SYNTHESIS OF 11-AZAPROSTAGLANDIN ANALOGS

Gerard P. Rozing¹, Henk de Koning^{*}, and Henderikus O. Huisman Laboratory of Organic Chemistry, University of Amsterdam Nieuwe Achtergracht 129, Amsterdam, The Netherlands

11-Deoxy-11-azaprostaglandin derivatives are obtained in eight steps, starting from methyl N-ethoxycarbonylglycinate and ethyl 4-tert-butoxycrotonate. The synthesis of an aza-analog of the "Corey-lactone" is described.

Recently several reports have appeared in the literature on the synthesis of prostaglandin analogs containing a nitrogen atom or atoms in ring positions 8 (2), 9 (3), 10 (4), 12 (5), or 8,12 (6). So far ll-azaprostaglandin analogs have not been described, and in this communication we wish to report our results in this field.

Ethoxycarbonylpyrrolidone 3 [IR (CHCl $_3$) 1760, 1720, 1690, 1670, 1630, and 1370 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 1.06 (s, C(CH $_3$) $_3$), 3.35 (d, 3 J $_8$,12 $^{=}$ 2Hz, C $_8$ -H of keto form (7)), 3.4 and 3.85 (m, C $_{13}$ -H), and 4.7 (m, C $_{12}$ -H of enol form)] was obtained in 46% yield by Michael reaction and subsequent Dieckmann cyclization (8) of methyl N-ethoxycarbonylglycinate 1 and methyl 4-tert-butoxycrotonate 2 (9) with sodium hydride in benzene at 80 $^{\circ}$ for 2 h. The potassium salt of β -keto-ester 3, prepared from 3 and potassium hydroxide in

ethanol, was dissolved in dimethyl sulfoxide and alkylated with ethyl 7-bromo-5-heptynoate (10) at ambient temperature. Demethoxy-carbonylation with lithium iodide in hexamethylphosphoric triamide at 120° afforded pyrrolidone 4 [46%; IR (CHCl₃) 1765, 1730, 1690, and 1370 cm⁻¹; 1 H NMR (CDCl₃) δ 1.12 (s, C(CH₃)₃), 1.75 (m, C₃-H), 2.1-2.4 (m, C₂-H, C₄-H), 2.45 (m, C₇-H), 2.6 (m, C₈-H), and 3.8 (m, C₁₀-H)] as an inseparable mixture of cis, trans isomers.

Cleavage of the <u>tert</u>-butyl ether in <u>4</u> with boron trifluoride etherate in acetic anhydride, followed by methanolysis in the presence of potassium carbonate and purification through a silica gel column, provided alcohol <u>5</u> [72%; IR (CHCl₃) 3450, 1765, 1730, and 1690 cm⁻¹; ¹H NMR (CDCl₃) $_{\delta}$ 1.75 (m, C₃-H), 2.10-2.45 (m, C₂-H, C₄-H), 2.55 (m, C₇-H), 3.68 (s, CO₂CH₃), and 3.9 (m, C₁₃-H)]. Partial hydrogenation of <u>5</u> over 5% palladium on barium sulfate in methanol at atmospheric pressure furnished <u>Z</u>-olefin <u>6</u> [85%; IR

(CHCl $_3$) 3450, 1765, 1730, and 1690 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 1.65 (m, C $_3$ -H), 3.65 (s, CO $_2$ CH $_3$), 3.85 (m, C $_1$ 3-H), and 5.42 (m, C $_5$ -H, C $_6$ -H)], while the saturated compound 7 [70%; IR (CHCl $_3$) 3450, 1765, 1730, and 1690 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 2.30 (t, J= 7 Hz, C $_2$ -H), 2.45 (m, C $_8$ -H), 3.68 (s, CO $_2$ CH $_3$), and 3.87 (m, C $_1$ 3-H)] was obtained upon hydrogenation over palladium on charcoal in methanol at 3-4 atm. Both compounds, 6 and 7, were obtained as an inseparable mixture of 10 cis and 11 trans isomers.

