

**Diastereoselective Michael Addition of
(4'S)-Ethyl-3-(2',2'-Dimethyl-1',3'-dioxolan-4'-yl) acrylate with Furyl
Lithium**

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Abstract: Michael addition of the chiral α,β -unsaturated ester **2** with furyl lithium proceeded in *syn* manner with excellent diastereoselectivity. The diastereoselectivity was not affected by the configuration of the double bond in **2**.

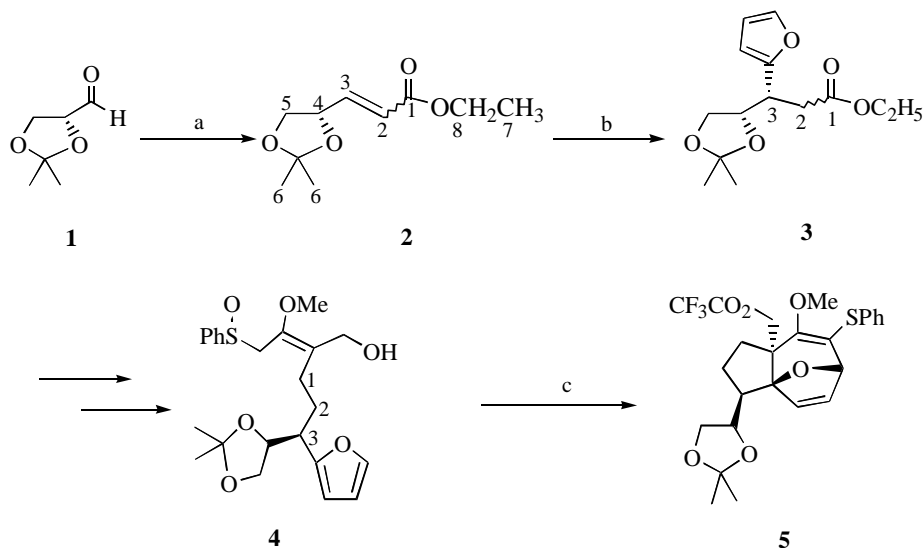
Keywords: Michael addition, configuration determination, *syn* addition.

Recently we reported that the diastereoselective intramolecular [4+3] cycloaddition of **4** led to the formation of cycloadduct **5** as the only isolated isomer in 50% yield. The chiral C-3 of compound **4** could be used to direct the stereochemical course of [4+3] cycloaddition enabling the control of the relative stereochemistry between three contiguous asymmetric centers in adduct **5**¹. The stereogenic center at C-3 of **4** was constructed by the asymmetric Michael addition of α, β -unsaturated ester **2** with furyl lithium. The configuration of C-3 of adduct **3** was deduced to be R based on the Leonard and others reports².

Most recently, we examined the configuration of the newly formed stereogenic C-3 in Michael addition of **2**, particularly with respect to the role of the α, β -unsaturated ester geometry in directing the sense and degree of the diastereoselection in the addition.

A mixture of α, β -unsaturated esters *Z*-**2** and *E*-**2** was prepared in excellent yield (98%) and 5:1 ratio by Wittig reaction of D-glyceraldehyde acetonide **1** with ylide **6** using methanol as reaction solvent. *Z*-**2** and *E*-**2** were easily separated by chromatography. The Michael addition was performed respectively by treatment of an ether solution of *Z*-**2** or *E*-**2** with furyl lithium at -78°C and quenching the resultant anion with aqueous NH_4Cl solution. It is noteworthy that both *Z*-**2** and *E*-**2** gave the same adduct **3** as the only product in 70% yield (**Scheme 1**).

