

## FLUORESCENT RHODAMINE-LABELED THYROID HORMONE DERIVATIVES

### Synthesis and binding to the thyroid hormone nuclear receptor

Sheue-yann CHENG, Norman L. EBERHARDT\*, Jacob ROBBINS, John D. BAXTER\* and Ira PASTAN<sup>†</sup>

*Clinical Endocrinology Branch, National Institute of Arthritis, Metabolism and Digestive Diseases, <sup>†</sup>Laboratory of Molecular Biology, National Cancer Institute, National Institutes of Health, Bethesda, MD 20014 and \*Endocrine Research Division, Department of Medicine, Biochemistry and Biophysics, and Metabolic Research Unit, University of California, San Francisco, CA 94143, USA*

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#### 1. Introduction

Studies of the transport, intracellular distribution and interaction of thyroid hormones with specific subcellular components have depended upon the use of radiolabeled hormones. Application of this methodology to the study of transport in intact cells, however, is limited because of the difficulties in differentiating transported hormone from bound hormone at the cell surface. Therefore, it is important to devise alternate techniques for studying the transport of thyroid hormones in intact cells. A highly sensitive technique termed 'video intensification microscopy' has been used [1–3] to study the binding and internalization of fluorescence-labeled peptide hormones and proteins in living cells. To apply this technique to thyroid hormones we have synthesized rhodamine-labeled derivations of iodothyronines directly to visualize their uptake, transport and distribution in cells.

To determine if the derivatives retained biological specificity, we studied their interaction with the thyroid hormone nuclear receptor [7]. A growing body of evidence indicates that many responses of thyroid hormones may be explained by their interaction with a chromatin-localized receptor which subsequently influences the expression of specific genes

(reviewed [4–6]). Studies from a number of laboratories indicate that the receptor binds  $T_3$ ,  $T_4$  and other thyroid hormone analogues in proportion to their biological activity [7–10].

#### 2. Materials and methods

$[^{125}\text{I}]T_3$  and  $[^{125}\text{I}]T_4$ , with spec. act. 1300  $\mu\text{Ci}/\mu\text{g}$  and 1000  $\mu\text{Ci}/\mu\text{g}$ , respectively, were obtained from New England Nuclear. Non-radioactive  $T_4$  and  $T_3$  were purchased from CalBiochem.  $T_0$  was kindly provided by Dr Paul Block, River Research, 4059 River Road, Toledo, OH 43614. TRITC (mixed isomers) was obtained from Research Organic, Cleveland, OH. Preparative and analytical thin-layer plates were from Whatman, New Jersey. Solvent systems for TLC were: (A) upper phase of ethyl acetate:methanol:water (5:2:3, v/v/v) which was adjusted to pH 10.5 with the addition of triethylamine; (B) upper phase of ethyl acetate:methanol:2 N  $\text{NH}_4\text{OH}$  (5:2:3, v/v/v); (C) hexane:*tert*-amyl alcohol:methanol:water (1:6:2:1; v/v/v/v). Absorption spectra were measured in a Cary 14 spectrophotometer.

##### 2.1. Synthesis of $[^{125}\text{I}]T_3$ - and $[^{125}\text{I}]T_4$ -rhodamine

The synthetic scheme for  $[^{125}\text{I}]T_3$ -rhodamine is illustrated in fig.1. In a typical synthesis, 3  $\mu\text{mol}$  each of TRITC and  $[^{125}\text{I}]T_3$  (0.5 Ci/mol) in 0.1 ml pyridine:water:triethylamine (9:1.5:0.1, v/v/v) were reacted for 1 h at room temperature. Purification was carried out by repeated preparative TLC in solvent

**Abbreviations:**  $T_3$ , 3,5,3'-tri-iodo-L-thyronine;  $T_4$ , 3,5,3',5'-tetra-iodo-L-thyronine;  $T_0$ , L-thyronine;  $[^{14}\text{C}]T_0$ , 4-hydroxy-phenoxy- $[^{14}\text{C}]$ tyrosine; TRITC, tetramethyl rhodamine isothiocyanate; TLC, thin-layer chromatography