Moffatt oxidation [dimethyl sulfoxide, l-cyclohexyl-3-(2-morpholinoethyl) carbodiimide metho-p-toluenesulfonate, trifluoroacetic acid, pyridine, and benzene] of alcohol $\underline{7}$ and subsequent reaction of the aldehyde, thus obtained, with the anion of dimethyl 2-oxoheptylphosphonate in tetrahydrofuran gave in 71% yield a mixture of enones $\underline{8}$ [IR (CHCl₃) 1760, 1730, 1700, 1680, and 1630 cm⁻¹; lh NMR (CDCl₃) δ 0.82 (t, J= 7 Hz, C₂₀-H), 2.25 (t, J= 7 Hz, C₂-H), 2.50 (t, J= 7 Hz, C₁₆-H), 3.63 (s, CO₂CH₃), 4.6 (m, C₁₂-H), 6.15 (d, J₁₃,14=16 Hz, C₁₄-H), and 6.60 (dd, J₁₃,14=16 Hz, J₁₂,13=8 Hz, C₁₃-H)] and $\underline{9}$ [lh NMR (CDCl₃) δ 5.0 (m, C₁₂-H), 6.19 (d, J₁₃,14=16 Hz, C₁₄-H), and 6.60 (dd, J₁₃,14=16 Hz, J₁₂,13=8 Hz, C₁₃-H)]. Configurational assignment of $\underline{8}$ and $\underline{9}$ was based upon the chemical shifts of the C₁₂-H's and the vicinal coupling constants J₈,12 (11), which were 4 Hz ($\underline{8}$) and 7 Hz ($\underline{9}$), in agreement with the corresponding spectral data of model compounds (12).

In order to achieve selective reduction of the C_{15} carbonyl function (13) the mixture of enones was treated with lithium trisec-butylborohydride in tetrahydrofuran at -78° . However, upon column chromatography on silica gel, besides some starting material, a fraction containing hydroxyketones $\underline{10}$ and $\underline{11}$, and a fraction containing $\underline{11}$ -aza-PGF₁ isomers $\underline{12}$ [45%; IR (CHCl₃) 3450, 1730, and

1680 cm⁻¹; 1 H NMR (CDCl₃) 3 0.9 (t, J= 7 Hz, C₂₀-H), 2.32 (t, J= 7 Hz, C₂-H), 3.7 (s, CO₂CH₃), and 5.62 (m, C₁₃-H, C₁₄-H)] were isolated. Preparative layer chromatography of the hydroxyketone fraction gave a C₁₅ epimeric mixture of l1-aza-PGE₁ analog <u>10</u> [18%; IR (CHCl₃) 3460, 1760, 1730, and 1690 cm⁻¹; 1 H NMR (CDCl₃) 3 0.9 (t, J= 7 Hz, C₂₀-H), 2.32 (t, J= 7 Hz, C₂-H), 3.7 (s, CO₂CH₃), 4.5 (m, 3 0.9 (t, 2 7 Hz, C₁₂-H), and 5.65 (m, C₁₃-H, C₁₄-H)], while enone <u>11</u> decomposed.