Scheme 1



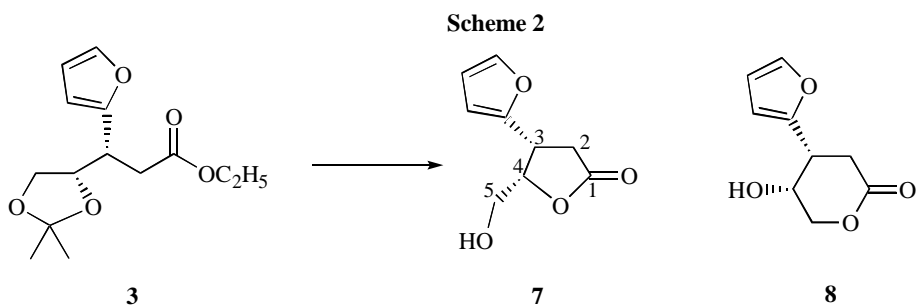
Reagents and reaction conditions:

a) $\text{Ph}_3\text{P}=\text{CHCO}_2\text{C}_2\text{H}_5$ (**6**), CH_3OH , r.t., overnight, 98%;

b) furyl lithium, Et_2O , -78°C , 2 h, 70%;

c) $(\text{CF}_3\text{CO})_2\text{O}$, 2,6-lutidine, CH_2Cl_2 , r.t., 4 h, 50%.

The stereochemical assignment was made after the conversion of **3** to conformationally restricted lactone **7** or **8** via acid-catalyzed transesterification (Scheme 2). In homonuclear decoupling experiment, irradiation on hydroxy proton of lactone changed multiplets at δ 3.52 corresponding to two protons at C-5 into doublets of doublets at 3.51 (dd, 1H, $J = 5.5$, $J = 12.6\text{Hz}$) and 3.61 (dd, 1H, $J = 3.8$, $J = 12.6\text{Hz}$), respectively. This compound also showed a characteristic IR-absorption band of γ -lactone at 1778 cm^{-1} . Therefore, the structure of lactone should be **7**.



Reagents and reaction conditions: *p*-TsOH, MeOH, reflux, overnight.

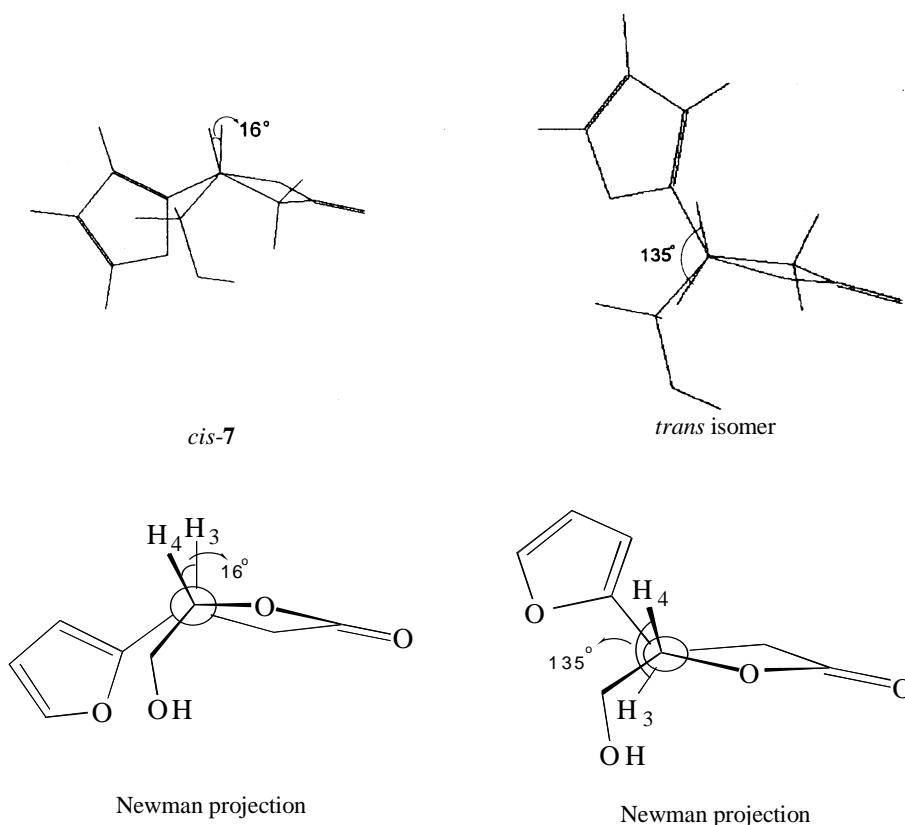
The stereochemistry of lactone **7** was determined based on the coupling constant of H-3,H-4 and *ab initio* calculation of the stable conformations of *cis* **7** and its *trans* isomer. Irradiation on H-2 simplified the signal of H-3 to a doublet at 3.97 (d, 1H, $J=6.9\text{Hz}$). The stable conformations of *cis* **7** and its *trans* isomer were obtained by the

