (206) Hamada, Y.; Shibata, M.; Shioiri, T. Tetrahedron Lett. 1985, 26,
- For reviews of macrocyclic stereocontrol strategy, see: (a) Vedejs, E.; Dolphin, J. M.; Gapinsku, D. M.; Mastalerz, H. In *Current* Trends in Organic Synthesis; Nozaki, H, Ed.; Pergamon Press: Oxford, 1983; pp 221-232. (b) Still, W. C. In Current Trends in Organic Synthesis; Nozaki, H., Ed.; Pergamon Press: Oxford, 1983; pp 233-246.
- (208) Kozikowski, A. P.; Stein, P. D. J. Org. Chem. 1984, 49, 2301.
  (209) Penning, T. D.; Djuric, S. W.; Haack, R. A.; Kalish, V. J.; Miyashiro, J. M.; Rowell, B. W.; Yu, S. S. Synth. Commun. 1990,
- 20, 307.

  (a) Bryostatin 1: Pettit, G. R.; Herald, C. L.; Doubek, D. L.; Herald, D. L.; Clardy, J.; Arnold, E. J. Am. Chem. Soc. 1982, 104, 6846. (b) Bryostatin 2: Pettit, G. R.; Herald, C. L.; Kamano, Y.; Gust, D.; Aoyagi, R. J. Nat. Prod. 1983, 46, 528. (c) Bryostatin 3: Pettit, G. R.; Herald, C. L.; Kamano, Y. J. Org. Chem. 1983, 48, 5354. (d) Bryostatin 4: Pettit, G. R.; Kamano, Y.; Herald, C. L.; Tozawa, M. J. Am. Chem. Soc. 1984, 106, 6768. (e) Bryostatins 5-7: Pettit, G. R.; Kamano, Y.; Herald, C. L.; Tozawa, M. Can. J. Chem. 1985, 63, 1204. (f) Bryostatin 8: Pettit, G. R.; Kamano, Y.; Aoyagi, R.; Herald, C. L.; Doubek, D. L.; Schmidt, J. M.; Rudloe, J. J. Tetrahedron 1985, 41, 985. (g) Bryostatin 9: Pettit, G. R.; Kamano, Y.; Herald, C. L. J. Nat. Prod. 1986, 49, 661. (h) Bryostatins 10 and 11: Pettit, G. R.; Kamano, Y.; Herald, C. L. J. Org. Chem. 1987, 52, 2848. (i) Bryostatins 12 and 13: Pettit, G. R.; Leet, J. E.; Herald, C. L.; Kamano, Y.; Boettner, F. E.; Baczynskyj, L.; Nieman, R. A. J. Org Chem. 1987, 52, 2854. (j) Bryostatins 14 and 15: Pettit, G. R.; Gao, F.; Sengupta, D.; Coll, J. C.; Herald, C. L.; Doubek, D. L.; Schmidt, J. M.; Van Camp, J. R.; Rudloe, J. J.; Nieman, R. A. Tetrahedron 1991, 47, 3601. (k) Review: Pettit, G. R. The Chemist 1989, 11. Pettit, G. R. Fortschritte (Prog. Chem. Org. Nat. Prod.) 1991, 57, 153 and references cited therein. (210) (a) Bryostatin 1: Pettit, G. R.; Herald, C. L.; Doubek, D. L.;
- 57, 153 and references cited therein.
- (212) (a) Masamune, S. Chimia 1988, 42, 210. (b) Masamune, S. Pure

- Appl. Chem. 1988, 60, 1587. (c) Blanchette, M. A.; Malamas, M. S.; Nantz, M. H.; Roberts, J. C.; Somfai, P.; Whritenour, D. C.; Masamune, S.; Kageyama, M.; Tamura, T. J. Org. Chem. 1989, 54, 2817. (d) Duplantier, A. J.; Nantz, M. H.; Roberts, J. C.; Short, R. P.; Somfai, P.; Masamune, S. Tetrahedron Lett. 1989, 20, 7037. (c) Kommun. M. Towley, T. Natz, M. H. P. 30, 7357. (e) Kageyama, M.; Tamura, T.; Nantz, M. H.; Roberts, J. C.; Somfai, P.; Whritenour, D. C.; Masamune, S. J. Am. Chem. Soc. 1990, 112, 7407.
- (213) (a) De Brabander, J.; Vanhessche, K.; Vandewalle, M. Tetrahedron Lett. 1991, 32, 2821. (b) De Brabander, J.; Vandewalle, M. Synlett 1994, 231. (c) De Brabander, J.; Vandewalle, M. Synthesia 1994, 255.

