## Photocycloaddition Reactions of 5,5-Dimethyl-3-(3-methylbut-3-en-1 ynyl)cyclohex-2-en-1-one

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Dedicated to Professor William C. Agosta on the occasion of his 75th birthday

The title cyclohexenone **1d** undergoes photodimerization selectively at the exocyclic  $C = C$  bond to give a 1 : 1 mixture of 1,2-dialkynyl-1,2-dimethylcyclobutanes 6 and 7. On irradiation in the presence of 2,3-dimethylbuta-1,3-diene, 1d affords bicyclo[8.4.0]tetradeca-1,2,3,7-tetraen-11-one 9. This – formal –  $(6 + 4)$ -cycloadduct undergoes quantitative isomerization to 3-cycloheptadienyl-2,5,5-trimethylcyclohex-2-enone 11 on treatment with basic silica gel.

1. Introduction. – Cyclohex-2-enones 1 undergo photodimerization to mixtures of (regioisomeric) tricyclo[6.4.0.0<sup>2,7</sup>]dodecanediones 2 and 3, the relative ratio of these tricyclic dimers depending on the polarity of the solvent  $[1-3]$ . This is exemplified in *Scheme 1* for 5,5-dimethylcyclohex-2-enone (1a) [4] or isophorone ( $=$  3,5,5-trimethylcyclohex-2-enone; 1b) [5] [6]. Both of these photoexcited enones also react efficiently with alkenes [7] to afford bicyclo<sup>[4.2.0]</sup>octan-2-ones **4**. We had previously shown [8] that replacing the Me group at  $C(3)$  of **1b** by an alkynyl group had a pronounced effect on the photocycloaddition of, e.g., **1c** to 2,3-dimethylbut-2-ene, as now the intermediate triplet 1,4-biradical undergoes selective 1,5-cyclization to afford cyclopentenyl carbene 5 as a reactive intermediate. We have now focussed on extending the conjugation of this  $C(3)$ -side chain by an additional C=C bond. Here, we report on the results of photocycloaddition reactions of 5,5-dimethyl-3-(3-methylbut-3-en-1-ynyl)cyclohex-2 enone (1d).

2. Results. – Irradiation (350 nm) of 0.5m solutions of 1d in either benzene or in MeCN afford 1:5:4 mixtures of three dimers 2d, 6, and 7, respectively (Scheme 2). The same product mixture is obtained when 1d is irradiated in the presence of a tenfold molar excess of 2,3-dimethylbut-2-ene, *i.e.*, cycloaddition of **1d** to this alkene is totally suppressed by photodimerization. Whereas the assignment of the constitution of 6 and 7 as 1,2-dialkynyl-1,2-dimethylcyclobutanes is unambiguous on the basis of their <sup>1</sup>H-NMR spectra, the determination of both constitution and configuration of the minor dimer 2d is not straightforward. Indeed, the assignment of the structural parameters for such tricyclic photodimers of cyclohexenones and related molecules by 1 H-NMR spectroscopy is only unequivocal for cyclic enones bearing no substituents on the olefinic C-atoms, by analysis of the corresponding  $AA'XX'$  patterns [9] [10]. For

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structure determination of the corresponding dimers of cyclic enones bearing a substituent on one of the olefinic C-atoms, a combination of enantioselective gas chromatography and GC/MS analysis [11] can be helpful, but only if these dimers are stable enough under these (thermal) conditions. In this context, it seemed appropriate to select  $^{13}$ C-NMR spectroscopy as possible tool for the differentiation of such dimers. Therefore, we prepared dimers 2b and 3b, and the new dimer 2c by irradiation of 1b and 1c (cf. Scheme 1), confirmed their structures by X-ray crystal-structure analysis, and recorded their 13C-NMR spectra. Comparison of the 13C-NMR parameters of these dimers with those of 2d allows the structural assignment of the latter.

