

## An Improved Synthesis of the Natural Product Isorhapontigenin

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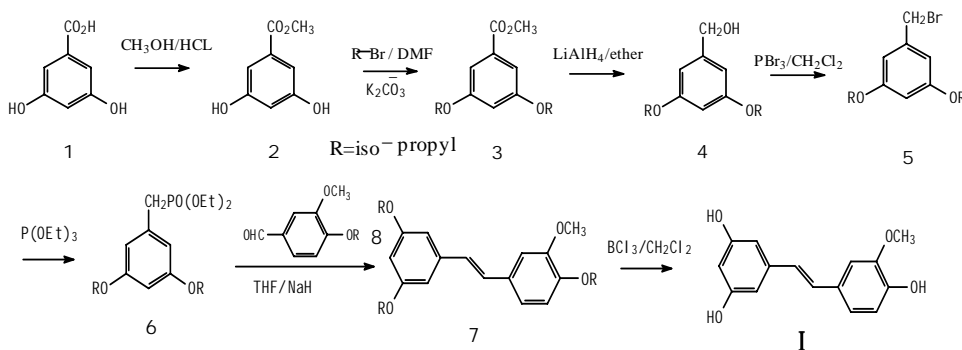
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**Abstract:** We have developed an alternative route to synthesize the natural product, isorhapontigenin. The synthetic product was characterized by IR and  $^1\text{H-NMR}$  in comparison with the corresponding natural product.

**Keywords:** Total synthesis, isorhapontigenin.

Isorhapontigenin (I) is a natural product with many biological activities, M. Lin *et al* first isolated it from the traditional Chinese herb, *G. parvifolium*<sup>1</sup>. The total synthesis of (I) has been reported<sup>2</sup> using Wittig reaction and TMS as the protecting group, the reaction gives both *E* and *Z*-stilbene, in an attempt to obtain enough sample of (I) for screening its bioactivities, we have developed an alternative synthetic route to prepare isorhapontigenin through a 7 steps reaction sequence<sup>3</sup>. However, the final deprotecting step is hard to operate and the yield is low. Herein we designed another route<sup>4</sup> as shown in **scheme 1**. Isopropyl groups were used to protect phenolic groups instead of benzyl groups and finally, the deprotection of isopropyl groups were carried out by  $\text{BCl}_3$ .

**Scheme 1** The synthetic route of isorhapontigenin



Starting from 3,5-dihydroxybenzoic acid **1**, the compound **2** was prepared in good yield by esterification, followed by protection of phenolic groups as iso-propyl ether

(compound **3**). Subsequent reduction by  $\text{LiAlH}_4$  to give compound **4**, through bromination and phosphonation, we obtained compound **6**.

Compound **6** reacted with the protected aldehyde **8** using Wittig-Horner reaction to give the precursor **7**. Finally, the protective iso-propyl groups were removed by  $\text{BCl}_3$  at  $0^\circ\text{C}$  under nitrogen. The product (**I**) was purified by column chromatography and characterized by IR, TLC and  $^1\text{H-NMR}$ . The spectral data were identical with that of the corresponding natural product.

### Acknowledgment

We are grateful to Professor M. Lin, who kindly provided the natural product.

### References and Notes

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2. E. Reimann, *Liebigs Ann. Chem.*, **1971**, 750, 109.
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5. For compound **7**, mp  $67-68^\circ\text{C}$ ;  $^1\text{H-NMR}$ , 300MHz,  $\text{DMSO-d}_6$   $\delta$  ppm: 7.189 (s, 1H, 4'-H), 7.040 (d, 1H, J=8.7Hz, 6-H), 6.921 (d, 1H, J=8.7Hz, 5-H), 6.666 (d, 2H, J=2.1Hz, 2', 6'-H), 6.295 (s, 1H, 2-H), 4.596 (m, 3H, -CH-Me<sub>2</sub>), 3.786 (s, 3H, OCH<sub>3</sub>), 1.246 (d, 12H, J=6Hz, CH<sub>3</sub>-3', 5'), 1.234 (d, 6H, J=6Hz, CH<sub>3</sub>-4) 7.148, 6.998, each 1H (d, J=16.5Hz,  $\alpha$ ,  $\beta$ -H); elemental analysis:  $\text{C}_{24}\text{H}_{32}\text{O}_4$ , calcd. C: 74.97%, H 8.39%, found C 74.87%, H 8.42% For compound **I**, mp  $195-197^\circ\text{C}$  (mp  $180-183^\circ\text{C}^1$ ;  $183-185^\circ\text{C}^3$ ); IR (KBr) 3327 (broad), 1608, 1518, 955, 847.  $^1\text{H-NMR}$ , 300MHz  $\text{CD}_3\text{COCD}_3\text{-d}_6$   $\delta$  ppm: 7.192 (d, 1H, J=1.8Hz, 2'-H), 6.982 (dd, 1H, J=8.1Hz, 1.8Hz, 6'-H), 6.792 (d, 1H, J=8.1Hz, 5'-H), 6.510 (d, 2H, J=2.1Hz, 2, 6-H), 6.240 (t, 1H, J=2.1Hz, 4-H), 6.996, 6.888, each 1H (d, J=16.5Hz,  $\alpha$ ,  $\beta$ -H), 8.423 (s, 2H, 3, 5-OH,  $\text{D}_2\text{O}$  exchangeable), 7.879 (s, 1H, 4'-OH,  $\text{D}_2\text{O}$  exchangeable), 3.877 (s, 1H, 3'-OCH<sub>3</sub>) its spectrum data and TLC is identical with the corresponding natural product.

Received 9 October 1999