

BIOLOGICAL ACTIVITY OF 1 β -HYDROXY-VITAMIN D₃

Implications for the steric structure of the active form of vitamin D

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1. Introduction

A number of compounds are known to have similar biological activity as the active form of vitamin D₃, the hormonal metabolite 1 α ,25-dihydroxy-vitamin D₃ (ref [1], 1 α ,25-(OH)₂D₃). Among these active materials are 5,6 *trans*-vitamin D₃ (B), dihydrotachysterol₃(C), 3-desoxy-1 α -hydroxy-vitamin D₃ (D) their 25-hydroxy analogues as well as the clinically useful 1 α -hydroxy-vitamin D₃ (A, 1 α -HOD₃). All these compounds possess the vitamin D₃ skeleton substituted at position 1 with a hydroxy group having the same configuration as 1 α -hydroxy-vitamin D₃ (these compounds are drawn up in the respective formulae as 1 β -hydroxy derivatives which is their actual configuration, since they all possess the 6,7-*s trans* conformation). However, nothing, has been reported on the activity of vitamin D₃ analogues having the hydroxy group at C-1 in the enantiomeric configuration. Thus in order to establish whether it is essential for the hydroxy function in vitamin D analogues to possess the α -configuration to elicit biological responses, it was necessary to synthesize the 1 β -hydroxy-vitamin D₃ (1 β -HOD₃) and to determine its activity.

2. Experimental

2.1. Biological assays

The biological response of the animals to vitamin D and its derivatives was assessed by daily dosing of the

steroids over seven days at which time the amounts of the vitamin and the analogues required to maintain an equivalent response was assessed [2]. This type of bioassay overcomes the problem of different time courses of response which arises in comparing the activity of the vitamin D₃ metabolites.

The intestinal response of the chicks to these steroids was assessed by measuring the changes they effected in intestinal calcium-binding protein (CaBP) levels [3] or by measuring the proportion of an oral dose of ⁴⁵Ca absorbed [4]. CaBP was determined by the Laurell method of immunoelectrophoresis using pure chick intestinal CaBP as standard. The degree of endochondrial calcification in rat tibia achieved by the steroids was determined by the radiographic method [5]. In a third type of assay the effect of these steroids on calcium mobilisation from bone of rats and chicks was assessed by recording the changes in plasma calcium levels effected by the steroids in animals on a low calcium diet.

2.2. Chemical synthesis

2.2.1. 1-keto-previtamin D₃ (G)

(i) A solution of 100 mg 1 α -hydroxy-vitamin D₃ (A) in 10 ml dry ethyl ether was treated at room temperature with 350 mg freshly prepared manganese dioxide for 6 h. The reaction mixture was filtered through a celite column and the filtrate was evaporated to dryness at room temperature. Chromatography on silica gel using an ethyl acetate: chloroform mixture (3:7) resulted in 35 mg 1-keto-previtamin D₃ (G). NMR spectrum:

δ ppm 0.74 (3H, s, 18-H), 1.69 (3H, s, 19-H), 5.99

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and 6.13 (2H, AB *q*, *J* 11 Hz, 6-H and 7-H), 5.51 (1H, *m*, 9-H) and 4.02 (1H, heptet, *J* 8.6 and 4.3 Hz, 3-H); ultraviolet λ_{\max} 287, 236 nm (ϵ 10 000, 9500).

(ii) A solution of 100 mg of 1 α -hydroxy-previtamin D₃ (F) in 10 ml ether was treated with 650 mg freshly prepared manganese dioxide for 6 h. Isolation as above resulted in 85 mg 1-keto-previtamin D₃ (G).

2.2.2. 1 β -Hydroxy-previtamin D₃ (H)

A solution of 50 mg 1-keto-previtamin D₃ (G) in 20 ml methanol was treated with 100 mg sodium borohydride at 0°C for 30 min, extracted with ether and washed with brine. The ethereal extract was dried over magnesium sulphate and evaporated at 0°C to dryness. The residue was chromatographed on silica gel. Elution with a mixture of ethyl acetate:chloroform (3:7) gave 35 mg of 1 β -hydroxy-previtamin D₃ (H). NMR spectrum:

δ ppm 0.70 (3H, *s*, 18-H), 1.70 (3H, *s*, 19-H), 5.56 (1H, *s*, 9-H), 5.78 and 5.94 (2H, AB *q*, *J* 11.5 Hz, 6-H and 7-H), 3.93 (1H, broad *s*, $w_{\frac{1}{2}}$ 11 Hz, 3-H) and 4.22 (1H, *m*, broad *s*, $w_{\frac{1}{2}}$ 10.5 Hz 1-H); ultraviolet, λ_{\max} 259 nm (ϵ 10 000).