As we already had observed [12] that cyclic enones with low triplet energies  $(<$  55 kcal/mol) undergo photoaddition to 'normal' alkenes, e.g., 2,3-dimethylbut-2ene, very inefficiently if at all, but in contrast very efficiently to dienes or unsaturated nitriles, we irradiated 1d in the presence of a tenfold molar excess of 2,3-dimethylbuta-1,3-diene (8; Scheme 3). NMR Analysis indicates the – quantitative – formation of two new products, 9 and 10, in a 4 : 1 ratio, respectively. Whereas the minor cycloadduct 10, which still contains a  $C\equiv C$  bond, can be isolated by chromatography, the major product 9, which shows no NMR signals for either acetylenic C-atoms or olefinic H-atoms, turns out to be very sensitive towards  $SiO<sub>2</sub>$ . Indeed, stirring of a CH<sub>2</sub>Cl<sub>2</sub> soln. of 9 over  $SiO<sub>2</sub>$ for ca. 30 min leads to its total decomposition. In contrast, on contact with  $Et_3N$ neutralized  $SiO<sub>2</sub>$ , 9 isomerizes quantitatively to 11 (*Scheme 3*).

3. Discussion. – The most striking feature in the photochemical behavior of 1d is the fact that the initial addition step occurs at the terminal olefinic C-atom of the side chain and not at the unsubstituted – i.e.,  $C(2)$  – C-atom of the cyclohexenone moiety, as



reflected by the preferential formation of  $(6 + 7)$  vs. 2d in the photodimerization and of 9 vs. 10 in the mixed cycloaddition. This feature can be best explained by comparing the relative stabilities of the resulting intermediates resulting from the addition of, e.g., an alkyl radical to a 1-acylhexa-1,5-dien-3-yne I. Whereas pent-4-en-2-yn-1-yl radical II (and the mesomeric penta-2,3,4-trien-1-yl radical  $\text{IIa}$  [13-15]) are additionally stabilized by the adjacent C=O group, this is not the case for  $III$  and  $III$ a, respectively

(Scheme 4). Indeed, this correlation between  $a$ ) stabilization of the biradical intermediate and  $b$ ) product ratio is already reflected by the fact that **1b** affords mixtures of 2b and 3b, depending on the polarity of the solvent, whereas 1c affords dimer 2c (stemming from a bis-propargyl biradical intermediate) selectively, both in polar and in nonpolar solvents.



Differentiation between 6 and 7, *i.e.*, the assignment of *trans*- and *cis*-configuration, stems from NMR analysis alone, as we were not able to obtain crystals suited for X-ray crystal-structure determination for neither of these dimers. Therefore, the AA'XX' patterns of the CH<sub>2</sub>CH<sub>2</sub> unit in the cyclobutane ring was simulated by SpinWorks [16], assuming the vicinal coupling constant  $3J$  for the *cis*-H-atoms to be by *ca*. 3 Hz larger than for the corresponding trans-H-atoms, and these results were then compared with the recorded data. Further evidence comes from 13C-NMR data of methylated cyclobutanes [17], where Me shifts for cis- ( $\delta$  15 ppm) and trans-1,2-dimethylcyclobutanes ( $\delta$  20 ppm) differ by ca. 5 ppm. The constitutional assignment (2d and not 3d) for the minor tricyclic dimer of 1d stems from comparison of the 13C-NMR chemical shifts of the cyclobutane C-atoms. Typical chemical shifts for the tertiary C-atoms of 2 are  $50 - 53$  ppm, and 62 ppm for the corresponding C-atoms in 3, while, for the quaternary C-atoms, the relative chemical shifts are inverted, i.e., 43 – 45 ppm for 2 and 37 – 38 ppm for 3 (cf. Exper. Part). The selective formation of the formal  $[6+4]$  photocycloadduct 9 from 1d and 2,3-dimethylbuta-1,3-diene (8) is unprecedented. Apparently, the two radical centres in the biradical intermediate (structure type  $\mathbf{I}/\mathbf{I}$ a; *cf. Scheme 4*) formed by addition of a methylidene unit of 8 to the (exocyclic) methylidene unit of 1d are located at bonding distance already, thus facilitating the second cyclization step, the  $(E)$ -configuration of the C(7)=C(8) bond reflecting the preferential s-trans-conformation of 8. Symmetrically substituted 1,1,4,4-tetraalkylbuta-1,2,3-trienes exhibit typical <sup>13</sup>C-NMR shifts [18] [19], the central olefinic C-atoms resonating at  $\delta$ 160 ppm, and the external ones at  $\delta$  110 ppm. Referring thereto, the values found for