- Synlett 1994, 231. (c) De Brabander, J.; Vandewalle, M. Synthesis 1994, 855.
  (214) (a) Roy, R.; Rey, A. W.; Charron, M.; Molino, R. J. Chem. Soc., Chem. Commun. 1989, 1308. (b) Roy, R.; Rey, A. W. Synlett 1990, 448. (c) Roy, R.; Rey, A. W. Can. J. Chem. 1991, 69, 62.
  (215) Hale, K. J.; Lennon, J. A.; Manaviazar, S.; Javaid, M. H.; Hobbs, C. J. Tetrahedron Lett. 1995, 36, 1359.
  (216) (a) Evans, D. A.; Carreira, E. M. Tetrahedron Lett. 1990, 31, 4703. (b) Evans, D. A.; Gauchet-Prunet, J. A.; Carreira, E. M.; Charette, A. B. J. Org. Chem. 1991, 56, 741.
  (217) Ohmori, K.; Suzuki, T.; Miyazawa, K.; Nishiyama, S.; Yamamura, S. Tetrahedron Lett. 1993, 34, 4981.
  (218) Munt, S. P.; Thomas, E. J. J. Chem. Soc., Chem. Commun. 1989, 480.

- (219) Only the revised synthetic plan will be described in detail here. For details of the earlier route, see refs 212a-c.
- (220) For details of the preparation of  $C_3-C_{10}$  segment 576, see the
- (220) For details of the preparation of C<sub>3</sub><sup>2</sup>-C<sub>10</sub> segment **376**, see the supplementary material in ref 212e.
  (221) Hanessian, S.; Lavallée, P. Can. J. Chem. 1**975**, 53, 2975.
  (222) Masamune, S.; Sato, T.; Kim, B.-M.; Wollmann, T. A. J. Am. Chem. Soc. **1986**, 108, 8279.
- (223) Anderson, R. J.; Henrick, C. A.; Rosenblum, L. D. J. Am. Chem. Soc. 1974, 96, 3654.
- (224) (a) Corey, E. J.; Katzenellenbogen, J. A.; Posner, G. H. J. Am. (224) (a) Corey, E. J.; Katzenenenbogen, J. A.; Posner, G. H. J. Am. Chem. Soc. 1967, 89, 4245. (b) Denmark, S. E.; Jones, T. K. J. Org. Chem. 1982, 47, 4595.
  (225) Hill, C. L.; Whitesides, G. M. J. Am. Chem. Soc. 1974, 96, 870.
  (226) Meyer, C. E.; Rose, W. C. J. Biol. Chem. 1936, 115, 721.
  (227) Fronza, G.; Fuganti, C.; Graselli, P.; Marinoni, G. Tetrahedron

- Lett. 1**979**, 3883.
- (228) Piers, E.; Morton, H. E. J. Org. Chem. 1980, 45, 4263.
  (229) Katsuki, T.; Lee, A. W. M.; Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Tuddenham, D.; Walker, F. J. J. Org. Chem. 1982, 47, 1373. (230) Oikawa, Y.; Tanaka, T.; Horita, K.; Yoshioka, T.; Yonemitsu, O.
- Tetrahedron Lett. 1984, 25, 5393
- (231) Tsunoda, T.; Suzuki, M.; Noyori, R. Tetrahedron Lett. 1980, 21,
- (232) Greck, C.; Grice, P.; Jones, A. B.; Ley, S. V. Tetrahedron Lett. 1987, 28, 5759.
- (233) (a) Masamune, S.; Bates, G. S.; Corcoran, J. W. Angew. Chem., Int. Ed. Engl. 1977, 16, 585. (b) Masamune, S.; Hirama, M.; Mori, S.; Ali, S. A.; Garvey, D. S. J. Am. Chem. Soc. 1981, 103, 1568. (c) Park, P.; Broka, C. A.; Johnson, B. F.; Kishi, Y. J. Am. Chem. Soc. 1987, 109, 6205.
- (234) Lavallée, P.; Ruel, R.; Grenier, L.; Bissonnette, M. Tetrahedron Lett. 1986, 27, 679.

