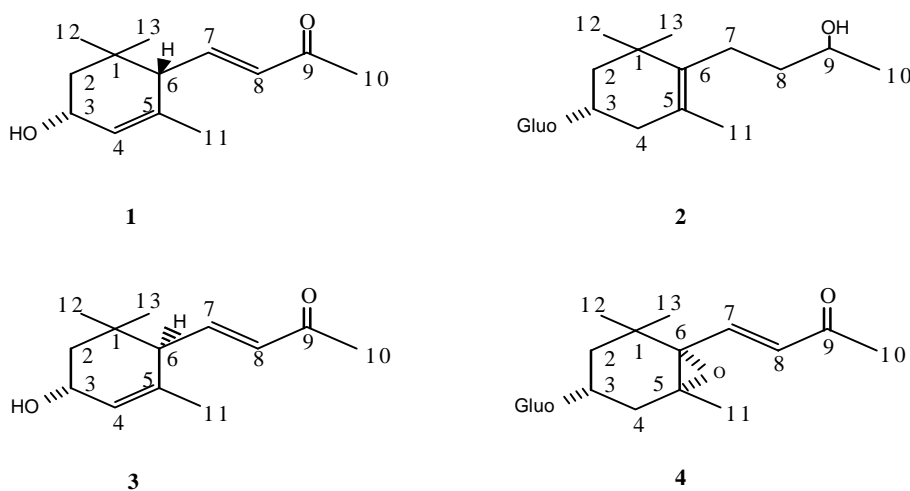


**Two New Ionone Derivatives from *Rhododendron przewalskii* Maxim.**Guo Qiang LI<sup>1,2\*</sup>, Zhong Jian JIA<sup>1</sup><sup>1</sup>Department of Chemistry, National Laboratory of Applied Organic Chemistry,  
Lanzhou University, Lanzhou 730000<sup>2</sup>Department of Applied Chemistry, Qingdao Institute of Chemical Technology, Qingdao 266042**Abstract:** Two new ionone derivatives, named rhododendrone and rhododendronside, were isolated from the alcoholic extract of the aerial parts of *Rhododendron przewalskii* Maxim. Their structures were elucidated on the basis of spectroscopic analysis**Keywords:** *Rhododendron przewalskii* Maxim., Ericaceae, rhododendrone, rhododendronside.

*Rhododendron przewalskii* Maxim. has been used as a folk medicine in China for the treatment of hypertension and coronary heart disease<sup>1</sup>. As a part of our continuing program on the study of plant-derived bioactive compounds, the chemical constituents of *R. przewalskii* Maxim. growing in Gansu were investigated. In the previous paper<sup>2</sup>, we reported eight components from this genus. The present paper deals with the structure elucidation of two new ionone derivatives named rhododendrone **1** and rhododendronside **2** isolated from the EtOH extract of the aerial parts of *R. przewalskii*.

**Figure 1** Structure of **1**, **2**, **3** and **4**

\*E-mail: ligq639@public.qd.sd.cn

Compound **1** was obtained as colorless oil. Its IR spectrum showed the absorption of a hydroxyl group at 3394 and a conjugated carbonyl group at 1671  $\text{cm}^{-1}$ . Its EI mass spectrum displayed a series of fragment ions:  $m/z$  193 ( $\text{M}-\text{CH}_3$ )<sup>+</sup>, 175 ( $\text{M}-\text{CH}_3-\text{H}_2\text{O}$ )<sup>+</sup>, 152, 125, 109 (100%), 91, 43. Its <sup>1</sup>H NMR spectrum showed the presence of four methyl groups ( $\delta$  0.88, s, 3H;  $\delta$  0.98, s, 3H;  $\delta$  1.63, s, 3H;  $\delta$  2.27, s, 3H), two conjugated olefinic proton ( $\delta$  6.08, d, 1H,  $J=17\text{Hz}$ ;  $\delta$  6.64, dd, 1H,  $J=16, 10\text{Hz}$ ), one trisubstituted olefinic proton ( $\delta$  5.59, m, 1H). In the <sup>13</sup>C NMR (DEPT) spectrum of **1**, thirteen carbon signals were assigned to four methyl carbons ( $\delta$  29.7, 27.2, 24.7, 22.6), one methylene carbon ( $\delta$  40.8), five methine carbons ( $\delta$  147.6, 135.4, 126.5, 66.4, 54.4) and three quaternary carbons ( $\delta$  198.2, 132.7, 35.0). Obviously, the signal at  $\delta$  198.2 were attributed to the conjugated carbonyl group. Signals at  $\delta$  147.6, 135.4, 132.7, 126.5 were due to the two conjugated olefinic carbons and the other two olefinic carbons, respectively. Thus the molecular composition of compound **1** was established as  $\text{C}_{13}\text{H}_{20}\text{O}_2$ . The molecular possessed four unsaturated factors, except for two olefinic bond and one carbonyl group, compound **1** might contain a ring structure in the molecular. Comparison with the <sup>13</sup>C NMR and <sup>1</sup>H NMR spectra data of the known compound **3** (3R, 6R)-3-hydroxyl- $\alpha$ -ionone<sup>3</sup> (Table 1), the structure of compound **1** was very similar to that of **3**, the significant differences between them were the chemical shifts of C-2, C-13 and H-6. Therefore, it is implied that **1** might be the stereo-isomer of **3**. It was reported<sup>3</sup> that the chemical shift of 5-Me proton were used to make the determination of the configuration at C-3. The proton signals of 5-Me of **1** and **3** were similar (Table 1). Hence the structural difference between **1** and **3** might just lie in the configuration of C-6. Thus compound **1** was identified to be (3R, 6S)-3-hydroxyl- $\alpha$ -ionone. It is a new compound named rhododendrone.