Because the above synthesis lacks stereoselectivity a new reaction scheme was designed, leading to an aza-analog of the well-known "Corey-lactone" (14). The synthesis starts from the keto-ester 13 (8), obtained in 53% yield from ethyl N-ethoxycarbonylglycinate and diethyl fumarate according to the procedure described for 3. Alkylation of the potassium salt of 13 with ethyl bromoacetate in dimethyl sulfoxide at ambient temperature, followed by decarboxylation with aqueous hydrochloric acid in acetic acid at $110^{\,\mathrm{O}}$ and re-esterification with ethanol in the presence of sulfuric acid, provided a mixture of pyrrolidones 14 [IR (CHCl₃) 1760, 1740, and 1690 cm⁻¹; 1 H NMR (CDCl₂) δ 2.7-3.0 (m, C₇-H, C₈-H)] in 40% yield after column chromatography on silica gel. Catalytic hydrogenation of 14 over platinum oxide in ethanol at 3-4 atm and subsequent chromatographic purification furnished the pure all-trans-hydroxy ester 15 [64%; IR (CHCl₂) 3500, 1720, and 1700 cm⁻¹; 1 H NMR (CDCl₃) δ 2.56 (m, C_7^{-H} , C_8^{-H}), 3.45 (dd, $J_{9,10\beta}^{=4}$ Hz, $J_{10\alpha,10\beta}^{=11.5}$ Hz, $C_{10\beta}^{-H}$), and 3.75 (br, exchangeable with D20, OH)]. The relative configuration was assigned after comparing 1H NMR spectra of 15 and its acetate 16 with those of model compounds (12).

Hydroxy-ester $\underline{15}$ was converted with methanesulfonyl chloride in pyridine into the corresponding mesylate $\underline{17}$ [58%; IR (CHCl $_3$) 1730, 1690, 1340, and 1160 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 2.55 (m, C $_8$ -H), 3.05 (m, C $_7$ -H; s, CH $_3$ SO $_2$), 3.7 (dd, J $_9$,10 $_8$ = 4 Hz, J $_1$ 0 $_{\alpha}$,10 $_8$ = 13 Hz, C $_1$ 0 $_8$ -H), and 5.05 (dt, J $_8$,9= J $_9$,10 $_8$ = 4 Hz, J $_9$,10 $_{\alpha}$ = 6 Hz, C $_9$ -H)], which could be transformed in high yield into the lactone acid $_1$ 9 [98%; IR (CHCl $_3$) 3400-2700, 1780, 1720, and 1700 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 5.10 (m, C $_9$ -H), and 6.0 (br, exchangeable with D $_2$ O, CO $_2$ H)] upon reaction with aqueous potassium hydroxide in tetrahydrofuran and subsequent acidification and heating at 60-80 $^{\circ}$ for 3 h. In order to obtain better analyzable 1 H NMR spectra, lactone acid 1 9 was treated with diazomethane to afford lactone methyl ester 2 0 [IR (CHCl $_3$) 1780, 1740, and 1690 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 2.52-3.10 (m, C $_7$ -H), 3.2 (m, C $_8$ -H), 3.65 (m, C $_1$ 0 $_{\alpha}$ -H), 3.78 (s, CO $_2$ CH $_3$), and 5.58 (m, C $_9$ -H)].

Reaction of mesylate $\underline{17}$ with tetraethylammonium acetate in refluxing acetone produced inverted acetate $\underline{18}$ [67%; IR (CHCl $_3$) 1740, 1690, and 1220 cm $^{-1}$; 1 H NMR (CDCl $_3$) δ 2.05 (s, CH $_3$ CO), 3.65-3.8 (m, C $_1$ O-H), and 5.35 (m, C $_9$ -H)] that could be converted into lactone ethyl ester $\underline{21}$ (85%) upon treatment with potassium hydroxide in aqueous ethanol and subsequent acidification and heating at ca. 70° .

Reduction of the carboxyl function in $\underline{19}$ with diborane in tetrahydrofuran afforded the "Corey-type" lactone alcohol $\underline{22}$ [50%; IR (CHCl₃) 3430, 1785, and 1690 cm⁻¹; 1 H NMR (CDCl₃) 8 1.26 (t, J= 7 Hz, NCO₂CH₂CH₃), 2.45-3.05 (m, C₇-H), 3.08 (m, C₈-H), 3.7 (br, exchangeable with D₂O, OH), 3.7 (m, C₁₃-H), 4.14 (q, J= 7 Hz, NCO₂CH₂CH₃), and 5.55 (m, C₉-H)] from which ll-azaprostaglandin analogs should be obtained by standard procedures (15).

