

Study on the Synthesis of Metabolite CM2 of Clausenamide

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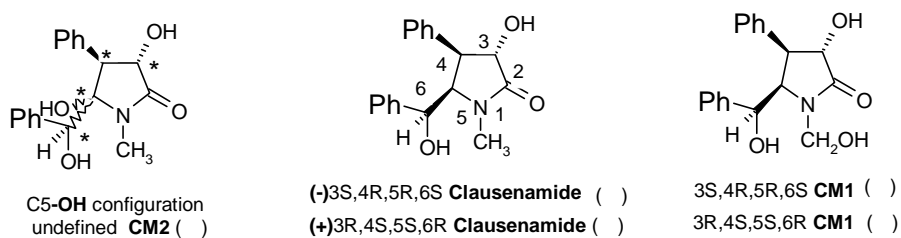
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Abstract: Synthesis of the optically active metabolite of clausenamide **CM2** (3, 5-dihydroxy-5-(α -hydroxybenzyl)-1-methyl-4-benzylpyrrolidin-2-one) from 3-O-acetyl- clausenamide was described.

Keywords: Metabolite, clausenamide, dehydration, dihydroxylation, deacylation.

(-) Clausenamide(I) possesses nootropic effect than its (+) isomer¹. The metabolites of these two optical enantiomers were almost the same, but the content of the metabolites **CM1**(III) and **CM2** (II) of (-) I were much higher than those of (+) I². In order to study the difference of bioactivity between the enantiomers, synthesis of optically active **CM1**(III) and **CM2** (II) has been undertaken. The preparation of **CM1** was reported in the previous paper³, here the synthesis of **CM2** was described.

Figure 1



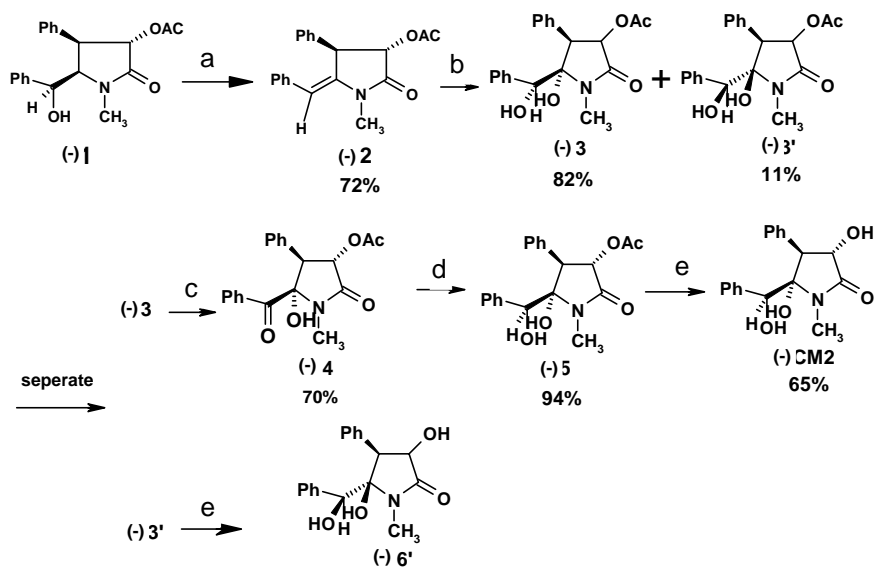
CM2 is the C5 hydroxylated product of of clausenamide (I), the hydroxylation could take place either at the same side (α -face) or at the opposite side (β -face) of the C5-H of (-) I. In any case the configuration of C6 should not be touched. But the direct introduction of the tertiary C5-OH is not an easy task. Therefore the synthetic route through hydroxylation of $\Delta^{5,6}$ (-) clausenamide (**2**) was designed for preparing the metabolite (-)**CM2** as shown in the **Scheme**.

(-)-3-O-acetyl-clausenamide(**1**)(mp 244-246°C, $[\alpha]_D^{14} = -167$ (c 0.106, CHCl₃)) was dehydrated under the condition of POCl₃/pyridine⁴ at ambient temperature to give (-)**2**

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(mp 119-120°C, $[\alpha]_D^{18} = -330$ (c 0.870, CHCl₃)). The NOE-DIFF indicated the double bond is in *trans* form.

Scheme



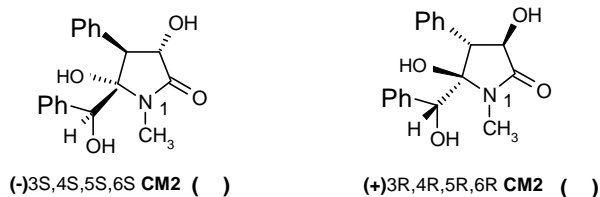
Reagents and conditions: a. POCl₃/Py, b. OsO₄/NMO/THF/Acetone, c. DMSO/oxalyl chloride/THF/TEA, d. NaBH₄/MeOH, e. SmI₂/MeOH

Compound (-) 2 was *cis*-dihydroxylated with OsO₄/NMO to give the main product (-) 3 (mp 125-128°C, $[\alpha]_D^{15} = -323$ (c 0.346, CH₃OH)) with 82% yield and the minor product (-) 3'. The NOE effect between the single sharp peak of C5-OH (δ , 5.66) and the the double peak of C4-H (δ , 3.76) indicated that C5-OH being at the same side (α -side) of C4-H, which means the *cis*-dihydroxylation mainly took place at the less hindered side as shown in **Scheme**. In this way the configuration of C6 was inverted from S as in (-) 1 to R. This most likely is not the right structure. The β -face dihydroxylated product (-) 3' would retained the C6 S configuration and the C5-OH was at the β -face. It was deacylated to give 6'. However the physical constants and spectral data of which did not coincide with what reported for (-) CM2.

From above experimental results, it is evident that (-) CM2 has the C5-OH at the α -face with S configuration of C6. This led to try *trans*-dihydroxylation of (-) 2 with various methods, but all were failed. A pathway through oxidation of the C6-OH in (-) 3 to ketone (-) 4 and then reduced to hydroxyl group was adopted on the basis of the reduction of (-) ketone of clausenamide [(-) clausenamidon] giving stereospecific S configuration⁵ of C6. (-) 3 was oxidized by Swern oxidation to yield α -hydroxyketone (-) 4 (mp 125-128°C, $[\alpha]_D^{14} = -310$ (c 0.360, CHCl₃)). Reduction of the ketone group of (-) 4 with NaBH₄ gave (-) 5 (mp 153-156°C, $[\alpha]_D^{18} = -31.9$ (c 0.455, CH₃OH)). Under the mild and neutral deacylation condition of SmI₂/MeOH⁶ an oil was obtained with identical physical constants and spectral data as reported for (-) CM2 ($[\alpha]_D^{18} = 53.6$ (c

0.497, CH₃OH). (+) Clausenamide($[\alpha]_D^{18} = +54.1$ (c 0.475, CH₃OH)) gave the enantiomer (+) **CM2** by the same process. According to the C5-OH substitution, the absolute configurations of (-)**CM2** and (+)**CM2** were assigned as (3S,4S,5S,6S) and (3R, 4R, 5R, 6R) respectively as shown in **Figure 2**.

Figure 2



Acknowledgment

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