

## Soluble Polymer-supported Synthesis of Indoles *via* Palladium-mediated Heteroannulation of Terminal Alkynes with *o*-Iodoanilines

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**Abstract:** A soluble polymer-supported synthesis of indoles *via* palladium-mediated heteroannulation of terminal alkynes with *o*-iodoanilines has been described. The protocol provides a useful tool for constructing combinatorial indole libraries.

**Keywords:** Indole, polyethylene glycol(PEG), liquid-phase.

Recently, the liquid-phase synthesis of small organic molecules has been a subject of intense research activity<sup>1</sup> since it profits from both the advantageous features of homogeneous solution chemistry and of solid-phase methods. Substituted indoles offer a high degree of structure diversity and have proven to be very important in medicinal chemistry. There have been a few solid-phase methods for the generation of indole-based combinatorial libraries<sup>2</sup>. As we know, however, so far little work has been done to construct indole derivatives using PEG as support on the liquid-phase.

In connection with our research on the PEG as soluble support in liquid-phase synthesis<sup>3</sup>, we wish to report here the first synthesis of indoles on PEG 4000 *via* palladium-mediated heteroannulation of terminal alkynes with *o*-iodoaniline derivatives.

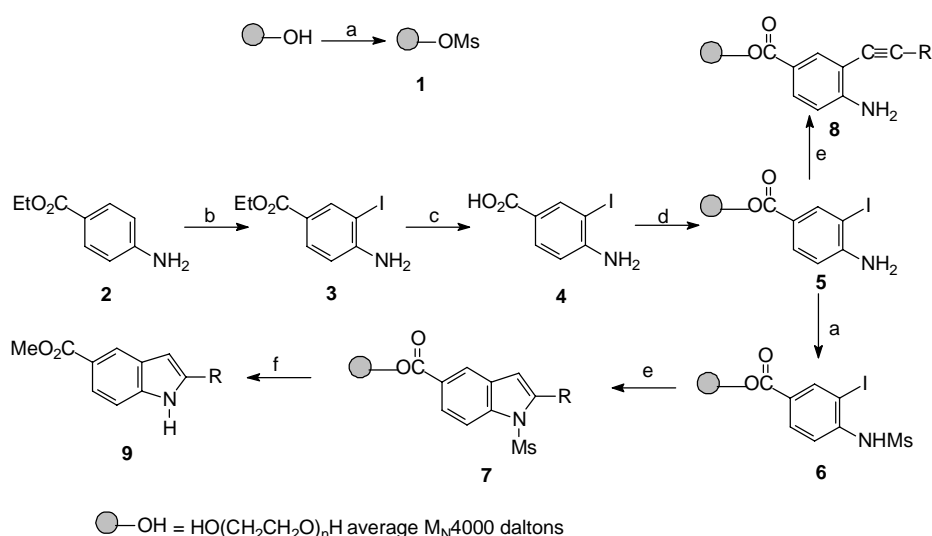
As shown in **Scheme 1**, ethyl 4-aminobenzoate **2** was treated with  $I_2/Ag_2SO_4$ <sup>4</sup>, followed by hydrolysis, to give 4-amino-3-iodobenzoic acid **4**. **4** was attached to the PEG 4000 support by reaction with the modified PEG support **1** in the presence of  $K_2CO_3$  in DMF at 65°C for 16 h. According to the Yamanaka's procedure for solution-phase synthesis of indole<sup>5</sup>, the polymer-supported 4-amino-3-iodobenzoate **5** was mesylated by  $MeSO_2Cl$ , and then reacted with a terminal alkyne in the presence of catalytic amounts of  $PdCl_2(PPh_3)_2$  and  $CuI$  in DMF and  $Et_3N$  at 80°C for 12 h to afford indole **7**. Under the same conditions, if the amino group of **5** was not mesylated *prior to* coupling with a terminal alkyne, an internal alkyne **8**, instead of 2-substituted indole **7**, was isolated. The aryl iodide **6** is coupled with a terminal alkyne to form the  $sp^2$ - $sp$  coupling product, which then undergoes an intramolecular cyclization to form indole ring **7**. In this process, activation of the amine is required<sup>5</sup>. Therefore, when the amine is activated by a strong electron-withdrawing group, sulfonyl, the  $sp^2$ - $sp$  coupling and the indole cyclization can occur in one pot under relatively mild conditions<sup>5</sup>.

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The cleavage of the indoles from the polymer was studied so that a N-unsubstituted indole would directly yield. It was found that transesterification of compounds **7** with methanol in the presence of CH<sub>3</sub>ONa at reflux resulted in 2-substituted indole **9** in excellent yield and purity (**Table 1**). It is important that the yields and purity for this protocol are actually much better than those for solid-phase synthesis<sup>2</sup>. The purity is enough for primary biological screening without further purification.

Scheme 1



Reagents and Conditions: a) MeSO<sub>2</sub>Cl, (C<sub>8</sub>H<sub>17</sub>)<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, RT; b) I<sub>2</sub>, Ag<sub>2</sub>SO<sub>4</sub>, EtOH; c) i) 4 mol/L NaOH, EtOH, ii) 5 Mol/L HCl; d) PEG-OMs, K<sub>2</sub>CO<sub>3</sub>, DMF, 65°C; e) RC≡CH, PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, CuI, Et<sub>3</sub>N, DMF, 80°C; f) CH<sub>3</sub>ONa, CH<sub>3</sub>OH.

**Table 1** Soluble polymer-supported synthesis of indoles<sup>i</sup>

Entry	R	M <sup>+</sup>	Yield <sup>ii</sup> (%)	Purity <sup>iii</sup> (%)
1	n-C <sub>4</sub> H <sub>9</sub>	231	87.4	99.2
2	n-C <sub>3</sub> H <sub>11</sub>	245	83.7	98.5
3	n-C <sub>6</sub> H <sub>13</sub>	259	81.9	97.8
4	Ph	251	86.5	98.7

i. The reaction was carried out with 1.5 g PEG bound molecule **6**, 5% mmol PaCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, 10% mmol CuI, 1.5 mmol corresponding terminal alkyne and 6 mmol Et<sub>3</sub>N in 6 mL DMF at 80°C for 12 h under nitrogen; ii. Yield of **9**, based on the loading level of polymer **6** and all the products are characterized by <sup>1</sup>H-NMR, MS and FT-IR; iii. Determined by GC-MS analysis.

In conclusion, we have developed a facile and efficient method for increasing diversity of combinatorial indole libraries. Further work is in progress to extend this method of liquid-phase synthesis to the preparation of 2,3-disubstituted indoles.

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