  (235) (a) Iguchi, S.; Nakai, H.; Hayashi, M.; Yamamoto, H. J. Org.
- Chem. 1979, 44, 1363. (b) Mori, Y.; Suzuki, M. Tetrahedron Lett. 1989, 30, 4383.
- (236) Van der Eycken, E.; De Wilde, H.; Deprez, L.; Vandewalle, M. Tetrahedron Lett. 1987, 28, 4759. Salzmann, T. N.; Ratcliffe, R. W.; Christensen, B. G.; Bouffard,
- F. A. J. Am. Chem. Soc. 1980, 102, 6161.
- (238) (a) Gais, H.-J.; Schmiedl, G.; Ball, W. A.; Bund, J.; Hellmann, G.; Erdelmeier, I. *Tetrahedron Lett.* **1988**, *29*, 1773. (b) Rehwinkel, H.; Skupsch, J.; Vorbrüggen, H. Tetrahedron Lett. 1988,
- (239) For an example of the use of such a reaction in a synthesis of a bryostatin segment, see 720 → 698 + 721 in Scheme 64 (ref 217), and 735 → 722 + 736 in Scheme 66 (ref 215).
  (240) Keck, G. E.; Abbott, D. E.; Boden, E. P.; Enholm, E. J. Tetrahedron Lett. 1984, 25, 3927.
  (241) Scrift D. M. Scrift P. M. Scrift P. J. Ord Chart. 1971.
- (241) (a) Seyferth, D.; Marmor, R. S.; Hilbert, P. J. Org. Chem. 1971 36, 1379. (b) Gilbert, J. C.; Weerasooriya, U. J. Org. Chem. 1979, 44, 4997. (c) Gilbert, J. C.; Weerasooriya, U. J. Org. Chem. 1982, 47, **1**837

- 47, 1837.
   (242) Roy, R.; Rey, A. W. Tetrahedron Lett. 1987, 28, 4935.
   (243) Cohen, S. G.; Khedouri, E. J. Am. Chem. Soc. 1961, 83, 4228.
   (244) Fox, C. M. J.; Ley, S. V. Org. Synth. 1987, 66, 108.
   (245) Kobayashi, S.; Mukaiyama, T. Chem. Lett. 1986, 1805.
   (246) For previous examples, see: (a) Reetz, M. T. Angew. Chem., Int. Ed. Engl. 1984, 23, 556. (b) Reetz, M. T. In Organotitanium Reagents in Organic Synthesis; Springer-Verlag: Berlin, 1986.
   (247) Reetz, M. T.; Kesseler, K. J. Chem. Soc., Chem. Commun. 1984, 1079
- (248) Bock, K.; Lundt, I.; Pedersen, C. Acta Chem. Scand. 1981, B35,
- (249) Szarek, W. A.; Zamojski, A.; Tiwari, K. N.; Ison, E. R. Tetrahedron Lett. 1986, 27, 3827.