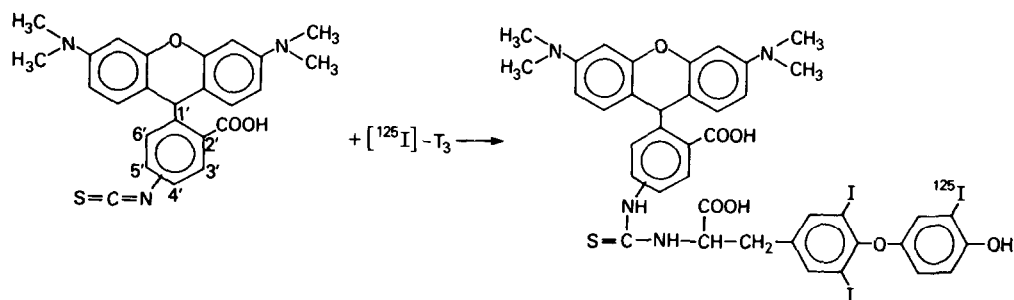


Fig.1. Synthetic scheme for the preparation of  $[^{125}\text{I}]\text{T}_3\text{-rhodamine}$ .

system A. The purple radiolabeled bands with  $R_F \sim 0.25$  and  $\sim 0.27$ , designated as  $\text{T}_3\text{-rhodamine-I}$  and  $\text{T}_3\text{-rhodamine-II}$ , respectively, were eluted with methanol. After evaporation of methanol, the dry residues were stored in the dark at  $-20^\circ\text{F}$  and were stable for periods of at least 6 months. The radiolabeled derivatives were used as markers for purification. The specific activity was maintained sufficiently low so that subsequent nuclear receptor binding assays utilizing  $[^{125}\text{I}]\text{T}_3$  were not affected.

$[^{125}\text{I}]\text{T}_4\text{-rhodamine}$  was synthesized and purified similarly except that the radiolabeled bands eluted had  $R_F \sim 0.13$  and  $\sim 0.15$  were designated as  $[^{125}\text{I}]\text{T}_4\text{-rhodamine-I}$  and  $[^{125}\text{I}]\text{T}_4\text{-rhodamine-II}$ , respectively.

## 2.2. Synthesis of $[^{14}\text{C}]\text{T}_0\text{-rhodamine}$

$[^{14}\text{C}]\text{T}_0$  was synthesized by coupling of 3,5-diiodo- $[^{14}\text{C}]\text{tyrosine}$  with 4-hydroxy-3,5-di-iodophenylpyruvic acid to form  $^{14}\text{C}$ -labeled  $\text{T}_4$  [11] followed by subsequent de-iodination.  $^{14}\text{C}$ -labeled  $\text{T}_4$  was first de-iodinated with a mixture of hypophosphorous acid and hydriodic acid to form  $^{14}\text{C}$ -labeled 3,5-di-iodothyronine [12] which was further de-iodinated to  $[^{14}\text{C}]\text{T}_0$  by catalytic hydrogenation in the presence of 5% Pd on alumina.

The method used to covalently link  $\text{T}_0$  to rhodamine was the same as described for the synthesis of  $\text{T}_3$ - or  $\text{T}_4$ -rhodamine. However, solvent system B was used in preparative TLC and the eluted band had  $R_F \sim 0.15$ . The unresolved 4' and 5' isomeric mixture of  $\text{T}_0$ -rhodamine was used in the nuclear receptor binding assays.

## 2.3. Binding to thyroid hormone nuclear receptors

The binding assays were carried out as detailed in [7].

## 3. Results and discussion

The thyroid hormones,  $\text{T}_3$  and  $\text{T}_4$ , and the biologically inactive  $\text{T}_0$  were attached through their amino groups to tetramethyl rhodamine via a thiourea linkage (fig.1). Since the starting material, TRITC, was commercially obtained as a mixture of 4' and 5' isomers, two isomeric derivatives of  $\text{T}_3$  would be expected. Two radiolabeled rhodamine-containing derivatives with similar chromatographic mobilities were observed during purification and were subsequently isolated. The slow- and fast-migrating derivatives,  $\text{T}_3\text{-rhodamine-I}$  and  $\text{T}_3\text{-rhodamine-II}$  (see section 2), respectively, presumably represent the attachment of  $\text{T}_3$  to the 4' or 5' position of rhodamine. The formation of the  $\text{T}_3$ -rhodamine conjugate was established by:

- Its chromatographic mobility which is lower than either of the initial reactants;
- The presence of  $\text{T}_3$  as determined by incorporation of  $^{125}\text{I}$  radioactivity,
- Its characteristic absorption spectrum which is consistent with the presence of rhodamine and  $\text{T}_3$  moieties. Purity was established by TLC on silica gel and cellulose plates in solvent system A and C. The absence of contamination by free hormone was confirmed by autoradiography and counting segments of the thin-layer chromato-

