DIMSYLSODIUM-INDUCED REARRANGEMENT OF ALKYLIMIDATES TO N-ALKYLLACTAMS

S<u>hinzo</u> K<u>ano</u>[™], T<u>sutomu</u> Y<u>okomatsu</u>, S<u>atoshi</u> H<u>ibino</u>, K<u>eiko</u> I<u>mamura</u>, and S<u>hiroshi</u> S<u>hibuya</u>

Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji,

Tokyo, Japan

Treatment of 1-ethoxy-3,4-dihydro-6,7-dimethoxyisoquinoline (1) with dimsylsodium yielded 2-ethyl-1,2,3,4tetrahydro-6,7-dimethoxy-1-oxoisoquinoline (2). Similarly, 2-ethoxy-4,5-dihydro-3<u>H</u>-pyrrole (3), 2-ethoxy-3,4,5,6tetrahydropyridine (5), 7-ethoxy-3,4,5,6-tetrahydro-2<u>H</u>azepine (7), and 7-ethoxy-3,4,5,6-tetrahydro-4-methyl-2<u>H</u>azepine (9) were allowed to react with dimsylsodium to give the corresponding N-ethyllactams (4), (6), (8), and (10), respectively, through 0 to N migration of ethyl group by the catalytic action of dimsylsodium.

A variety of N-alkylamide have been prepared by the rearrangement of O-alkylimidates by many research groups¹⁾, in an apparently uncatallyzed thermal rearrangement, frequently treated as a special case of the Chapman-Mumm rearrangement. The temperatures required for the rearrangement of alkyl group are generally quite high. Catalysis by Lewis and Brönsted acid²⁾, alkyl halide³⁾, or metal salt⁴⁾

<u>-1319</u>

have been useful for the O to N migration of the alkyl group of imidates. Following examples would suffice as illustrations.





However, there is no survey on the base catalyzed migration of alkyl group leading to N-alkylamides. We have found that N-alkyllactams were formed from alkylimidates by the catalytic action of dimsylsodium. We wish to report these results in this paper.

1-Ethoxy-3,4-dihydro-6,7-dimethoxyisoquinoline (1), prepared from 1,2,3,4-tetrahydro-6,7-dimethoxy-1-oxoisoquinoline by treatment with triethyloxonium fluoroborate, was allowed to react with dimsylsodium in THF-DMSO at room temperature for 1 hr resulted in formation of 2-ethy1-1,2,3,4-tetrahydro-6,7-dimethoxy-1-oxoisoquinoline (2), mp 81-83° (MeOH-Et $_2$ O), in quantitative yield. The spectral data [IR $(CHC1_3)$ cm⁻ 1630 (-N-CO), NMR δ (CDC1₃) 1.43 (3H, t, J=7 Hz, CH₃CH₂),

-1320-

3.90 (6H, s, $2xOCH_3$), 4.22 (2H, q, CH_3CH_2), 6.63 (1H, s, 5-H), 7.17 (1H, s, 8-H), mass ($\underline{m/e}$) 235 (M^+), 220 (M^+ -15), 207 (M^+ -28), 178 (M^+ -57)] were in good agreement with the assigned structure. Similar results were also obtained by the use of alicyclic imidates. The reaction of 2-ethoxy-4,5-dihydro-3<u>H</u>-pyrrole (3), 2-ethoxy-3,4,5,6tetrahydropyridine (5), 7-ethoxy-3,4,5,6-tetrahydro-2H-azepine (7), and 7-ethoxy-3,4,5,6-tetrahydro-4-methyl-2<u>H</u>-azepine (9) with dimsylsodium afforded 1-ethyl-2-pyrrolidone (4)⁵⁾, 1-ethyl-2-piperidone (6)⁶⁾, 1-ethylcaprolactam (8)⁷⁾, and 1-ethyl-hexahydro-5-methyl-2<u>H</u>azepin-2-one (10), respectively, in a similar fassion, in good yield.



 $(R=C_2H_5)$

The ethyl migration was not observed on standing imidates in a solution of DMSO without dimsylsodium. It is of interest that Oalkylimidates were converted to N-alkyllactams by a catalytic action of dimsyl anion under very mild conditions. The application and limitation of this reaction are under further investigation.

References

C. G. McCarty and L. A. Garner, <u>The Chemistry of The Amidines and Imidates</u>, ed., by S. Patai, John Wiley & Sons (London, New York, Sydney, Toronto), <u>1975</u>, p. 189.
A. W. Chapman, <u>J. Chem. Soc.</u>, <u>1927</u>, 1743.
W. Wislicenus and M. G. Schmidt, <u>Chem. Ber</u>., <u>33</u>, 1470 (1900).
H. F. Stewart and R. P. Seibert, <u>J. Org. Chem</u>., <u>33</u>, 4560 (1968).
J. W. Ralls, <u>J. Org. Chem</u>., <u>26</u>, 66 (1961).

6. O. V. Schicku, <u>Brit.</u>, 919,404, Feb. 27, 1963. [<u>C. A. 59</u>, p. 1600a (1963)].

7. T. Duong, R. H. Prager, A. D. Ward, and D. I. B. Kerr, <u>Aust. J.</u> <u>Chem.</u>, 29, 2651 (1976).

Received, 9th June, 1977