- (250) Danishefsky, S.; Kobayashi, S.; Kerwin, J. F., Jr. J. Org. Chem. **1982**, 47, 1981.
- Demuth, M.; Palomer, A.; Sluma, H.-D.; Dey, A. K.; Krüger, C.;
- Demoun, M., Falomer, A.; Siuma, H.-D.; Dey, A. K.; Krüger, C.; Tsay, Y.-H. Angew. Chem., Int. Ed. Engl. 1986, 25, 1117.
  Sharpless, K. B.; Amberg, W.; Bennani, Y. L.; Crispino, G. A.; Hartung, J.; Jeong, K.-S.; Kwong, H.-L.; Morikawa, K.; Wang, Z.-M.; Xu, D.; Zhang, X.-L. J. Org. Chem. 1992, 57, 2768.
  Xu, D.; Crispino, G. A.; Sharpless, K. B. J. Am. Chem. Soc. 1992, 114, 7570.
- (254) Caldwell, C. G.; Rupprecht, K. M.; Bondy, S. S.; Davis, A. A. J.
- (254) Caldwell, C. G.; Rupprecht, K. M.; Bondy, S. S.; Davis, A. A. J. Org. Chem. 1990, 55, 2355.
  (255) Gao, Y.; Hanson, R. M.; Klunder, J. M.; Ko, S. Y.; Masamune, H.; Sharpless, K. B. J. Am. Chem. Soc. 1987, 109, 5765.
  (256) For previous examples of the use of such synthons, see: (a) Birch, A. J. J. Chem. Soc. 1944, 430. (b) Birch, A. J.; Fitton, P.; Smith, D. C. C.; Steere, D. E.; Stelfox, A. R. J. Chem. Soc. 1963, 2209. (c) Kirkemo, C. L.; White, J. D. J. Org. Chem. 1985, 50, 1316. (d) Evans, D. A.; Sjogren, E. B. Tetrahedron Lett. 1986, 27, 3119. (e) Bringmann, G.; Künkel, G.; Geuder, T. Synlett. 1990, 256.
  (257) Note that for those members of the bryostatin family bearing a
- (257) Note that for those members of the bryostatin family bearing a substituent at C20, Evans suggested that the geometry of the unsaturated ester at  $C_{21}$  might be controlled in a less elaborate manner by minimization of  $A_{1,3}$  allylic interactions in a simple Wittig reaction, in analogy to literature precedents for substituted cyclohexanones. Thus, Wittig reaction of 2-hydroxycyclohexanone (i) was reported to give only the (E)- $\alpha$ , $\beta$ -unsaturated ester ii (eq 1). See: Garner, P.; Ramakanth, S. J. Org. Chem.

1987, 52, 2629. Such an approach has recently been attempted by Hale et al., during the synthesis of a bryostatin 1 C-ring segment, but negligible stereoselectivity was observed (735 -722 + 736 in Scheme 66). See ref 215.

(258) Krishnamurthy, S. J. Org. Chem. 1981, 46, 4628.

(259) Details of the synthesis of advanced intermediate 755 have not yet been published.

- (260) (a) Ganem, B.; Ikota, N.; Muralidharan, V. B.; Wade, W. S.; Young, S. D.; Yukimoto, Y. J. Am. Chem. Soc. 1982, 104, 6787. (b) Bartlett, P. A.; Maitra, U.; Chouinard, P. M. J. Am. Chem. Soc. 1986, 108, 8068.
- (261) Gustafson, K.; Roman, M.; Fenical, W. J. Am. Chem. Soc. 1989, *111*, 7519.
- (262) Rychnovsky, S. D.; Skalitzky, D. J.; Pathirana, C.; Jensen, P. R.; Fenical, W. J. Am. Chem. Soc. 1992, 114, 671.
- (263) Note that fermentation of the deep sea bacterium has proved unreliable and macrolactin A is no longer available in significant yield by this means (see ref 262). Total synthesis of the
- yield by this means (see ref 262). Total synthesis of the macrolactins thus appears increasingly desirable.

  (264) Benvegnu, T.; Schio, L.; Le Floc'h, Y.; Grée, R. Synlett 1994, 505.

  (265) Donaldson, W. A.; Bell, P. T.; Wang, Z.; Bennett, D. W. Tetrahedron Lett. 1994, 35, 5829.