**Table 1** <sup>1</sup>H and <sup>13</sup>C NMR (DEPT) data for compound **1** (400MHz,  $\text{CDCl}_3$ ,  $\delta$ ppm)

position	<b>1</b>	<b>3</b>	<b>1</b>	<b>3</b>
1			35.0	33.9
2	1.38 (1H, dd, $J=13.4, 10\text{Hz}$ ) 1.71 (1H, dd, $J=13.5, 6.0\text{Hz}$ )	1.38 (1H, dd, $J=13.5, 6.4\text{Hz}$ ) 1.81 (1H, dd, $J=13.5, 6.0\text{Hz}$ )	40.8	43.9
3	4.25 (1H, m)	4.24 (1H, m)	66.4	65.5
4	5.59 (1H, m)	5.60 (1H, m)	126.5	125.9
5			132.7	133.3
6	2.28 (1H, d, $J=10\text{Hz}$ )	2.47 (1H, d, $J=10\text{Hz}$ )	54.4	54.4
7	6.64 (1H, dd, $J=16, 10\text{Hz}$ )	6.49 (1H, dd, $J=16, 10\text{Hz}$ )	147.6	147.0
8	6.08 (1H, d, $J=17\text{Hz}$ )	6.07 (1H, d, $J=17\text{Hz}$ )	135.4	135.4
9			198.2	198.0
10	2.27 (3H, S)	2.23 (3H, S)	27.0	27.2
11	1.63 (3H, S)	1.59 (3H, S)	22.4	22.6
12	0.88 (3H, S)	0.86 (3H, S)	29.1	29.7
13	0.98 (3H, S)	1.00 (3H, S)	27.0	24.7

**Table 2**  $^{13}\text{C}$ NMR (DEPT) data of compound **2** (100MHz,  $\text{CD}_3\text{OD}$ ,  $\delta$  ppm)

C	<b>2</b> (DEPT)	<b>4</b> (DEPT)	C	<b>2</b> (DEPT)	<b>4</b> (DEPT)
1	38.8 (C)	35.5 (C)	11	30.2 ( $\text{CH}_3$ )	29.0 ( $\text{CH}_3$ )
2	47.5 ( $\text{CH}_2$ )	44.7 ( $\text{CH}_2$ )	12	23.2 ( $\text{CH}_3$ )	25.4 ( $\text{CH}_3$ )
3	73.3 (CH)	71.4 (CH)	13	20.0 ( $\text{CH}_3$ )	20.0 ( $\text{CH}_3$ )
4	40.7 ( $\text{CH}_2$ )	37.8 ( $\text{CH}_2$ )	G-1	102.4	103.2
5	125.0 (C)	67.0 (C)	G-2	75.2	75.2
6	138.6 (C)	69.8 (C)	G-3	78.1	78.7
7	25.5 ( $\text{CH}_2$ )	142.9 (CH)	G-4	71.7	71.7
8	39.8 ( $\text{CH}_2$ )	133.3 (CH)	G-5	77.8	78.3
9	69.1 (CH)	197.0 (C)	G-6	62.8	62.8
10	28.8 ( $\text{CH}_3$ )	27.7 ( $\text{CH}_3$ )			