**9** ( $\delta$  112 ppm for C(1), 116 ppm for C(4), 156 ppm for C(3), and 158 ppm for C(2)) give indeed a very good fit. Interestingly, cyclodeca-1,2,3-triene has already been synthesized [20], but no <sup>13</sup>C-NMR spectroscopy was available at that time. Such [3]cumulenes are known to dimerize to radialenes and to react readily with atmospheric oxygen [18]. It is, therefore, not surprising that attempts to purify 9 by chromatography failed. As for the minor cycloadduct 10, the relative configuration of the Me and the methylethenyl group on the cyclobutane ring is unknown. Finally, the base-catalyzed isomerization  $9 \rightarrow 11$  is intriguing, since very few reports on carbanionic rearrangements are found in the literature, and only one of them is classified as 'ringstrain-promoted' [21]. A tentative mechanistic interpretation involves an addition/ elimination sequence followed by a (formal) 1-methylpropa-1,2-diene  $\rightarrow$  buta-1,3diene tautomerization involving intermediates **IV–VI**, the isomerization of a  $(\beta, \gamma)$ unsaturated) 3-methylidenecyclohexanone to a (fully conjugated) cyclohexenone representing the driving force (Scheme 5).





## Experimental Part

1. General. Photolyses were conducted in a Rayonet RPR-100 photoreactor equipped with (16) 350 nm lamps and solvents of spectrophotometric grade. Column chromatography (CC): silica gel 60 (Merck; 230–400 mesh). Gas chromatography (GC): 30-m SE-30 capillary column. <sup>1</sup>H- and <sup>13</sup>C-NMR Spectra (including two-dimensional plots): Bruker WM-500; at 500.13 and 125.8 MHz, resp.,  $\delta$  in ppm, J in Hz. GC/EI-MS: at 70 eV. X-Ray analyses: Bruker APEX CCD three-circle diffractometer at 153 K with Mo $K_a$  radiation ( $\lambda$  0.71073 Å).

2. Starting Materials. 3,5,5-Trimethylcyclohex-2-enone (1b) is commercially available. 3-(3,3-Dimethylbut-1-ynyl)-5,5-dimethylcyclohex-2-enone (1c) was synthesized according to [22]. 5,5-Dimethyl-3-(3-methylbut-3-en-1-ynyl)cyclohex-2-enone (1d) has been prepared by a Pd/Ag-catalyzed vinyl triflate/1- (trimethylsilyl)alkyne coupling in 94% yield [23], but no data of this compound have been reported. We obtained 1d from 3-ethoxy-5,5-dimethyl-cyclohex-2-enone and the Grignard derivative of 2-methylbut-1-en-3-yne in analogy to **1c** [22] in 45% yield after CC ( $Et_2O/pentane 1:1$ ) as colorless oil. <sup>1</sup>H-NMR

 $(CDCl_3)$ : 6.20  $(t, J=1.5)$ ; 5.43, 5.38 (br. s, 2 H); 2.35  $(d, J=1.5, 2$  H); 2.21 (s, 2 H); 1.95 (br. s, 3 H); 1.07  $(s, 6H)$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 199.3  $(s)$ ; 153.2 $(s)$ ; 141.5  $(s)$ ; 131.7  $(d)$ ; 124.7  $(t)$ ; 100.8  $(s)$ ; 88.0  $(s)$ ; 51.4  $(t)$ ; 44.7 (t); 34.1 (s); 28.5 (q); 23.4 (q). EI-MS: 188 (79,  $M^{+}$ ), 132 (100).

