

Oxidation of the Norditerpenoid Alkaloids Isotalatizidine and 6-Epiforsticine

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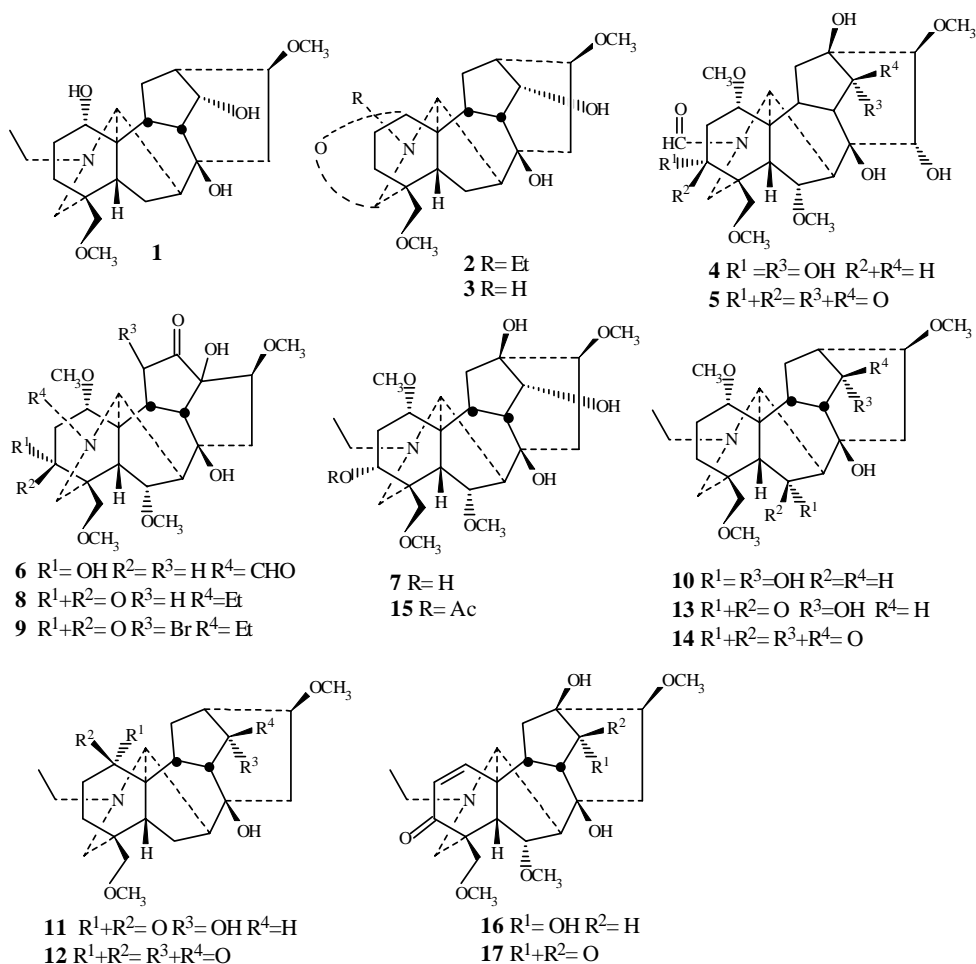
Abstract: Oxidation of **1** with KMnO₄ in acetone-H₂O (1:1) for 1h gave nevadenine **2** (38%). But, **3** was formed in 98% yield when prolonging time and raising temperature (40°C). Reaction of **1** and **10** with Conforth reagent afforded the ketones **11** (55%), **12** (30%), and **13** (13%), **14** (20%), respectively. While oxidation of **7** and **15** with a variety of the oxidizing agents gave **17** only in low 20% of yield besides the minor **16**. In addition, **1** was converted to **2** by the fungi *Curvularia lunata*.

Keywords: Norditerpenoid alkaloid, oxidation, isotalatizidine, 6-epiforsticine, *Curvularia lunata*.

In previous works on modifications of norditerpenoid alkaloids, we reported some important reactions such as *N*-deethylation and preparations of the imines and 7, 17-*seco* derivatives¹. Now we wish to report oxidation of the norditerpenoid alkaloids isotalatizidine and 6-epiforsticine.