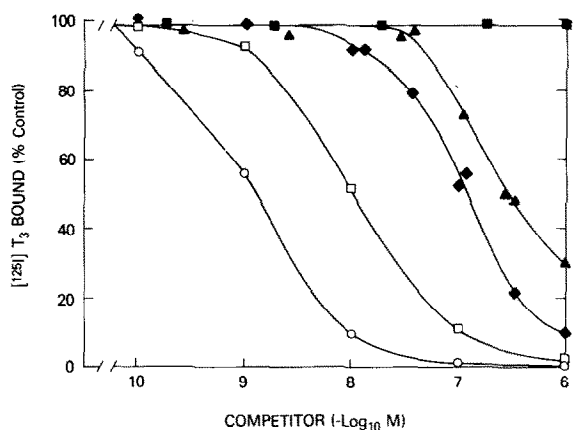


Fig.2. Competition of [ $^{125}$ I] $T_3$  binding to the solubilized thyroid hormone nuclear receptor by unlabeled  $T_3$  (circles),  $T_4$  (open squares),  $T_3$ -rhodamine-II (solid diamonds),  $T_4$ -rhodamine-II (solid triangles) and  $T_0$ -rhodamine (solid squares). The nuclear receptor [7] was solubilized from purified rat liver nuclei and was incubated with [ $^{125}$ I] $T_3$  at 1 nM final conc. with varying concentrations of the indicated competitors for 2 h at 22°C. Bound hormone was separated from free hormone by passage of the incubation mixtures over 2.0 ml Sephadex G-25 columns and subsequent determination of bound radioactivity. The individual data points represent the average of duplicate determinations.

grams. The purity of [ $^{125}$ I] $T_4$ - and [ $^{14}$ C] $T_0$ -rhodamine was established similarly.

Figure 2 shows the results of the competition of [ $^{125}$ I] $T_3$  binding to the receptor by unlabeled  $T_3$ ,  $T_4$ ,  $T_3$ -rhodamine-II,  $T_4$ -rhodamine-II and  $T_0$ -rhodamine. These data indicate that  $T_3$ - and  $T_4$ -rhodamine-II both bind to the receptor with affinities  $\sim 2$  orders of magnitude lower than  $T_3$ . The calculated [7]  $K_d$  values for the binding of  $T_3$ -rhodamine-II and  $T_4$ -rhodamine-II to the receptor are 20.3 nM and 70.8 nM, respectively. Thus, the derivatives bind to the receptor at concentrations which can be reasonably achieved in intact cells. The data also indicate that  $T_3$ -rhodamine-II binds to the receptor more avidly than  $T_4$ -rhodamine-II, although the difference in the affinities of the two derivatives is somewhat smaller than with the parent hormones (fig.2) [7–10]. In addition,  $T_0$ -rhodamine does not interact with the receptor over the concentration range employed in these studies. Since  $T_0$  is biologically inactive, we conclude that the binding of

the thyroid hormone-rhodamine derivatives to the receptor results from biospecific interactions.

The slow- and fast-migrating  $T_3$ - and  $T_4$ -rhodamine derivatives possess slightly different affinities for the receptor (table 1). The fast-migrating derivatives have higher affinities than the slow-migrating derivatives. Although the exact linkage position for each of the derivatives is unknown, the data indicate that the orientation of rhodamine with respect to the hormone has a minor influence on the binding affinities of the derivatives to the receptor.

The synthesis of fluorescein-labeled  $T_4$  has been reported [13]; however, its physico-chemical properties and biological specificity have not been examined. Furthermore, fluorescein has a relatively faster fluorescence bleaching rate than rhodamine thereby decreasing the sensitivity of its detection by 'video intensification microscopy'.

Our data indicate that the rhodamine-labeled derivatives of thyroid hormones interact with nuclear thyroid hormone receptors in a manner which reflects the binding specificities of the parent hormones. These data therefore indicate that these derivatives may be useful in studies of thyroid hormone interaction with intact cells.

Table 1  
Apparent equilibrium dissociation constants for the binding of rhodamine-labeled thyroid hormones to the nuclear receptor

Rhodamine-labeled thyroid hormones	$K_d$ (nM) <sup>a</sup>
$T_3$ -rhodamine-II	20.3
$T_3$ -rhodamine-I	38.3
$T_4$ -rhodamine-II	70.8
$T_4$ -rhodamine-I	203

<sup>a</sup> The dissociation constants were calculated from the equation:

$$K_d = C_{50} \cdot K_T / (T + K_T)$$

where  $K_d$  = apparent dissociation constant of the competitor,  $C_{50}$  = concentration of competitor required to produce 50% inhibition,  $K_T$  = dissociation constant of  $T_3$ , and  $T$  = concentrations of [ $^{125}$ I] $T_3$  utilized in the assay. The receptor preparation utilized in these studies had a  $T_3$   $K_d$  of 0.221 nM as determined by Scatchard analysis

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