3. Photodimerizations. Ar-Degassed solns. of 1 are irradiated (concentration, solvent, duration, and workup as described).

3.1. Photodimerization of 1b in  $H_2O$ . A soln. of 18.22 mg (0.13 mmol) of 1b in  $H_2O$  (2 ml) is irradiated for 48 h. The white precipitate is filtered, dried, and recrystallized from Et<sub>2</sub>O to afford 12.7 mg (71%) of cis-transoid-cis-5,5,7,8,10,10-hexamethyltricyclo[6.4.0.0<sup>2,7</sup>]dodecane-3,12-dione (2b). M.p. 159°.  ${}^{1}H\text{-NMR (CDCl}_3)$ : 2.81 (s, 2 H); 2.47 (d, J = 14.8, 2 H); 2.08 (d, J = 14.2, 2 H); 2.06 (dd, J = 2.4, 14.9,  $2H$ ); 1.36 (dd, J = 2.4, 14.2, 2 H); 1.17 (s, 6 H); 1.10 (s, 6 H); 0.93 (s, 6 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 209.9 (s); 53.2(d); 51.4 (t); 45.1 (s); 43.1 (t); 35.9 (s); 33.5 (q); 27.5 (q); 23.5 (q).

X-Ray Crystal-Structure Determination of  $2b^1$ ). Pale colorless blocks (0.07  $\times$  0.17  $\times$  0.52 mm) from Et<sub>2</sub>O, C<sub>18</sub>H<sub>28</sub>O<sub>2</sub>, M<sub>r</sub> 276.40; monoclinic, space group C2/c, Z = 8, a = 12.3453(15), b = 17.248(2), c = 15.982(2)  $\dot{A}$ ,  $\beta = 106.307(2)$ °,  $V = 3266.2(7)$   $\dot{A}$ <sup>3</sup>,  $D_x = 1.124$  g cm<sup>-3</sup>.

3.2. Photodimerization of 1b in Cyclohexane. A soln. of 138 mg (1 mmol) of 1b in cyclohexane (2ml) is irradiated for 48 h. After evaporation of the solvent, the crystalline residue is washed with cyclohexane and recrystallized from Et<sub>Q</sub> to afford 60.6 mg (88%) of cis-transoid-cis-1,5,5,7,11,11-hexamethyltricy $clo[6.4.0.0<sup>2.7</sup>] do de cane-3.9-dione (3b)$ . M.p. 207°. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 2.50 (s, 2 H); 2.26 (dd, J = 2.4, 16.2,  $2H$ ; 2.11 (d, J = 14.2, 2 H); 1.87 (d, J = 16.2, 2 H); 1.79 (dd, J = 2.4, 14.2, 2 H); 1.20 (s, 6 H); 1.03 (s, 6 H); 0.87 (s, 6 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 212.5 (s); 62.4 (d); 53.8 (t); 52.7 (t); 37.6 (s); 33.9 (s); 31.7 (q); 28.5 (q); 27.5 $(q)$ .

X-Ray Crystal-Structure Determination of  $3b<sup>1</sup>$ ). Pale colorless needles (0.02  $\times$  0.14  $\times$  0.31 mm) from Et<sub>2</sub>O, C<sub>18</sub>H<sub>28</sub>O<sub>2</sub>, M<sub>r</sub> 276.40; triclinic, space group P1, Z = 1, a = 5.7811(12), b = 8.0588(17), c = 9.103(2) Å,  $\alpha = 84.951(4), \beta = 76.606(4), \gamma = 73.635(4)^\circ, V = 395.73(15) \text{ Å}^3, D_x = 1.160 \text{ g cm}^{-3}.$ 