  (266) (a) Laabassi, M.; Grée, R. Tetrahedron Lett. 1988, 29, 611. (b) Nunn, K.; Mosset, P.; Grée, R.; Saalfrank, R. W. Angew. Chem., Int. Ed. Engl. 1988, 27, 1188. (c) Benvegnu, T.; Martelli, J.; Grée, R. Tetrahedron Lett. 1990, 31, 3145. (d) Donaldson, W. A.; Craig, R.; Spanton, S. Tetrahedron Lett. 1992, 33, 3367. (e) Regions. R.; Spanton, S. Tetrahedron Lett. 1992, 33, 3967. (e) Review: Grée, R. Synthesis 1989, 341.
- (267) (a) Monpert, A.; Martelli, J.; Grée, R.; Carrié, R. Tetrahedron Lett. 1981, 22, 1961. (b) Djedaini, F.; Grée, D.; Martelli, J.; Grée, R.; Leroy, L.; Bolard, J.; Toupet, L. Tetrahedron Lett. 1989, 30,
- (268) For a description of the  $\psi$ -endo/ $\psi$ -exo nomenclature, see: Clinton, N. A.; Lillya, C. P. J. Am. Chem. Soc. 1970, 92, 3058
- (269) Kappes, D.; Gerlach, H.; Zbinden, P.; Dobler, M. Helv. Chim. Acta 1990, 73, 2136.
  (270) Narasaka, K.; Morikawa, A.; Saigo, K.; Mukaiyama, T. Bull.
- Chem. Soc. Jpn. 1977, 50, 2773.
- (271) Epimerization of this type, under acidic conditions, has been observed previously. See: Grée, D.; Grée, R.; Lowinger, T. B.; Martelli, J.; Negri, J. T.; Paquette, L. A. J. Am. Chem. Soc. 1992, *114*, 8841.
- (272) Tomooka, K.; Okinaga, T.; Suzuki, K.; Tsuchihashi, G. Tetrahedron Lett. 1987, 28, 6335. (273) Donaldson, W. A.; Tao, C.; Bennett, D. W.; Grubisha, D. S. J.
- Org. Chem. 1991, 56, 4563.
  (274) Tao, C.; Donaldson, W. A. J. Org. Chem. 1993, 58, 2134.
- (275) Donaldson, W. A.; Shang, L.; Rogers, R. D. Organometallics 1994, 13, 6 and references cited therein.
- 13, 6 and references cited therein.

  (276) For a recent review of the amphidinolides covering the congeners A-H, see: Kobayashi, J.; Ishibashi, M. Chem. Rev. 1993, 93, 1753. See also: (a) amphidinolide J: Kobayashi, J.; Sato, M.; Ishibashi, M. J. Org. Chem. 1993, 58, 2645. (b) amphidinolide K: Ishibashi, M.; Sato, M.; Kobayashi, J. J. Org. Chem. 1993, 58, 6928. (c) amphidinolide L: Tsuda, M.; Sasaki, T.; Kobayashi, J. J. Org. Chem. 1994, 59, 3734. (d) amphidinolide M: Kobayashi, J.; Yamaguchi, N.; Ishibashi, H. J. Org. Chem. 1994, 59,

- 4698. (e) amphidinolide N: Ishibashi, M.; Yamaguchi, N.; Sasaki,
- T.; Kobayashi, J. J. Chem. Soc., Chem. Commun. 1994, 1455. (277) Determination of relative stereochemistry: (a) amphidinolide A: see Kobayashi, J.; Ishibashi, M.; Hirota, H. J. Nat. Prod. 1991, 54, 1435 and ref 276a. (b) amphidinolide K: see ref 276b.
- (278) Determination of absolute stereochemistry: (a) amphidinolide B: see Ishibashi, M.; Ishiyama, H.; Kobayashi, J. Tetrahedron Lett. 1994, 35, 8241. (b) amphidinolide J: see ref 276a. (c) amphidinolide L: see ref 276c.

(279) O'Connor, S. J.; Williard, P. G. Tetrahedron Lett. **1989**, 30, 4637. (280) Boden, C.; Pattenden, G. Synlett **1994**, 181.

- (281) Note that while the structures depicted in ref 279 correspond to those expected from the use of (R)-methyl 3-hydroxy-2-methylpropionate, the starting material was erroneously described in the text as being the S enantiomer.
- (282) Brown, H. C.; Ayyangar, N. R.; Zweifel, G. J. Am. Chem. Soc. 1964, 86, 397.
- Miyaura, N.; Ishiyama, T.; Sasaki, H.; Ishikawa, M.; Satoh, M.; Suzuki, A. *J. Am. Chem. Soc.* **1989**, *111*, 314.