REFERENCES

- 1 Taken in part from the doctorate thesis of G.P. Rozing, University of Amsterdam, 1977.
- 2 G. Bolliger and J.M. Muchowski, <u>Tetrahedron Letters</u>, 1975, 2931; J.W. Bruin, H. de Koning, and H.O. Huisman, <u>ibid.</u>, 1975, 4599; M.A. Ondetti and M.E. Condon, <u>Chem.Abstr.</u>, 1976, <u>85</u>, 5487d, Ger.Offen. 2,536,382; J. Himizu, S. Saijo, K. Noguchi, M. Wada, Y. Harigaya, and O. Takaichi, <u>ibid.</u>, 1976, <u>85</u>, 12375lh, Japan. Kokai 76 01,461; R.J. DeFranco, R.M. Scribner, <u>ibid.</u>, 1976, <u>85</u>, 192546g, U.S. 3,975,399; A. Jean, C. Pigerol, L.P. Eymard, and J. Simiand, <u>ibid.</u>, 1977, <u>86</u>, 55278c, Ger.Offen. 2,612,114; J. Himizu, S. Harigaya, M. Wada, S. Saijo, K. Noguchi, and O. Takaiti, <u>ibid.</u>, 1977, <u>86</u>, 72433s, Ger.Offen. 2,618,176.
- 3 G.P. Rozing, T.J.H. Moinat, H. de Koning, and H.O. Huisman, Heterocycles, 1976, 4, 719; G.P. Rozing, H. de Koning, and H.O. Huisman, ibid., 1976, 5, 325.

- 4 R. Aries, Chem.Abstr., 1976, 84, 121288t, Fr.Demande 2,258,376;
 K. Kühlein, A. Linkies, and D. Reuschling, Tetrahedron Letters,
 1976, 4463; D. Reuschling, M. Mitzlaff, and K. Kühlein, ibid., 1976,
 4467; P.A. Zoretic and F. Barcelos, ibid., 1977, 529.
- 5 R.M. Scribner, <u>Tetrahedron Letters</u>, 1976, 3853; J.C. Lapierre Armande and U.K. Pandìt, <u>ibìd</u>., 1977, 897; F. Cassidy and G. Wootton, Chem. <u>Abstr.</u>, 1976, 85, 192576s, Ger. Offen. 2,552,312.
- 6 R.M. Scribner, <u>Chem.Abstr.</u>, 1974, <u>80</u>, 47986t, Ger.Offen. 2,323,193; <u>ibid.</u>, 1975, <u>83</u>, 97288z, Ger.Offen. 2,451,160.
 - 7 Prostaglandin numbering throughout this communication.
 - 8 R. Kuhn and G. Osswald, Chem.Ber., 1956, 89, 1423.
- 9 Prepared by ${\rm BF_3}$. ${\rm H_3PO_4}$ catalyzed reaction of methyl 4-hydroxy-crotonate and isobutene.
- 10 J. Martel and E. Toromanoff, <u>Chem.Abstr.</u>, 1972, <u>76</u>, 24712d, Ger.Offen. 2,121,361.
- 11 These coupling constants were proved unambiguously by double resonance.
- 12 G.P. Rozing, H. de Koning, and H.O. Huisman, to be published.
- 13 A. Hamon, B. Lacoume, G. Pasquet, and W.R. Pilgrim, <u>Tetrahedron</u>
 Letters, 1976, 211; T.S. Burton, M.P.L. Caton, E.C.J. Coffee,
- T. Parker, K.A.J. Stuttle, and G.L. Watkins, <u>J.Chem.Soc.</u>, <u>Perkin</u>
 Trans. I, 1976, 2550.
- 14 E.J. Corey, N.M. Weinshenker, T.K. Schaaf, and W. Huber, J.Amer.Chem.Soc., 1969, 91, 5675.
- 15 Conversion of lactone alcohol $\underline{22}$ into 11-azaprostaglandins is currently under investigation.

Received, 18th May, 1977