3.3. Photodimerization of 1c in MeCN. A soln. of 25 mg (0.12 mmol) of 1c in MeCN (1 ml) is irradiated for 24 h. After evaporation of the solvent, the residue is washed with cyclohexane and recrystallized from Et<sub>2</sub>O to afford 15.7 mg (64%) of cis-transoid-cis-7,8-bis(3,3-dimethylbut-1-ynyl)-5,5,10,10-tetramethyltricyclo[6.4.0.0<sup>2,7</sup>]dodecane-3,12-dione (2c). M.p. 150°. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 2.95 (s,  $2H$ ; 2.41 (d, J = 15.0, 2 H); 2.30 (d, J = 14.5, 2 H); 2.15 (dd, J = 2.0, 15.0, 2 H); 1.80 (dd, J = 2.0, 14.5, 2 H); 1.19 (s, 18 H); 1.10 (s, 6 H); 0.93 (s, 6 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 208.0 (s); 96.2 (s); 80.9 (s); 51.3 (t); 50.5 (d); 44.0 (t); 42.9 (s); 35.5 (s); 31.0 (q); 30.5 (q); 27.5 (s) 26.9 (q).

X-Ray Crystal-Structure Determination of  $2c^1$ ). Pale colorless blocks (0.19  $\times$  0.24  $\times$  0.50 mm) from Et<sub>2</sub>O, C<sub>28</sub>H<sub>40</sub>O<sub>2</sub> + 0.5 (C<sub>2</sub>H<sub>3</sub>N) + 0.5 (C<sub>2</sub>H<sub>3</sub>N), M<sub>r</sub> 449.65; triclinic, space group P1, Z = 2, a = 10.1044(9),  $b = 11.7421(11), c = 15.5705(14), \lambda, \alpha = 68.906(2), \beta = 84.965(2), \gamma = 65.878(2)^\circ, V = 1569.3(2), \lambda^3, D_x = 0.001(25)$  $0.952$  g cm<sup>-3</sup>.

3.4. Photodimerization of 1d in Benzene. A soln. of 376 mg  $(2 \text{ mmol})$  1d in benzene  $(10 \text{ ml})$  is irradiated for 24 h up to 90% conversion. After evaporation of the solvent, the residue (from  $\rm{^1H\text{-}NMR}$ : 10% of 1d, 9% of 2d, 45% of 6, and 36% of 7) is subjected to CC (SiO<sub>2</sub>; pentane/Et<sub>2</sub>O 1:1) to afford first 1d  $(25 \text{ mg})$ .

The second fraction (26.5 mg, 7.7%) contains pure cis-transoid-cis-7,8-bis(3-methylbut-3-en-1-ynyl)- 5,5,10,10-tetramethyltricyclo[6.4.0.0<sup>2,7</sup>]dodecane-3,12-dione (2**d**) as light yellow oil. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 5.24 (s, 2 H); 5.22 (s, 2 H); 3.06 (s, 2 H); 2.45 (d, J = 15.1, 2 H); 2.36 (d, J = 13.9, 2 H); 2.18 (dd, J = 2.0, 15.1, 2 H); 2.01 (dd, J = 2.0, 13.9, 2 H); 1.87 (s, 6 H); 1.13 (s, 6 H); 1.07 (s, 6 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 207.5  $(s)$ ; 126.3  $(s)$ ; 121.5  $(t)$ ; 90.7  $(s)$ ; 86.9  $(s)$ ; 51.2 $(t)$ ; 49.7  $(d)$ ; 44.2 $(t)$ ; 43.4  $(s)$ ; 35.2 $(s)$ ; 31.5  $(q)$ ; 25.8  $(q)$ ;  $24.9 (q)$ .