  (a) Kauffmann, T.; Joussen, R. *Chem. Ber.* **1977**, *110*, 3930. (b) Eisch, J. J.; Behrooz, M.; Suresh, K. D. *J. Organomet. Chem.*
- 1985, 285, 121. (285) Musich, J. A.; Rapoport, H. J. Am. Chem. Soc. 1978, 100, 4865. (286) Trost, B. M.; Lynch, J.; Renaut, P.; Steinman, D. H. J. Am. Chem.
- Soc. 1986, 108, 284.
- (287) Cohen, N.; Banner, B. L.; Lopresti, R. J.; Wong, F.; Rosenberger, M.; Liu, Y.-Y.; Thom, E; Liebman, A. A. J. Am. Chem. Soc. 1983, 105, 3661.
- (288) Glattfeld, J. W. E.; Forbrich, L. R. J. Am. Chem. Soc. 1934, 56,
- (289) Barbero, A.; Cuadrado, P.; Fleming, I.; González, A. M.; Pulido, F. J. J. Chem. Soc., Perkin Trans. 1 1992, 327.
- (290) Farina, V.; Krishnan, B. J. Am. Chem. Soc. 1991, 113, 9585.
  (291) Schmitz, F. J.; Gunasekera, S. P.; Yalamanchili, G.; Hossain, M. B.; van der Helm, D. J. Am. Chem. Soc. 1984, 106, 7251. Note that the stereochemistry of C<sub>17</sub> was incorrectly drawn in this reference. A congener, 13-deoxytedanolide, has been isolated from the Japanese marine sponge Mycale adhaerens. See ref 196b.
- (292) (a) Kashman, Y.; Groweiss, A.; Shmueli, U. Tetrahedron Lett. 1980, 21, 3629. (b) Spector, I.; Shochet, N. R.; Kashman, Y.; Groweiss, A. Science 1983, 219, 493. (c) Groweiss, A.; Shmueli, U.; Kashman, Y. J. Org. Chem. 1983, 48, 3512. (d) Kashman, Y.; Groweiss, A.; Lidor, R.; Blasberger, D.; Carmely, S. Tetrahedron 1985, 41, 1905. (e) Kashman, Y.; Lidor, R.; Blasberger, D.; Carmely, S. Tetrahedron Lett. 1986, 27, 1367. (f) Blasberger, D.; Green, D.; Carmely, S.; Spector, I.; Kashman, Y. Tetrahedron Lett. 1987, 28, 459. (g) Blasberger, D.; Carmely, S.; Cojocaru, M.; Spector, I.; Shochet, N. R.; Kashman, Y. Justus Liebigs Ann. Chem. 1989, 1171. (h) Spector I.; Shochet, N. R.; Blasberger, D.; Kashman, Y. Cell Motil. Cytoskl. 1989, 13, 127.
- (293) Latrunculin A has more recently been found in the Pacific nudibranch Chromodoris elisabethina and in the Fijian sponge Spongia mycofijiensis. See, respectively: (a) Okuda, R. K.; Scheuer, P. J. Experientia 1985, 41, 1355. (b) Kakou, Y.; Crews, P.; Bakus, G. J. J. Nat. Prod. 1987, 50, 482.
- (294) Note, however, that a macrodiolide had already been obtained from a marine source. Aplasmomycin, isolated from a strain of Streptomyces griseus found in shallow sea mud, was reported in 1976. See ref 142.
- (295) Note that the structures originally reported for  $\boldsymbol{C}$  and  $\boldsymbol{D}$  were reversed.
- (296) (a) Coné, M.; Brenner, S. L.; Spector, I.; Korn, E. D. FEBS Lett. 1987, 213, 316. (b) Schatten, G.; Schatten, H.; Spector, I.; Cline, C.; Paweletz, N.; Simerly, C.; Petzelt, C. Exp. Cell Res. 1986, 166, 191.
- (297) (a) Zibuck, R.; Liverton, N. J.; Smith, A. B., III. J. Am. Chem. Soc. 1986, 108, 2451. (b) Smith, A. B., III; Noda, I.; Remiszewski, S. W.; Liverton, N. J.; Zibuck, R. J. Org. Chem. 1990, 55, 3977. (c) Smith, A. B., III; Leahy, J. W.; Noda, I.; Remiszewski, S. W.; Liverton, N. J.; Zibuck, R. J. Am. Chem. Soc. 1992, 114, 2995.
  (298) (a) White, J. D.; Kawasaki, M. J. Am. Chem. Soc. 1990, 127.
- 4991. (b) White, J. D.; Kawasaki, M. J. Org. Chem. 1992, 57, 5292.
- (299) Heathcock, C. H. In Asymmetric Synthesis; Morrison, J. D., Ed.; Academic Press: Orlando, FL, 1984; Vol. 3, p 111 and references cited therein.
- For examples of macrolactonization procedures involving carboxyl activation, see *inter alia*: (a) ref 47. (b) ref 48. (c) ref 233a.
- (301) A similar enhancement was used by White et al. in their synthesis of boromycin: see ref 144e.
- synthesis of boromycin: see ref 144e.