Oxidation of norditerpenoid alkaloids having an 1 α -hydroxyl group with reagents such as KMnO₄, K₃Fe(CN)₆, OsO₄, Ag₂O and NBS afforded compounds with an *O*, *N*-mixed acetal systems². Thus, treatment of isotalatizidine **1** with KMnO₄ in acetone-H₂O (5:1) for 1 h gave the known compound nevadenine **2**³ in 38% yield, the products greatly depended upon reaction condition. Compound **3**⁴ was formed in 98% yield when prolonging the reaction time and raising the temperature (40°C). The NMR spectrum of **3** showed signals at δ_{H} 4.07 (1H, s); δ_{C} 82.8 (d) for an *O*, *N*-mixed acetal moiety but absence of an *N*-ethyl group. K. Wiesner, *et al*⁵ reported a method for preparation of **6** starting from **4** by CrO₃/pyridine oxidation *via* an acyloin rearrangement of product **5**. This led us to obtain **9** starting from **7** through oxidation followed by bromination of **8**. Model reaction of **1** and **10**⁶ with Conforth reagent/ pyridine⁷ yielded the ketones **11**⁸ (55%), **12**⁹ (30%), and **13**¹⁰ (13%), **14**¹¹ (20%), respectively. But, oxidation of pseudoaconine **7** and 3-acetyl pseudoaconine **15** with a variety of reagents such as CrO₃/pyridine, Jones' or Conforth reagents, PDC/CH₂Cl₂, afforded the desired product **17**¹² in low yield (20%), besides the minor compound **16**¹³, probably due to hindrance of the 13-hydroxyl group. Treatment of the diketone **17** with 1% NaOH in methanol at

room temperature for 1 day yielded the complex products instead of the desired compound **8**.



In addition, we have found that isotalizidine **1** was converted to neviradine **2** by the fungi *Curvularia lunata*.