Compound **2** was obtained as colorless oil; EIMS of **2** exhibited a series of ion fragments at  $m/z$  194 ( $\text{M-Glu-H}_2\text{O}$ )<sup>+</sup>, 177 ( $\text{M-Glu-2H}_2\text{O}$ )<sup>+</sup>, 161 ( $\text{M-Glu-2H}_2\text{O-CH}_3$ )<sup>+</sup>, 136, 121 (100%). Acidic hydrolysis of **2** yielded one glucose and one aglycone, and thus **2** was a glycoside confirmed by six carbon signals ( $\delta$  102.4, 75.2, 78.1, 71.7, 77.8, 62.8). Its IR spectrum showed the absorption of a hydroxyl (3433) and an olefinic bond ( $1636\text{ cm}^{-1}$ ). The  $^{13}\text{C}$ NMR and DEPT showed the presence of four  $\text{CH}_3$ , five  $\text{CH}_2$ , seven CH and three C. Then the molecular formula of compound **2** was deduced to be  $\text{C}_{19}\text{H}_{34}\text{O}_7$ , which could be a single ring glucoside except for one glucose and one olefinic bond. The aglycone ( $\text{C}_{13}\text{H}_{23}\text{O}_2$ ) of **2** might possess the similar skeleton of **1** ( $\text{C}_{13}\text{H}_{20}\text{O}_2$ ). In  $^1\text{H}$ NMR spectrum of **2**, an anomeric proton signal was found at  $\delta$  4.43 (d, 1H,  $J=7.8\text{Hz}$ ), therefore, the glucoside linkage should be in  $\beta$  configuration. The signals at  $\delta$  1.04 (s, 3H), 1.06 (s, 3H), 1.17 (d, 3H,  $J=6.1\text{Hz}$ ) and 1.64 (s, 3H) were due to four methyl groups, and obviously, the signal at  $\delta$  1.17 and 1.64 were due to 9-Me protons and 5-Me protons. The absence of the olefinic proton signals suggested that the olefinic bond of **2** was quartersubstituted which was confirmed by two quaternary carbon signals ( $\delta$  125.0, 138.6) of  $^{13}\text{C}$ NMR spectrum data of **2**. Comparison with the known compound 1,1,5-trimethyl-3-O-( $\beta$ -D-glucopyranosyl)-5 $\alpha$ ,6 $\alpha$ -epoxy-6-(butene-1-one-3)-cyclohexane<sup>4</sup> **4**, the obvious difference lied in C-5, C-6, C-7, C-8 (Table 2), that meant there might be some structural varities at C-5, C-6, C-7, C-8 of **2**, olefinic group might placed for the epoxy group of **4** in C-5 and C-6 position, the  $-\text{CH}_2-\text{CH}_2-$  group might replaced for the olefinic bond group of **4** in C-7 and C-8, CHOH group replaced for the carbonyl group of **4** in C-9, which were coincident with the  $^{13}\text{C}$ NMR data of **2** and **4**. The remaining problem was the glucoside position. Analyzing the  $^{13}\text{C}$ NMR spectrum data of the similar compounds<sup>5,6</sup>. It was found that the chemical shifts C-5 and C-9 often appeared at  $\delta$  64.0 and 69.0 in the ionone compounds. However, the formation of glucoside in C-3 or C-9 will make the chemical shift to be changed to downfield 8 ppm approximately. The carbon signals attributed to C-3 and C-9 of **2** were 73.3 and 69.2, respectively, therefore, the glucose moiety should be connected to C-3. Compound **2** was thus identified as 1,1,5 trimethyl-3-O-( $\beta$ -D-glucopyranosyl)-6-(butanol-3)-cyclohexene-5. It is a new compound named rhododendronside.

### References

1. H. R. Yang, S. X.. Wang, *Acta Botanica. Sinica.*, **1978**, 20 (4),355.
2. Z. J. Jia, G. Q. Li, *Chinese Traditional and Herbal Drugs*, **1996**, 27 (5) 262.
3. D. A. Hartman, M. E. Pontones, V.F. Kloss, *J. Nat. Prod.* **1988**, 51 (5), 947.
4. M. Toshio, U. Akira , T. Nohuo, *Chem. Pharm Bull.* **1987**, 35 (3), 1109.
5. M. Toshio, U. Akira , T. Nohuo, *Chem .Pharm. Bull.* **1987**, 35 (9), 3713.
6. Y.H. Gong, “<sup>13</sup>CNMR Chemical Shifts of Narural Organic Compounds”, 1rded., Yunnan Scientific and Technologic Press, Yunnan,**1986**, P.47.

Received 8 April, 2002