  (302) For leading references on dialkylboron triflate-mediated aldol reactions of ketones, see: (a) Mukaiyama, T.; Inoue, T. Chem. Lett. 1976, 559. (b) Inoue, T.; Uchimaru, T.; Mukaiyama, T. Chem. Lett. 1977, 153. (c) Inoue, T.; Mukaiyama, T. Bull. Chem. Soc. Jpn. 1980, 53, 174. (d) Evans, D. A.; Vogel, E.; Nelson, J. V. J. Am. Chem. Soc. 1979, 101, 6120. (e) Evans, D. A.; Nelson, J. V.; Vogel, E.; Taber, T. R. J. Am. Chem. Soc. 1981, 103, 3099. (f) Masamune, S.; Mori, S.; Van Horn, D.; Brooks, D. W. Tetrahedron Lett. 1979, 1665. (g) Hirama, M.; Masamune, S.

- Tetrahedron Lett. 1979, 2225. (h) Van Horn, D. E.; Masamune, S. Tetrahedron Lett. 1979, 2229. (i) Hirama, M.; Garvey, D. S.; Lu, L. D. L.; Masamune, S. Tetrahedron Lett. 1979, 3937.
- (303) A similar equilibration was previously observed in the synthesis of the talaromycins. See: (a) Smith, A. B., III; Thompson, A. S. J. Org. Chem. 1984, 49, 1469. See also: (b) Mrozik, H.; Eskola, P.; Arison, B. H.; Albers-Schönberg, G.; Fisher, M. H. J. Org. Chem. 1982, 47, 489.
- C<sub>13</sub> epimerizations in latrunculins have also been observed by Kashman. See refs 292c and 292f.
- (305) (a) Yoshimura, J.; Yamaura, M.; Suzuki, T.; Hashimoto, H. Chem. Lett. 1983, 1001. (b) Williams, R. M.; Armstrong, R. M.; Dung, J.-S. J. Med. Chem. 1985, 28, 733.
- (306) Williams, R. M.; Kwast, E. Tetrahedron Lett. 1989, 30, 451 and references cited therein.
- (307) White, J. D.; Jensen, M. S. Tetrahedron Lett. 1992, 33, 577.
- (308) Mori, K.; Ikunaka, M. Tetrahedron 1984, 40, 3471.
- (309) For a discussion of the stereochemistry of the Wittig reaction of allylic phosphoranes, see: Barlow, L.; Pattenden, G. J. Chem. Soc., Perkin Trans. 1 1976, 1029.
- (310) Büchi, G.; Pawlak, M. J. Org. Chem. 1975, 40, 100.
- (311) Buchta, E.; Andrée, F. Justus Liebigs Ann. Chem. 1961, 640,
- (312) For earlier examples involving addition of ketone enolates to 893, followed by intermolecular and intramolecular Wittig reactions, see respectively: (a) Fuchs, P. L. Tetrahedron Lett. 1974, 4055. (b) Martin, S. F.; Desai, S. R. J. Org. Chem. 1978, 43, 4673. (313) Alderdice, M.; Spino, C.; Weiler, L. Tetrahedron Lett. 1984, 25,
- 1643.
- (314) Spino, C.; Weiler, L. Tetrahedron Lett. 1987, 28, 731.
- Sonoda, N.; Yamamoto, G.; Natsukawa, K.; Kondo, K.; Murai, S. Tetrahedron Lett. 1975, 1969.