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References and Notes

1. F. P. Wang, J. Z. Fang, Z. B. Li, J. S. Yang, B. G. Li, *Chin. Chem. Lett.*, **1999**, *10*, 375; F. P. Wang, Z. B. Li, J. Z. Fang, B. G. Li, *ibid.*, **1999**, *10*, 453; F. P. Wang, J. Z. Fang, X. X. Jian, B. G. Li, *ibid.*, **1999**, *10*, 379; F. P. Wang, J. S. Yang, *J. Nat. Prod.*, to be submitted.
2. F. P. Wang, X. T. Liang, Chemistry of the diterpenoid alkaloids, in “*The Alkaloids: Chemistry and Pharmacology*” (G. A. Cordell, ed.), vol. 42, Academic Press, pp. 152-247, **1992**.
3. A. G. Gonzalez, G. de la Fuente, T. Orribo, R. D. Acosta, *Heterocycles*, **1985**, *23*, 2979.
4. **3**: colorless rhombic crystals, mp. 283-285°C (acetone- n-hexane). C₂₁H₃₁NO₅ (EIMS+¹³C-NMR). ¹H NMR (200 MHz, CDCl₃) δ : 3.31, 3.33 (each 3H, s, 2×OCH₃), 4.07 (1H, s, H-19), 4.20 (1H, t, J=4.8 Hz, H-14 β). ¹³C NMR (50 MHz) δ : 69.1 (1), 23.0 (2), 22.3 (3), 42.8 (4), 36.9 (5), 25.9 (6), 43.8 (7), 72.0 (8), 45.5 (9), 53.5 (10), 46.5 (11), 26.9 (12), 38.5 (13), 75.2 (14), 39.5 (15), 81.6 (16), 55.6 (17), 74.0 (18), 82.8 (19), 56.3 (16'), 59.2 (18'). EIMS m/z (%): 377 (M⁺, 32), 346 (M-31, 100).
5. K. Wiesner, M. Gotz, D. C. Simmons, L. R. Fowler, *Coll. Czech. Chem. Commun.*, **1963**, *28*, 2462.
6. F. P. Wang, Z. B. Li, J. J. Chen, *J. Nat. Prod.*, in press.
7. B. S. Chung, H. K. Lee, *J. Nat. Prod.*, **1986**, *49*, 1074.
8. **11**: white amorphous powder, C₂₃H₃₅NO₅ (EIMS+¹³C NMR). ¹H NMR (200 MHz, CDCl₃) δ : 1.04 (3H, t, J=7.0 Hz, NCH₂CH₃), 2.93 (1H, s, H-17), 3.29, 3.31 (each 3H, s, 2×OCH₃), 3.85 (1H, br.s, disappeared with D₂O, HO-14 α), 4.21 (1H, t, J=4.4 Hz, H-14 β). ¹³C NMR (50 MHz) δ : 215.0 (1), 40.8 (2), 35.4 (3), 38.4 (4), 49.6 (5), 25.3 (6), 45.8 (7), 73.5 (8), 46.3 (9), 39.5 (10), 59.9 (11), 31.2 (12), 39.0 (13), 75.2 (14), 41.0 (15), 81.7 (16), 64.8 (17), 78.2 (18), 53.9 (19), 48.7 (NCH₂-), 13.2 (NCH₂CH₃), 56.1 (16'), 59.3 (18'). EIMS m/z (%): 405 (M⁺, 12), 390 (100), 374 (M-31, 75), 349 (43).
9. **12**: white amorphous powder, C₂₃H₃₃NO₅ (EIMS+¹³C NMR). ¹H NMR (200 MHz, CDCl₃) δ : 1.07 (3H, t, J=7.2 Hz, NCH₂CH₃), 3.06, 3.13 (each 1H, ABq, J=9.0 MHz, H₂-18), 3.29 (6H, s, 2×OCH₃), 3.67 (1H, m, W1/2=17.6 Hz, H-16 α). ¹³C NMR (50 MHz) δ : 214.4 (1), 40.8 (2), 35.2 (3), 38.5 (4), 49.2 (5), 25.4 (6), 46.1 (7), 81.3 (8), 53.8 (9), 37.6 (10), 59.7 (11), 28.7 (12), 46.4 (13), 216.3 (14), 39.7 (15), 86.0 (16), 65.6 (17), 78.2 (18), 53.7 (19), 48.8 (NCH₂-), 13.3 (NCH₂CH₃), 56.0 (16'), 59.3 (18'). EIMS m/z (%): 403 (M⁺, 78), 388 (100), 372 (11).
10. **13**: white amorphous powder, C₂₄H₃₇NO₆ (¹H- and ¹³C-NMR). ¹H NMR (200 MHz, CDCl₃) δ : 1.06 (3H, t, J=7.0 Hz, NCH₂CH₃), 3.29, 3.32, 3.34 (each 3H, s, 3×OCH₃), 4.03 (1H, t, J=5.0 Hz, H-14 β). ¹³C NMR (50 MHz) δ : 84.3 (1), 25.7 (2), 32.4 (3), 37.9 (4), 55.9 (5), 217.7 (6), 58.3 (7), 73.1 (8), 47.7 (9), 45.8 (10), 46.7 (11), 27.1 (12), 37.0 (13), 74.6 (14), 36.2 (15), 81.4 (16), 62.5 (17), 76.9 (18), 53.5 (19), 49.1 (NCH₂-), 13.3 (NCH₂CH₃), 56.2 (1'), 56.6 (16'), 59.2 (18').
11. **14**: white amorphous powder, C₂₄H₃₅NO₆ (¹H- and ¹³C-NMR). ¹H NMR (200 MHz, CDCl₃) δ : 3.32, 3.32, 3.33 (each 3H, s, 3×OCH₃). ¹³C NMR (50 MHz) δ : 84.7 (1), 25.6 (2), 32.1 (3), 38.0 (4), 55.7 (5), 218.6 (6), 58.2 (7), 81.8 (8), 55.7 (9), 43.7 (10), 47.2 (11), 24.8 (12), 46.0 (13), 212.3 (14), 34.8 (15), 84.0 (16), 62.9 (17), 76.6 (18), 53.5 (19), 49.0 (NCH₂-), 13.2 (NCH₂CH₃), 56.2 (1'), 56.2 (16'), 59.2 (18').
12. **17**: white amorphous powder, C₂₄H₃₃NO₇ (¹H- and ¹³C-NMR). ¹H NMR (200 MHz, CDCl₃) δ : 0.98 (3H, t, J=7.0 Hz, NCH₂CH₃), 3.29, 3.42, 3.50 (each 3H, s, 3×OCH₃), 4.28 (1H, d, J=6.2 Hz, H-6 β). 6.45, 6.28 (each 1H, ABq, J=10.2 Hz, H-1 and H-2). ¹³C NMR (50 MHz) δ : 147.3 (1), 132.5 (2), 200.2 (3), 50.1 (4), 48.6 (5), 83.6 (6), 50.7 (7), 86.1 (8), 57.4 (9), 40.1 (10), 51.2 (11), 41.1 (12), 87.9 (13), 215.7 (14), 39.7 (15), 86.8 (16), 60.1 (17), 71.8 (18), 48.8 (19), 48.7 (NCH₂-), 12.8 (NCH₂CH₃), 58.1 (6'), 58.4 (16'), 59.0 (18').
13. **16**: white amorphous powder, C₂₄H₃₁NO₇ (¹H- and ¹³C-NMR). ¹H NMR (200 MHz, CDCl₃) δ : 1.00 (3H, t, J=7.0 Hz, NCH₂CH₃), 3.23, 3.42, 3.50 (each 3H, s, 3×OCH₃), 4.10 (1H, t, J=4.6

Hz, $14\beta\text{-H}$), 6.28, 6.45 (ABq, $J=10.2$ Hz, H-2, H-1). ^{13}C NMR (50 MHz) δ : 147.7 (1), 131.7 (2), 200.6 (3), 49.1 (4), 48.6 (5), 81.5 (6), 52.9 (7), 74.6 (8), 48.7 (9), 37.8 (10), 50.9 (11), 38.1 (12), 76.1 (13), 79.2 (14), 43.3 (15), 82.8 (16), 61.3 (17), 72.0 (18), 51.3 (19), 48.7 (NCH₂-), 12.9 (NCH₂CH₃), 57.9 (6'), 58.1 (16'), 59.0 (18')

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