2.2.3. 1 β -Hydroxy-vitamin D₃ (E)

A solution of 30 mg 1 β -hydroxy-previtamin D₃ (H) in 10 ml iso-octane was heated under a nitrogen atmosphere at 60°C for 4 h. The solvent was evaporated to dryness and the residue was chromatographed

on silica gel. Elution with ethyl ether gave a yield of 25 mg 1 β -hydroxy-vitamin D₃ (E).

2.3. Other chemicals and methods

1 α -(OH)D₃ was synthesised as described previously [6] and vitamin D₃ was a crystalline product obtained from Sigma.

Plasma calcium levels were determined by atomic absorption spectroscopy and protein was determined spectrophotometrically.

Animals, Rhode Island Red \times Light Sussex chicks (1 day old) were raised on a vitamin D-deficient diet and used after 4 weeks when they were deficient [7]. Piebald weanling rats were raised on a low phosphate vitamin D-deficient diet for two weeks by which time they were rachitic. In the appropriate experiments groups of animals were placed on a low calcium vitamin D-deficient diet. The steroids were dissolved in propylene glycol and administered orally at appropriate levels.

3. Results and discussion

In none of the vitamin D-sensitive biological responses used in this study could an effect with 1 β -(OH)D₃ be obtained. As much as 10 μ g of this steroid did not produce any detectable amount of CaBP in chick intestine. Since 0.5 μ g of vitamin D₃ gives a readily measurable amount of this protein (table 1) the potency of this compound in stimu-

Table 1
Relative effectiveness of vitamin D₃, 1 α - and 1 β -hydroxy-vitamin D₃ in chick intestinal CaBP synthesis and rat endochondral calcification

| Compound | Dose level (μ g) | CaBP levels (μ g/mg protein) | Dose level (ng) | Calcification score |
|-------------------------------|-----------------------|-----------------------------------|-----------------|---------------------|
| Vitamin D ₃ | 0.5 | 9.2 | 50 | 3.4 |
| | 1.0 | 24.6 | 100 | 4.3 |
| | 2.0 | 34.0 | 200 | 6.9 |
| 1 α (OH)D ₃ | 2.0 | 35.9 | 50 | 2.6 |
| | 4.0 | 42.5 | 100 | 5.3 |
| | 10.0 | ND | 200 | 0 |

Vitamin D-deficient birds and rats were dosed daily so as to receive the amounts indicated over 7 days. 24 h later the birds were killed and intestinal cytosol prepared for estimation of CaBP (3). ND, no observable precipitation line whatsoever. The rats were X-rayed after 7 days and 14 days and the degree of healing assessed. The scores are those obtained after 14 days

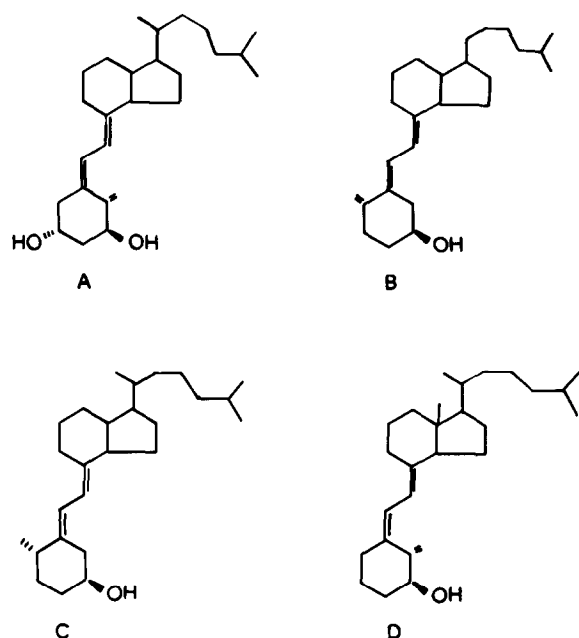


Fig.1. Structures of 1 α -HOD₃ (A); 5,6-*trans*-vitamin D₃ (B); dihydrotachysterol₃ (C); 3-desoxy-1 α -hydroxyvitamin D₃ (D).

