SCIADOLINE, A NEW TYPE OF BISBENZYLISOQUINOLINE ALKALOID

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Sciadoline, a new alkaloid from <u>Sciadotenia toxifera</u> Krukoff and A. C. Smith (Menispermaceae), has been assigned structure II. It is the first example of a head-to-tail bisbenzylisoquinoline base which contains a completely aromatic isoquinoline unit.

We recently reported the isolation of the new bisbenzylisoquinoline alkaloid sciadenine (I) from a countercurrent fraction of the bases from <u>Sciadotenia toxifera</u> Krukoff and A. C. Smith.¹⁾ Another base fraction from this plant has now yielded the related new alkaloid sciadoline (II), the first example of a head-to-tail bisbenzylisoquinoline in which one of the head units is a completely aromatic isoquinoline system.

Sciadoline (II) crystallized from chloroform-acetone as colorless needles, mp 225-228° (decomp.), $[\alpha]_D^{22} = +46°$ (c = 0.46, CHCl₃). The mass spectrum of II indicates the composition $C_{36}H_{34}N_2O_6$, showing a strong molecular ion at m/e 590, and fragment ions at m/e 589 (base peak), 576, 575, 483, 296, 295.5, 295, 204 and 190. The nmr spectrum of II (CDCl₃) showed one N-methyl at § 2.35 (3 H) and three methoxyls at 3.55 (3 H), 3.83 (3 H) and 4.02 (3 H). In the aromatic region, two pyridine-type protons appeared as a low-field

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AB quartet (J = 5.5 Hz) centered at 7.48 and 8.40, in addition to a oneproton singlet at § 7.05 and nine additional protons in the range § 5.50-6.85. The uv absorption spectrum of II showed maxima (log e) at 275 (3.94) and 335 nm (3.78), and shoulders at 283 (3.88) and 326 nm (3.77); the spectrum resembles that of the simple isoquinoline alkaloid crykonisine (III).²⁾ The isoquinoline chromophore of II shows a marked bathochromic shift not only in acid (323 and 368 nm) but also in base (300 and 373 nm), indicating the presence of a phenolic hydroxy1 in the isoquinoline nucleus. Acetic anhydride in pyridine effected acetylation of this hydroxy1 with the formation of the 0-monoacetate (IV), mp 236-237°, ir (KBr) 5.57 μ , mass spectrum m/e 632 (M⁺, base peak). No deuterium was introduced into II by D₂0 under basic conditions, showing that no aromatic proton is present ortho to the phenolic function.³⁾

Reaction of II with excess diazoethane afforded O-ethylsciadoline (V), mp 192-194^o (decomp.), m/e 618 (M^+ , base peak). Reductive cleavage of V with sodium in liquid ammonia gave, as the only isolable product, R-armepavine (VI), identical with an authentic sample.⁴)

The results described above lead to the assignment of structure II to sciadoline, which is therefore a more highly unsaturated analog of its companion alkaloid sciadenine (I). Sciadoline is also the first example of a head-to-tail bisbenzylisoquinoline alkaloid which contains a fully aromatic isoquinoline system. A head-to-head bisbenzylisoquinoline structure containing an aromatic isoquinoline unit has been proposed for the alkaloid thalfine.⁵⁾

| Compound | <u>II</u> | IV | V | VII |
|-----------------------------------|-----------|-----------------------------------|--|------------------------|
| с ₃ -н ^а | 8.40 | 8.55 | 8.41 | 8.48 |
| с ₄ -н ^а | 7.48 | 7.55 | 7.47 | 7.48 |
| с ₅ -н ^ь | 7.05 | 7.10 | 7.02 | 7.03 |
| Aromatic ^C protons | 5.50~6.85 | 5.58 ~ 6.92 | 5.55 ~ 6.85 | 5.5416.80 |
| ^C 6-OMe ^b | 4.02 | 3.93 | 3.90 | 3.95 |
| C ₇ -OR | | 1.82 (<u>Me</u> CO) ^b | 0.98 (С <u>н</u> 3Сн ₂) ^d | 3.63 (Me) ^b |
| с _б '-Оме ^b | 3.83 | 3.84 | 3.80 | 3.83 |
| С ₇ '-ОМе ^b | 3.55 | 3.55 | 3.53 | 3.57 |
| C2'-NMe b | 2.35 | 2.43 | 2.40 | 2.42 |

Nmr Values ($\delta,\ \text{CDCl}_3)$ for Sciadoline and Its Derivatives

a: doublet, $J = 5.5 \sim 6.0$ Hz, b: singlet, c: multiplet

d: triplet, J = 7 Hz

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REFERENCES

K. Takahashi, M. J. Mitchell, and M. P. Cava, <u>Heterocycles</u>, 1976, 4, 471.
Sheng-Teh Lu, <u>J. Pharm. Soc. (Japan</u>), 1967, 87, 1278.
G. W. Kirby and L. Ogunkoya, <u>J. Chem. Soc</u>., 1965, 6914.
M. P. Cava and A. Afzali, <u>J. Org. Chem.</u>, 1975, 40, 1553.
S. Abdizhabbarova, Z. F. Ismailov, and S. Yu. Yunusov, <u>Khim. Prir. Soedin.</u>, 1968, 4, 330, and 1970, 6, 279.

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