- (316) Maclaren, J. A. Aust. J. Chem. 1968, 21, 1891. (317) Logue, M. W.; Teng, K. J. Org. Chem. 1982, 47, 2549. (318) Tapiolas, D. M.; Roman, M.; Fenical, W.; Stout, T. J.; Clardy, J. J. Am. Chem. Soc. 1991, 113, 4682.
- (319) Other medium-ring marine lactones have also been obtained from marine sources. For instance, ascidiatrienolide A (iii) was isolated from the marine ascidian *Didemnum candidum* (see: Lindquist, N.; Fenical, W. Tetrahedron Lett. 1989, 30, 2735). However, ascidiatrienolide A does not show any appreciable biological activity, and is thus outside the scope of this review, although it should be noted that crude extracts from Didemnum candidum did display strong in vitro inhibition of the enzyme phospholipase  $A_2$ . Note that ascidiatrienolide A was originally misassigned by Lindquist and Fenical as the 9-membered lactone iv, but the structure of the natural product was later revised to the 10-membered lactone iii as a result of total synthesis of both iii and iv by Holmes and co-workers (see: Congreve, M. S.; Holmes, A. B.; Hughes, A. B.; Looney, M. G. J. Am. Chem. Soc. 1993, 115, 5815).

- (320) Buszek, K. R.; Sata, N.; Jeong, Y. J. Am. Chem. Soc. 1994, 116,
- (321) McWilliams, J. C.; Clardy, J. J. Am. Chem. Soc. 1994, 116, 8378.
   (322) Nicolaou, K. C.; McGarry, D. G.; Somers, P. K.; Kim, B. H.;
   Ogilvie, W. W.; Yiannikouros, G.; Prasad, C. V. C.; Veale, C. A.; Hark, R. R. J. Am. Chem. Soc. 1990, 112, 6263.

  (a) Corey, E. J.; Nicolaou, K. C. J. Am. Chem. Soc. 1974, 96,
- 5614. (b) Gerlach, H.; Thalmann, A. Helv. Chim. Acta 1974, 57,
- (324) Corey, E. J.; Brunelle, D. J.; Stork, P. J. Tetrahedron Lett. 1976,
- (325) Koppenhoefer, B.; Schurig, V. Org. Synth. 1988, 66, 160.
  (326) Hart, D. W.; Blackburn, T. F.; Schwartz, J. J. Am. Chem. Soc. 1975, 97, 679.
- Takai, K.; Hotta, Y.; Oshima, K.; Nozaki, H. Tetrahedron Lett. 1978, 2417
- (328) (a) Narasaka, K.; Soai, K.; Aikawa, Y.; Mukaiyama, T. Bull. Chem. Soc. Jpn. 1976, 49, 779. (b) Urech, R. J. Chem. Soc., Chem. Commun. 1984, 989. (c) Asaoka, M.; Takei, H. Tetrahe-
- dron Lett. 1987, 28, 6343.
  (329) (a) Herrmann, J. L.; Schlessinger, R. H. J. Chem. Soc., Chem. Commun. 1973, 711. (b) Grieco, P. A. Synthesis 1975, 67.
- (330) Veysoglu, T.; Mitscher, L. A. Tetrahedron Lett. 1981, 22, 1299.
  (331) (a) Uchida, K.; Utimoto, K.; Nozaki, H. J. Org. Chem. 1976, 41, 2941. (b) Uchida, K.; Utimoto, K.; Nozaki, H. Tetrahedron 1977,
- 33, 2987. (a) Miller, R. B.; McGarvey, G. J. Org. Chem. 1978, 43, 4424. (b) Miller, R. B.; McGarvey, G. J. Org. Chem. 1979, 44, 4623.

CR941118Z