lating CaBP synthesis must be assessed as less than 5% of vitamin D₃. Experience with other compounds with very low vitamin D activity suggests that the activity of 1 β -(OH)D₃ is less than 1% of the activity of the parent vitamin. Similar findings were obtained in the other bioassay systems so that for example 0.25 μ g of the 1 β -(OH)D₃ failed to produce any effect on calcification in the rat or on elevation of plasma calcium in rat or chick. In both of these latter systems a good response was obtained to increasing amounts of both vitamins D₃ and 1 α -(OH)D₃.

In the course of this work the 1-keto-vitamin D₃ compound was prepared, but it was found to be unstable and to undergo rapid re-arrangement to 1-keto-previtamin D₃. The biological activity of the 1-keto-vitamin D₃ compound was also low in that 0.5 μ g was unable to increase the proportion of an oral dose of ⁴⁵Ca absorbed whereas 125 ng of vitamin D₃ gave a measurable response.

Since the discovery of 1,25-(OH)₂D₃ as the active form of vitamin D₃ there have been several studies in the different systems responsive to it of the features of the molecule necessary for biological activity (see ref. [1] for recent review). There are several convincing pieces of

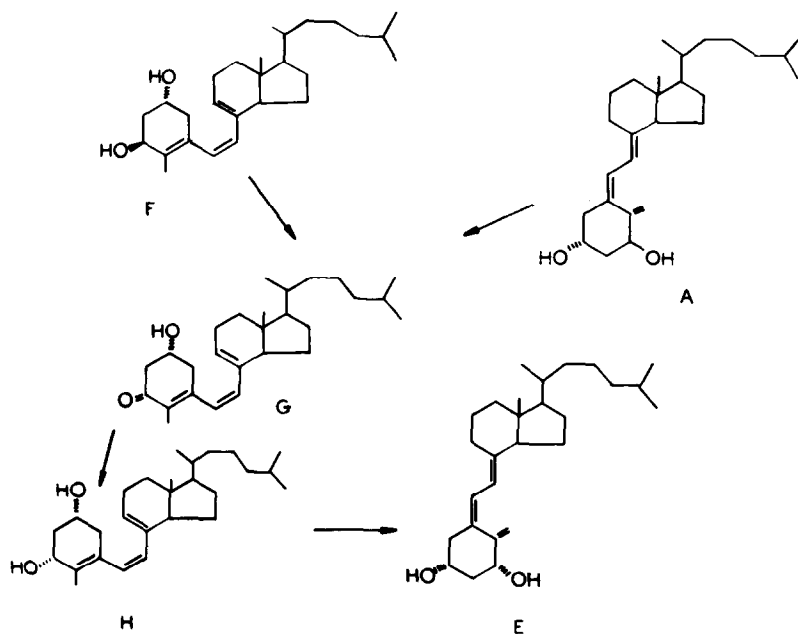


Fig.2. Routes for synthesis of 1 β -HOD₃. 1 β -hydroxyvitamin D₃ (E); 1 α -hydroxy-previtamin D₃ (F); 1-keto-previtamin D₃ (G); 1 β -hydroxy-previtamin D₃ (H).

information that the natural form of the hormone has the hydroxyl group at C-1 in the α -position. Thus the insertion of this hydroxyl group by the 1-hydroxylase to form 1,25-(OH)₂D₃ involves loss of ³H specifically located in the 1 α -position of the substrate [7]. Furthermore chemical syntheses have been described which yield specifically 1 α 25-(OH)₂D₃ and the biological potency of this material corresponds completely with the biologically produced substance [8]. The synthesis of 1 β -(HO)D₃ described here and its very low biological potency confirms that the natural 1,25-(OH)₂D₃ has an almost absolute requirement for the 1-hydroxyl group to be in the α -position (i.e., above the plane of the A ring of the vitamin D molecule).

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