The third fraction (35 mg, 11%) contains pure trans-1,2-bis[2-(5,5-dimethyl-3-oxocyclohex-1-en-1 yl)ethynyl]-1,2-dimethylcyclobutane (6). M.p. 49°. <sup>1</sup>H-NMR (C<sub>6</sub>D<sub>6</sub>): 6.41 (t, J = 1.7, 2 H); 2.15 (AA'XX',  $J_{AX} \approx 11, J_{AA'} = J_{XX'} \approx 7, J_{AX} \approx 10, 2 \text{ H})$ ; 1.97 (s, 4 H); 1.93 (d, J = 1.7, 4 H); 1.66 (AA'XX',  $J_{AX} \approx 11, J_{AA'}$  =  $J_{XX} \approx 7, J_{AX} \approx 10, 2 \text{ H}$ ); 1.55 (s, 6 H); 0.65 (s, 12 H). <sup>13</sup>C-NMR (C<sub>6</sub>D<sub>6</sub>): 196.9 (s); 141.5 (s); 132.1 (d); 102.9  $(s)$ ; 84.6  $(s)$ ; 50.9  $(t)$ ; 44.5  $(t)$ ; 40.2  $(s)$ ; 32.5  $(s)$ ; 31.7  $(t)$ ; 27.8  $(q)$ ; 25.2  $(q)$ .

<sup>1)</sup> CCDC-672011, -672012, and -672013 contain the supplementary crystallographic data for 2b, 3b, and 2c, resp. This data can be obtained free of charge *via* www.ccdc.cam.ac.uk/data\_request/cif.

The last fraction (60 mg, 18%) contains pure cis-1,2-bis[2-(5,5-dimethyl-3-oxocyclohex-1-en-1yl)ethynyl]-1,2-dimethylcyclobutane (7). M.p. 51°. <sup>1</sup>H-NMR (C<sub>6</sub>D<sub>6</sub>): 6.49 (t, J = 1.7, 2 H); 2.26 (AA'XX',  $J_{AX} \approx 11, J_{AA'} = J_{XX'} \approx 10, J_{AX} \approx 7, 2 \text{ H}$ ); 2.11 (d, J = 1.7, 4 H); 2.02 (s, 4 H); 1.59 (AA'XX',  $J_{AX} \approx 11, J_{AA'} =$  $J_{XX} \approx 10, J_{AY} \approx 7, 2 \text{ H}$ ; 1.12 (s, 6 H); 0.74 (s, 12 H). <sup>13</sup>C-NMR (C<sub>6</sub>D<sub>6</sub>): 196.8 (s); 140.5 (s); 131.1 (d); 105.1 (s); 90.5 (s); 51.0 (t); 44.8 (t); 40.7 (s); 33.5 (s); 32.2 (t); 27.9 (q); 22.0 (q).

4. Photocycloaddition of 1d to 2,3-Dimethylbuta-1,3-diene (8). A soln. containing 94 mg (0.5 mmol) of 1d and 164 mg (2 mmol) of 8 in benzene (5 ml) is irradiated for 4 h. After evaporation of the solvent, <sup>1</sup>H-NMR analysis indicates the residue (140 mg) to be a 4:1 mixture of 9 and 10. The following data of (7E)-4,7,8,13,13-pentamethylbicyclo[8.4.0]tetradeca-1,2,3,7-tetraen-11-one (9) stem directly from the crude photolysate. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 3.08 (dd,  $J = 3.0, 11.0, H - C(10)$ ); 2.92 (dd,  $J = 11.0, 14.0,$  $H_{ax}-C(9)$ ; 2.87 (dt, J = 7.5, 13.5,  $H_{ax}-C(6)$ ); 2.45, 2.22 (AB, J = 14.0, 2 H – C(14)); 2.38, 2.29 (AB, J =  $13.5$ ,  $2 H - C(12)$ );  $2.34$  (m,  $H_{eq} - C(6)$ );  $2.05$ ,  $1.95$  (m,  $2 H - C(5)$ );  $1.82$  (s, Me $-C(8)$ );  $1.67$  (s, Me $-C(7)$ ); 1.62 (s, Me-C(4)); 1.08, 0.89 (s, 2 Me-C(13)). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 209.5 (s, C(11)); 158.1 (s, C(2)); 156.5 (s, C(3)); 12 7.2 (s, C(8)); 12 7.0 (s, C(7)); 116.5 (s, C(4)); 111.4 (s, C(1)); 54.5 (t, C(12)); 48.7 (d, C(10)); 48.5 (t, C(14)); 34.9 (s, C(13)); 31.9 (t, C(6)); 30.5 (q, Me-C(13)); 30.2 (t, C(9)); 29.5 (t, C(5)); 26.5 (q,  $Me-C(13)$ ; 23.9 (q,  $Me-C(8)$ ); 17.8 (q,  $Me-C(7)$ ); 16.4 (q,  $Me-C(4)$ ).

