

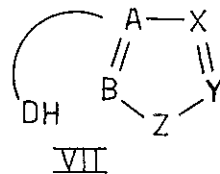
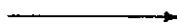
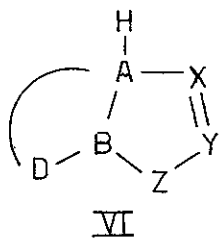
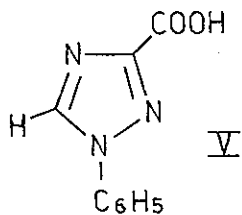
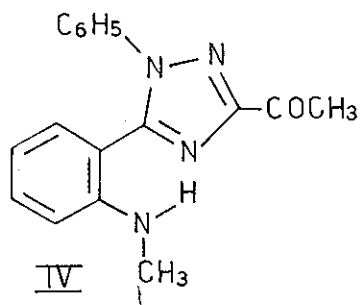
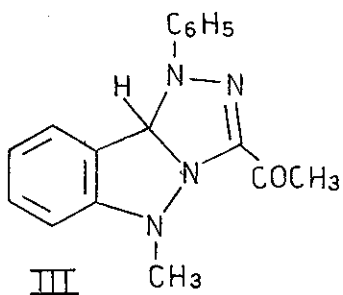
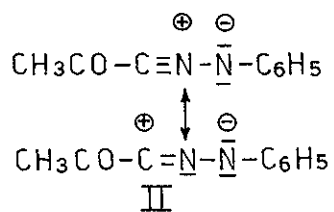
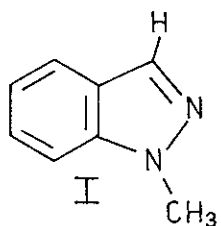
## A 1,3-DIPOLAR CYCLOADDITION REACTION OF 1-METHYLINDAZOLE

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A 1,3-dipolar cycloaddition reaction of 1-methylindazole with C-acetyl-N-phenyl-nitrilimine was performed. The cycloadduct could be used to synthesize 1,2,4-triazole derivatives.

In the framework of our studies<sup>1</sup> on the behaviour of heterocyclic systems towards nitrilimines, we have carried out the reaction between 1-methylindazole (I) and C-acetyl-N-phenyl-nitrilimine (II), prepared in situ. Treatment of (I) with stoichiometric amount of  $\alpha$ -chloro-(N-phenylhydrazono)acetone and three-fold excess of triethylamine (THF, 20 days, room temperature) gave a reaction mixture which, after triethylamine hydrochloride being filtered off and the solvent evaporated under vacuum, afforded 1-phenyl-3-acetyl-5-methyl-1,9b-dihydro-5H-1,2,4-triazol- $[4,3b]$ indazole (III), mp 134-135°, in 40% yield. The structure of the cycloadduct (III) was assigned on the basis of elemental analysis, spectroscopic data and chemical transformation. An examination of the nmr spectrum  $[(C_6D_6)]$   $\delta$ : 2.17 (3H, s, -CO-CH<sub>3</sub>), 3.26 (3H,



s,  $-\text{N}-\text{CH}_3$ ), 6.61 (1H, s, 8a-H), 6.35-7.10 (9H, m, Ar-H)] revealed large shielding effect on 9b-H and  $-\text{N}-\text{CH}_3$  protons with respect to 3-H and  $-\text{N}-\text{CH}_3$  protons of (I) ( $\Delta\delta = -1.30$  and  $-0.27$  respectively). This shielding effect substantiated the absence of ring current clearly supporting the structure proposed. The carbonyl frequency [ $\nu_{\text{max}}$  (nujol mull)  $1666 \text{ cm}^{-1}$ ] observed is also consistent with (III). A sample of (III) with catalytic amount of conc. HCl (THF, 24 hr, room temperature) gave, after evaporation of the solvent under vacuum, 1-phenyl-3-acetyl-5-[o-methylaminophenyl]-1,2,4-triazole (IV), in almost quantitative yield. The structure of (IV), mp  $201-202^\circ$ , was deduced on the basis of the following evidence: ir  $\nu_{\text{max}}$  (nujol mull)  $3322, 1690 \text{ cm}^{-1}$ ; nmr (DMSO- $d_6$ )  $\delta$ : 2.63 (3H, s,  $-\text{CO}-\text{CH}_3$ ), 2.68 (3H, d,  $-\text{N}-\text{CH}_3$ ,  $J=4.5$ ), 6.03 (1H, br q, NH,  $J=4.5$ ), 6.20-7.70 (4H, m, Ar-H), 7.39 (5H, s,  $\text{N}-\text{C}_6\text{H}_5$ ); uv  $\text{EtOH}_{\text{max}}$  nm (log  $\epsilon$ ): 221sh (4.37), 258sh (4.06), 3.44 (3.65). Moreover, a suspension of (IV) in a warm solution of sodium hydroxide treated with  $\text{KMnO}_4^*$ , afforded 1-phenyl-1,2,4-triazol-3-carboxylic acid (V)<sup>2</sup>.

The results obtained give further information on the behaviour, as dipolarophiles, of the nitrogen heteroaromatic compounds putting in evidence the tendency of C=N bond of 1-methylindazole (I) to react towards nitrilimines so as the C=N bond of some mono<sup>3</sup> and diazines<sup>4</sup>, azomethines<sup>5</sup>, oximes<sup>6</sup>, etc.. As expected, the unsubstituted nitrogen controls the orientation of the dipole. Furthermore the transformation of (III) into (IV) furnishes an interesting alternative route to synthesize 1,2,4-triazole derivatives. This behaviour is similar to that of other non aromatic cycloadducts<sup>7,8</sup>

which, by treatment with acids, give aromatic compounds through an elimination reaction, e.g. scheme (VI → VII).

#### REFERENCES AND FOOTNOTES

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- 8 Likewise, we have observed that cycloadduct obtained from N-methylindole and diphenyl-nitrilimine gives 1,3-diphenyl-4-methylaminophenyl pyrazole by treatment with HCl in ethanolic solution at room temperature, through a fission of D-B bond (unpublished results). A different behaviour showed cycloadducts obtained from N-substituted indole derivatives and C-acetyl or C-carbethoxy-N-phenyl-nitrilimines. In these cases, in fact, under similar experimental conditions we have observed the cleavage of the B-Z bond<sup>1a,1d</sup>.

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