

## Resolution of DL-Pantolactone with Ethyldiamine Bridged Dimer Permethy $\beta$ -Cyclodextrin as GC Stationary Phase

Dan Ni TAO<sup>1</sup>, Xue Yan SHI<sup>2</sup>, Jun Ling GU<sup>1</sup>, Ruo Nong FU<sup>1\*</sup>

<sup>1</sup> College of Chemical Engineering and Materials Science, Beijing Institute of Technology, Beijing 100081

<sup>2</sup> Department of Application Chemistry, Chinese Agriculture University, Beijing 100094

**Abstract:** Ethyldiamine bridged dimer permethy- $\beta$ -CD and other cyclodextrins were used as GC stationary phase to resolute DL-pantolactone. It is found that this CD dimer has a good selectivity for DL-pantolactone.

**Keywords:** Cyclodextrin dimer, gas chromatography, stationary phase.

Cyclodextrin dimers (CD dimers) are synthesized by linking two single cyclodextrins with a difunctional spacer. They have been used as enzymatic models for catalytic purpose in the last 20 years, but this separation behaviors as GC stationary phase have been investigated recently<sup>1</sup>. Ethyldiamine bridged dimer permethyl- $\beta$ -CD is synthesized only by linking two single cyclodextrins with an ethyldiamine molecular, which is first used as GC stationary phase. DL-Pantolactone, DL-dihydro-3-hydroxy-4,4-dimethyl-2(3H)-furanone, was found in the degradation product of pantothenic acid from liver by Willimas in 1940. It is an important intermediate in the synthesis of pantothenic acid, D-pantothenyl alcohol and D-pantothenic acid calcium salt, which are usually used in biochemical study<sup>2</sup>. So far the chiral resolution of DL-pantolactone by gas chromatography has not been reported. In this work, the column coated by ethyldiamine bridged dimer permethyl- $\beta$ -CD exhibited good selectivity for DL-pantolactone.

### Column Preparation

Fused-silica capillary tubes (10 m $\times$ 0.25 mm I.D., Yong Nian Optical Fibre factory, Hebei province, China) were roughened by depositing sodium chloride onto their inner wall, the columns were then statically coated at 35 $^{\circ}$ C with 0.5% (w/v) solution of the stationary phase in dichloromethane. Following with conditioning under a slow nitrogen flow at 40 $^{\circ}$ C, 80 $^{\circ}$ C, 120 $^{\circ}$ C, 160 $^{\circ}$ C for 1h each and finally at 200 $^{\circ}$ C for 5h.

### Column Evaluation

A Model 3700 gas chromatography (Beijing Analytical Instrument Factory, Beijing,

China) equipped with a capillary split injection system and flame-ionization detector (FID) was used. Carrier gas was high purity nitrogen. The injection split ratio was 72:1. The injector and detector temperature were 250°C and 260°C respectively.

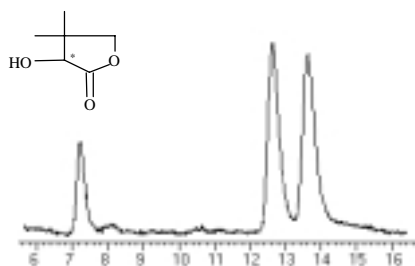
## Results and Discussion

Six columns coated by various CDs stationary phases were chosen. **Table 1** lists the relative retention ( $\alpha$ ) and resolution (R) values of DL-pantolactone on them. **Figure 1** and **Figure 2** show that DL-pantolactone was separated by ethyldiamine bridged dimer permethyl- $\beta$ -CD and heptakis (2,6-di-o-benzyl-3-o-allyl)- $\beta$ -CD bonded polysiloxane. Ethyldiamine bridged dimer permethyl- $\beta$ -CD linking two single cyclodextrins with ethyldiamine spacer gave better selectivity for DL-pantolactone than the rest.

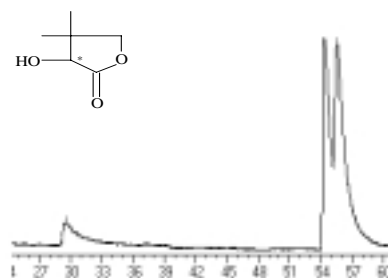
**Table 1** The relative retention ( $\alpha$ ) and resolution (R) values of resolving DL-pantolactone

Stationary phase	T/°C	R	$\alpha$
Ethyldiamine bridged dimer permethyl- $\beta$ -CD	140	0.943	1.09
Heptakis (2,6-di-o-benzyl-3-o-allyl)- $\beta$ -CD bonded polysiloxane	120	0.510	1.05
Mono (2,6-di-o-pentyl-3-o-allyl)-hexakis (2,6-di-o-pentyl-3-o-methyl)- $\beta$ -CD	120	0.000	1.00
Mono (2,6-di-o-pentyl-3-o-propanyl-3')-hexakis (2,6-di-o-pentyl-3-o-methyl)- $\beta$ -CD	120	0.000	1.00
2,6-di-o-pentyl-3-o-benzyl- $\beta$ -CD	120	0.000	1.00
Tri (2,6-di-o-pentyl-3-o-allyl)-hexakis (2-o-pentyl-2-o-allyl-6-o-penylcarbamate)- $\beta$ -CD	120	0.000	1.00

**Figure 1** chromatogram of DL-pantolactone on column 1 at 140°C



**Figure 2** Chromatogram of DL-pantolactone column 2 at 120°C.



## Reference

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