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optimized geometry of lactones by use of Hartree-Fock method with Gaussian 94 program (**Figure 1**). The computational result indicated that in *cis*-form **7** the H₃-C₃-C₄-H₄ dihedral angle was 16° and in *trans*-form, it was 135°. According to Karplus equation, the observed coupling constant ($J = 6.9$ Hz) of H-3,H-4 is consistent with *cis* stereochemistry. In literature, Abraham *et al.*³ reported that the *trans* coupling constant of H-3,H-4 in D-ribo-1→4-lactone was <0.5 Hz and the *cis* coupling constant of H-2-H-3 was 5.5 Hz. Therefore, the absolute configuration of C-3 in compound **3** is assigned as R-form, and our initial, tentative assignment of stereochemistry of compounds **3** and **5**^{1b} were thus verified.

Figure 1 The stable conformations and Newman projections for *cis*-**7** and its *trans* isomer.



In summary, the configuration of C-3 in Michael adduct **3** is assigned as R based on the value of coupling constant between H-3 and H-4 in the conformationally restricted lactone **7** with homonuclear decoupling experiment. The results demonstrated that Michael addition of α,β -unsaturated ester **2** with furyl lithium proceeded in *syn* manner and the geometry of α,β -unsaturated ester had not significant effect on stereoselectivity in Michael addition.

Acknowledgment

This work was partially supported by the National Natural Science Foundation of China.

References and Notes

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4. Selected data for compound E-2: $^1\text{H NMR}$ (300Hz, CDCl_3 , δ ppm): 1.28 (t, 3H, $J = 7.0\text{Hz}$, H-7), 1.38(s, 3H, H-6), 1.42(s, 3H, H-6), 3.60(dd, 1H, $J = 7.1\text{Hz}$, 8.2Hz, H-5), 4.10-4.25(m, H-3, H-5, H-8), 4.62(m, 1H, H-4), 6.05(d, 1H, $J = 15.6\text{Hz}$, H-2), 6.28(dd, 1H, $J = 5.7$, 15.6Hz, H-3).
5. Selected data for compound Z-2: $^1\text{H NMR}$ (300Hz, CDCl_3 , δ ppm): 1.29 (t, 3H, $J = 7.2\text{Hz}$, H-7), 1.38(s, 3H, H-6), 1.45(s, 3H, H-6), 3.61(dd, 1H, $J = 6.9\text{Hz}$, 8.2Hz, H-5), 4.12(q, 2H, $J = 7.2\text{Hz}$, H-8), 4.36(dd, 1H, $J = 7.2\text{Hz}$, 8.2Hz, H-5), 5.48(m, 1H, H-4), 5.83(d, 1H, $J = 11.8\text{Hz}$, H-2), 6.33(dd, 1H, $J = 6.9\text{Hz}$, 11.8Hz, H-3).
6. Selected data for compound 7: IR(film): 3419, 2916, 1778, 1506, 1175, 1043, 1014, 737 cm^{-1} ; $^1\text{H NMR}$ (300MHz, CDCl_3 , δ ppm): 1.82(br, 1H, OH), 2.86(dd, 1H, $J = 9.06$, 17.3Hz, H-2), 2.96(dd, 1H, $J = 7.97$, 17.3Hz, H-2), 3.52(m, 2H, H-5), 3.95(m, 1H, H-3), 4.75(m, 1H, H-4); 6.21(d, 1H, $J = 1.9\text{Hz}$, H-3'), 6.35(m, 1H, H-4'), 7.39(d, 1H, $J = 1.4\text{Hz}$, H-5'); HREIMS for $\text{C}_9\text{H}_{10}\text{O}_4$: calcd: 182.0579; found: 182.0556.

Received 1 December, 2000