THE REACTION OF INDOLES WITH SUCCINIMIDO-SULFONIUM SALTS I. A CONVENIENT SYNTHESIS OF INDOLE-3-THIOETHERS

K<u>uniyuki</u> T<u>omita</u>^{*}, A<u>tsusuke</u> T<u>erada</u>, and R<u>yuji</u> T<u>achikawa</u> Central Research Laboratories, Sankyo Co., Ltd., Shinagawa-ku, Tokyo

Reaction of indoles with succinimido-dialkyl- or alkylarylsulfonium chlorides, prepared from dialkyl or alkylaryl sulfides and N-chlorosuccinimide, afforded indole-3-dialkyl- or alkylarylsulfonium chlorides which gave 3-alkylthio- or arylthioindoles on pyrolysis.

Succinimido-dimethylsulfonium chloride, prepared from Nchlorosuccinimide and dimethylsulfides,¹⁾ have been known to react with some nucleophiles such as amines, phenols, alcohols, carbanions and enamines.²⁾ In this paper, we report a new reaction of the sulfonium salts with indole compounds.

To a solution of succinimido-diethylsulfonium chloride in dichloromethane or chloroform, prepared by the addition of diethylsulfide (Ia) to a solution of N-chlorosuccinimide at $0^{\circ}C$, was added equimolar amount of indole (IIa) in the same solvent at $-20^{\circ}C$ under an atmosphere of nitrogen. The temperature of the mixture was raised gradually to $20^{\circ}C$ during 1 hour. After removal of the solvent, 3-diethylsulfoniumindole (IIIa) and succinimide were obtained quantitatively: (IIIa); mp 148-149[°]C

- 729 -

(decomp. methanol-tetrahydrofuran), NMR (60 Mc, in D_20); § 1.19 (6H, t, J=7 Hz, $-CH_2-CH_3$), 3.65 (4H, q, J=7 Hz, S⁺- CH_2-CH_3), 7.25-7.90 (4H, m, aromatic protons), 8.15 (1H, s, indole α -H). When the salt (IIIa) was heated at 150°C in nitrogen atmosphere or refluxed in xylene for 30 min, 3-ethylthioindole (IVa) was obtained as a light yellow oil in 80% yield: NMR (in CDCl₃); § 1.16 (3H, t, J=8 Hz, $-CH_2-CH_3$), 2.70 (2H, q, J=8 Hz, S- CH_2-CH_3), 7.10-7.80 (5H, m, aromatic protons), 8.2 (1H, br., N<u>H</u>). IR^{1iq}: 3430 cm⁻¹ (NH). Mass spectrum; m/e 177 (M⁺). Likewise, with some other dialkylsulfides (Ib-e), 3-alkylthioindoles (IVb-g) were prepared in good yields via the intermediate (IIIb-g). (Scheme 1)

The reaction of indole with the succinimido-sulfonium salt of tetrahydrothiophene gave the corresponding 3-sulfonium chloride quantitatively which afforded 3- ω -chloro-n-butylthioindole on pyrolysis in 90% yield: NMR (in CDCl₃); δ 1.70 (4H, m, $-C\underline{H}_2-C\underline{H}_2-$), 2.63 (2H, t, J=7 Hz, S- $C\underline{H}_2-C\underline{H}_2-$), 3.43 (2H, t, J=7 Hz, $-CH_2-C\underline{H}_2-Cl$), 7.05-7.80 (6H, m, aromatic protons and N<u>H</u>). This finding suggests that the indole-3-thioethers (IV) are formed by the nucleophilic attack of the Cl⁻ on the α carbons of the alkyl groups in the sulfonium salts (III). Therefore phenylmethylsulfide presented 3-phenylthioindole, mp 151-3°C, in 50% overall yield as the sole product, expectedly. The usual way for the synthesis of indole-3-alkylthioethers is the alkylation of indolethiols^{3a)} or the direct formation by the indole synthesis.^{3b)} Indole-3-arylthioethers have been formed by the latter method.^{3c)} The present reaction

- 730 -

provides a simpler and more efficient way to approach both alkyl- and arylthioethers.

Synthetic application of this new reaction will be shown in the subsequent papers.



Scheme 1

Ia	$R=C_2H_5$	IIa K'=R''=H	III,IVa R=C2H5	R'=R''=H
ъ	CH 3	b R*=H, R**=CH3	ъ св _з	18
c	nC ₃ H7	c R'=CH3, R''=H	e nC ₃ H7	18
à	nC4H9		â nC ₄ H ₉	n
е	пС ₈ Н ₁₇		e nC ₈ H ₁₇	**
			f C ₂ H ₅	R'=H, R''=CH3
			g "	R'=CH3, R''=H

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