The crude photolysate is stirred over  $SiO<sub>2</sub>$  in CH<sub>2</sub>Cl<sub>2</sub> for 3 h, followed by filtration, washing with CH<sub>2</sub>Cl<sub>2</sub>, and evaporation of the solvent. CC (SiO<sub>2</sub>; CH<sub>2</sub>Cl<sub>2</sub>/toluene 9:1) affords 8 mg (6%) of 4,4,7trimethyl-6-(3-methylbut-3-en-1-ynyl)-7-(1-methylethenyl)bicyclo[4.2.0]octan-2-one (10) as colorless oil.  ${}^{1}H\text{-NMR } (C_{6}D_{6})$ : 5.26, 5.00 (s, 2 H); 4.77, 4.52 (s, 2 H); 3.15 (dd, J = 9.5, 10.0); 2.11 (d, J = 14.2); 2.07 (s,  $2H$ );  $2.05(m, 1 H)$ ;  $1.87(d, J = 14.2)$ ;  $1.73(s, 3 H)$ ;  $1.68(m, 1 H)$ ;  $1.61(s, 3 H)$ ;  $1.33(s, 3 H)$ ;  $1.07(s, 3 H)$ ; 0.73 (s, 3 H). <sup>13</sup>C-NMR (C<sub>6</sub>D<sub>6</sub>): 208.5 (s); 150.1 (s); 126.2 (s); 121.2 (t); 110.1 (t); 92.9 (s); 86.5 (s); 51.5 (t); 49.1 (s); 45.1 (d); 42,5 (s); 41.7 (t); 34.8 (s); 32.1 (t); 31.5 (q); 30.6 (q); 26.5 (q); 24.9 (q); 19.5 (q).

5. Isomerization of 9 to 11. The crude photolysate (cf. above) is stirred for 3 h over  $SiO<sub>2</sub>$  pretreated with 1% Et<sub>3</sub>N in CH<sub>2</sub>Cl<sub>2</sub>. After filtration and evaporation of the solvent, the residue is subjected to CC (SiO<sub>2</sub> pretreated with 1% Et<sub>3</sub>N in CH<sub>2</sub>Cl<sub>2</sub>; CHCl<sub>3</sub>) to afford 70 mg (55%) of 3-(6,7-dimethyl-3methylidenecyclohepta-1,6-dien-1-yl)-2,2,5-trimethylcyclohex-2-enone  $(11)$  as colorless oil. <sup>1</sup>H-NMR  $(CDCl_3)$ : 6.08 (s, 1 H); 4.89 (s, 1 H); 4.78 (s, 1 H); 2.40 (m, 2 H); 2.33 (s, 2 H); 2.31 (m, 2 H); 2.25 (s, 2 H); 1.87 (s, 3 H); 1.82 (s, 3 H); 1.78 (s, 3 H); 0.99 (s, 6 H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 198.9 (s); 153.1 (s); 145.1  $(s)$ ; 142.2  $(s)$ ; 135.9  $(s)$ ; 131.1  $(s)$ ; 126.9  $(s)$ ; 122.1  $(d)$ ; 111.1  $(t)$ ; 52.1  $(t)$ ; 42.9  $(t)$ ; 34.9  $(t)$ ; 32.5  $(s)$ ; 32.1  $(t)$ ; 28.1 (q); 21.1 (q); 14.7 (q); 12.2 (q).

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