STRUCTURE OF

A NEW BISBENZYLISOQUINOLINE ALKALOID, LINDOLDHAMINE1)

Sheng-Teh Lu*and Ih-Sheng Chen School of Pharmacy, Kaohsiung Medical College, Taiwan

The structure of lindoldhamine, isolated from the leaves of <u>Lindera oldhamii</u> Hemsl.(Lauraceae), has been established as (1), by spectral data and chemical degradations.

In the previous paper $^{1)}$, we have reported that an unknown alkaloid with mp $173-176^{\circ}$ has been separated from the leaves of Lindera oldhamii Hemsl.(Lauraceae). This substance is a bisbenzylisoquinoline alkaloid, named lindoldhamine(1), with $C_{34}H_{36}N_{2}O_{6}$, colorless fine needles, mp $183-186^{\circ}$, $[\propto]_{D}^{33}+35^{\circ}$ (c=1.0, EtoH), λ EtoH 205, 220 sh, and 280 nm(log £ 4.65, 4.39, and 3.91), m/e $568(M^{+})$, 178(base peak), when it was chromatographed on neutral alumina using methanol as eluent and recrystallized from a mixture of EtoH, $Me_{2}CO$ and $CHCl_{3}$. The nmr spectrum δ ($CF_{3}COOH$) showed two methoxyl groups at 3.98(6H, s.) and complicated signal due to eleven aromatic protons at 6.79-7.54 region.

In the nmr spectrum $\boldsymbol{\delta}$ (CDCl₃) of N,N-dimethyllindoldhamine (2) with $[\alpha]_D^{21}$ -85°(c=0.65, CHCl₃) afforded by treating 1 with formalin and NaBH₄, the chemical shifts of two N-methyl groups at 2.42(3H, s) and 2.47(3H, s), two methoxyl groups at 3.80(6H, s), the eleven aromatic protons at 6.22-7.09 region, and three phenolic hydroxyl groups at 5.58(3H, broad s), respectively, suggested a bisbenzyltetrahydroisoquinoline alkaloid with one "tail to tail" diphenyl ether linkage²⁾.

Permanganate oxidation of N,N,O,O,O-pentaethyllindoldhamine

(3) diethobromide, yielded by treating lindoldhamine(1) with
ethanolic KOH and EtBr, furnished 4-ethoxy-3,4'-oxydibenzoic acid

(4) as colorless sands, mp 273-275°.

Cleavage reaction of the diphenyl ether linkage of 0,0,0-trimethyllindoldhamine(5), prepared by methylation of 1 with diazomethane, with sodium in liquid NH₃ afforded R-(-)-N-nor-0-methylarmepavine³⁾(6), whose oxalate was colorless needles with mp $214-215^{\circ}$ (swelling), $\left[\alpha\right]_{D}^{31}$ -38°(c=0.9, MeOH), which was identified by direct comparison with the sample obtained by methylation of R-(+)-N-norarmepavine⁴⁾(7) with diazomethane, as nonphenolic base part, and R-(+)-N-norarmepavine⁴⁾(7) as colorless needles, mp 152-153°, $\left[\alpha\right]_{D}^{32}$ +22°(c=1.0, CHCl₃) as phenolic base part. Cleavage reaction of the diphenyl ether linkage of 0,0,0-triethyllindoldhamine(8), prepared by ethylation of 1 with diazoethane, with sodium in liquid NH₂ gave R-0,0-diethylcoclau-

rine(9), whose oxalate was colorless fine needles with mp $224-225^{\circ}$ (swelling), which was identified by direct comparison with the sample prepared by ethylation of dl-coclaurine⁵⁾ with diazoethane, as non-phenolic base part, and R-7-0-ethylcoclaurine(10) as colorless prisms, mp $142-143^{\circ}$, nmr δ (CDCl₃): $1.42(3H, t, J=7Hz, OCH_2CH_3)$, $3.82(3H, s, OCH_3)$, $4.04(2H, ABq, J_1=7Hz, J_2=14Hz, OCH_2CH_3)$, 4.52(2H, broads, OH and NH), 6.56, 6.66(eachlh, eachs, ArH), 6.59, 6.98(eachle 2H, eachd, J=8Hz, ArH), whose ethylation sample formed by treating of 10 with diazoethane was identical with 9, as phenolic base part.

Therefore, the structure of lindoldhamine has been offered as (1).

- (1) $R_1 = R_2 = H$
- (2) $R_1 = Me$, $R_2 = H$
- (3) $R_1 = R_2 = Et$
- (5) $R_1 = H$, $R_2 = Me$
- (8) $R_1 = H$, $R_2 = Et$

- (6) $R_1 = R_2 = Me$
- (7) R₇=Me, R₂=H
- (9) $R_1 = R_2 = Et$
- (10) $R_1 = Et$, $R_2 = H$

ACKNOWLEDGEMENT We are grateful to Emeritus Prof. M. Tomita, Kyoto University, for his direction and supply of 4-ethoxy-3,4'-oxydibenzoic acid. We also thank Prof. Y. Inubushi, Kyoto university, and Assistant Prof. H. Ishii, Chiba University, for the elemental analyses and the nmr and ms measurements. The research was supported by the National Science Council of the Republic of China.

REFERENCES

- S. T. Lu, S.J. Wang, P. H. Lai, T. M. Lin, and L. C. Lin,
 J. Pharm. Soc. Japan, 1972, 92, 910.
- 2) M. Tomita, T. Shingu, K. Fujitani, and H. Furukawa, Chem. Pharm. Bull., 1965, 13, 921.
- 3) M. Tomita, E. Fujita, and F. Murai, J. Pharm. Soc. Japan, 1951, 71, 1035.
- 4) T. H. Yang and S. T. Lu, J. Pharm. Soc. Japan, 1963, 83, 22.
- 5) S. T. Lu, <u>J. Pharm. Soc. Japan</u>, 1963, <u>33</u>, 19.

Received, 25th February, 1976